

Supporting Information

Room Temperature Asymmetric Pd-Catalyzed Methoxycarbonylation of Norbornene. Highly Selective Catalysis and HP-NMR Studies

Carolina Blanco^[a], Cyril Godard,^{*[a]} Ennio Zangrando,^[b] Aurora Ruiz^[a] and Carmen Claver^{*[a]}

[a] Departament de Química Física i Inorgànica,
Universitat Rovira i Virgili, Marcel·li Domingo s/n,
43007, Tarragona, Spain.

[b] Dipartimento di Scienze Chimiche e Farmaceutiche,
Università di Trieste, via Licio Giorgieri 1,
34127 Trieste, Italy.

Contents

SI. 1 Experimental Section.....	3
SI. 2 Characterization of Palladium Complexes.....	3
Figure 1. ^1H NMR spectrum of $[\text{PdCl}_2(\mathbf{1})_2]$ 1a	3
Figure 2. $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum of $[\text{PdCl}_2(\mathbf{1})_2]$ 1a	3
Figure 3. ^1H NMR spectrum of $[\text{PdCl}_2(\mathbf{4})]$ 4a	4
Figure 4. $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum of $[\text{PdCl}_2(\mathbf{4})]$ 4a	4
Figure 5. ^1H NMR spectrum of 4-(H⁺)₂	5
Figure 6. $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum of 4-(H⁺)₂	5
Figure 7. ^1H NMR spectrum of $[\text{PdBr}(\mathbf{7})]$ 7a	6
Figure 8. $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum of $[\text{PdBr}(\mathbf{7})]$ 7a	6
SI. 3 X-ray data	7
Table 1. Crystal data for structures 4a and 7a	7

SI. 1 Experimental Section

General Methods

All reactions were carried out under an argon atmosphere using Standard Schlenk techniques. All solvents were carefully purified by appropriate procedures. PdCl₂ was purchased from Johnson Matthey Inc. and used without further purification. All other reagents were used as received from commercial suppliers. CD₂Cl₂, THF-d₈, CDCl₃ were stored over molecular sieves under nitrogen. Air sensitive compounds were stored under argon.

NMR data

¹H, ¹³C{¹H}, ³¹P{¹H} spectra were recorded on a Varian Mercury 400 spectrometer (400.14, 100.63, and 161.98 MHz respectively). Chemical shifts were reported relative to tetramethylsilane for ¹H and ¹³C{¹H} as internal standard. ³¹P chemical shifts were referenced to an external 85% H₃PO₄ sample.

SI. 2 Characterization of Palladium Complexes

Figure 1. ¹H NMR spectrum of complex [PdCl₂(1)₂] 1a (CD₂Cl₂, 298K)

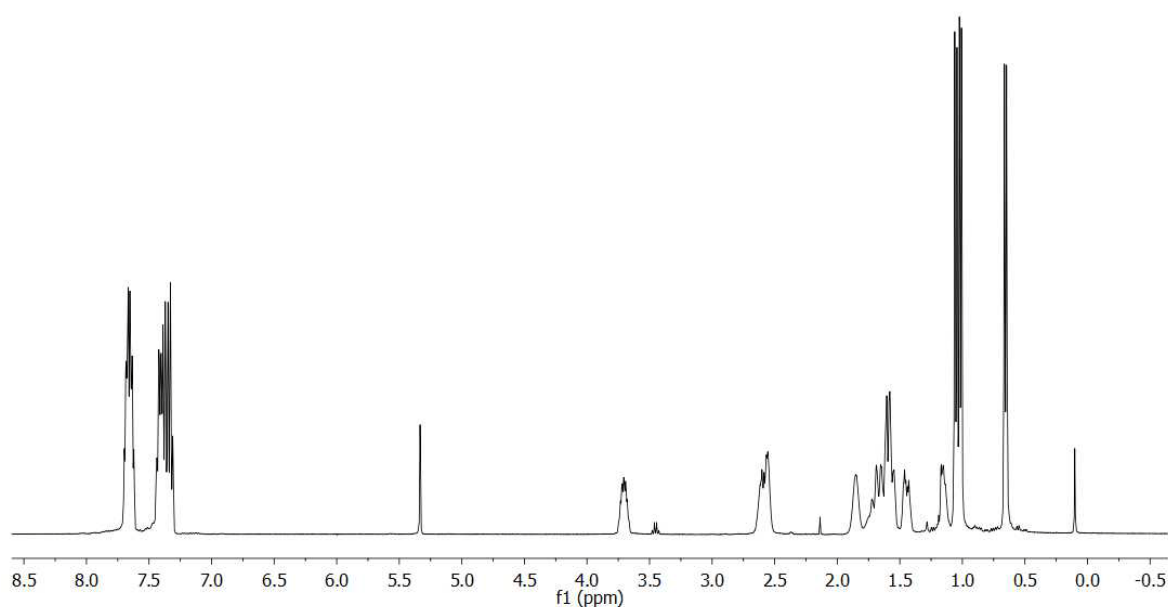


Figure 2. ³¹P{¹H} NMR spectrum of complex 1a (CD₂Cl₂, 298K)

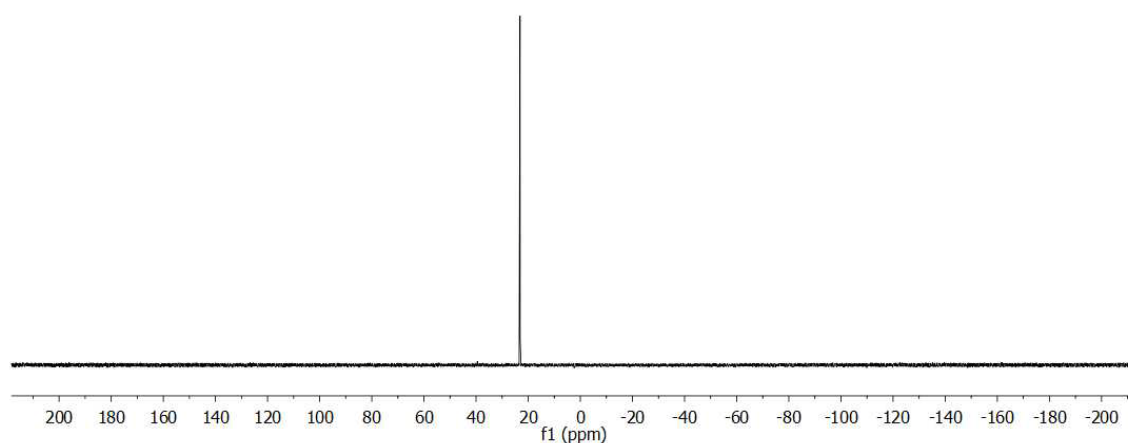


Figure 3. ^1H NMR spectrum of $[\text{PdCl}_2(\text{4})]$ **4a** (CD_2Cl_2 , 298K)

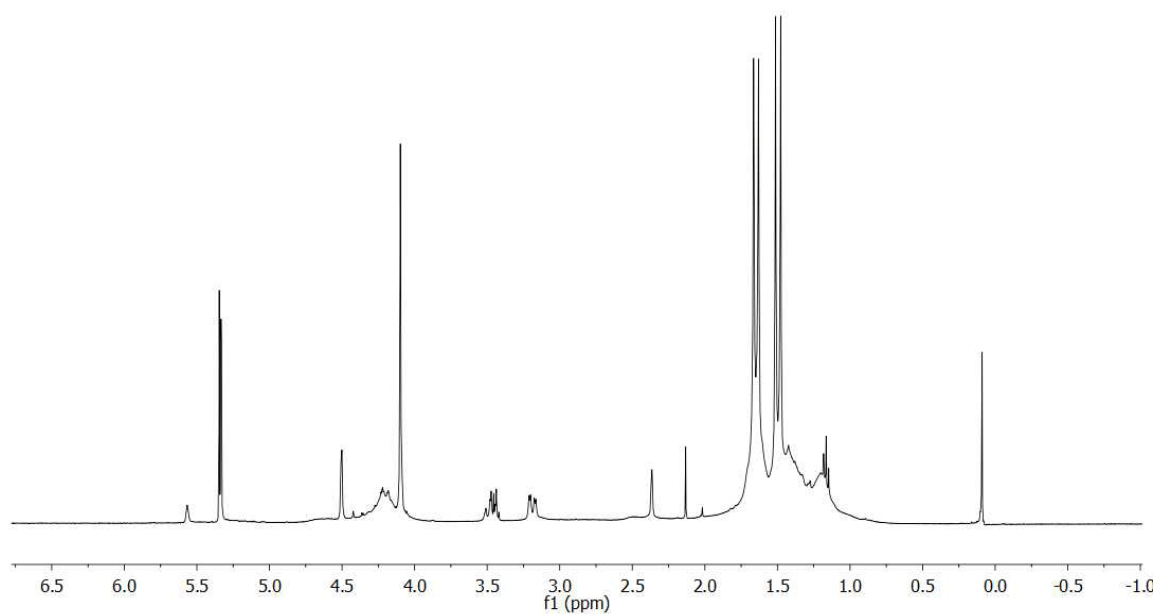


Figure 4. $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum of $[\text{PdCl}_2(\text{4})]$ **4a** (CD_2Cl_2 , 298K)

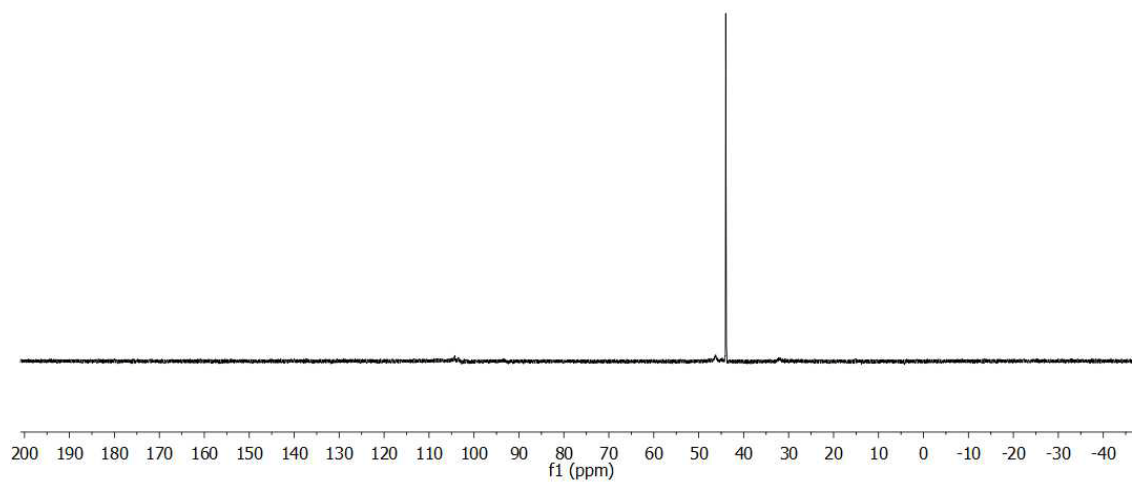


Figure 5. ^1H NMR spectrum of $4\text{-(H}^+)_2$ (THF- d_8 , 298K)

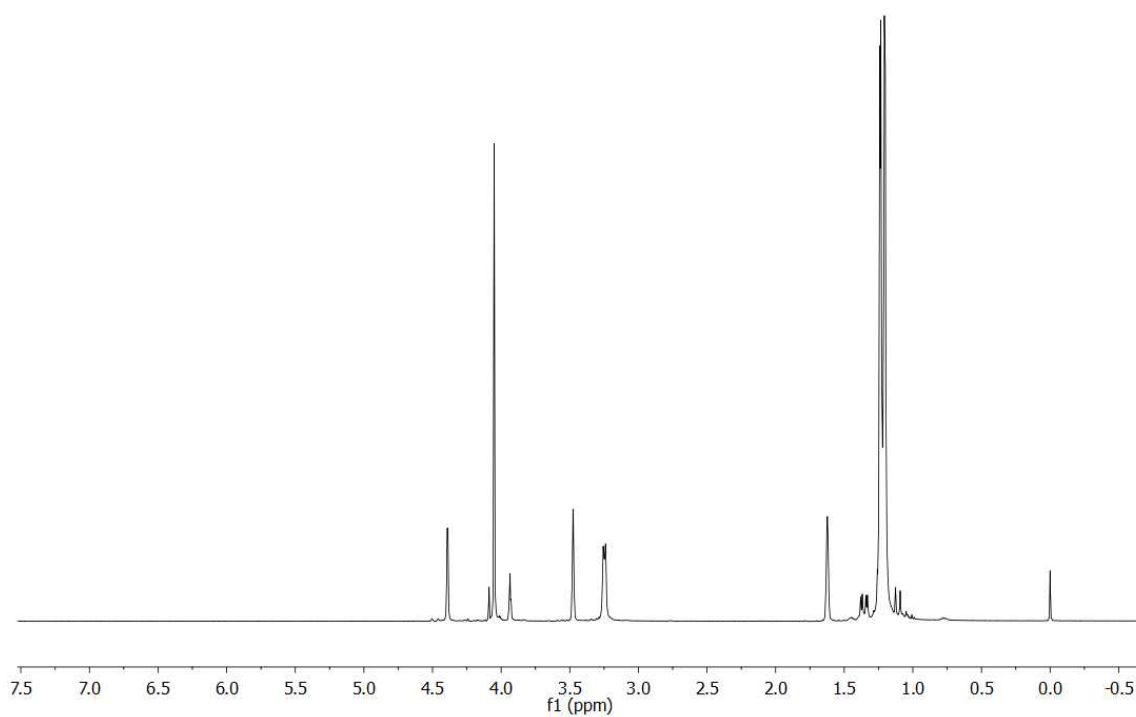


Figure 6. $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum of $4\text{-(H}^+)_2$ (THF- d_8 , 298K)

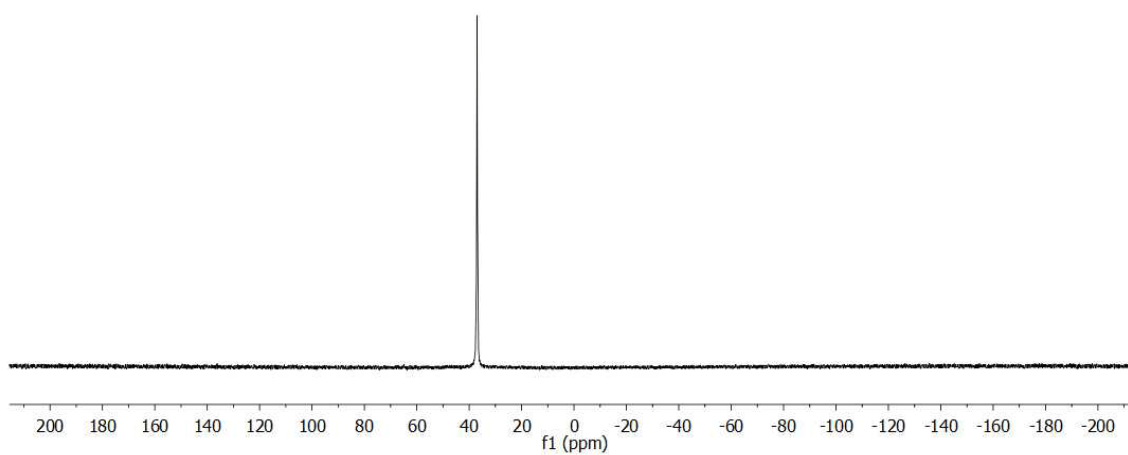


Figure 7. ^1H NMR spectrum of $[\text{PdBr}(7)]$ **7a** (CD_2Cl_2 298K)

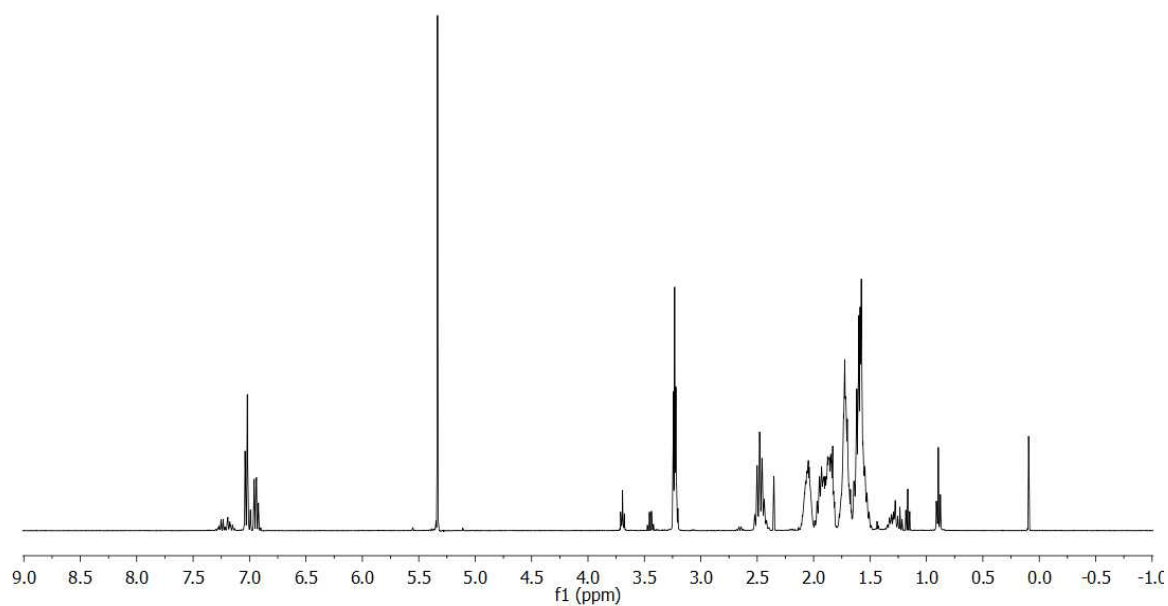
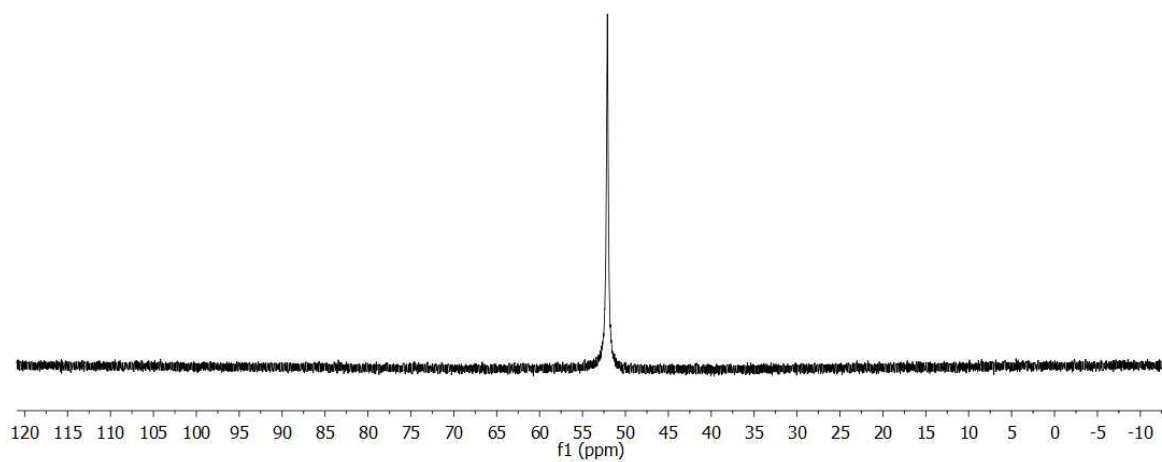


Figure 8. $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum of $[\text{PdBr}(7)]$ **7a** (CD_2Cl_2 298K)



SI. 3 X-ray data

Table 1. Crystal data for structures **4a** and **7a**

	4a	7a
Empirical Formula	C ₂₈ H ₄₈ Cl ₂ FeP ₂ Pd. C ₄ H ₈ O	C ₂₈ H ₄₃ BrP ₂ Pd
M	751.86	627.87
T/K		293(2)
$\lambda/\text{\AA}$		0.71073
Crystal system	Monoclinic	Orthorhombic
Space group	<i>P</i> 2 ₁ /n	<i>P</i> bca
<i>a</i> / \AA	11.558(3)	15.151(3)
<i>b</i> / \AA	14.646(3)	17.180(4)
<i>c</i> / \AA	20.721(4)	21.361(4)
$\beta/^\circ$	94.29(3)	90.0
Unit cell volume/ \AA^3	3497.8(13)	5560(2)
D _{calcd} /g cm ⁻³	1.428	1.500
Z	4	8
$\mu(\text{Mo K}\alpha)/\text{mm}^{-1}$	1.196	2.234
F(000)	1568	2576
Theta range for data collection	1.70 - 25.02	2.62 - 27.10
Absorption correction		refdelf
Refinement method		Full-matrix least squares on <i>F</i> ²
Data/parameter	6015/352	5864/289
Final R indices [<i>I</i> >2 σ (<i>I</i>)]	R ₁ =0.0510, wR ₂ = 0.1159	R ₁ =0.0395, wR ₂ = 0.0959
R indices (all data)	R ₁ = 0.1008, wR ₂ = 0.1359	R ₁ = 0.852, wR ₂ = 0.1079
Goodness-of-fit on <i>F</i> ²	0.894	0.860
Largest diff peak and hole (e/ \AA^3)	0.668, -0.406	0.382, -0.697