Supplementary Material for:

Hydroformylation by Pt-Sn compounds from N-heterocyclic stannylenes

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$PtCl(SnCl{Me_2Si(NAr)_2})(PPh_3)_2 (2a)$

Preparative Scale: A solution of $[Me_2Si{NAr}_2]Sn (1) (158 mg, 0.30 mmol)$ in toluene was added drop wise to a suspension of $PtCl_2(PPh_3)_2 (240 mg, 0.30 mmol)$ in toluene. The solution went clear and orange after complete addition and stirring for 10 min. The reaction mixture was stirred for a further 18 h. The solution was concentrated, filtered and stored at -20 °C, affording orange crystals of **2a**. Yield = 256 g (67 %). Note, these crystals readily desolvate under vacuum to give a yellow powder.

NMR Scale: A yellow solution of $[Me_2Si\{NAr\}_2]Sn (1) (30.4 mg, 0.06 mmol, 2 equiv.) in C₆D₆ was added to a NMR tube fitted with a J. Youngs tap that had previously been charged with PtCl₂(PPh₃)₂(22.8 mg, 0.03 mmol). A clear orange solution formed. NMR spectra were consistent with PtCl{(Me₂Si{NAr}₂SnCl)}(PPh₃)₂ ($ **2a**), with signals present for one equivalent of unreacted**1**.

Anal. Calcd. For C₆₂H₇₀Cl₂N₂P₂PtSiSn (*1317.96*): C 56.50, H 5.35, N 2.13 %. Found: C 56.62, H 5.26, N 2.05 %. ¹H NMR (C₆D₆, 400 MHz): δ 7.42 (m, 6H, *o*-PP*h*₃), 7.36 (d, 4H, *m*-C₆*H*₃), 7.28 (m, 6H, *o*-PP*h*₃), 7.21 (t, 2H, *p*-C₆*H*₃), 6.93-6.74 (m, 12H, *m*-PP*h*₃), 6.51 (m, 6H, *p*-PP*h*₃), 5.05 (br, 2H, CHMe₂), 4.64 (br, 2H, CHMe₂), 1.81-1.21 (br, 24H, CHMe₂), 0.66 (s, 3H, SiMe₂), 0.43 (s, 3H, SiMe₂). ¹³C{¹H} NMR (C₆D₆, 100 MHz): δ 149.0, 143.7, 135.0, 133.0, 131.1, 130.9, 129.5, 128.1, 123.9, 123.3 (C₆H₃ and PP*h*₃), 27.4 (CHMe₂), 26.6 (br, CHMe₂), 5.0, 4.2 (SiMe₂). ³¹P{¹H} NMR (C₆D₆, 162 MHz): δ 31.7 (d, ²J_{PP} = 15 Hz, ²J_{119SnP} = 3522 Hz, ²J_{117SnP} = 3367 Hz, ¹J_{PtP} = 2597 Hz), 19.9 (d, ²J_{PP} = 15 Hz, ²J_{119SnP} = 136 Hz, ¹J_{PtP} = 3886 Hz). ¹¹⁹Sn{¹H} NMR (C₆D₆, 85 MHz): δ -167 (dd, ²J_{SnP} = 139 Hz, ²J_{SnP} = 3523 Hz, ¹J_{SnPt} = 15436 Hz). ¹⁹⁵Pt{¹H} NMR (C₆D₆, 85 MHz): δ -4695 (dd, ¹J_{PtP} = 2596 Hz, ¹J_{PtP} = 3884 Hz).

$PtCl(SnCl{Me_2Si(NAr)_2})(COD) (2b)$

NMR Scale: Compound **2b** was prepared as for **2a**, using $[Me_2Si{NAr}_2]Sn$ (**1**) (19.3 mg, 0.037 mmol) and $PtCl_2(COD)$ (13.7 mg, 0.037 mmol). NMR spectra were consistent with $PtCl(Me_2Si(NAr)_2SnCl)(COD)$ (**2b**).

¹H NMR (C₆D₆, 400 MHz): δ 7.25 (d, ³*J*_{HH} = 7.1 Hz, 4H, *m*-C₆*H*₃), 7.12 (t, ³*J*_{HH} = 7.1 Hz, 2H, *p*-C₆*H*₃), 5.28 (br s, ²*J*_{PtH} = 37.0 Hz, 2H, COD-C*H*), 4.85 (br s, ²*J*_{PtH} = 67.0 Hz, 2H, COD-C*H*), 4.65 (sept, 4H, ³*J*_{HH} = 6.7 Hz C*H*Me₂), 1.56 (d, ³*J*_{HH} = 6.7 Hz 12H, CH*Me*₂), 1.49 (d, ³*J*_{HH} = 6.7 Hz, 12H, CH*Me*₂), 1.19 (m, 4H, COD-C*H*₂), 0.99 (m, 4H, COD-C*H*₂), 0.55 (s, 3H, Si*Me*₂), 0.47 (s, 3H, Si*Me*₂). ¹³C{¹H} NMR (C₆D₆, 100 MHz): δ 148.1, 141.7, 124.4, 124.1 (*C*₆H₃), 121.9 (COD-*C*H), 87.7 (COD-*C*H), 32.0, 27.6 (COD-*C*H₂), 27.2 (*C*HMe₂), 26.6 (br, CH*Me*₂), 4.4, 4.0 (Si*Me*₂). ¹¹Sn{¹H} NMR (C₆D₆, 149 MHz): δ -221.7 (¹*J*_{PtSn} = 21576 Hz). ¹⁹⁵Pt{¹H} NMR (C₆D₆, 85 MHz): δ -3669.3 (¹*J*_{Pt117Sn} = 20618 Hz, ¹*J*_{Pt119Sn} = 21569 Hz).

$[Pt(\mu-Cl){(Me_2Si{NAr}_2)SnCl}(PEt_3)]_2 (3)$

NMR Scale: Compound **3** was prepared as for **2a**, using $[Me_2Si\{NAr\}_2]Sn$ (**1**) (15.4 mg, 0.029 mmol) and $[PtCl(\mu-Cl)(PEt_3)]_2$ (11.5 mg, 0.015 mmol). NMR spectra were consistent with $[Pt(\mu-Cl)\{(Me_2Si\{NAr\}_2)SnCl\}(PEt_3)]_2$ (**3**).

¹H NMR (C₆D₆, 400 MHz): δ 7.06 (d, ³*J*_{HH} = 7.5 Hz, 4H, *m*-C₆*H*₃), 6.97 (t, ³*J*_{HH} = 7.5 Hz, 2H, *p*-C₆*H*₃), 4.37 (br, 4H, C*H*Me₂), 1.45 (br, 12H, CH*Me*₂), 1.37 (d, ³*J*_{HH} = 6.2 Hz 12H, CH*Me*₂), 1.11 (m, 6H, PC*H*₂CH₃), 0.73 (s, 3H, Si*Me*₂), 0.65 (dt, 9H, ³*J*_{HH} = 7.5 Hz, ³*J*_{HP} = 18.0 Hz, PCH₂C*H*₃), 0.23 (s, 3H, Si*Me*₂). ³¹P{¹H} NMR (C₆D₆, 162 MHz): δ 17.5 (s, ²*J*_{PSn} = 178 Hz, ¹*J*_{PPt} = 3761 Hz). ¹¹⁹Sn{¹H} NMR (C₆D₆, 149 MHz): δ -402 (d, ²*J*_{PSn} = 180 Hz, ¹*J*_{SnPt} = 26999 Hz). ¹⁹⁵Pt{¹H} NMR (C₆D₆, 85 MHz): δ -4058 (d, ¹*J*_{PtP} = 3742 Hz).

$Pt[Me_2Si{N(Ar)(SnCl_2)}_2][(Me_2Si{NAr}_2)Sn]_2 (4)$

NMR Scale: Compound **4** was prepared as for **2a**, using $[Me_2Si{NAr}_2]Sn$ (**1**) (81.0 mg, 0.15 mmol, 4 equiv.) and PtCl₂ (10.0 mg, 0.038 mmol). The sample was heated at 60 °C for 3 days, during which time the product crystallized as red crystals.