

### Electronic Supplementary Information

## A Novel Mixed Valent $\text{Cu}^{\text{II}}$ - $\text{Cu}^{\text{I}}$ 2D Framework Made of a Hydrazone and $\mu$ -SCN Bridged Metallacyclic Loops Cross-linked by $\mu_3$ -SCN Chains

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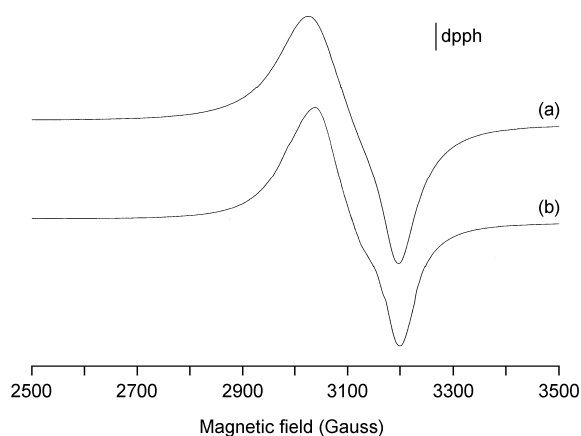


Figure F1. X-band EPR spectra of the polycrystalline complex: (a) at RT, (b) at 100 K. Diphenylpicrylhydrazyl (dpph) is the standard field marker ( $g = 2.0036$ ).

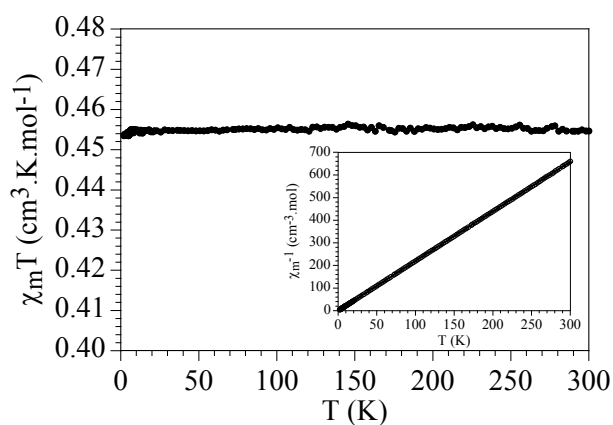


Figure F2. Thermal variation of the  $\chi_m T$  product for the complex. Inset shows the Curie plot. Solid line is the best fit to the Curie-Weiss law

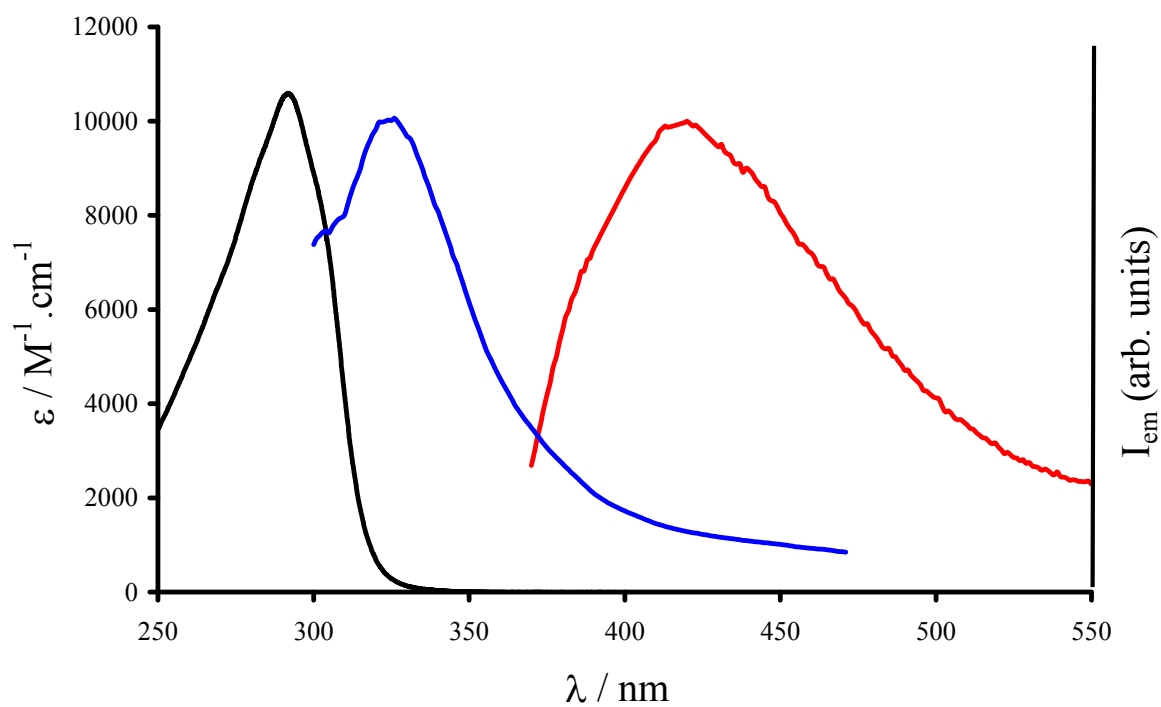


Figure F3. UV-Vis absorption spectrum of LH in ethanol ( $c = 1.18 \times 10^{-4} \text{ M}$ , black) and emission spectra of LH in ethanol ( $\lambda_{\text{exc}} = 270 \text{ nm}$ , blue) and of the mixed valent copper complex in the solid state ( $\lambda_{\text{exc}} = 338 \text{ nm}$ , red)

Table T1 Bond length (Å) and bond angles (°) for the complex

Bond lengths		Bond angles	
Cu1-N1	2.015(4)	N1-Cu1-N2	80.76(14)
Cu1-N2	1.925(3)	N1-Cu1-N4	102.58(16)
Cu1-N4	1.931(4)	N2-Cu1-N4	164.90(16)
Cu1-O1	1.983(3)	O1-Cu1-N1	158.00(14)
Cu1-S1	2.6444(17)	O1-Cu1-N2	78.62(13)
Cu2-N3 <sup>ii</sup>	2.033(3)	O1-Cu1-N4	95.47(15)
Cu2-N3 <sup>iv</sup>	2.033(3)	S1-Cu1-O1	93.41(11)
Cu2-S2	2.4249(14)	S1-Cu1-N1	96.39(12)
Cu2-S2 <sup>v</sup>	2.4249(14)	S1-Cu1-N2	96.20(11)
Cu3-N5 <sup>i</sup>	1.938(5)	S1-Cu1-N4	98.03(13)
Cu3-N5 <sup>iii</sup>	1.938(5)	N3 <sup>ii</sup> -Cu2-N3 <sup>iv</sup>	116.10(13)
Cu3-S2	2.4292(15)	S2-Cu2-S2 <sup>v</sup>	107.05(5)
Cu3-S2 <sup>v</sup>	2.4292(15)	S2-Cu2-N3 <sup>ii</sup>	101.51(11)
Cu1...Cu2	5.5938(6)	S2-Cu2-N3 <sup>iv</sup>	115.44(10)
Cu1...Cu3	5.8449(6)	S2 <sup>v</sup> -Cu2-N3 <sup>ii</sup>	115.44(10)
Cu2...Cu3	2.8903(12)	S2 <sup>v</sup> -Cu2-N3 <sup>iv</sup>	101.51(11)
		N5 <sup>i</sup> -Cu3-N5 <sup>iii</sup>	116.7(2)
		S2-Cu3-S2 <sup>v</sup>	106.77(6)
		S2-Cu3-N5 <sup>i</sup>	109.17(17)
		S2-Cu3-N5 <sup>iii</sup>	107.32(16)
		S2 <sup>v</sup> -Cu3-N5 <sup>i</sup>	107.32(16)
		S2 <sup>v</sup> -Cu3-N5 <sup>iii</sup>	109.17(17)

Symmetry codes: (i) 1/2-x, 1/2-y, 1-z; (ii) 1/2-x, 3/2-y, 1-z; (iii) 1/2+x, 1/2-y, 1/2+z; (iv) 1/2+x, 3/2-y, 1/2+z; (v) 1-x, y, 3/2-z.

Table T2 Bond valence values for copper centers in the complex

Bond type	Bond distance ( $R_{ij}$ ) (Å)	$R_0$ (Å)	$b$ (Å)	Bond valence ( $S_{ij}$ )			BVS ( $N_i$ )
				Cu1	Cu2	Cu3	
Cu1-N1	2.015(4)	1.713	0.37	0.4421			2.1597
Cu1-N2	1.925(3)	1.713	0.37	0.5638			
Cu1-N4	1.931(4)	1.713	0.37	0.5548			
Cu1-O1	1.983(3)	1.655	0.37	0.4121			
Cu1-S1	2.6444(17)	2.024	0.37	0.1869			
Cu2-N3 <sup>ii</sup>	2.033(3)	1.571	0.37		0.2869		0.9788
Cu2-N3 <sup>iv</sup>	2.033(3)	1.571	0.37		0.2869		
Cu2-S2	2.4249(14)	1.834	0.37		0.2025		
Cu2-S2 <sup>v</sup>	2.4249(14)	1.834	0.37		0.2025		
Cu3-N5 <sup>i</sup>	1.938(5)	1.571	0.37			0.3708	1.1418
Cu3-N5 <sup>iii</sup>	1.938(5)	1.571	0.37			0.3708	
Cu3-S2	2.4292(15)	1.834	0.37			0.2001	
Cu3-S2 <sup>v</sup>	2.4292(15)	1.834	0.37			0.2001	