

Supporting Information for

**Solvothermal synthesis and characterization of thioindate-thioantimonates with transition-metal complexes: the first examples of the incorporation of transition metal ions into In-S-Sb framework**

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Table S1 Some important bond lengths (Å) and angles (°) for **1-3**.

<b>1</b>			
Sb1-S	2.454(5)-2.460(5)	Sb2-S	2.417(5)-2.447(5)
In1-S	2.445(5)-2.492(5)	In2-S	2.422(4)-2.491(5)
Ni1-N	2.03(15)-2.17(5)	S-Sb1-S	94.76(18)-96.66(18)
S3-Sb2-S	91.31(18)-100.8(2)	S-In1-S	105.20(19)-113.78(18)
S-In2-S	103.14(18)-117.29(17)	N-Ni1-N	77.4(10)-172.6(9)
<b>2</b>			
Sb1-S	2.4212(17)-2.4288(18)	Sb2-S	2.4263(19)-2.4638(19)
Sb3-S	2.4311(19)-2.447(2)	In1-S	2.4212(19)-2.4942(19)
In2-S	2.4379(19)-2.470(2)	In3-S	2.4328(17)-2.4721(19)
Ni1-N	2.188(7)-2.217(6)	Ni2-N	2.155(14)-2.29(2)
S-Sb1-S	92.75(6)-98.89(7)	S-Sb2-S	91.26(7)-95.92(7)
S-Sb3-S	90.87(7)-101.22(8)	S-In1-S	101.85(6)-123.84(7)
S-In2-S	103.41(7)-114.16(7)	S-In3-S	107.09(7)-112.56(6)
N-Ni1-N	77.5(3)-168.4(3)	N-Ni2-N	79.1(9)-173.4(6)
<b>3</b>			
Sb1-S	2.413(6)-2.435(6)	In1-S	2.463(6)-2.484(7)
Co1-N	2.07(2)- 2.21(2)	Co1-S	2.318(6)
S-Sb1-S	90.6(2)- 99.5(2)	S-In1-S	97.68(18)-119.5(2)
N-Co1-N	78.0(8)-130.0(9)	N-Co1-S	95.5(5)-173.7(5)

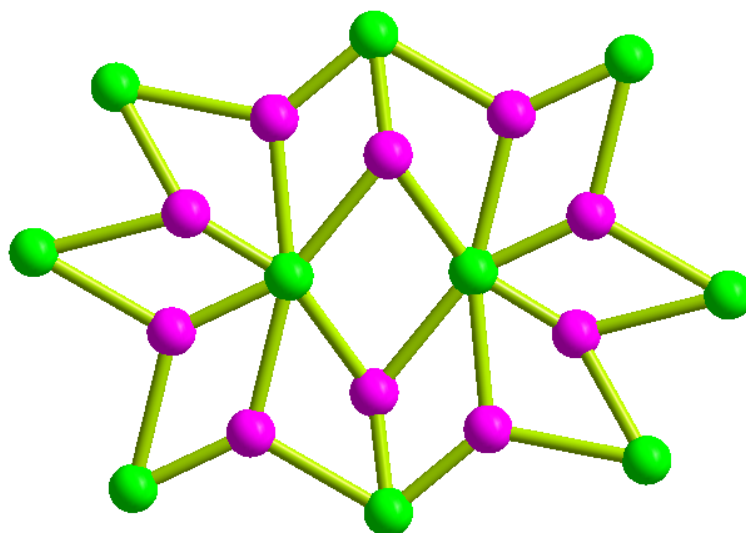


Figure S1 Topology of the  $[\text{In}_2\text{Sb}_2\text{S}_7]^{2n-}_n$  layer in **1** (green/pink:  $[\text{In}_2\text{S}_7]/[\text{SbS}_3]$  nodes).

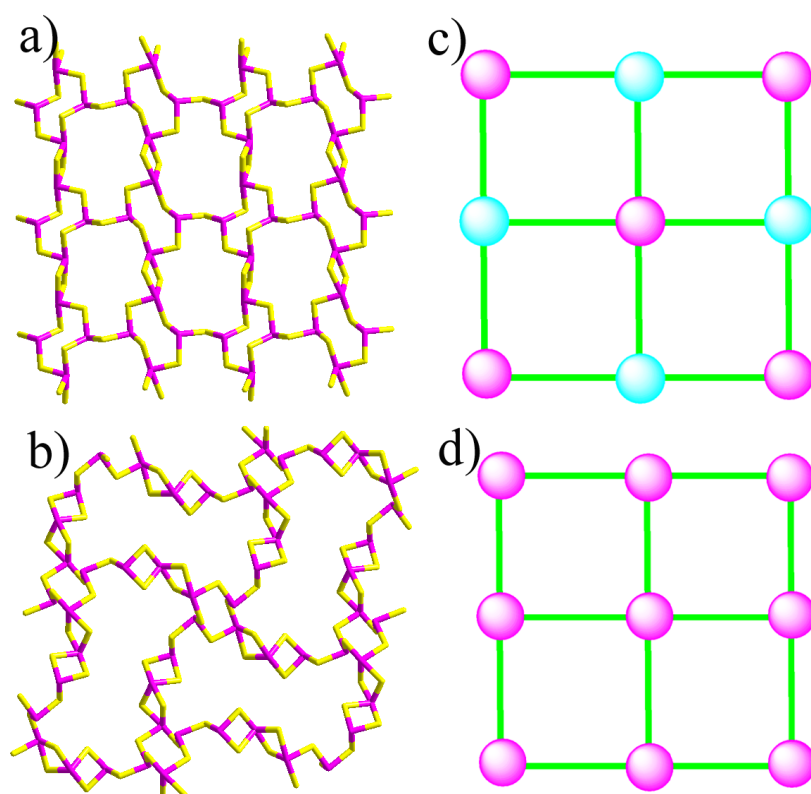


Figure S2 View of 2-D  $[\text{Sb}_4\text{S}_7]^{2n-}_n$  layers (a,b). Topology of the  $[\text{Sb}_4\text{S}_7]^{2n-}_n$  layers (c, blue/pink:  $[\text{Sb}_2\text{S}_5]/[\text{Sb}_2\text{S}_6]$  nodes; d, pink:  $[\text{Sb}_4\text{S}_9]$  nodes).

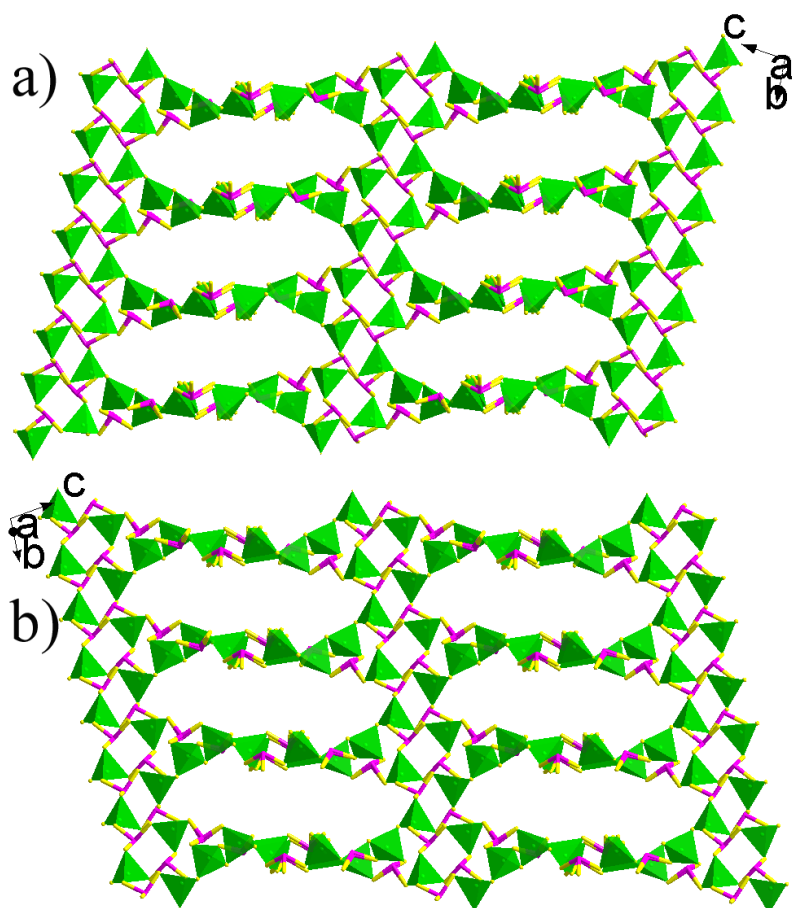
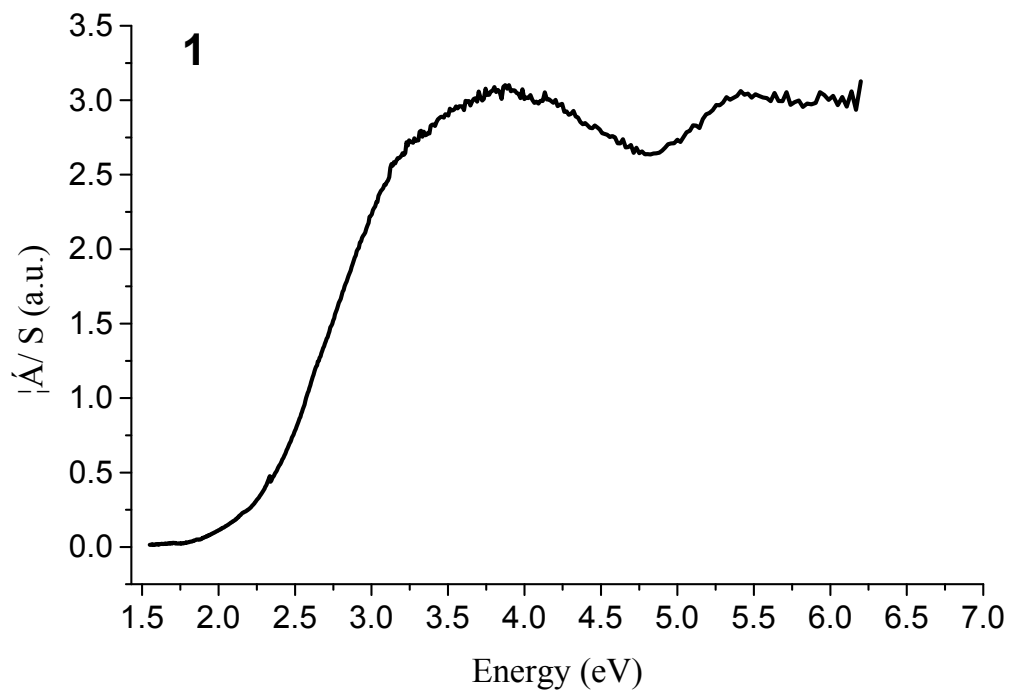


Figure S3 View of 3-D In-S-Sb open-framework in **2** along  $[1\bar{1}0]$  and  $[1\bar{1}0]$ .



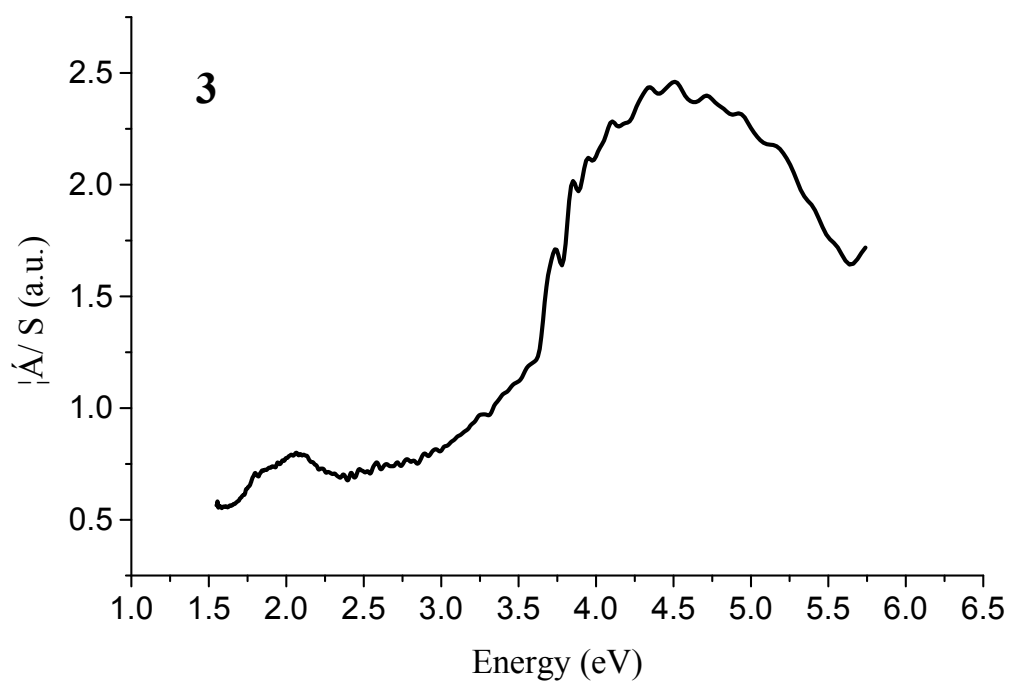
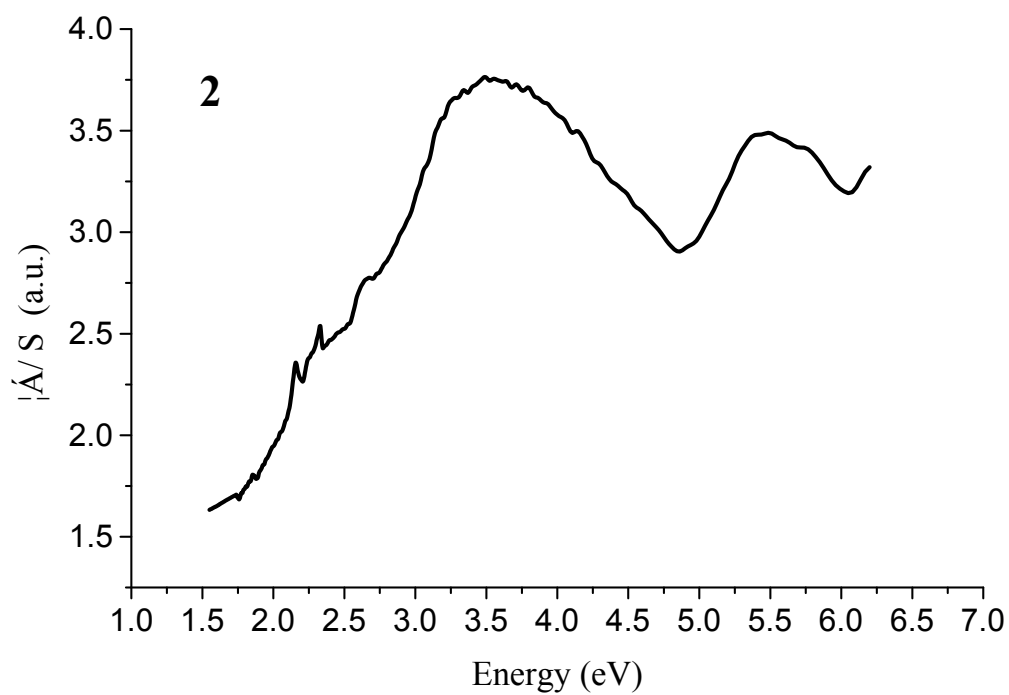


Figure S4 The solid-state UV/Vis absorption spectra of **1-3**.

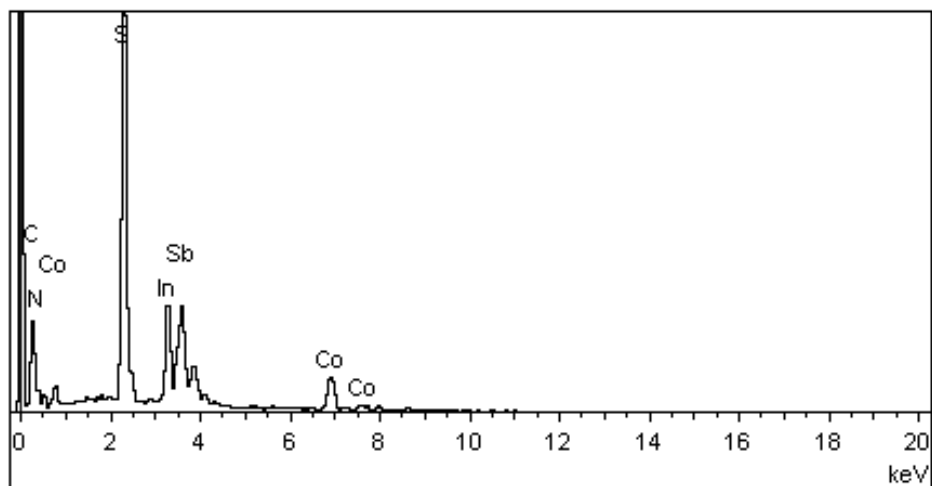


Figure S5 EDS spectrum of **3**.

The energy dispersive spectroscopy (EDS) analyses show that molar ratios of In:Sb:Co:S are 1.05:1.03:1.07:4.05 for **3**, consistent with the stoichiometry.

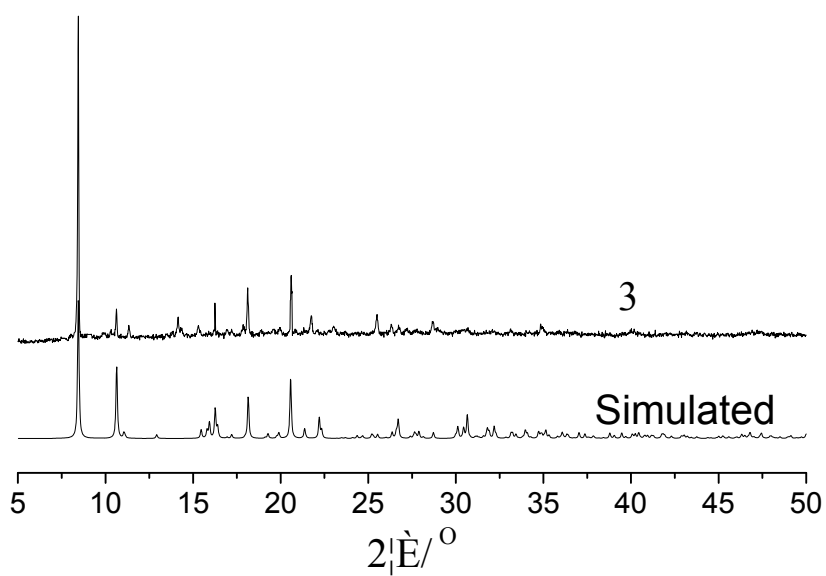


Figure S6 Simulated and experimental powder XRD patterns of **3** (The simulated XRD pattern is slightly different from the measured one. It means that **3** is sensible to the moisture and air).