

Appendix II (Fluorescence quenching of BSA by copper(II) complexes)

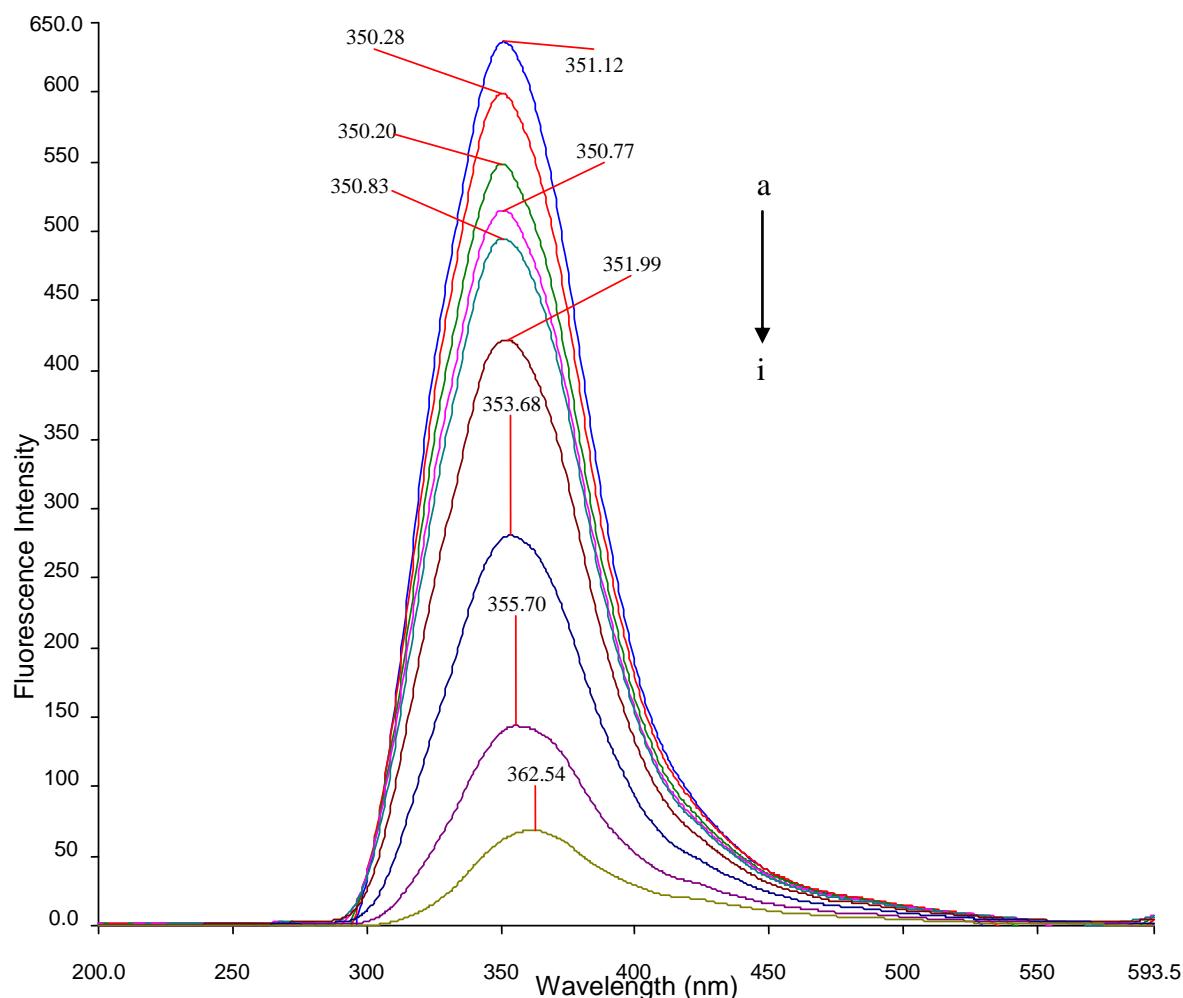


Figure S4.4(I) Fluorescence quenching of 0.24 mM BSA by increasing concentration [0 (a), 2 (b), 4 (c), 10 (d), 20 (e), 40 (f), 100 (g), 200 (h) and 400 (i) μ M] of D- [Cu(phen)(5MeOCA)(H₂O)NO₃] 4. (First trial)

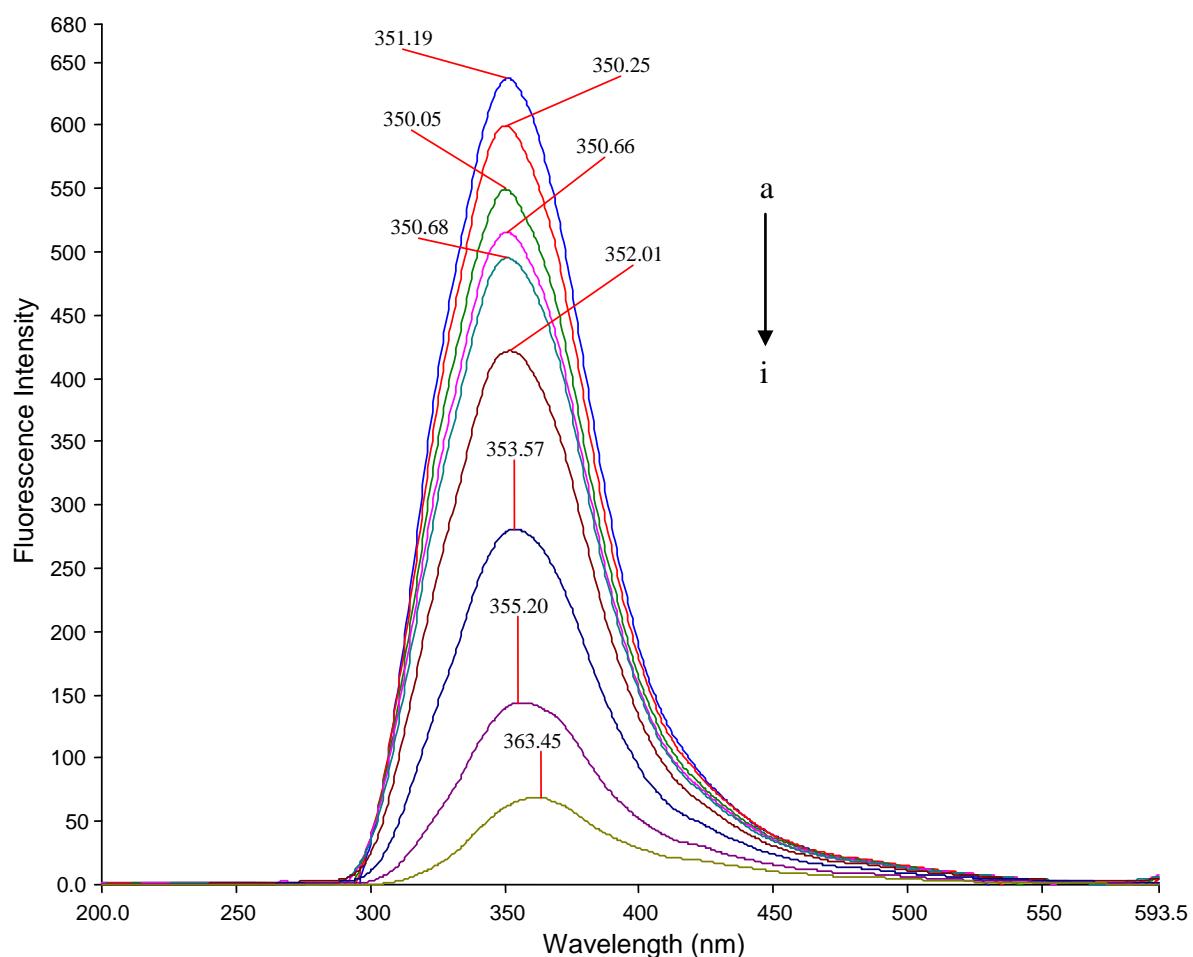


Figure S4.4(II) Fluorescence quenching of 0.24 mM BSA by increasing concentration [0 (a), 2 (b), 4 (c), 10 (d), 20 (e), 40 (f), 100 (g), 200 (h) and 400 (i) μ M] of D-
[Cu(phen)(5MeOCA)(H₂O)NO₃] **4**. (Second trial)

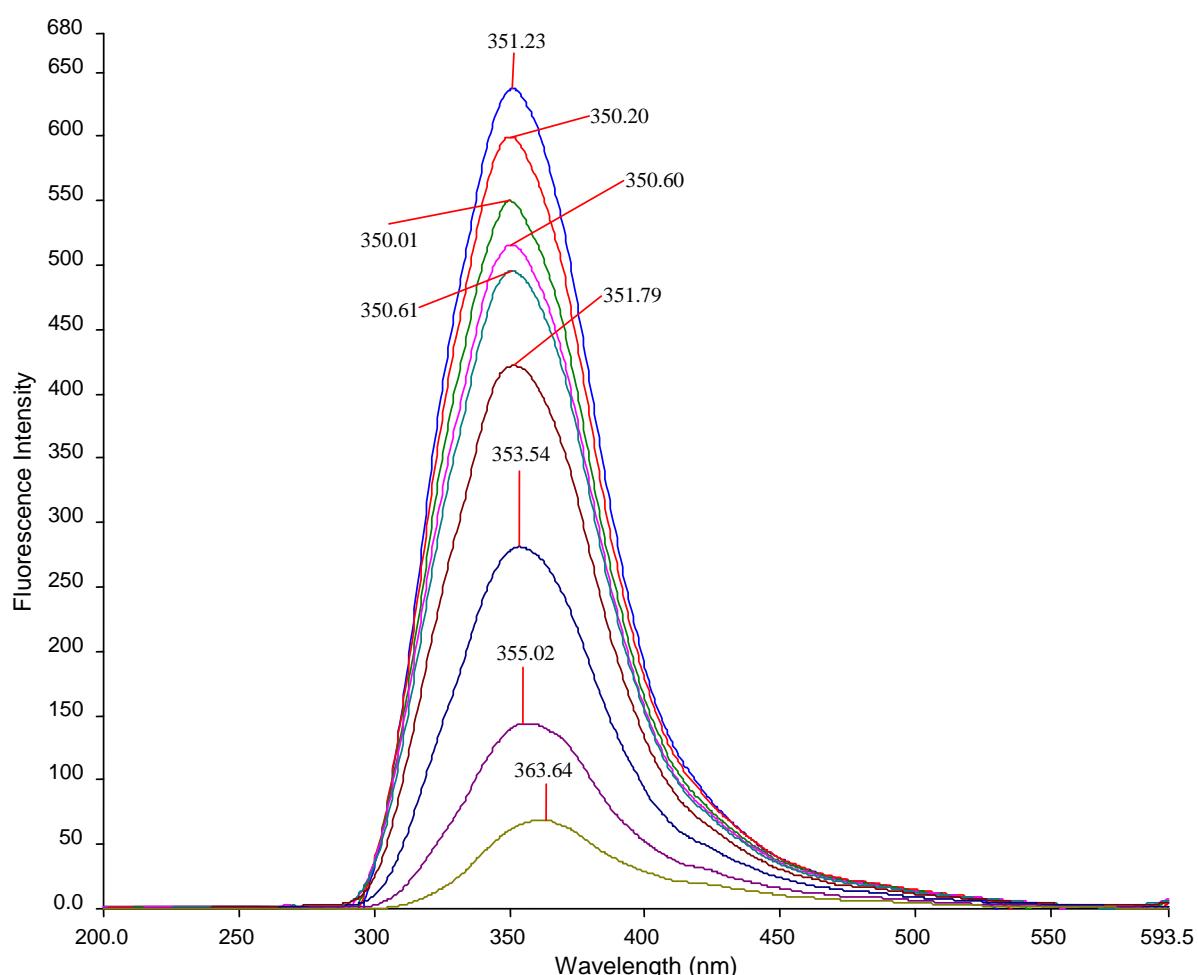


Figure S4.4(III) Fluorescence quenching of 0.24 mM BSA by increasing concentration [0 (a), 2 (b), 4 (c), 10 (d), 20 (e), 40 (f), 100 (g), 200 (h) and 400 (i) μ M] of D-[Cu(phen)(5MeOCA)(H₂O)NO₃] **4**. (Third trial)

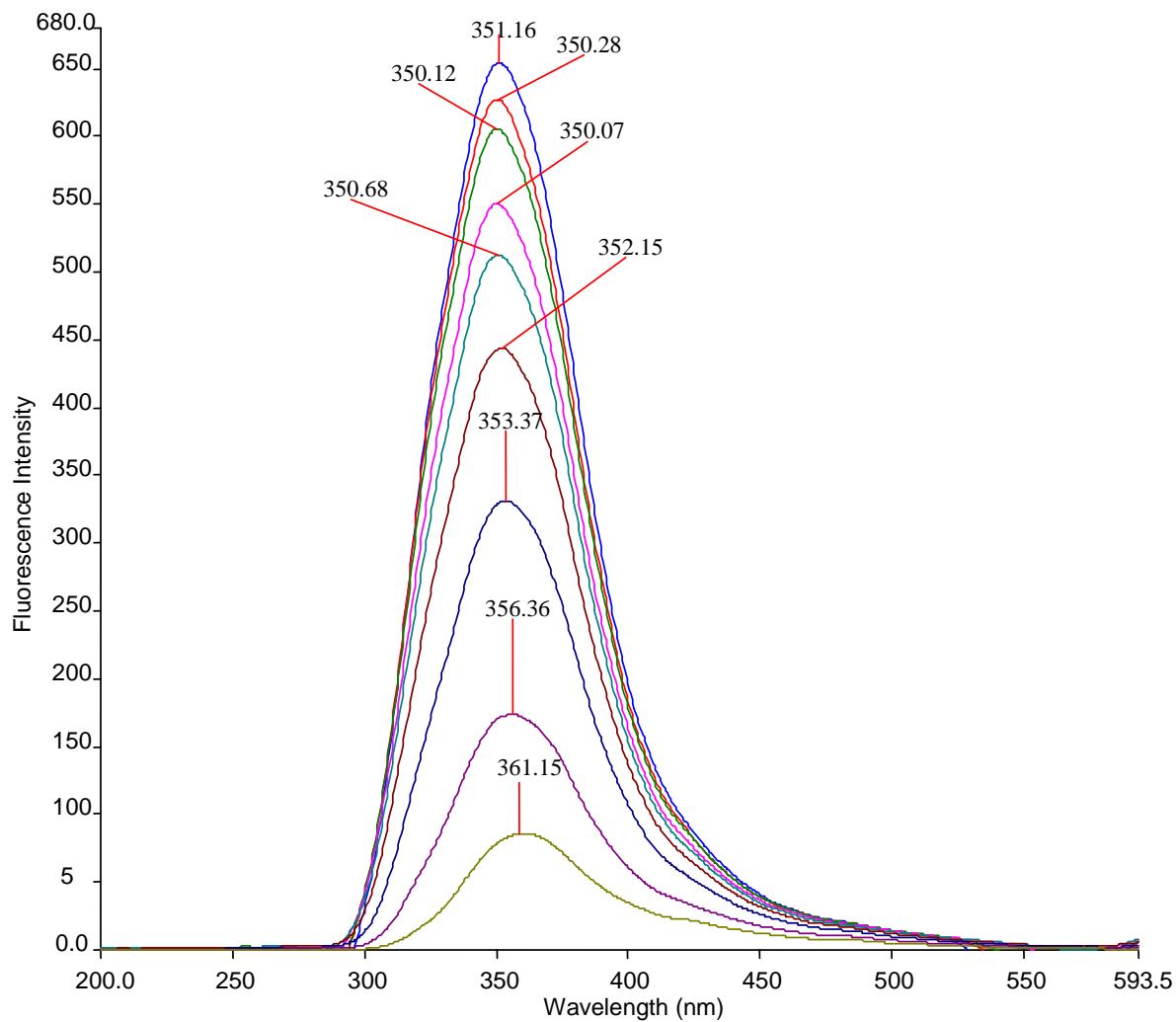


Figure S4.3(I) Fluorescence quenching of 0.24 mM BSA by increasing concentration [0 (a), 2 (b), 4 (c), 10 (d), 20 (e), 40 (f), 100 (g), 200 (h) and 400 (i) μ M] of L-[Cu(phen)(5MeOCA)(H₂O)NO₃] **3**. (First trial)

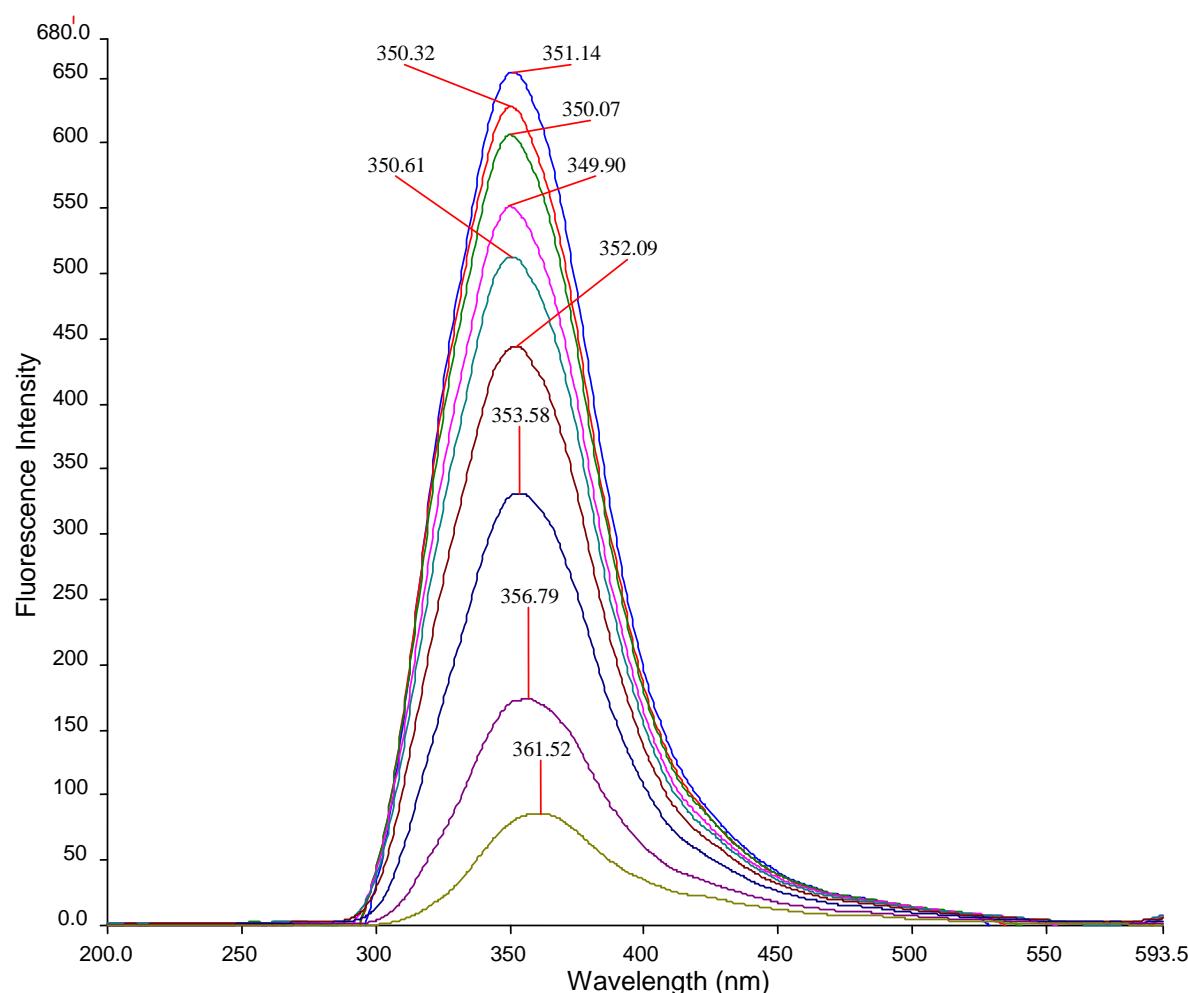


Figure S4.3(II) Fluorescence quenching of 0.24 mM BSA by increasing concentration [0 (a), 2 (b), 4 (c), 10 (d), 20 (e), 40 (f), 100 (g), 200 (h) and 400 (i) μM] of $L\text{-}[\text{Cu}(\text{phen})(5\text{MeOCA})(\text{H}_2\text{O})\text{NO}_3]$ (3.. (Second trial)

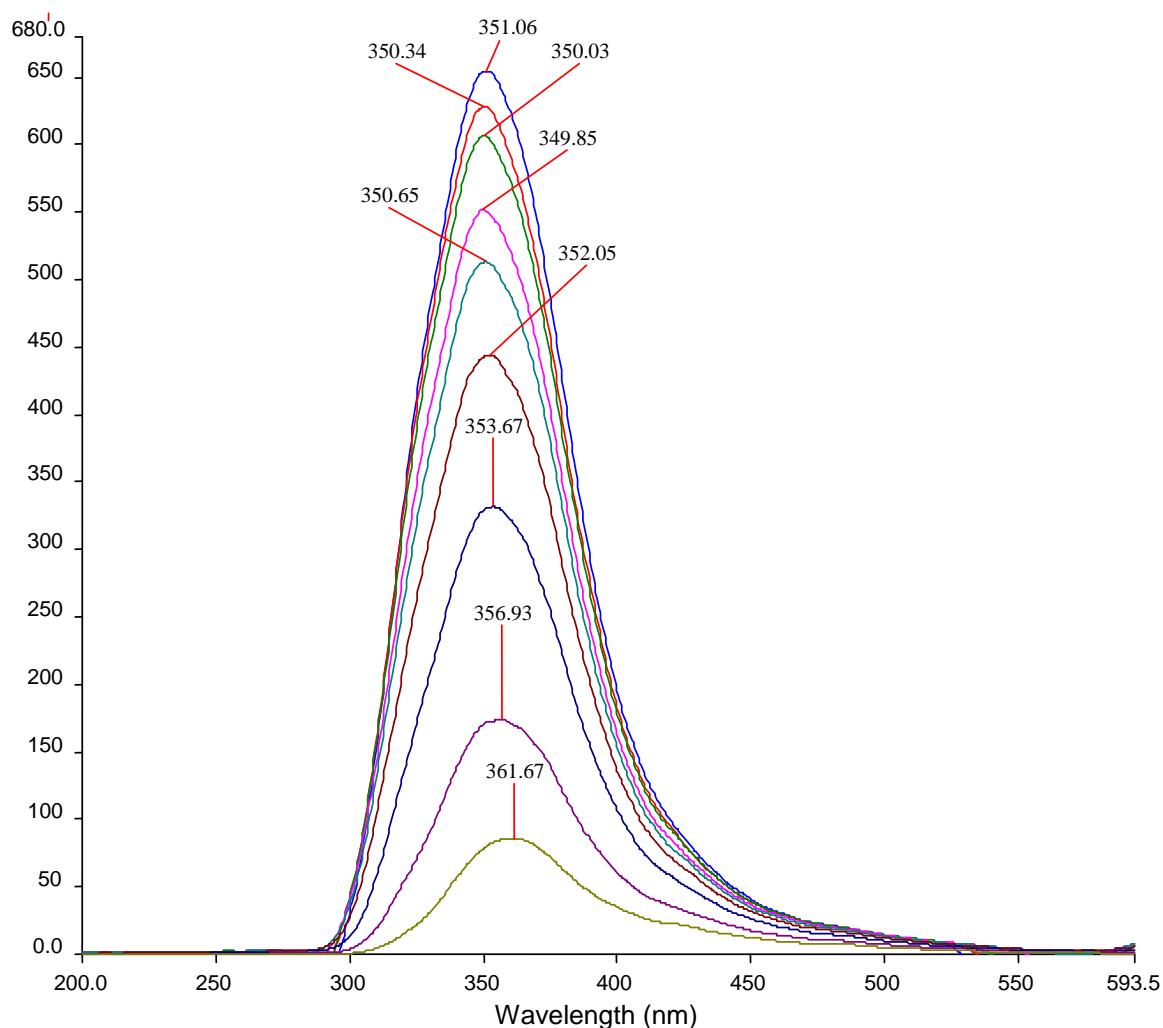


Figure S4.3(III) Fluorescence quenching of 0.24 mM BSA by increasing concentration [0 (a), 2 (b), 4 (c), 10 (d), 20 (e), 40 (f), 100 (g), 200 (h) and 400 (i) μ M] of L-[Cu(phen)(5MeOCA)(H₂O)NO₃] **3**. (Third trial)

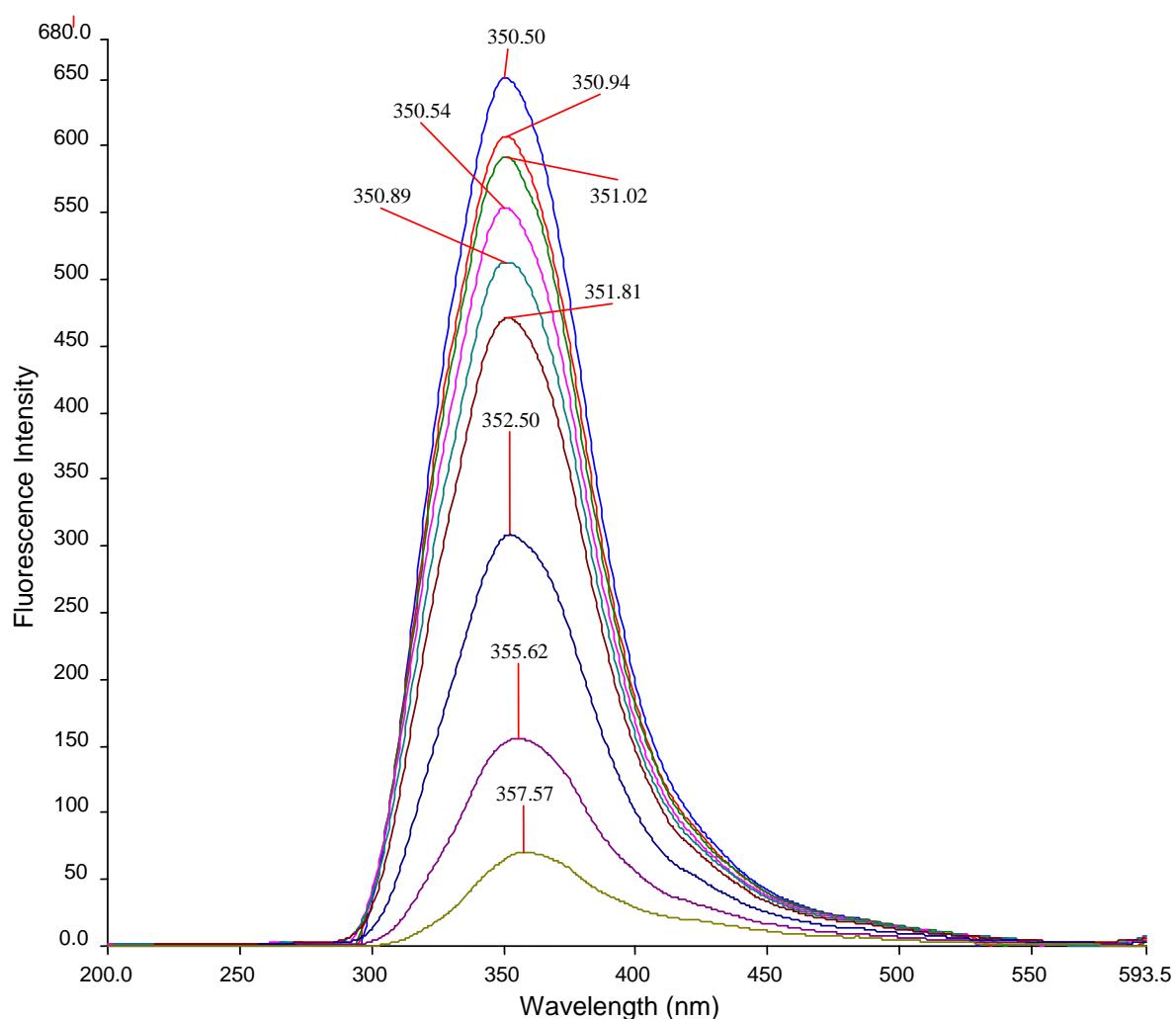


Figure S4.2(I) Fluorescence quenching of 0.24 mM BSA by increasing concentration [0 (a), 2 (b), 4 (c), 10 (d), 20 (e), 40 (f), 100 (g), 200 (h) and 400 (i) μ M] of $[\text{Cu}(\text{phen})(\text{D-threo})(\text{H}_2\text{O})\text{NO}_3]$ **2**. (First trial)

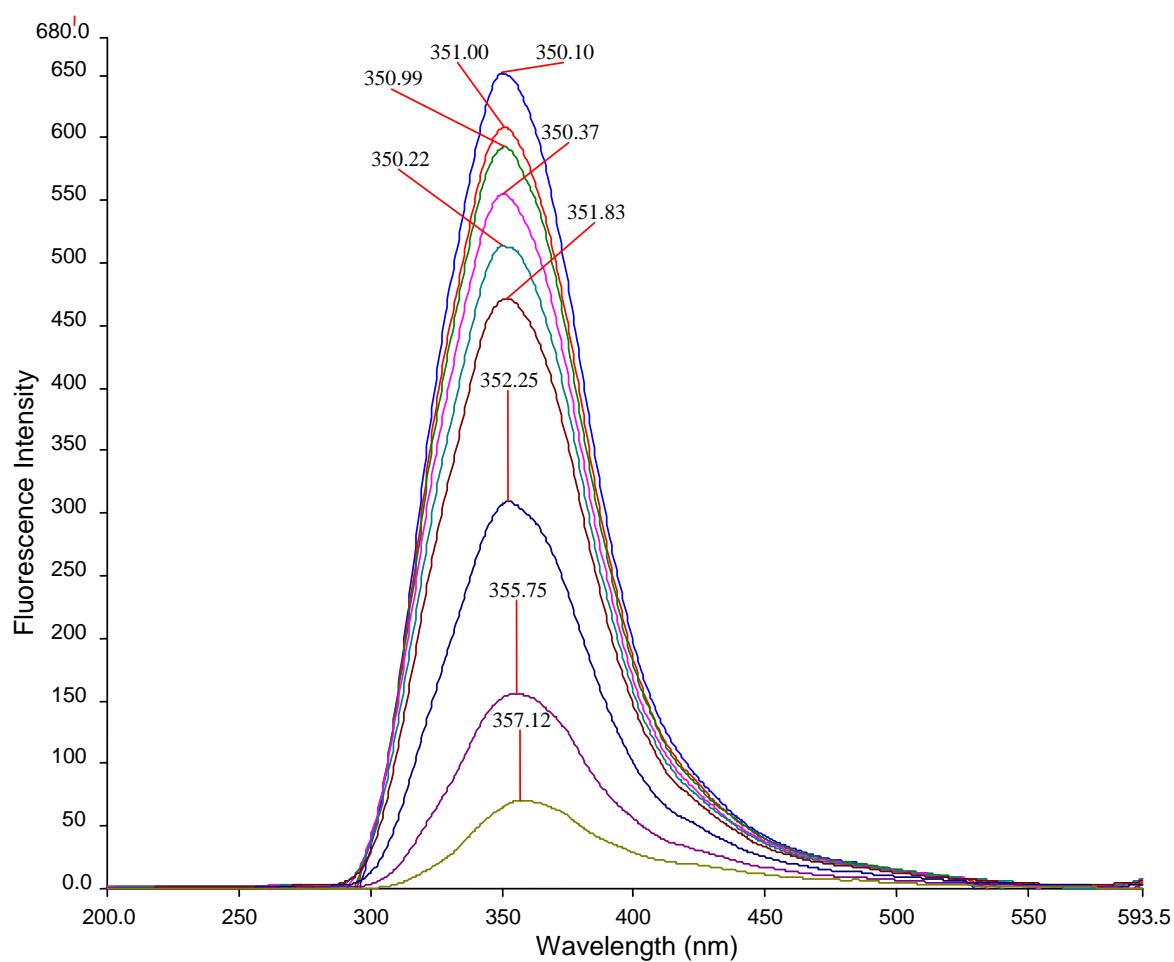


Figure S4.2(II) Fluorescence quenching of 0.24 mM BSA by increasing concentration [0 (a), 2 (b), 4 (c), 10 (d), 20 (e), 40 (f), 100 (g), 200 (h) and 400 (i) μ M] of $[\text{Cu}(\text{phen})(\text{D-threo})(\text{H}_2\text{O})\text{NO}_3]$ **2**. (Second trial)

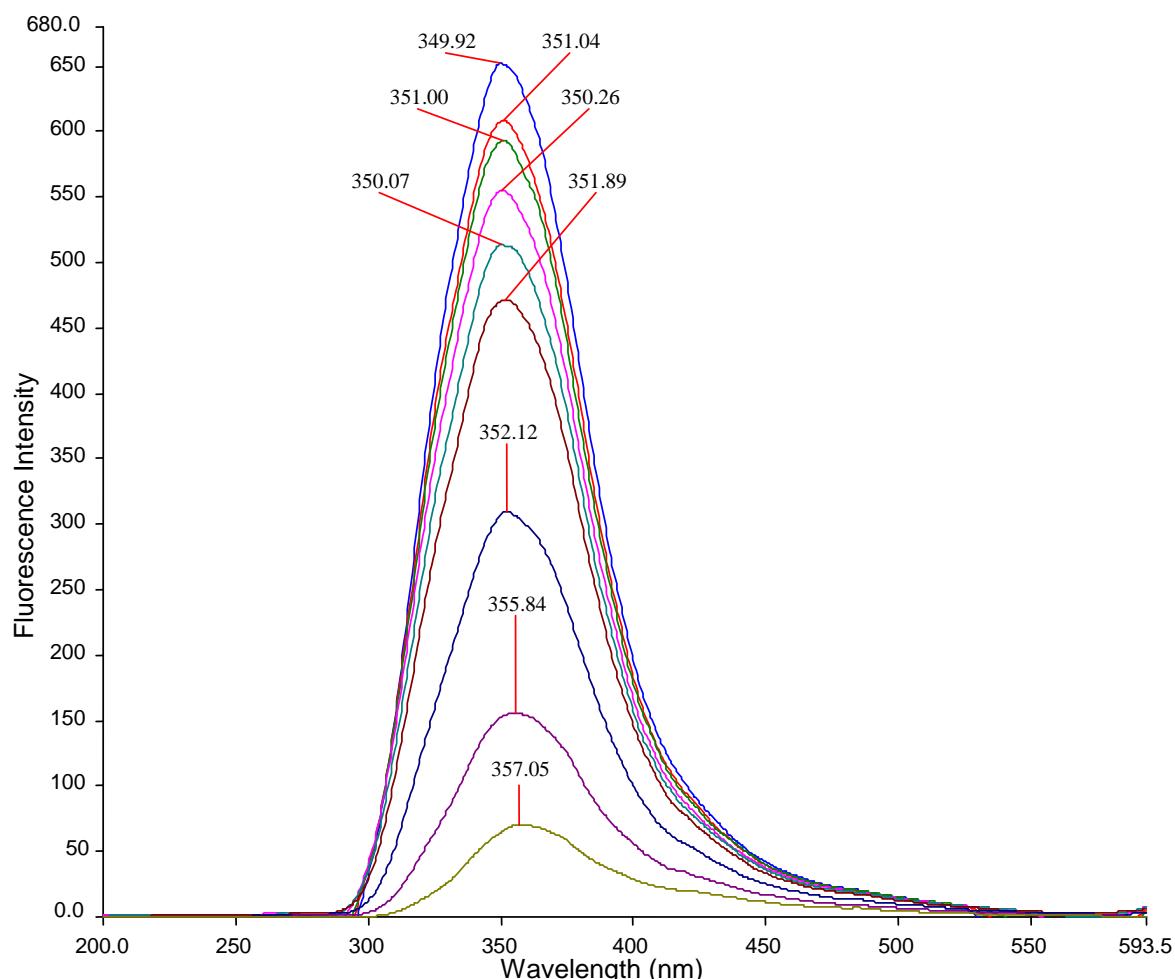


Figure S4.2(III) Fluorescence quenching of 0.24 mM BSA by increasing concentration [0 (a), 2 (b), 4 (c), 10 (d), 20 (e), 40 (f), 100 (g), 200 (h) and 400 (i) μ M] of $[\text{Cu}(\text{phen})(\text{D-threo})(\text{H}_2\text{O})\text{NO}_3]$ **2**. (Third trial)

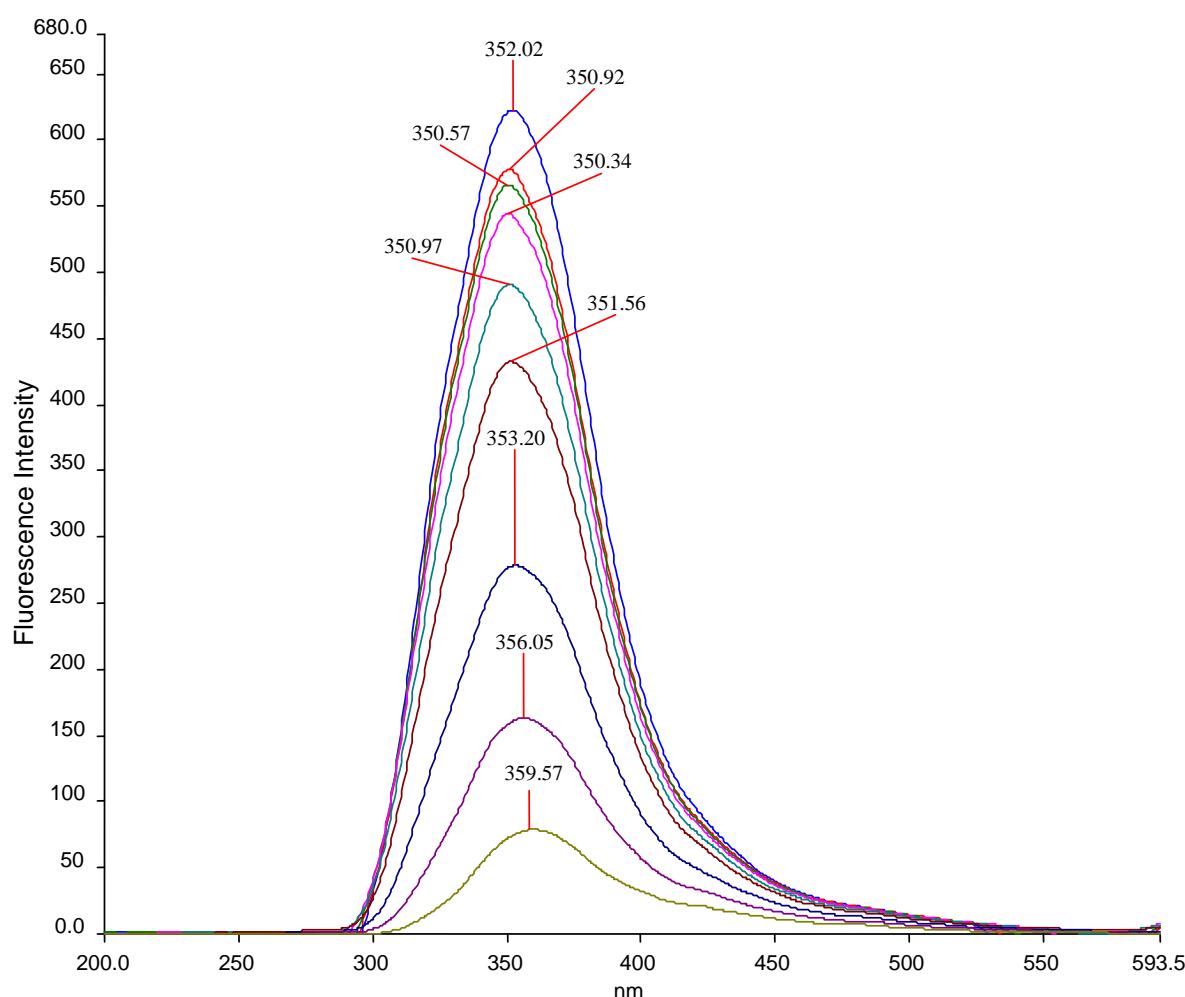


Figure S4.1(I) Fluorescence quenching of 0.24 mM BSA by increasing concentration [0 (a), 2 (b), 4 (c), 10 (d), 20 (e), 40 (f), 100 (g), 200 (h) and 400 (i) μ M] of $[\text{Cu}(\text{phen})(\text{L-threo})(\text{H}_2\text{O})\text{NO}_3]$ **1**. (First trial)

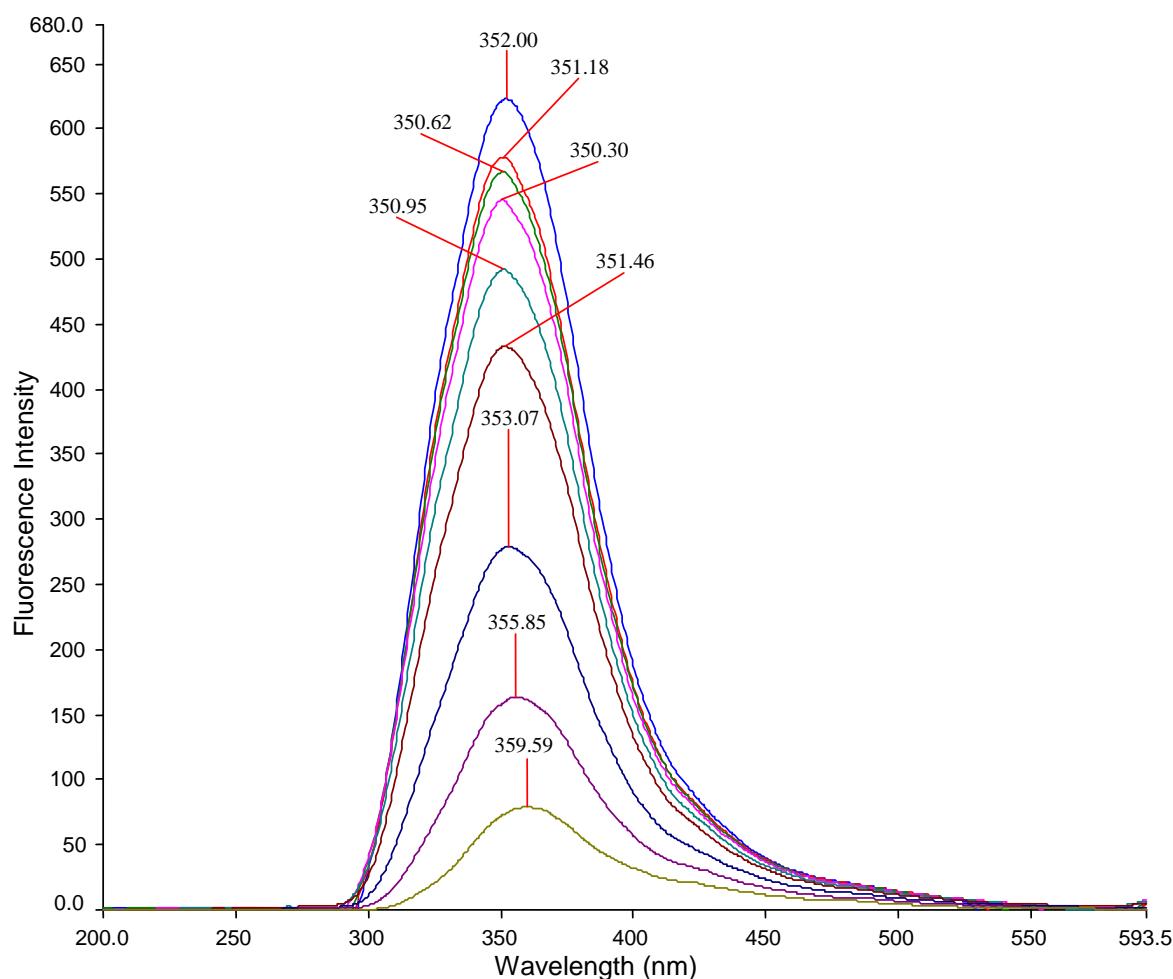


Figure S4.1(II) Fluorescence quenching of 0.24 mM BSA by increasing concentration [0 (a), 2 (b), 4 (c), 10 (d), 20 (e), 40 (f), 100 (g), 200 (h) and 400 (i) μ M] of $[\text{Cu}(\text{phen})(\text{L-threo})(\text{H}_2\text{O})\text{NO}_3]$ **1**. (Second trial)

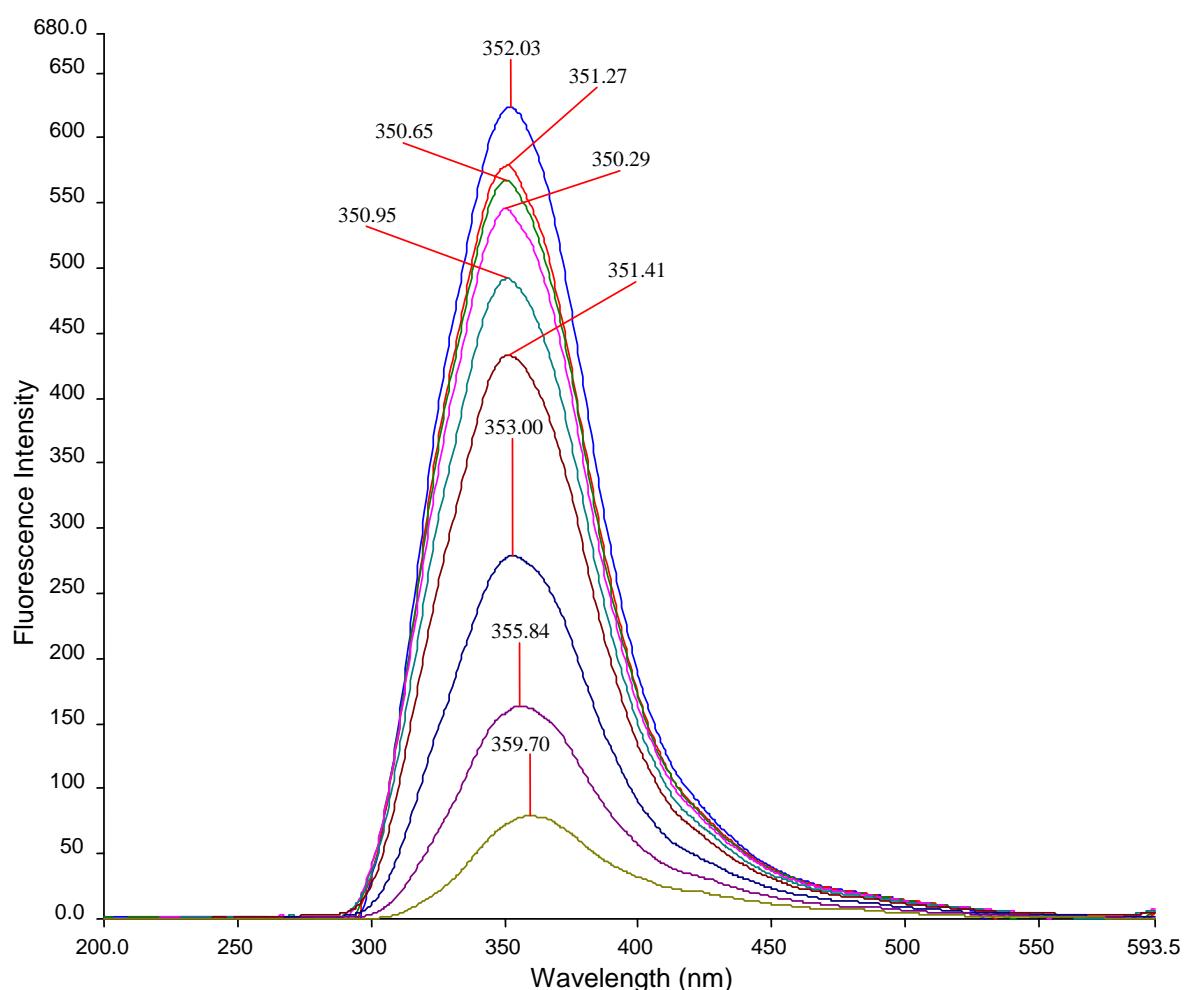


Figure S4.1(III) Fluorescence quenching of 0.24 mM BSA by increasing concentration [0 (a), 2 (b), 4 (c), 10 (d), 20 (e), 40 (f), 100 (g), 200 (h) and 400 (i) μM] of $[\text{Cu}(\text{phen})(\text{L-threo})(\text{H}_2\text{O})\text{NO}_3]$ **1**. (Third trial)