

Supporting Information

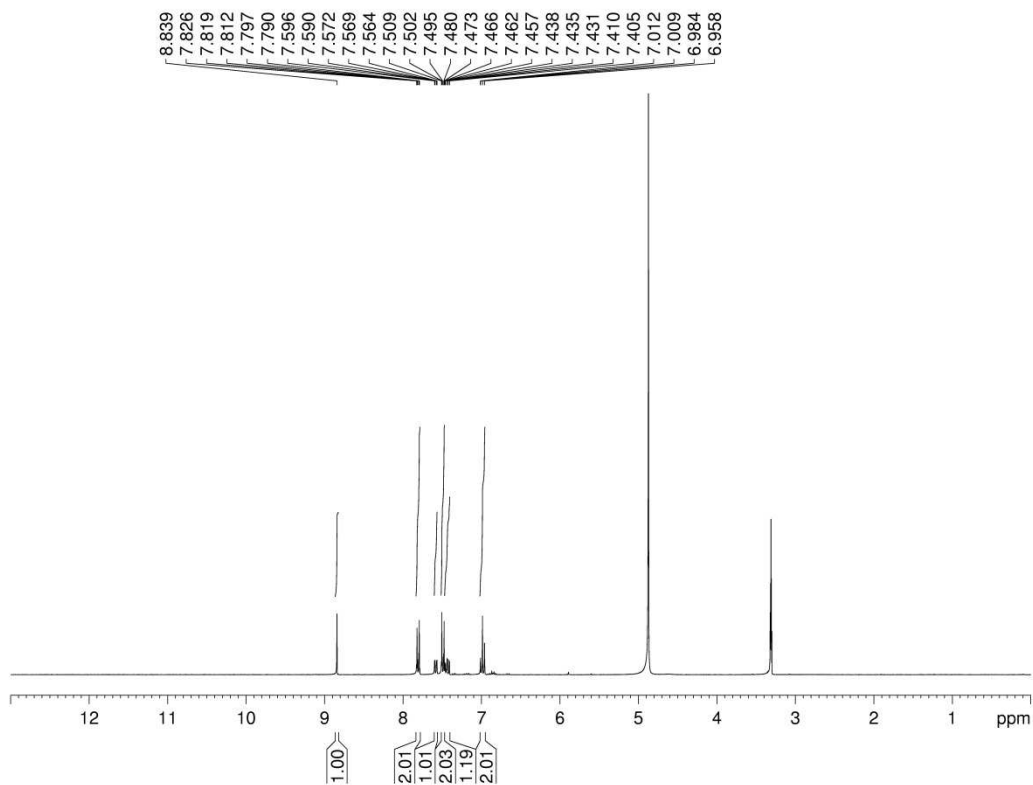
Ligand dissociation/recoordination in fluorescent ionic zinc–salicylideneimine compounds: synthesis, characterization, photophysical properties, and ¹H NMR studies

Ho-Wen Chiang, Yo-Ting Su and Jing-Yun Wu*

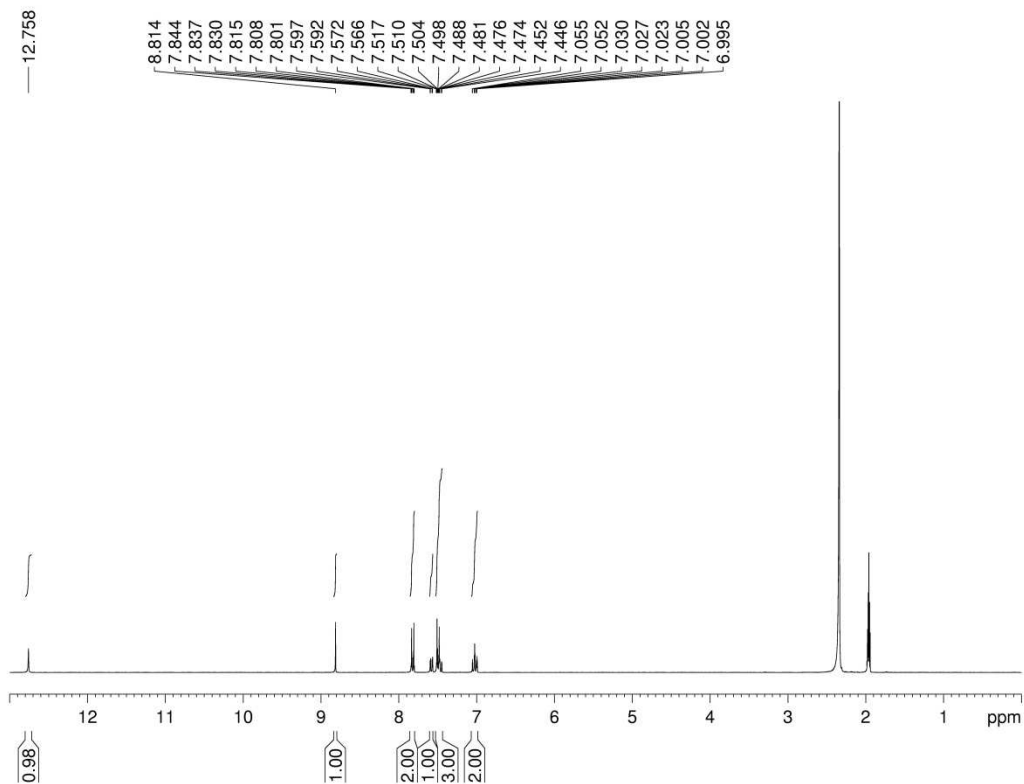
Department of Applied Chemistry, National Chi Nan University, Nantou 545, Taiwan. Fax:

+886-49-2917956; Tel: +886-49-2910960-4918; E-mail: jyunwu@ncnu.edu.tw

(a)



(b)



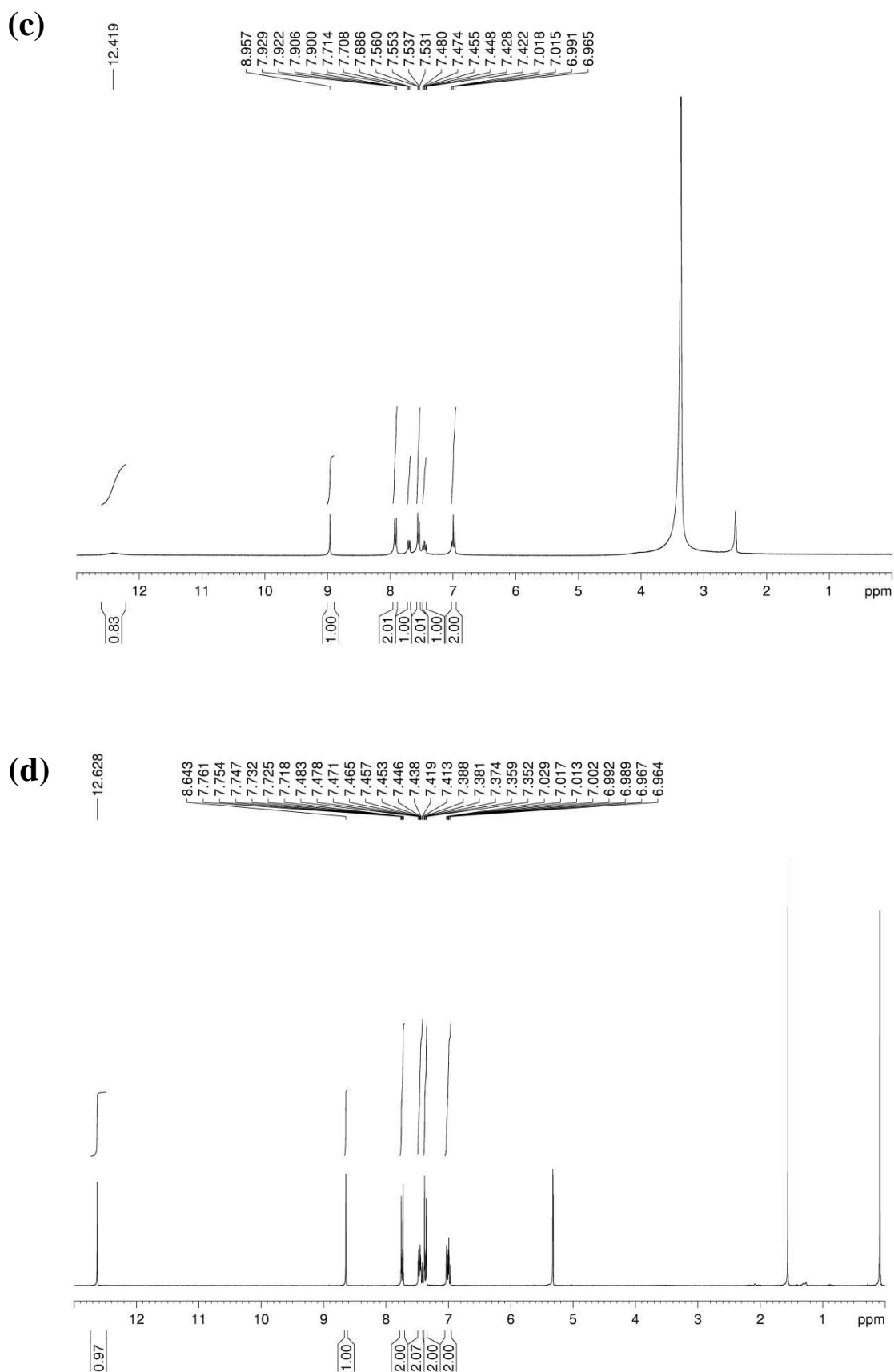
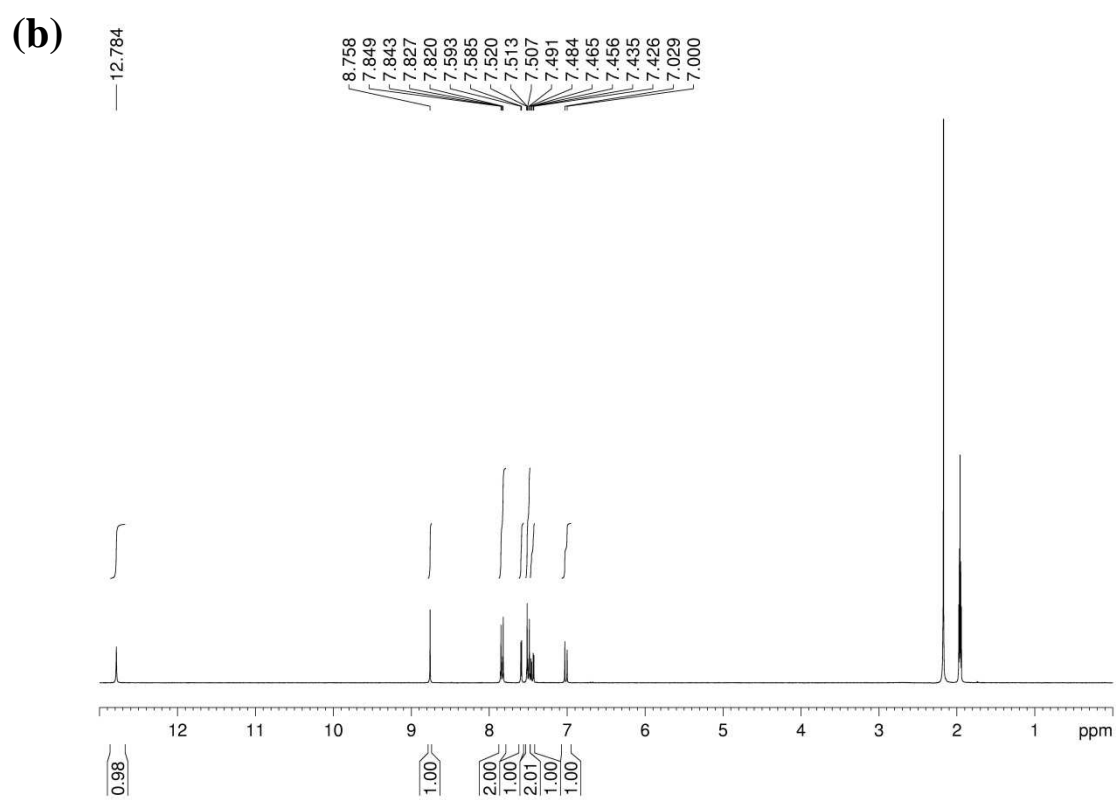
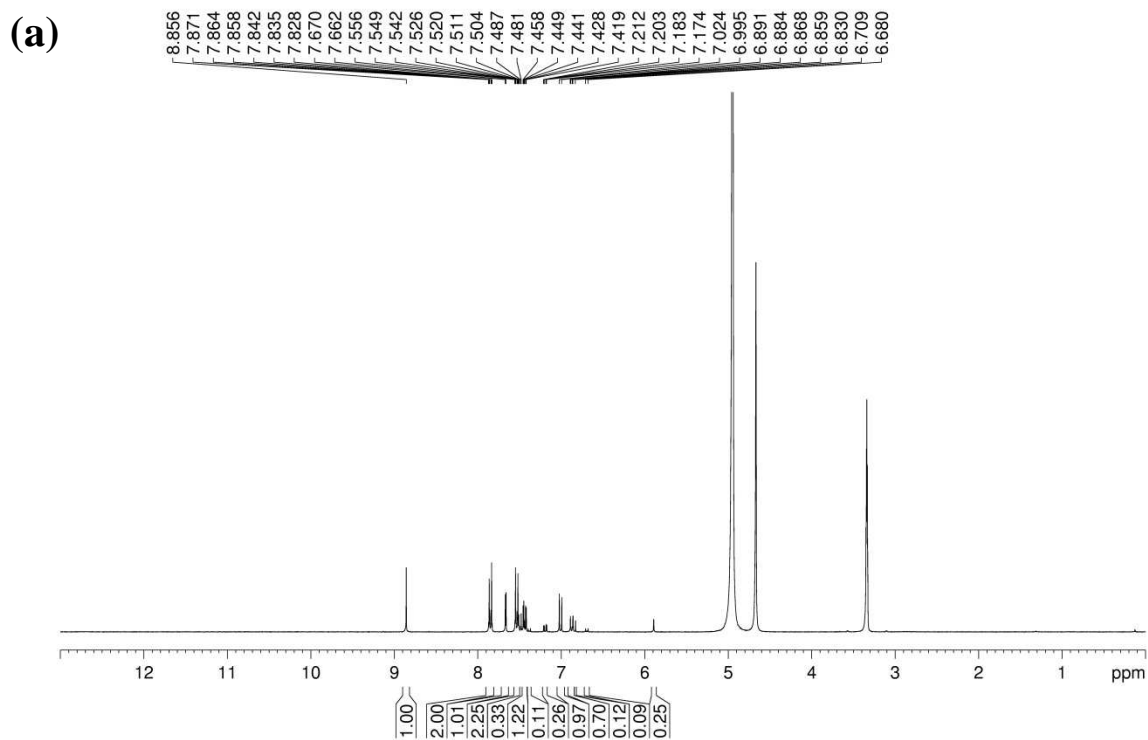


Fig. S1 ^1H NMR spectra of 4-(salicylideneimino)benzointrile ($\text{Hsal}^{\text{H}}\text{-4-CN}$) in (a) $\text{MeOH-}d_4$, (b) $\text{ACN-}d_3$, (c) $\text{DMSO-}d_6$, and (d) $\text{DCM-}d_2$.



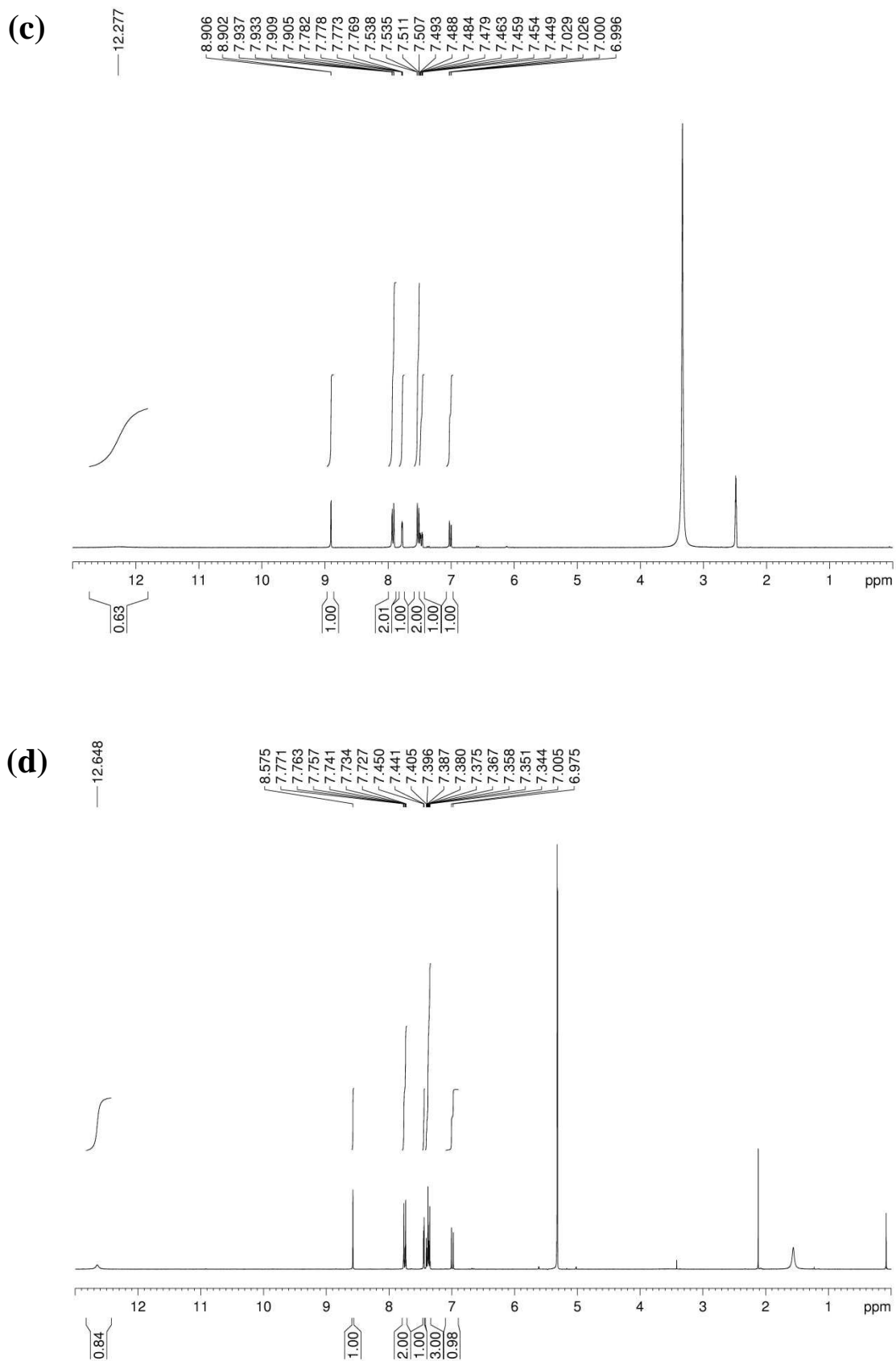
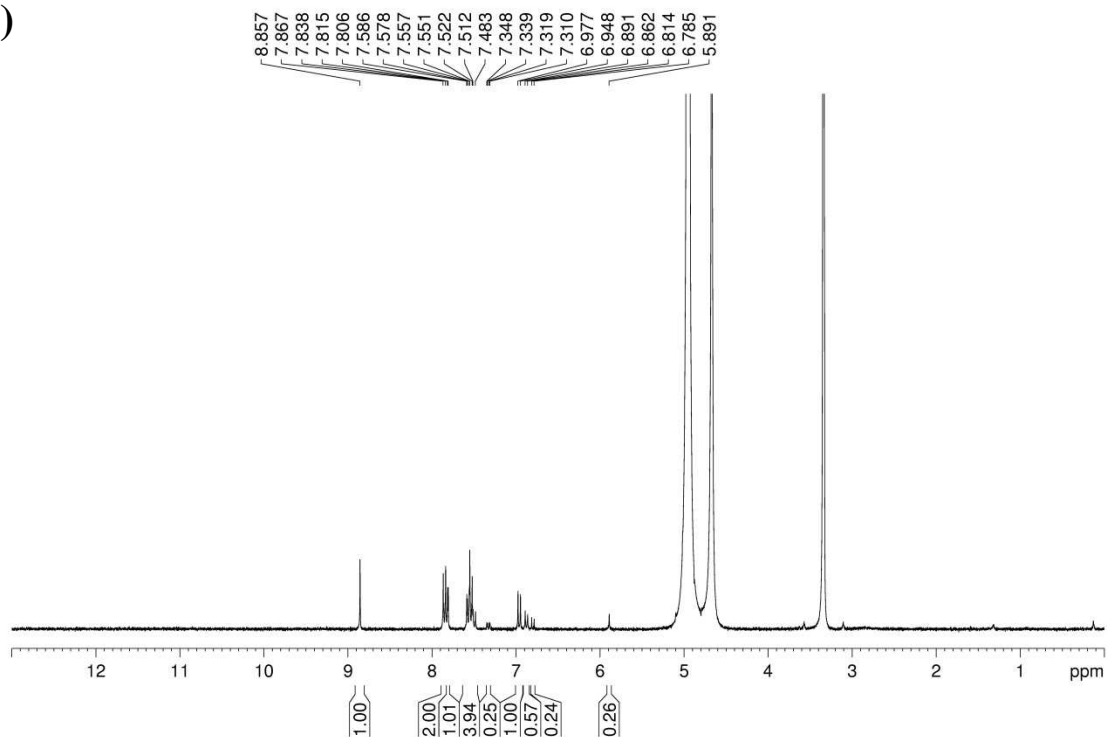
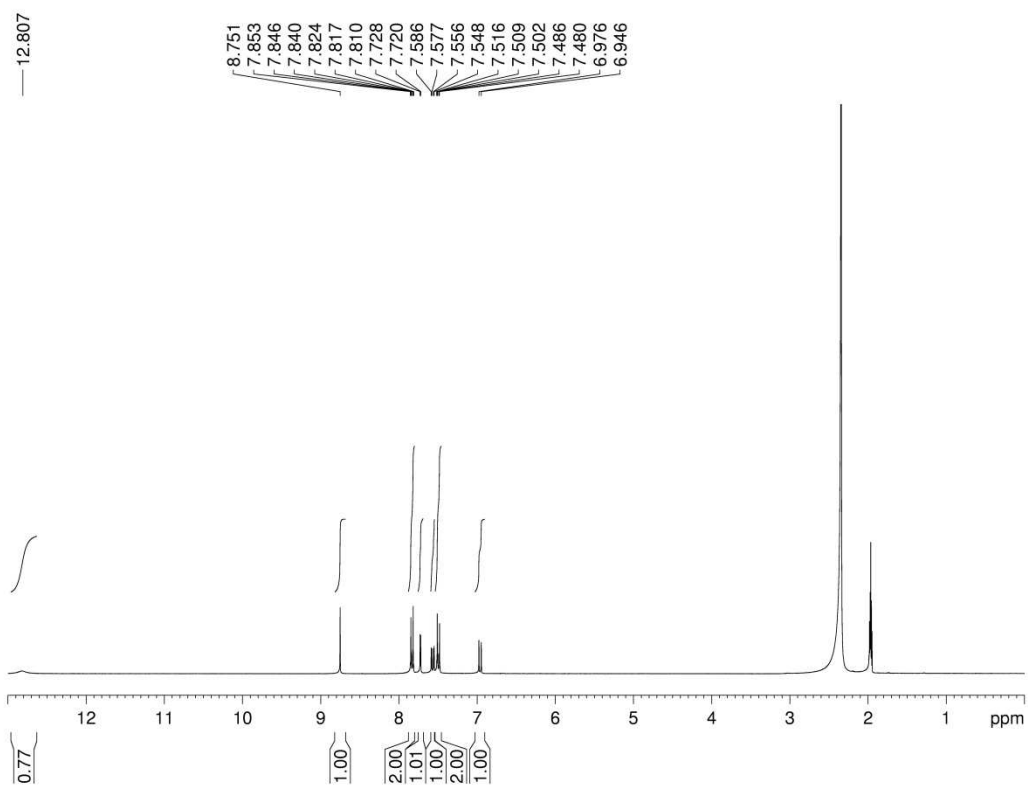


Fig. S2 ^1H NMR spectra of 4-(5'-chlorosalicylideneimino)benzonitrile ($\text{Hsal}^{\text{Cl}}\text{-4-CN}$) in (a) $\text{MeOH-}d_4$, (b) $\text{ACN-}d_3$, (c) $\text{DMSO-}d_6$, and (d) $\text{DCM-}d_2$.

(a)



(b)



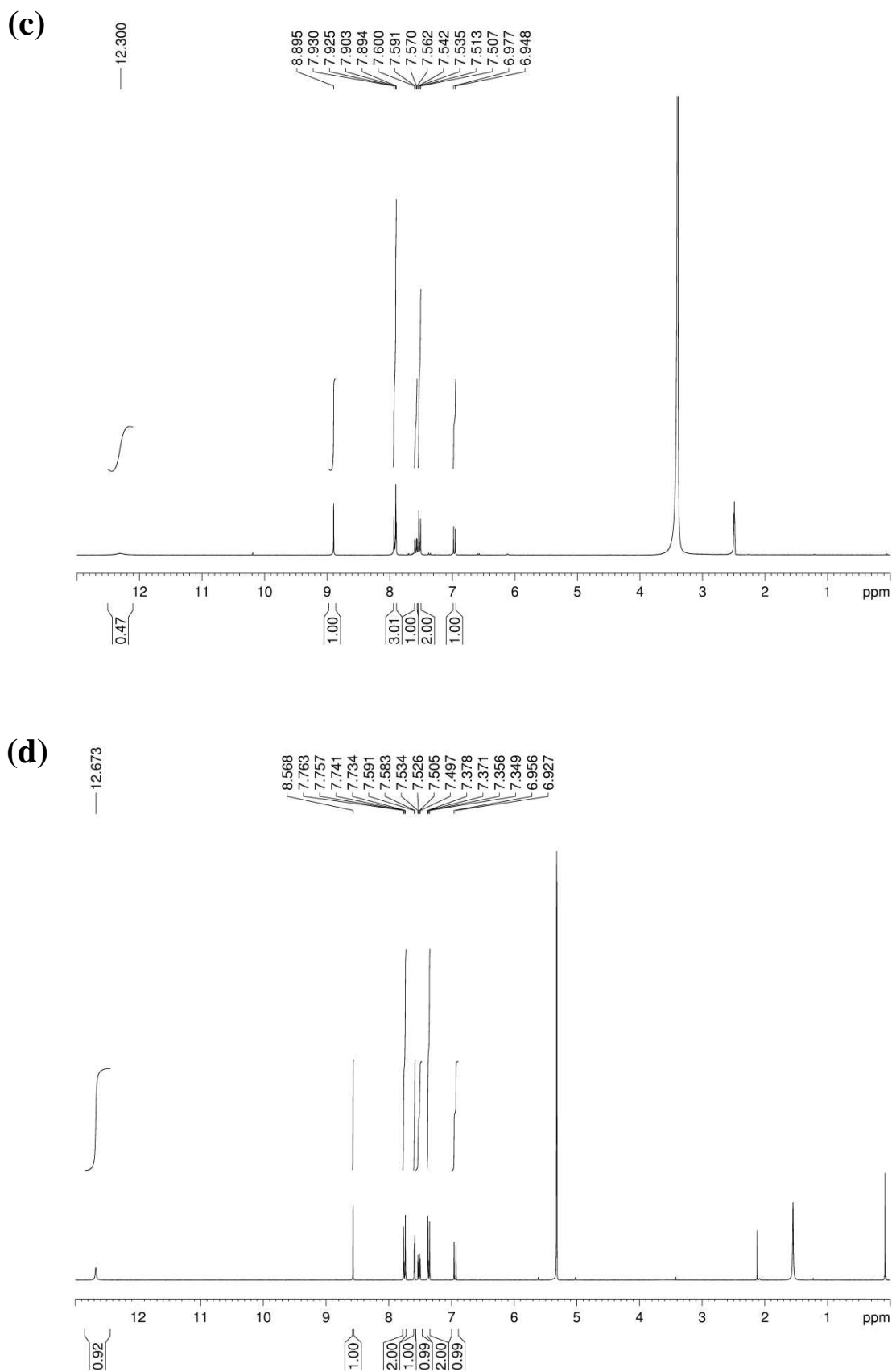
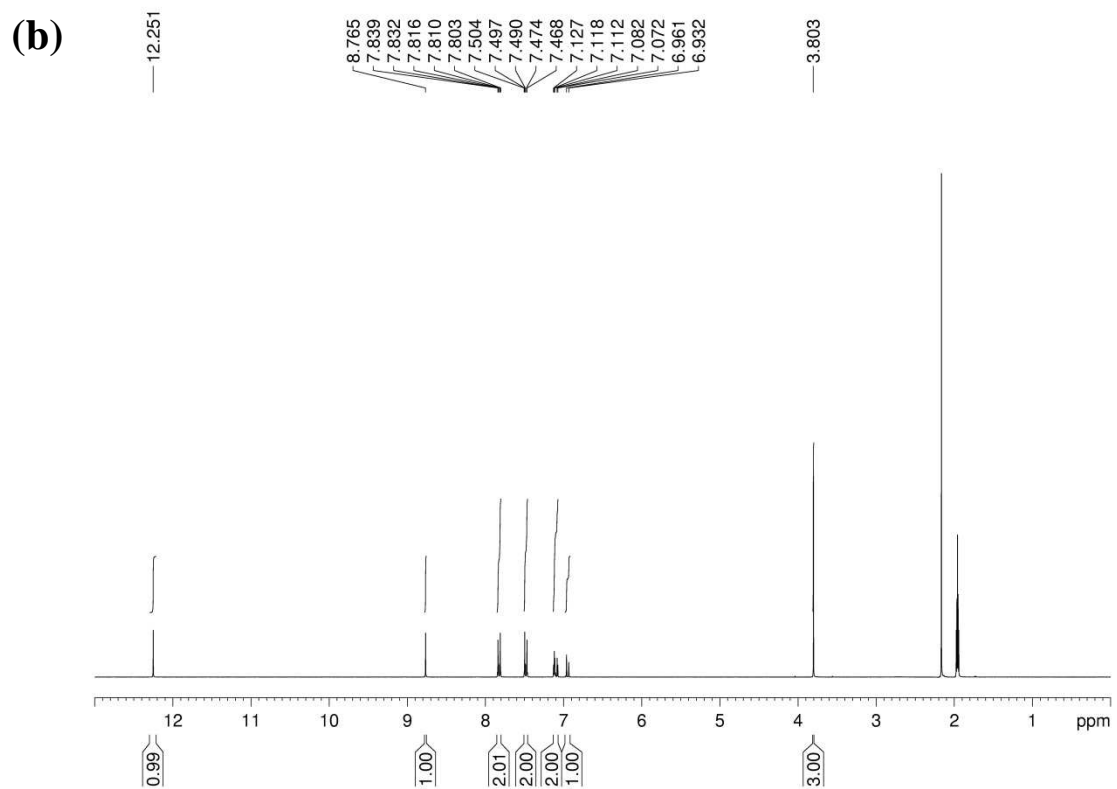
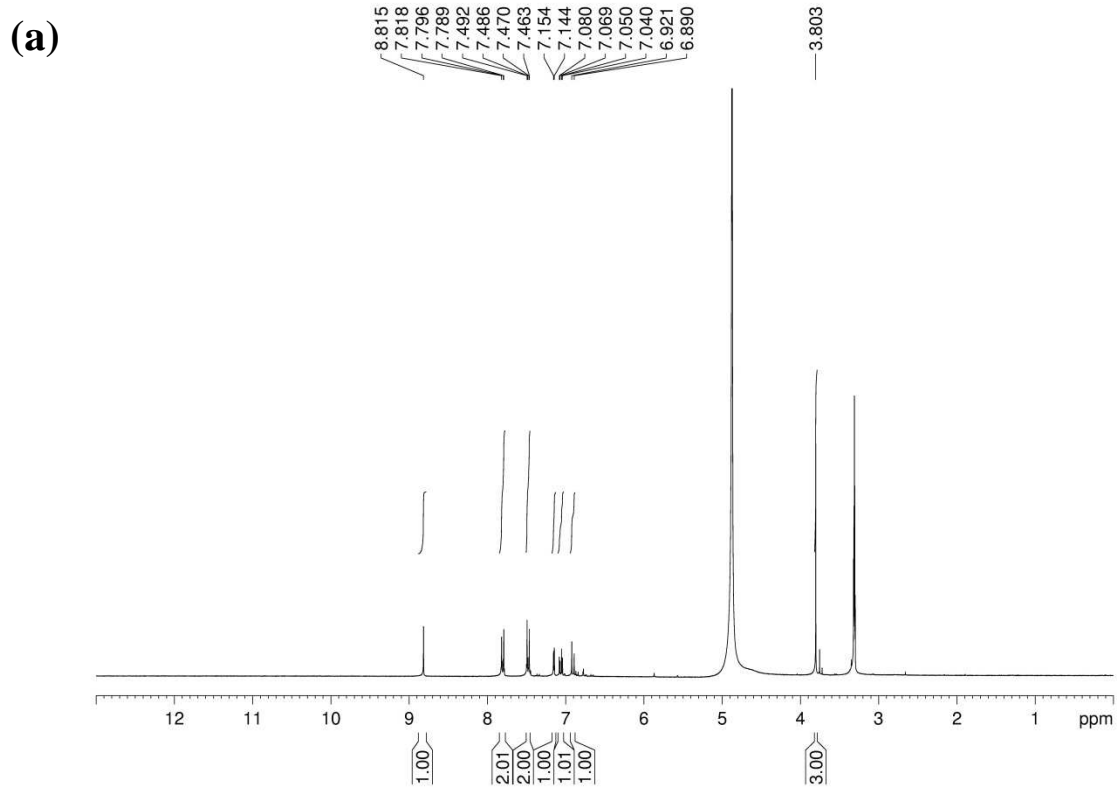


Fig. S3 ^1H NMR spectra of 4-(5'-bromosalicylideneimino)benzotrile ($\text{Hsal}^{\text{Br}}\text{-4-CN}$) in (a) $\text{MeOH-}d_4$, (b) $\text{ACN-}d_3$, (c) $\text{DMSO-}d_6$, and (d) $\text{DCM-}d_2$.



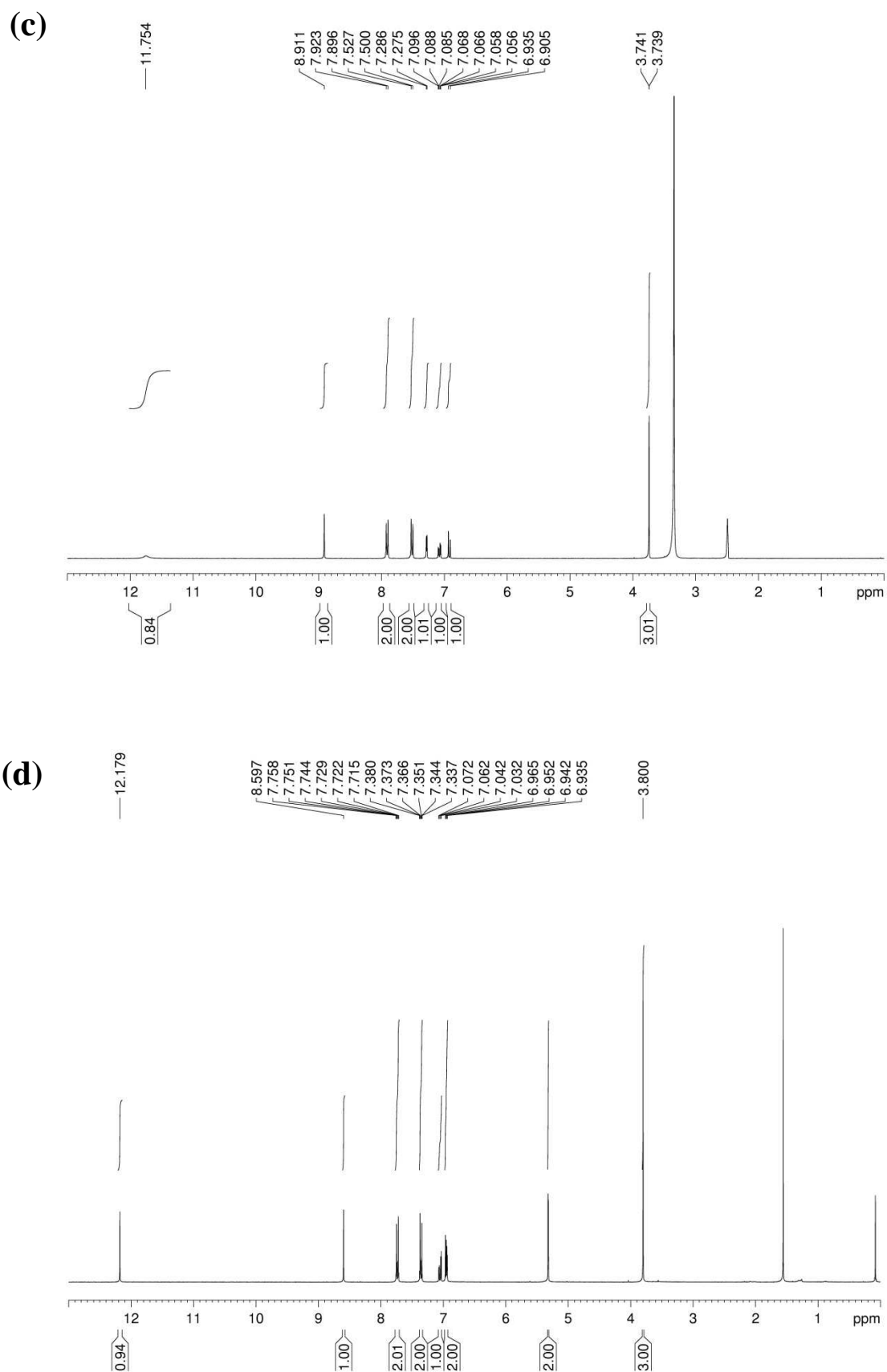
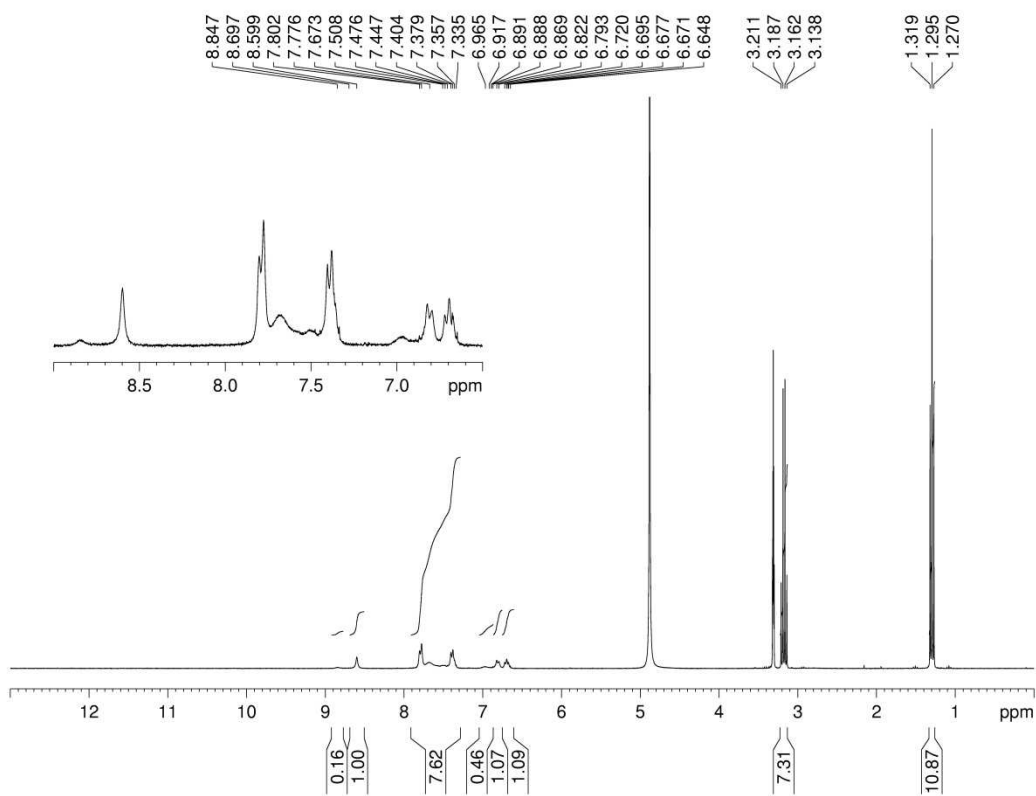
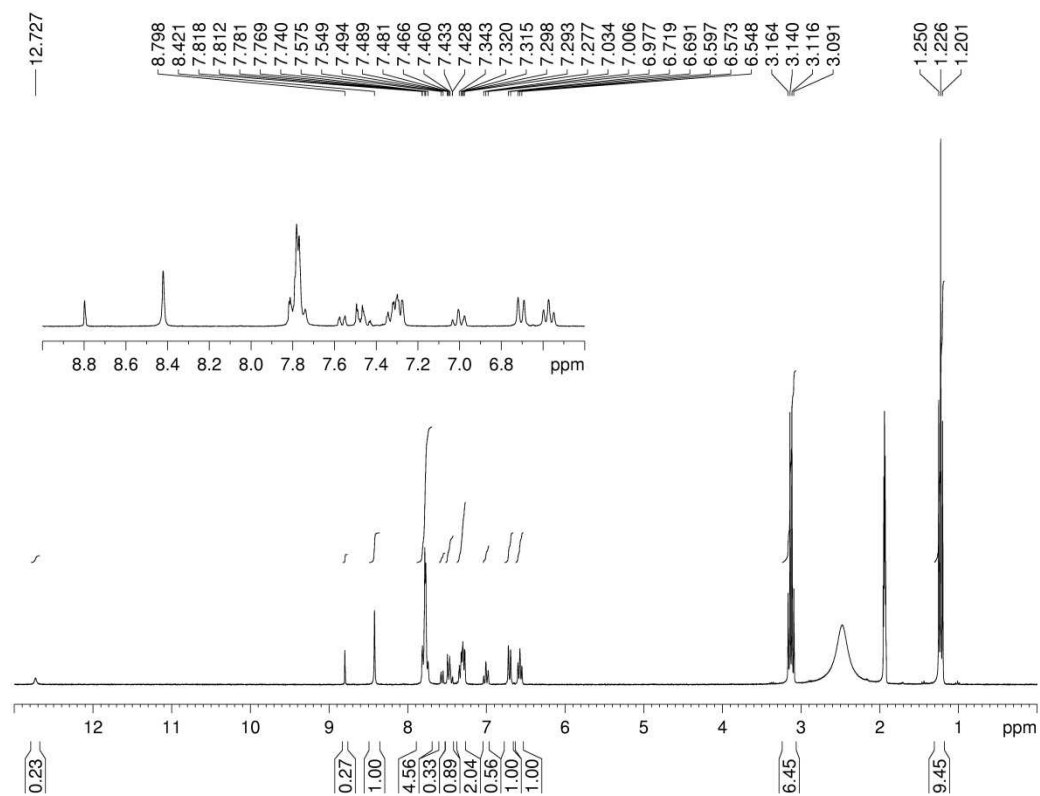


Fig. S4 ^1H NMR spectra of 4-(5'-methoxysalicylideneimino)benzonitrile (Hsal^{OMe}-4-CN) in (a) MeOH- d_4 , (b) ACN- d_3 , (c) DMSO- d_6 , and (d) DCM- d_2 .

(a)



(b)



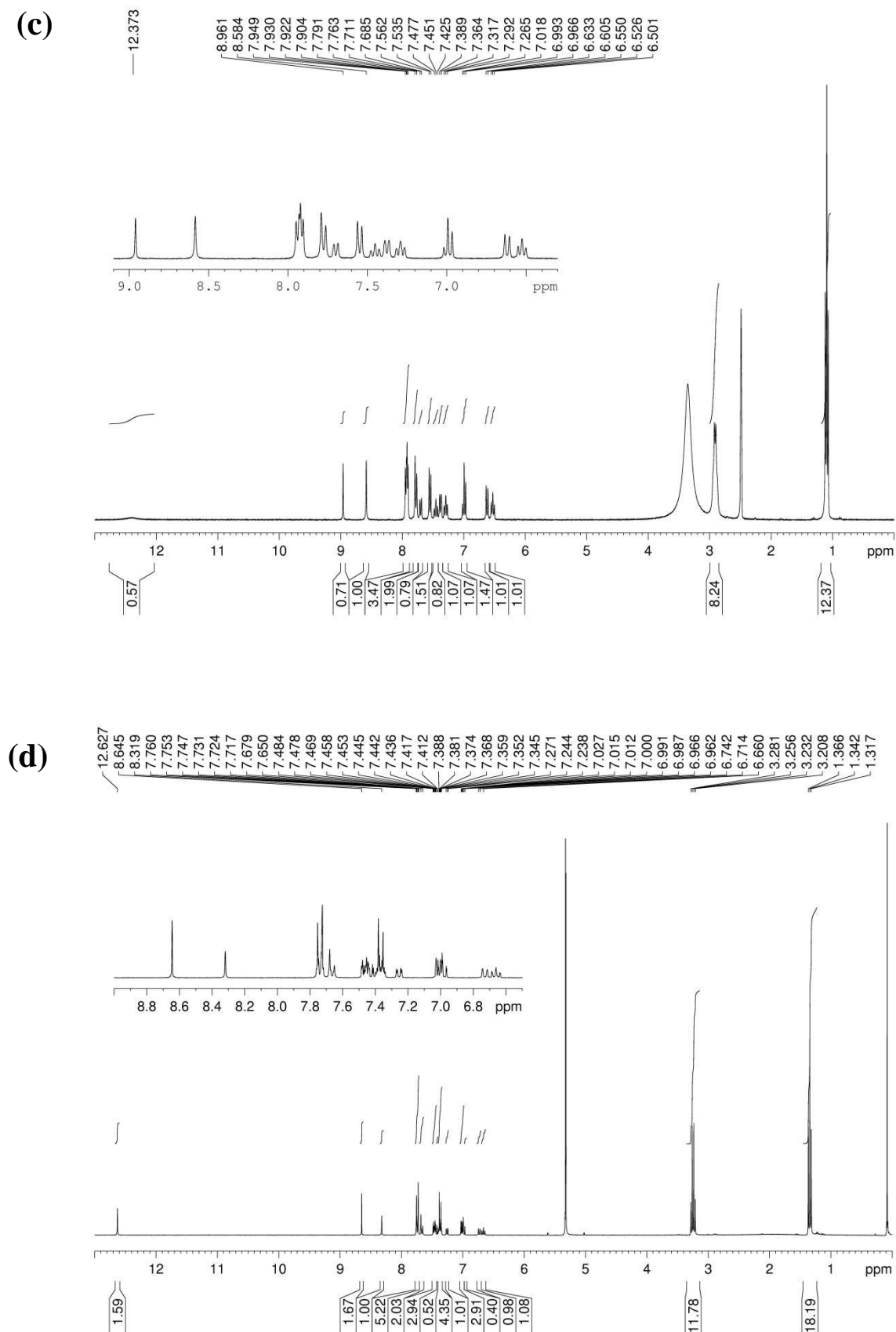
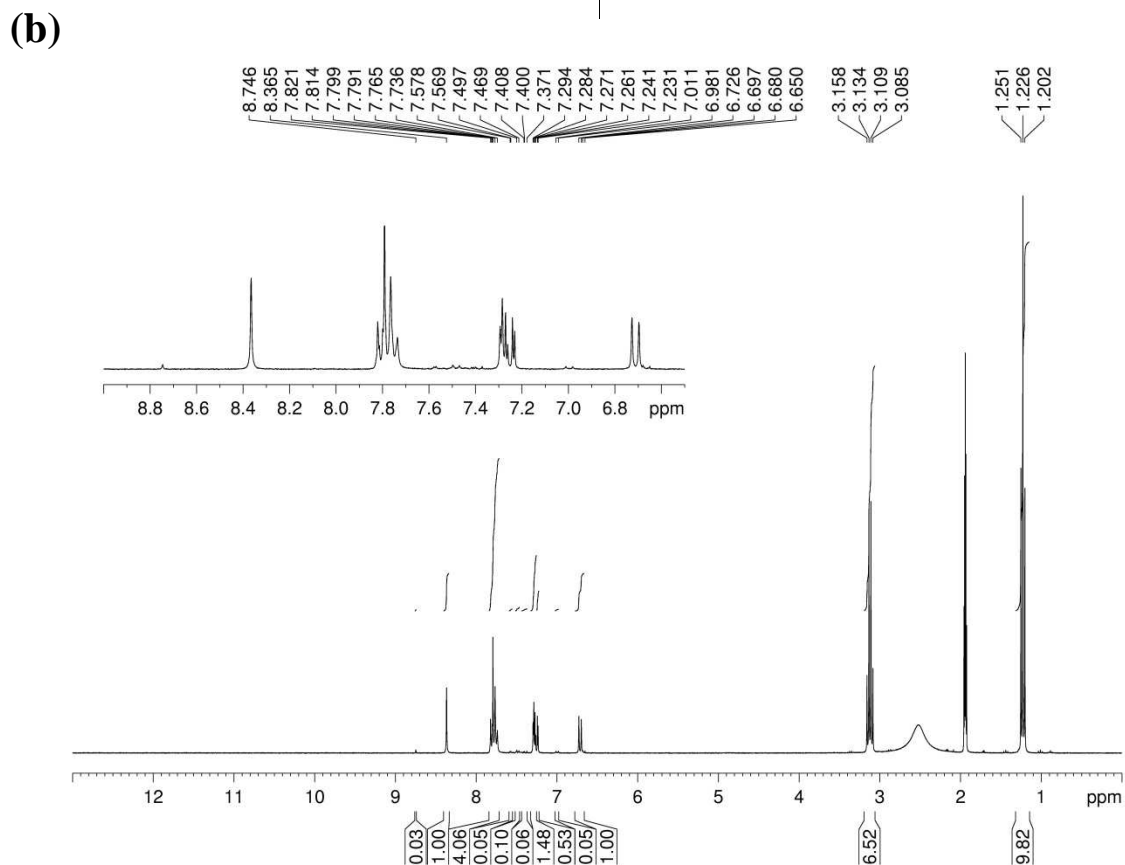
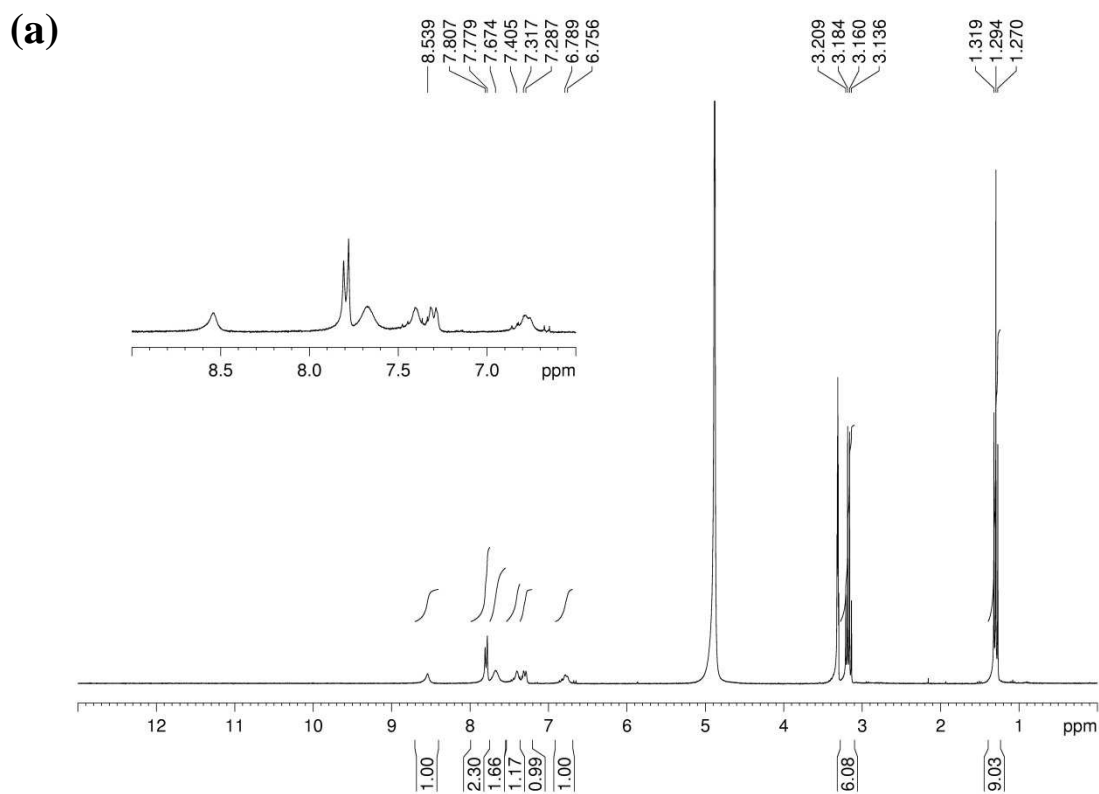


Fig. S5 ^1H NMR spectra of $[\text{HNEt}_3][\text{Zn}(\text{sal}^{\text{H}}\text{-4-CN})\text{Cl}_2]$ (**1**) in (a) $\text{MeOH-}d_4$, (b) $\text{ACN-}d_3$, (c) $\text{DMSO-}d_6$, and (d) $\text{DCM-}d_2$.



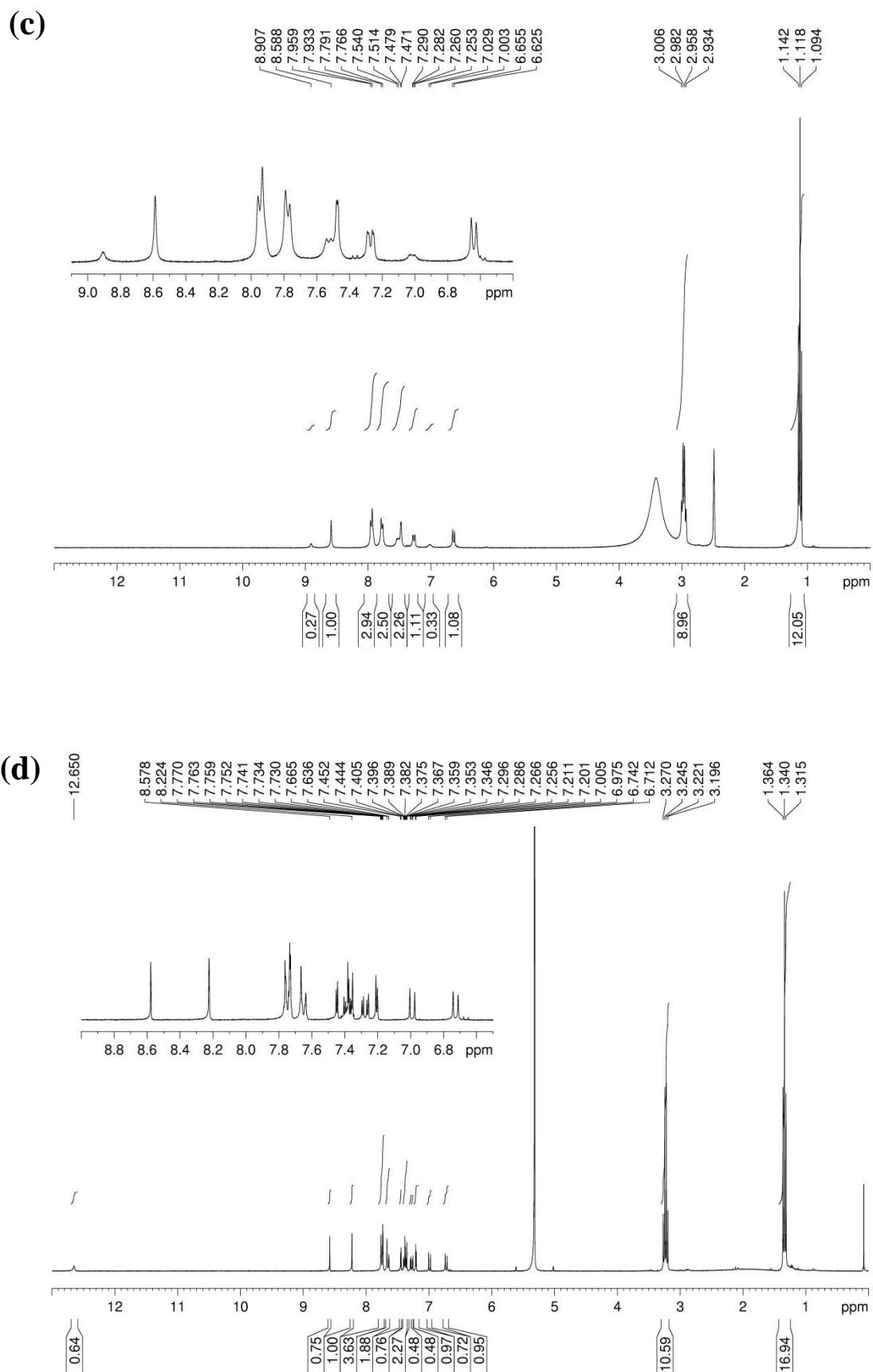
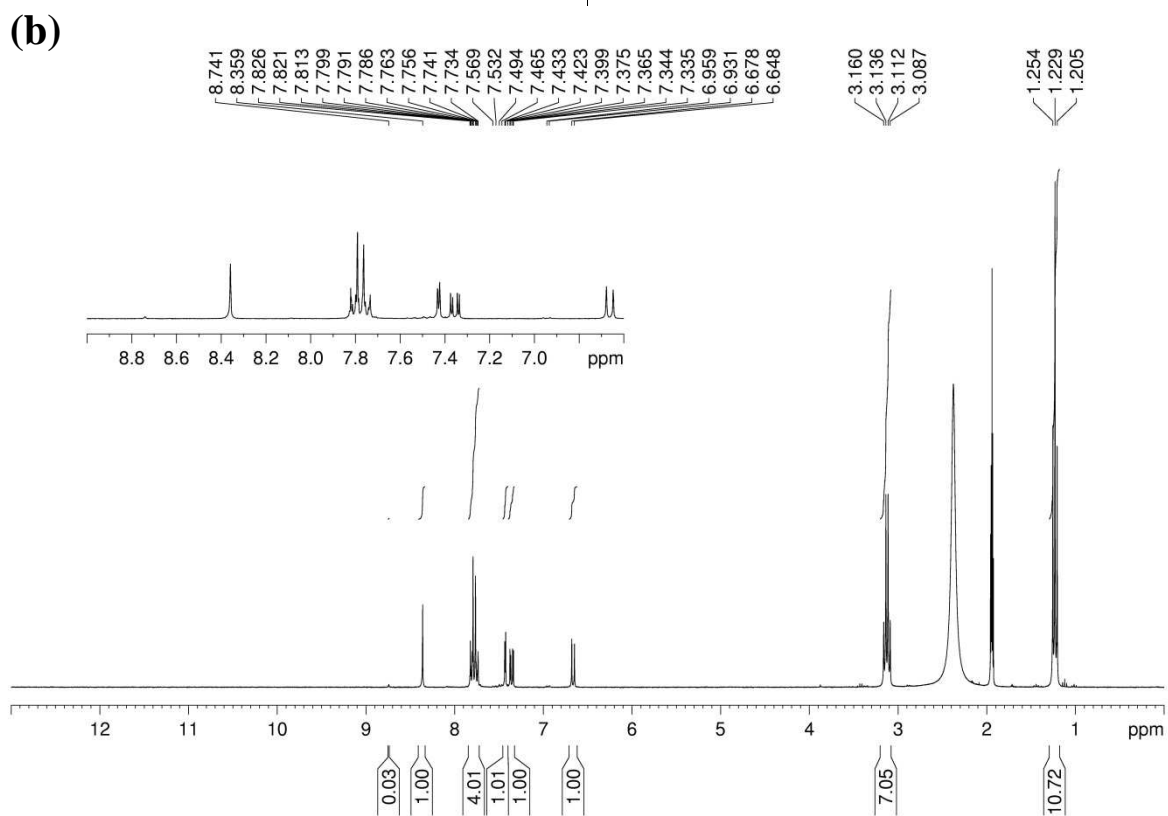
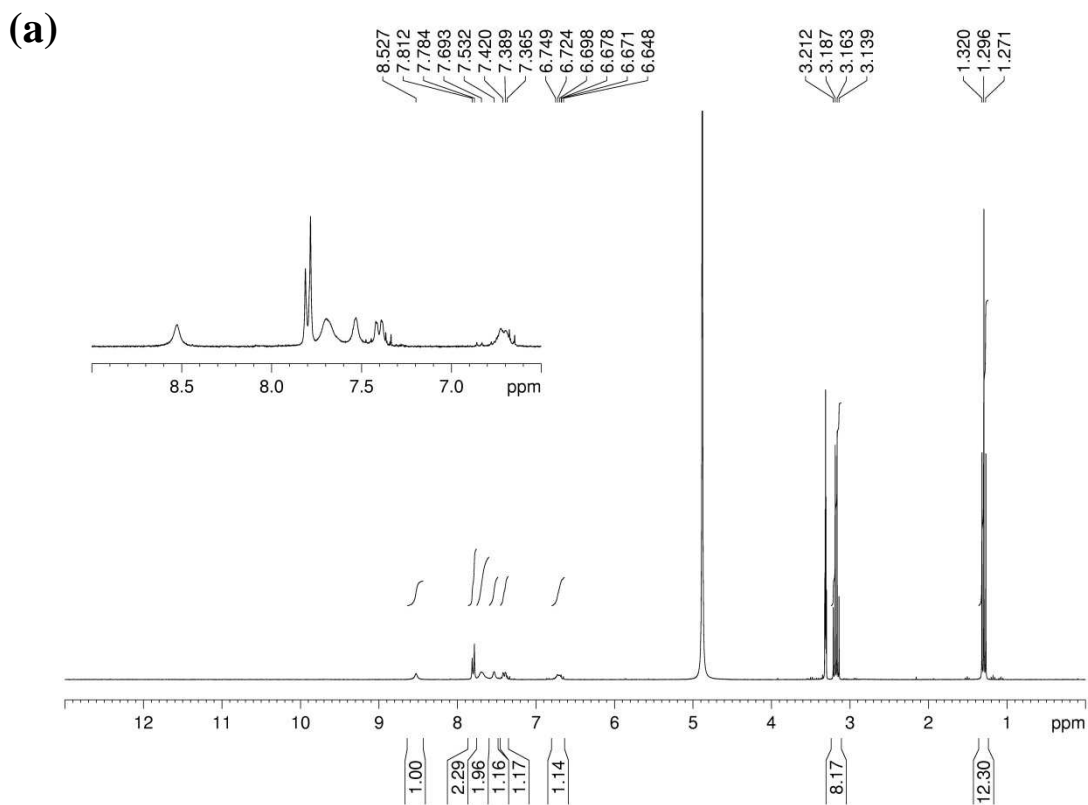


Fig. S6 ^1H NMR spectra of $[\text{HNEt}_3][\text{Zn}(\text{sal}^{\text{Cl}^-}\text{-4-CN})\text{Cl}_2]$ (**2**) in (a) $\text{MeOH-}d_4$, (b) $\text{ACN-}d_3$, (c) $\text{DMSO-}d_6$, and (d) $\text{DCM-}d_2$.



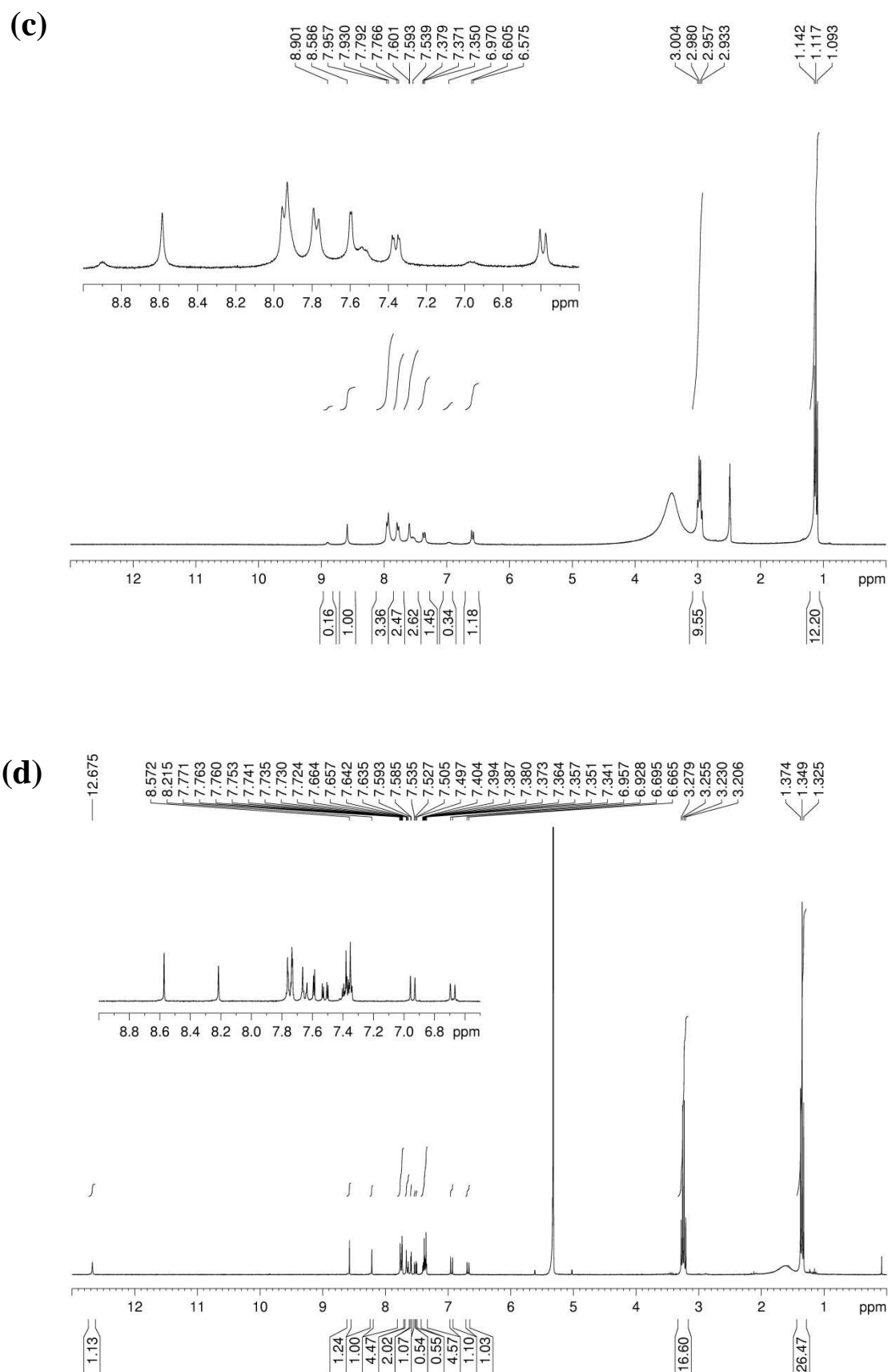
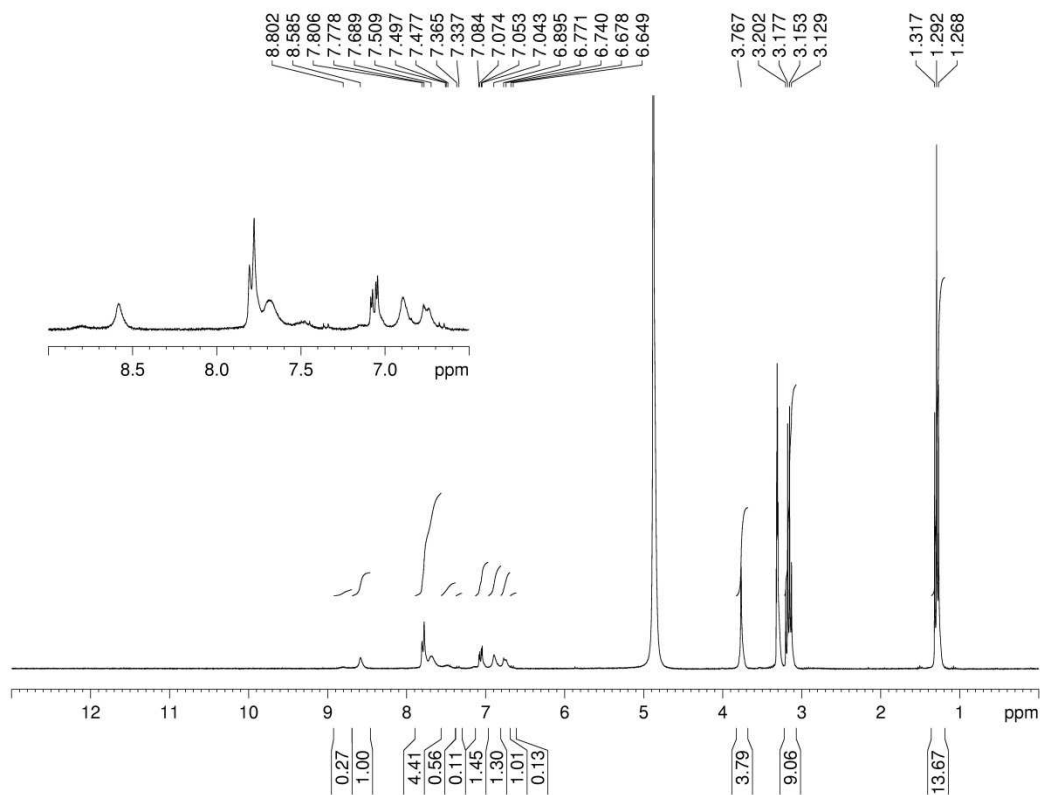
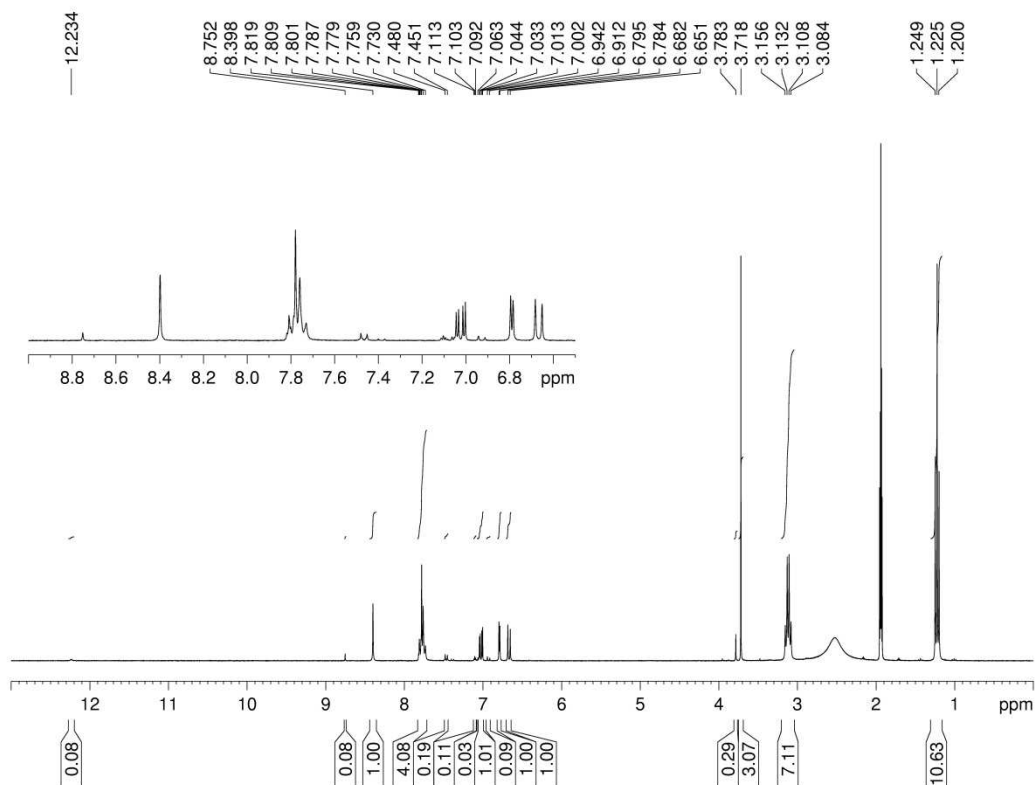


Fig. S7 ^1H NMR spectra of $[\text{HNEt}_3][\text{Zn}(\text{sal}^{\text{Br-4-CN}})\text{Cl}_2]$ (**3**) in (a) $\text{MeOH-}d_4$, (b) $\text{ACN-}d_3$, (c) $\text{DMSO-}d_6$, and (d) $\text{DCM-}d_2$.

(a)



(b)



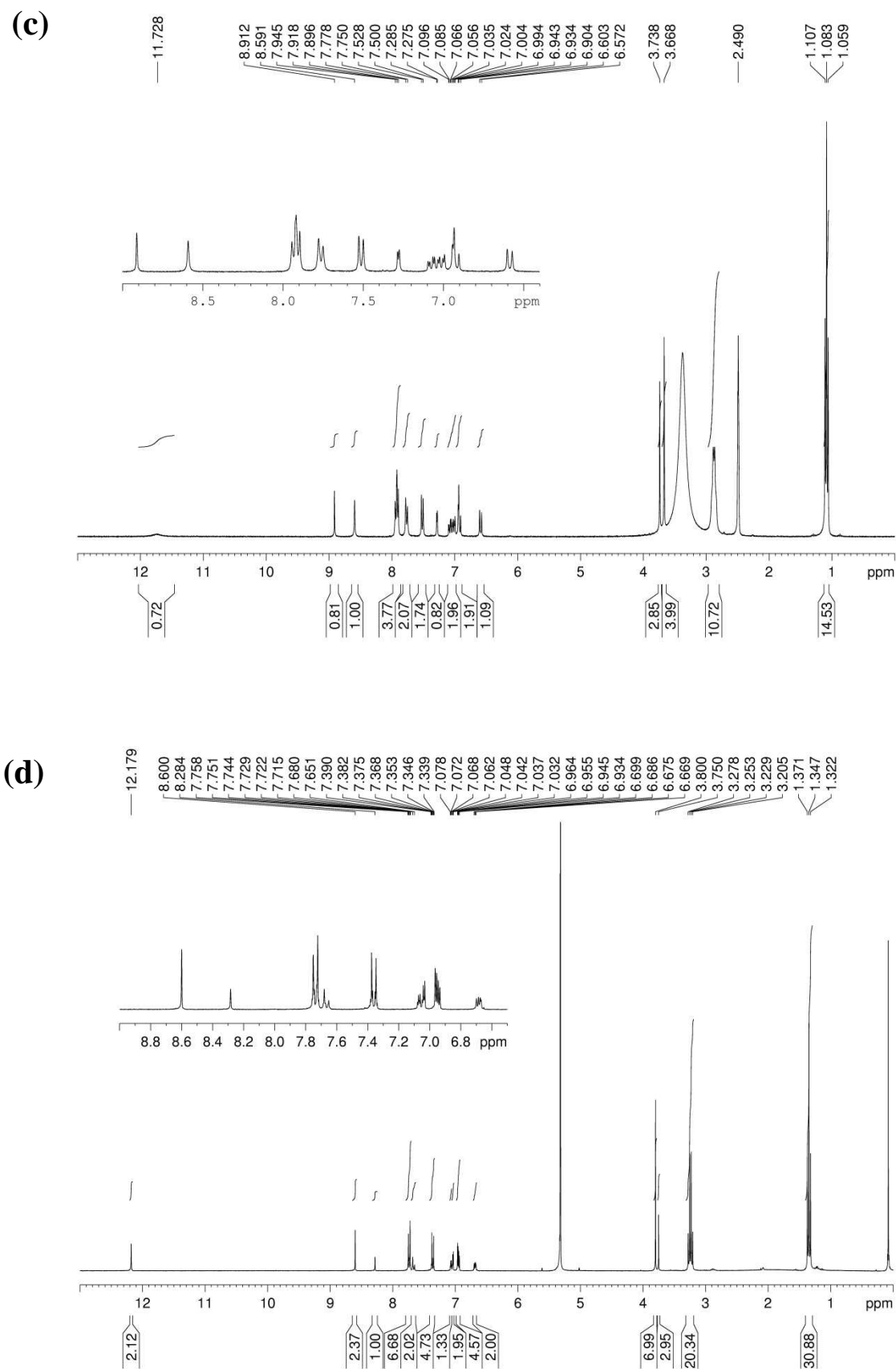


Fig. S8 ^1H NMR spectra of $[\text{HNEt}_3][\text{Zn}(\text{sal}^{\text{OMe}}\text{-4-CN})\text{Cl}_2]$ (**4**) in (a) $\text{MeOH-}d_4$, (b) $\text{ACN-}d_3$, (c) $\text{DMSO-}d_6$, and (d) $\text{DCM-}d_2$.

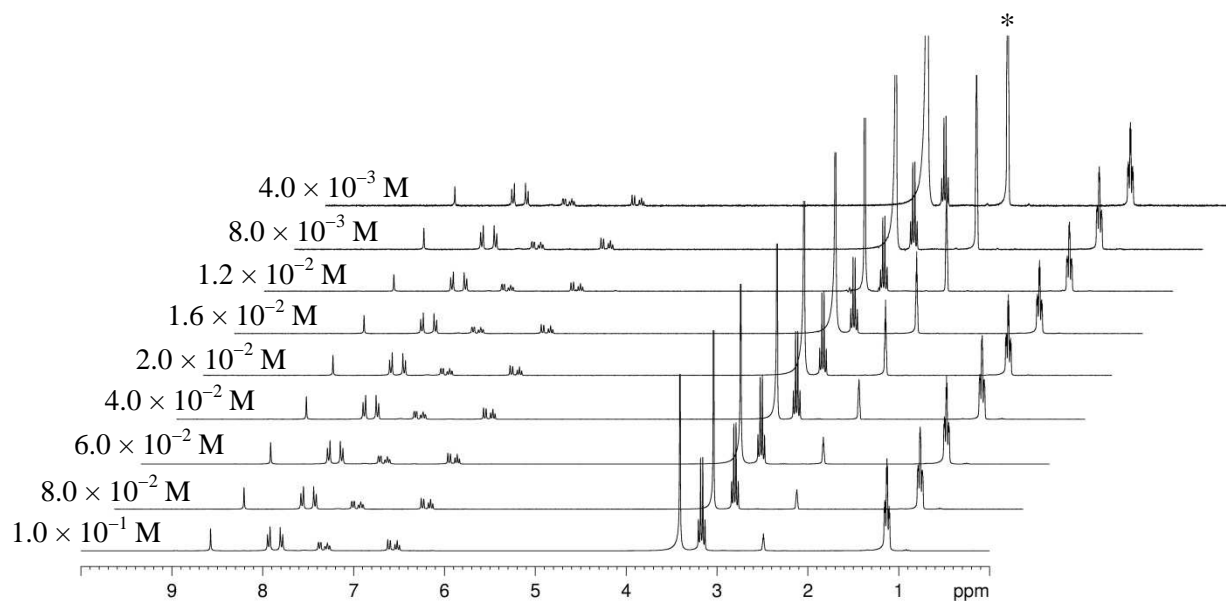


Fig. S9 Concentration independence of ^1H NMR spectra of $[\text{NEt}_4][\text{Zn}(\text{sal}^{\text{H}}\text{-4-CN})\text{Cl}_2]$ (5) in $\text{DMSO-}d_6$ at room temperature.

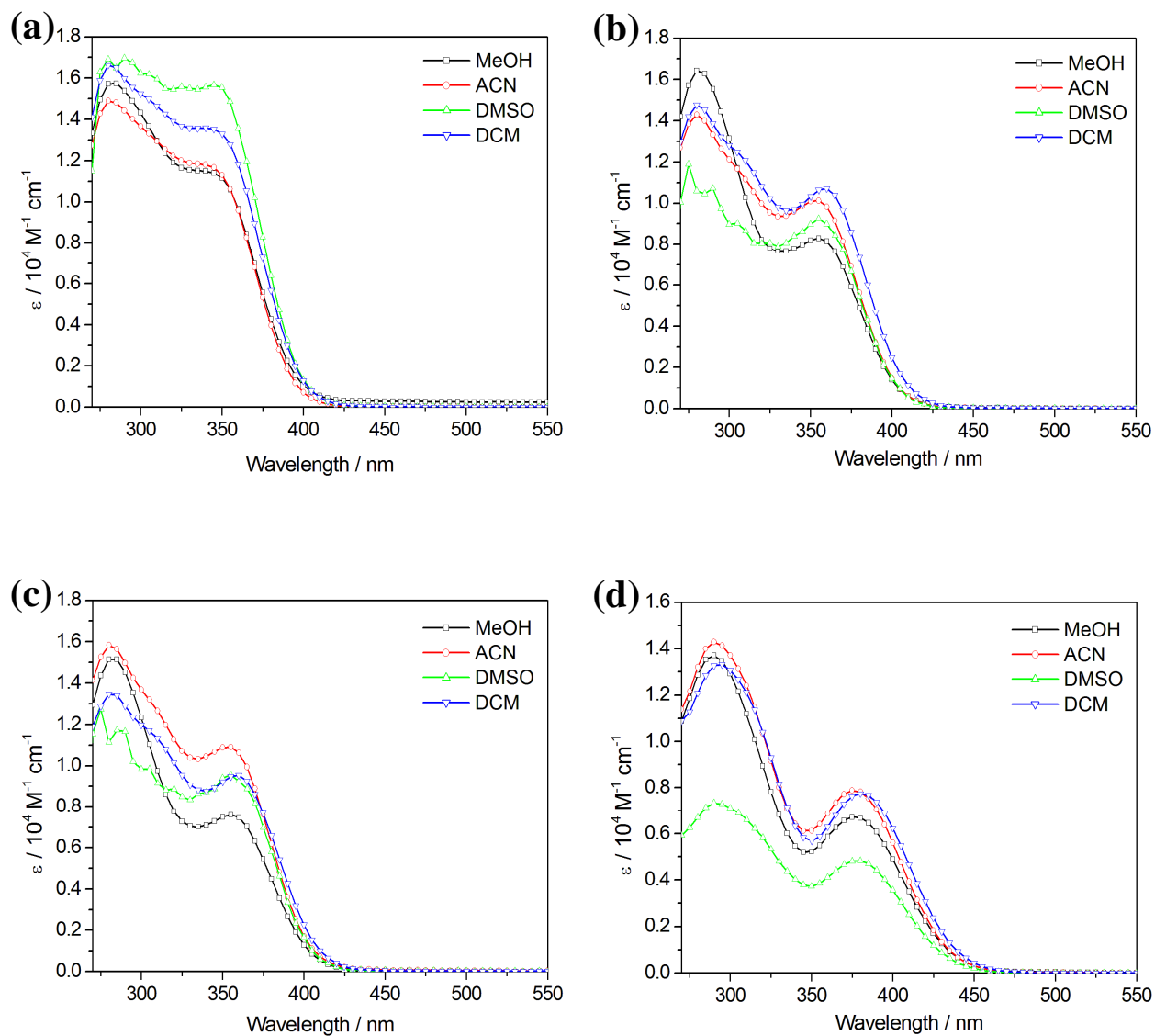


Fig. S10 UV-vis absorption spectra of (a) Hsal^H-4-CN, (b) Hsal^{Cl}-4-CN, (c) Hsal^{Br}-4-CN, and (d) Hsal^{OMe}-4-CN in various solvents (100 μM) at room temperature.

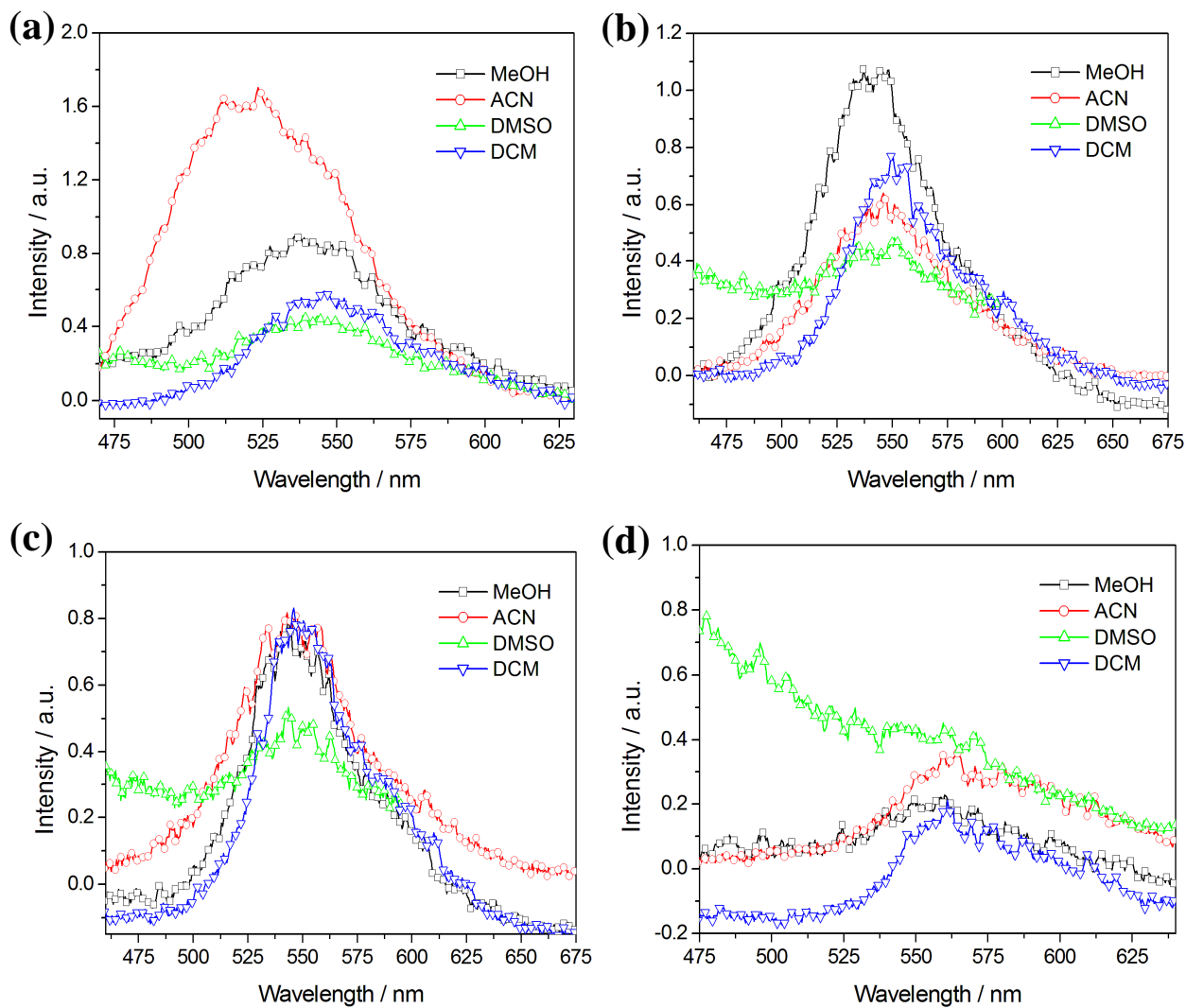


Fig. S11 Fluorescence spectra of (a) Hsal^H-4-CN, (b) Hsal^{Cl}-4-CN, (c) Hsal^{Br}-4-CN, and (d) Hsal^{OMe}-4-CN in various solvents (100 μ M) at room temperature.

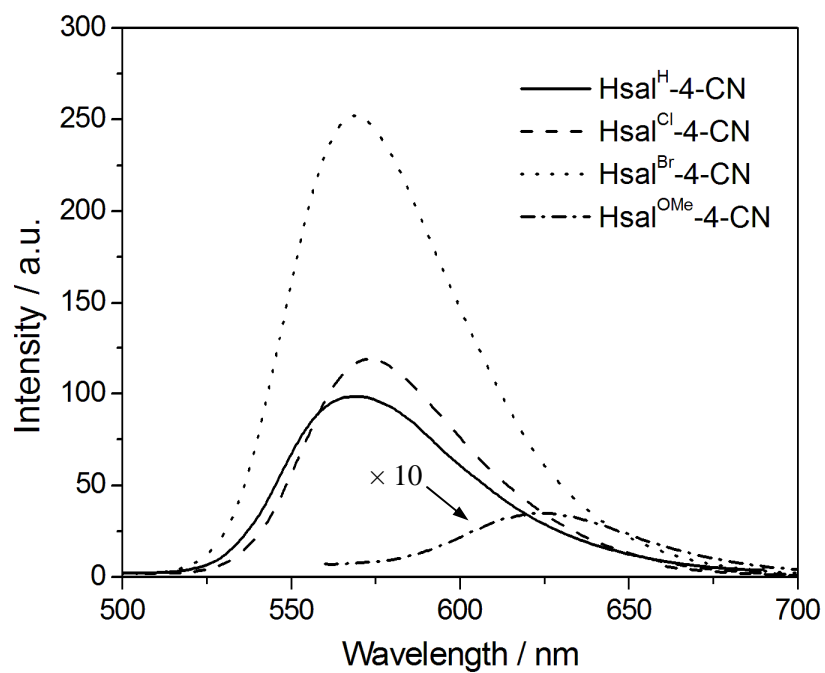


Fig. S12 Solid-state fluorescence spectra of Hsal^R-4-CN (R = H, Cl, Br, and OMe) ligands at room temperature.

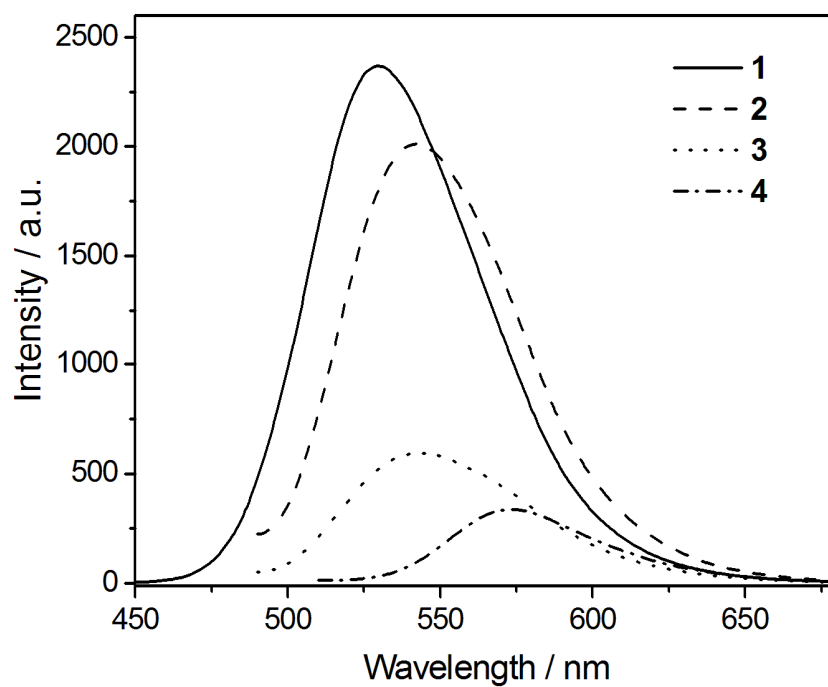


Fig. S13 Solid-state fluorescence spectra of compounds **1–4** at room temperature.

Table S1 Selected bond lengths (Å) and angles (°) and dihedral angle θ (°) between the salicylideneimine phenyl ring and the benzonitrile phenyl ring for zinc–salicylideneimine compounds **1–5**

Compound				4		5
	1	2	3	Molecule I	Molecule II	
Zn1/2–O1/3	1.9598(11)	1.9551(12)	1.9555(15)	1.956(4)	1.950(4)	1.9345(16)
Zn1/2–N1/4	2.0471(13)	2.0473(13)	2.0393(17)	2.031(5)	2.054(5)	2.0360(19)
Zn1/2–Cl1/3	2.2452(4)	2.2287(5)	2.2293(8)	2.241(2)	2.244(2)	2.2646(6)
Zn1/2–Cl2/4	2.2300(4)	2.2313(6)	2.2233(8)	2.232(2)	2.233(2)	2.2418(7)
O1/3–Zn1/2–N1/4	94.00(5)	95.50(5)	95.50(5)	95.7(2)	93.6(2)	95.42(7)
O1/3–Zn1/2–Cl1/3	114.73(4)	110.15(4)	108.52(5)	112.43(17)	115.91(17)	111.87(5)
O1/3–Zn1/2–Cl2/4	111.68(3)	108.50(4)	109.91(5)	107.42(14)	106.15(15)	112.47(5)
N1/4–Zn1/2–Cl1/3	106.93(4)	118.18(4)	106.55(5)	106.63(17)	105.46(17)	102.96(6)
N1/4–Zn1/2–Cl2/4	115.45(4)	106.83(4)	118.21(6)	118.00(18)	121.81(18)	119.33(5)
Cl1/3–Zn1/2–Cl2/4	112.728(16)	115.60(2)	115.92(3)	115.15(8)	112.95(8)	113.28(3)
θ	26.5	1.9	2.2	36.5	6.3	22.5

Table S2 Dissociation ratio (R_D) of **1** in DMSO- d_6 with different concentrations at room temperature^a

Concentration (M)	Dissociation Ratio (R_D)
4.0×10^{-3}	52.2
8.0×10^{-3}	43.8
1.2×10^{-2}	40.5
1.6×10^{-2}	36.7
2.0×10^{-2}	32.4
4.0×10^{-2}	29.1
6.0×10^{-2}	25.4
8.0×10^{-2}	21.9
1.0×10^{-1}	18.0

^a $R_D = I_{CH=N(HL)} / \{I_{CH=N(HL)} + I_{CH=N([Zn(L)Cl_2]^-)}\} \times 100$, where $I_{CH=N(HL)}$ = relative integrated area of the azomethine proton (CH=N) signal of free ligand form and $I_{CH=N([Zn(L)Cl_2]^-)}$ = relative integrated area of the azomethine proton (CH=N) signal of complex form.