Electronic Supplementary Information (ESI)

A cationic and ferromagnetic hexametallic Mn(III) Single-Molecule Magnet based on the salicylamidoxime ligand

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Preparation of 1. All manipulations were performed under aerobic conditions, using materials and solvents as received. $Mn(ClO_4)_2 \cdot 6H_2O$ (0.249 g, 6.88 mmol) was dissolved with continuous stirring in EtOH (20 mL), then H₂N-saoH₂ (0.102g, 6.70 mmol) was added followed by pyridine (1mL, 12.4 mmol) and NEt₃ (0.05 mL, 3.58 mmol). The dark green final solution was left to evaporate in a fumehood at room temperature. Dark green crystals were formed in 1 day, which were suitable for X-ray diffraction studies. Yield: 72%. Anal. Cald. (found) for $C_{76}H_{78}O_{24}N_{18}Cl_2Mn_6$ (1): C, 45.0 (45.5); H, 3.9 (4.1); N, 12.4 (12.3) %. IR peaks (ATR/cm⁻¹): 1597, 1086, 1036, 1017, 878, 700, 673, 638, 628.

A)



B)



C)



Figure S1. (A) Perspective view of the unidimensional arrangement of $[Mn_6(\mu_3-O)_2(H_2N-sao)_6(py)_6(EtOH)_2]^{2+}$ cations and ClO_4^- anions in **1** through H-bonding interactions (dashed lines). Solvent molecules have been omitted for clarity. [Colour code: pink, Mn; red, O; blue, N; black, C; green, Cl]. (B) Expansion of this interaction into 2 dimensions, highlighting the H-bonded square sheets. (C) View of the interdigitated Mn_6 cluster cations in the third dimension.



Figure S2. Two *J* coupling exchange model used for the fit of compound 1.



Figure S3. Plot of the reduced magnetisation (*M*/Nµ_B vs. *H*/T) for **1** in 4, 5, 6 and 7 T fields and temperatures 2–7 K. The solid lines represent the best fit of the experimental data with S = 12, $g = 1.98 \pm 0.02$ and $D_{\text{cluster}} = -0.34 \pm 0.02$ cm⁻¹.