

Supporting Information

Synthesis of β -diketiminato-ligated bimetallic and monometallic lanthanide amide complexes and their reactivity with isoprene and AlMe_3

Song Sun,^a Hao Ouyang,^a Yunjie Luo,^{*,b} Yong Zhang,^a Qi Shen,^a and
Yingming Yao^{*,a}

^a Key Laboratory of Organic Synthesis of Jiangsu Province, College of Chemistry, Chemical Engineering and Materials Science, Dushu Lake Campus, Soochow University, Suzhou 215123, People's Republic of China

^b Organometallic Chemistry Laboratory, Ningbo Institute of Technology, Zhejiang University, Ningbo 315100, P. R. China

* To whom correspondence should be addressed.

E-mail: yaoym@suda.edu.cn; lyj@nit.zju.edu.cn

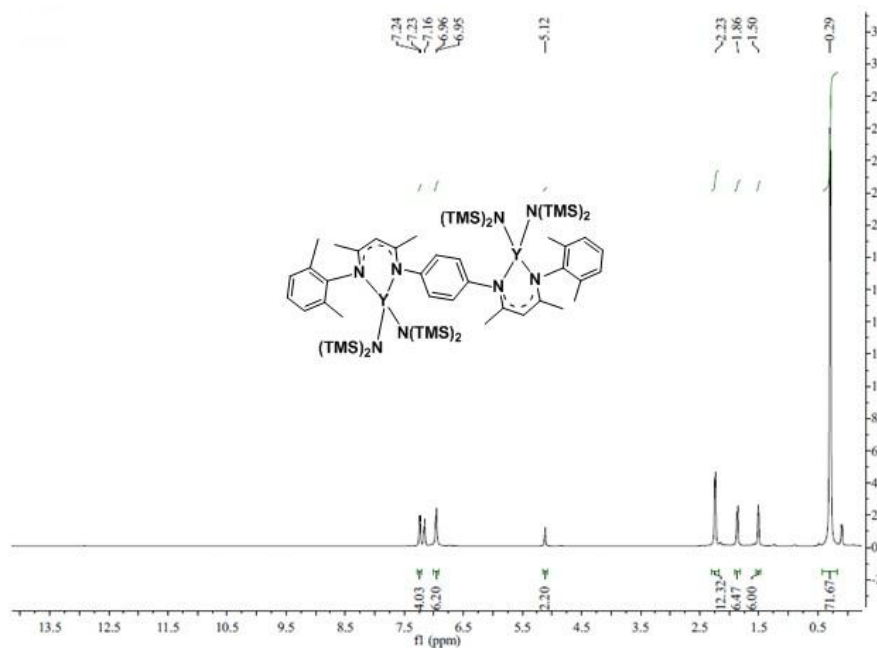


Figure S1. ¹H NMR spectrum of complex **1** (400 MHz, C₆D₆)

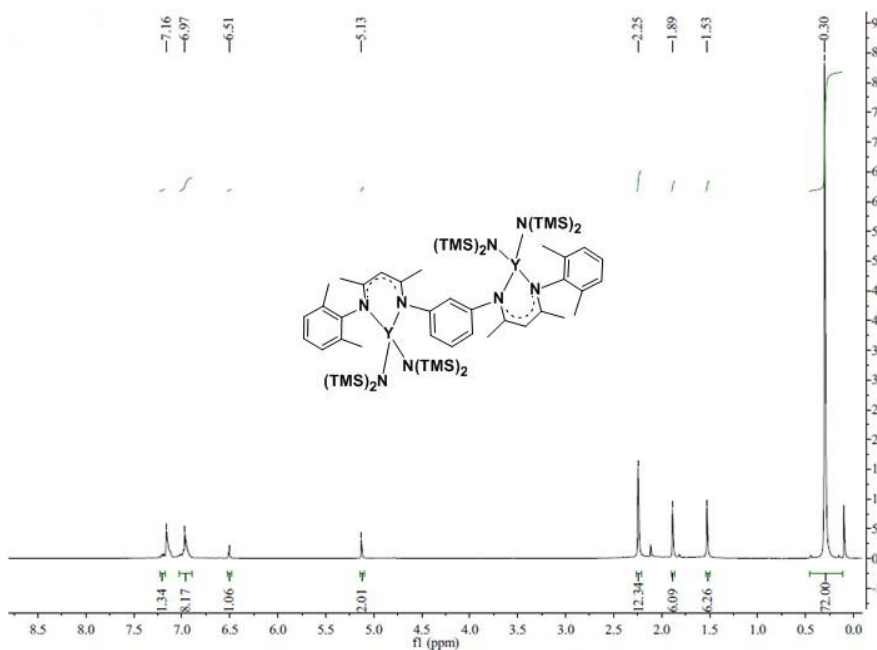


Figure S2. ¹H NMR spectrum of complex **3** (400 MHz, C₆D₆)

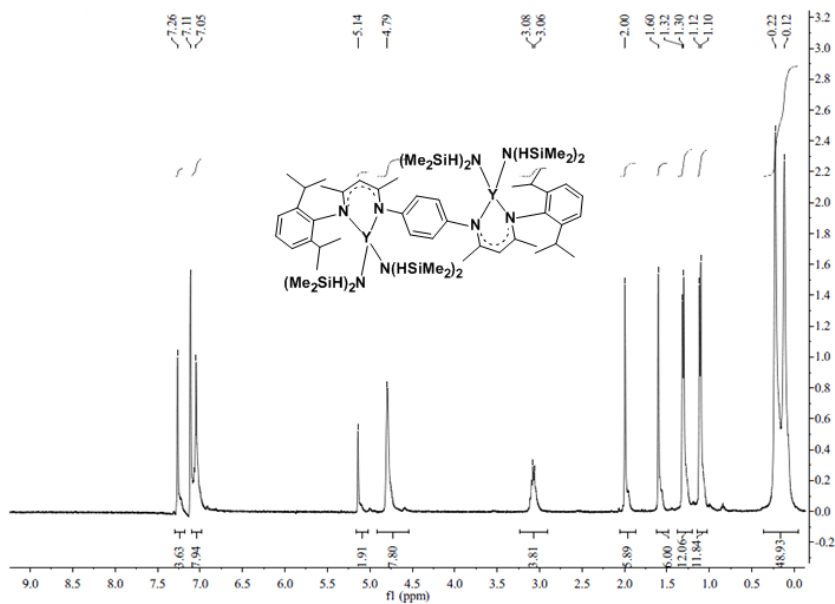


Figure S3. ^1H NMR spectrum of complex **4** (400 MHz C_6D_6)

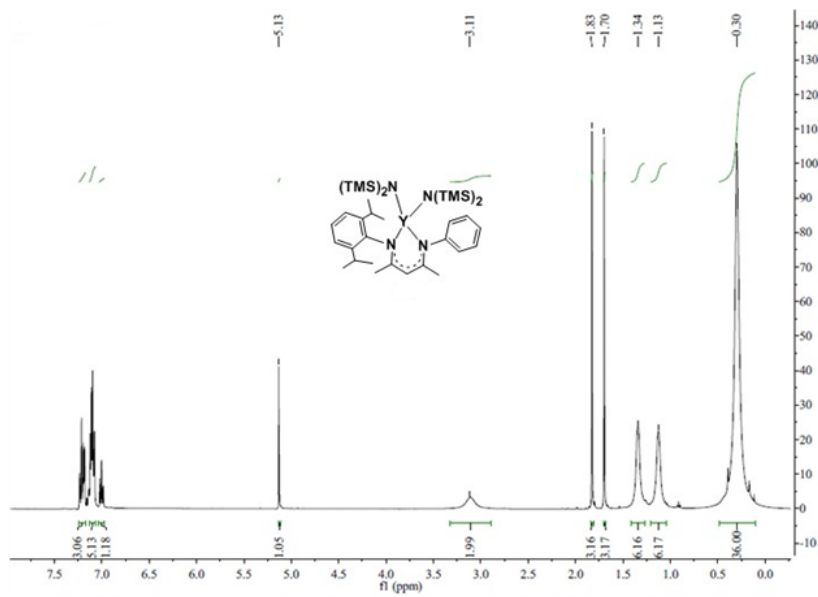


Figure S4. ^1H NMR spectrum of complex **6** (400 MHz C_6D_6)

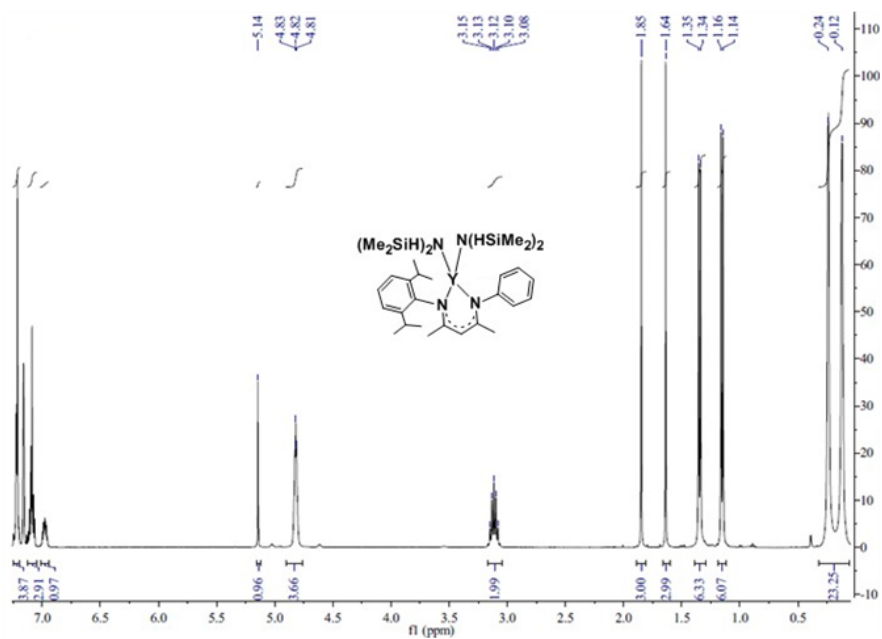


Figure S5. ^1H NMR spectrum of complex **7** (400 MHz, C_6D_6)

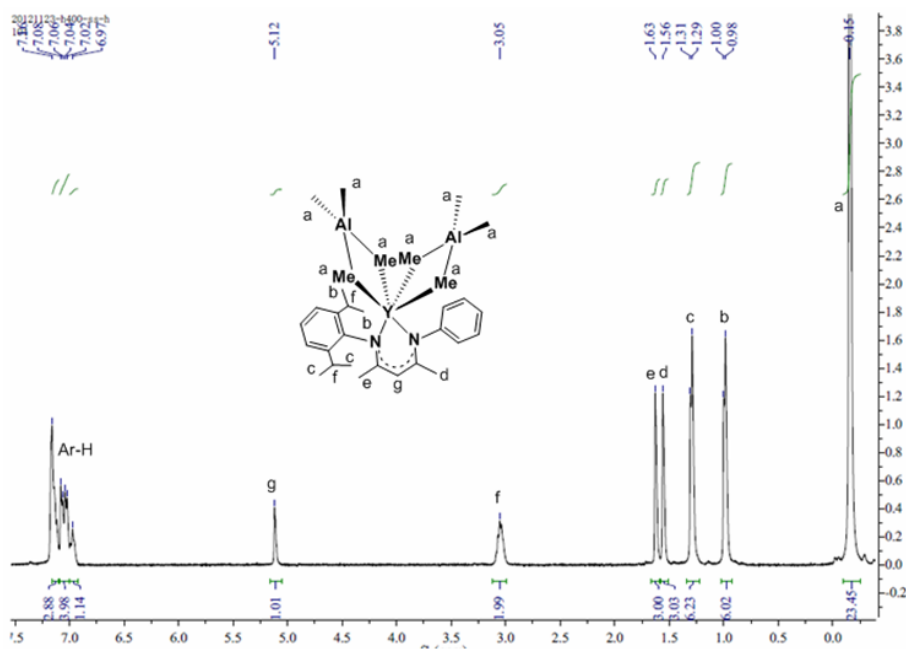


Figure S6. ^1H NMR spectrum of complex **8** (400 MHz, C_6D_6)

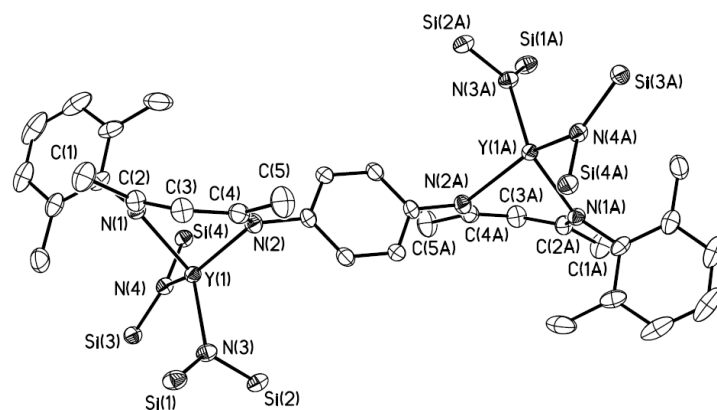


Figure S7. ORTEP diagram of complex **1** showing an atom numbering scheme. Thermal ellipsoids are drawn at the 30% probability level, and hydrogen atoms are omitted for clarity.

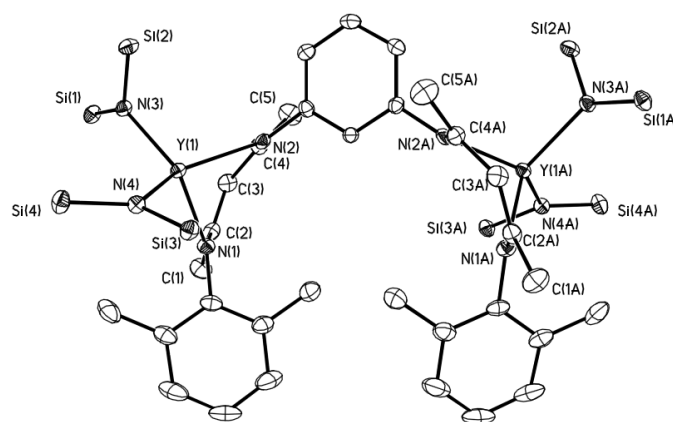


Figure S8. ORTEP diagram of complex **3** showing an atom numbering scheme. Thermal ellipsoids are drawn at the 30% probability level, and hydrogen atoms are omitted for clarity.

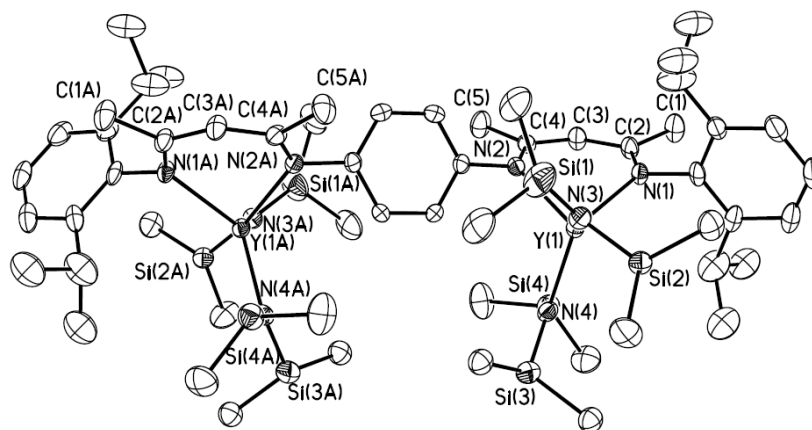


Figure S9. ORTEP diagram of complex **4** showing an atom numbering scheme. Thermal ellipsoids are drawn at the 30% probability level, and hydrogen atoms are omitted for clarity.

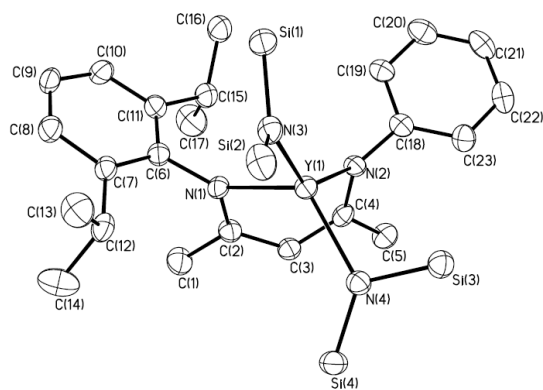


Figure S10. ORTEP diagram of complex **6** showing an atom numbering scheme. Thermal ellipsoids are drawn at the 30% probability level, and hydrogen atoms are omitted for clarity.

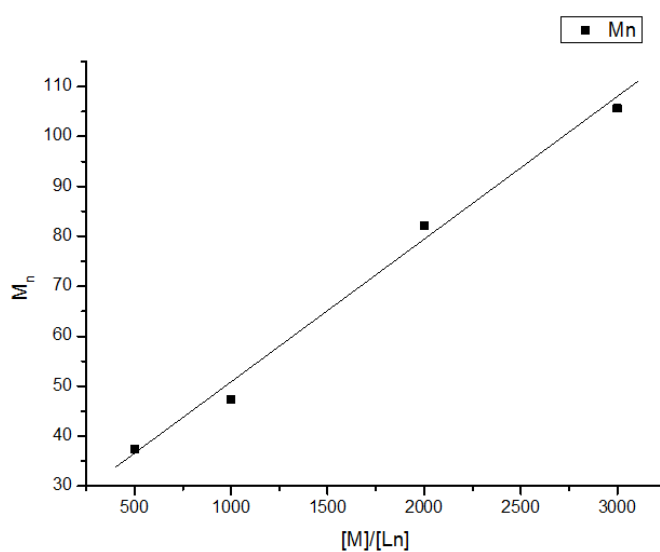


Figure S11. Polymerization of isoprene with complex **4** under activation of organic broate and trialkylaluminium. Relationship between the number-averaged molecular weight (M_n) and the molar ratio of monomer to initiator.

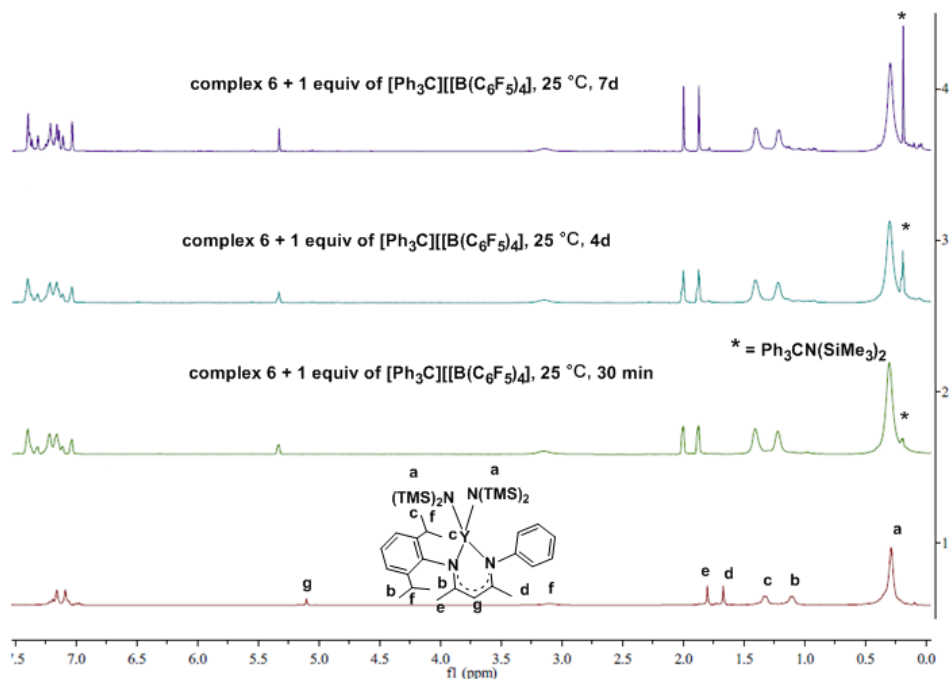


Figure S12. Combined ^1H NMR spectrum of the reaction of complex **6** + 1 equiv of $[\text{Ph}_3\text{C}][\text{B}(\text{C}_6\text{F}_5)_4]$, 25 °C, after 30 min, (2), 4 days (3), 7 days (4). (400 MHz, $\text{C}_6\text{D}_5\text{Cl}+\text{C}_6\text{D}_6$)

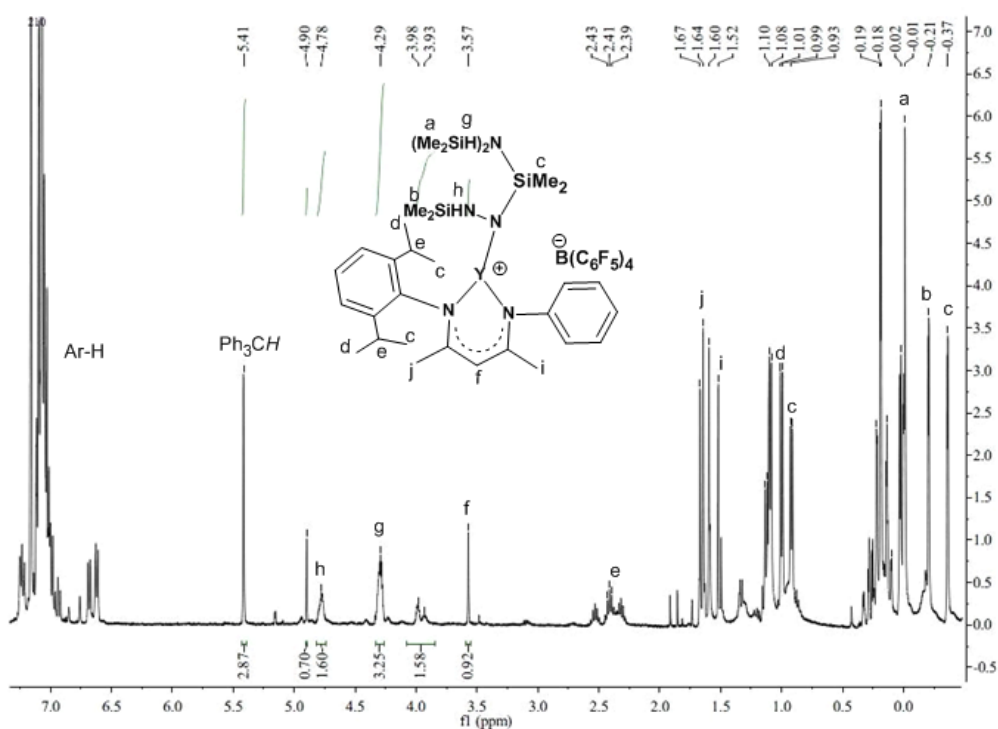


Figure S13. ^1H NMR spectrum of the reaction of complex **7** + 1 equiv of $[\text{Ph}_3\text{C}][\text{B}(\text{C}_6\text{F}_5)_4]$, 25 °C, after 30 min. (400 MHz, $\text{C}_6\text{D}_5\text{Cl}+\text{C}_6\text{D}_6$)