

## SUPPORTING INFORMATION

### **Ring-Opening Homo- and Co-polymerization reaction of $\epsilon$ -Caprolactone and Lactides by Salalen Aluminum Complexes**

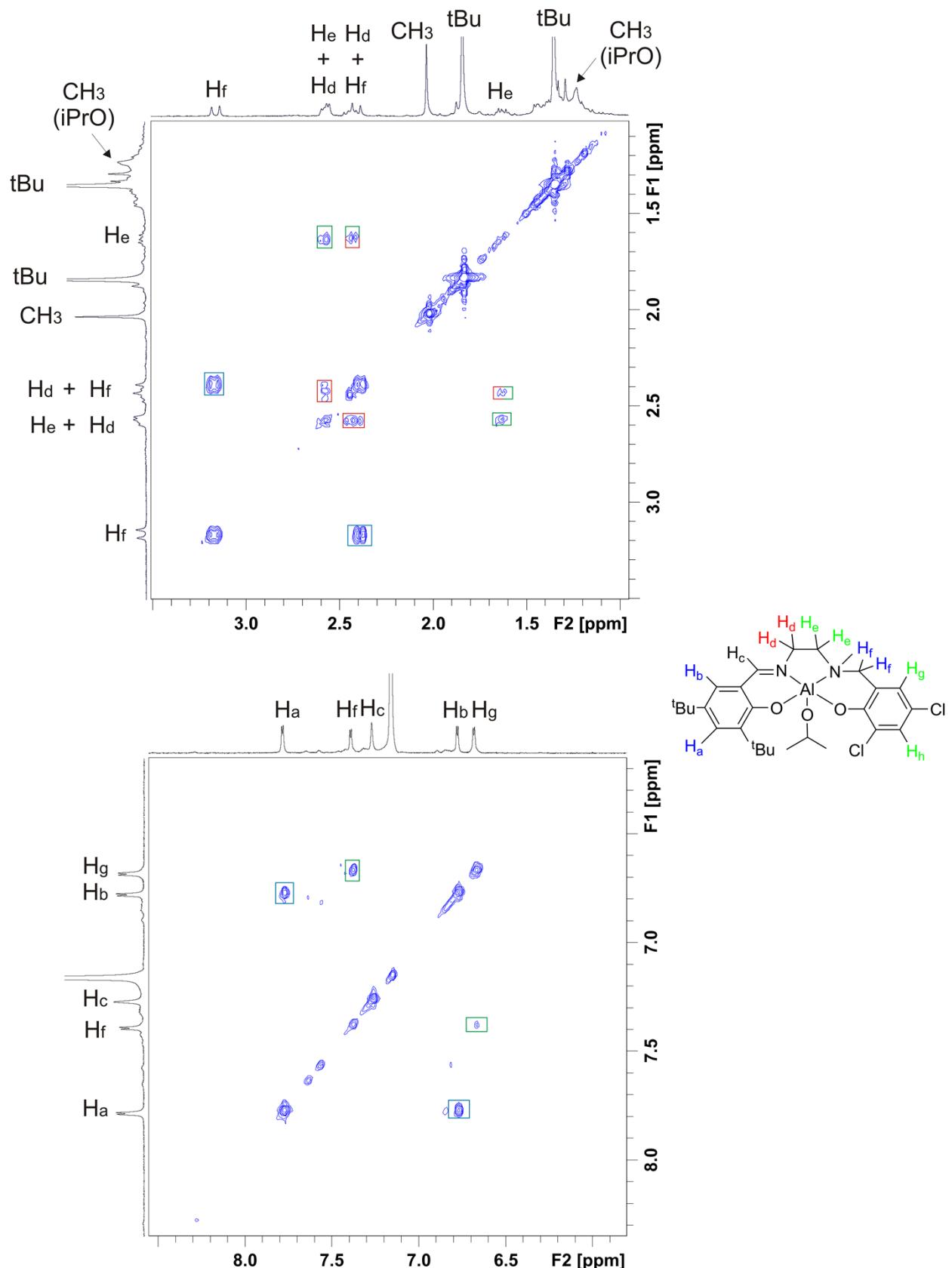
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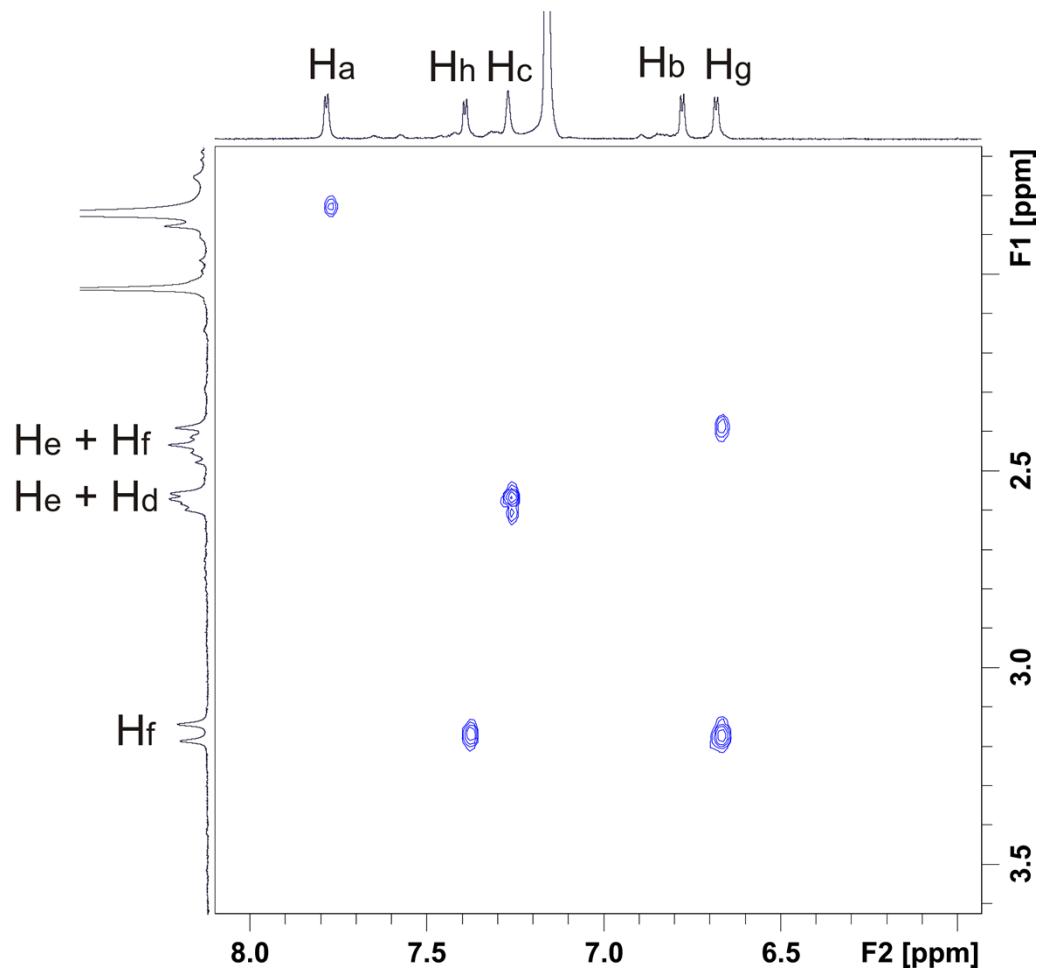
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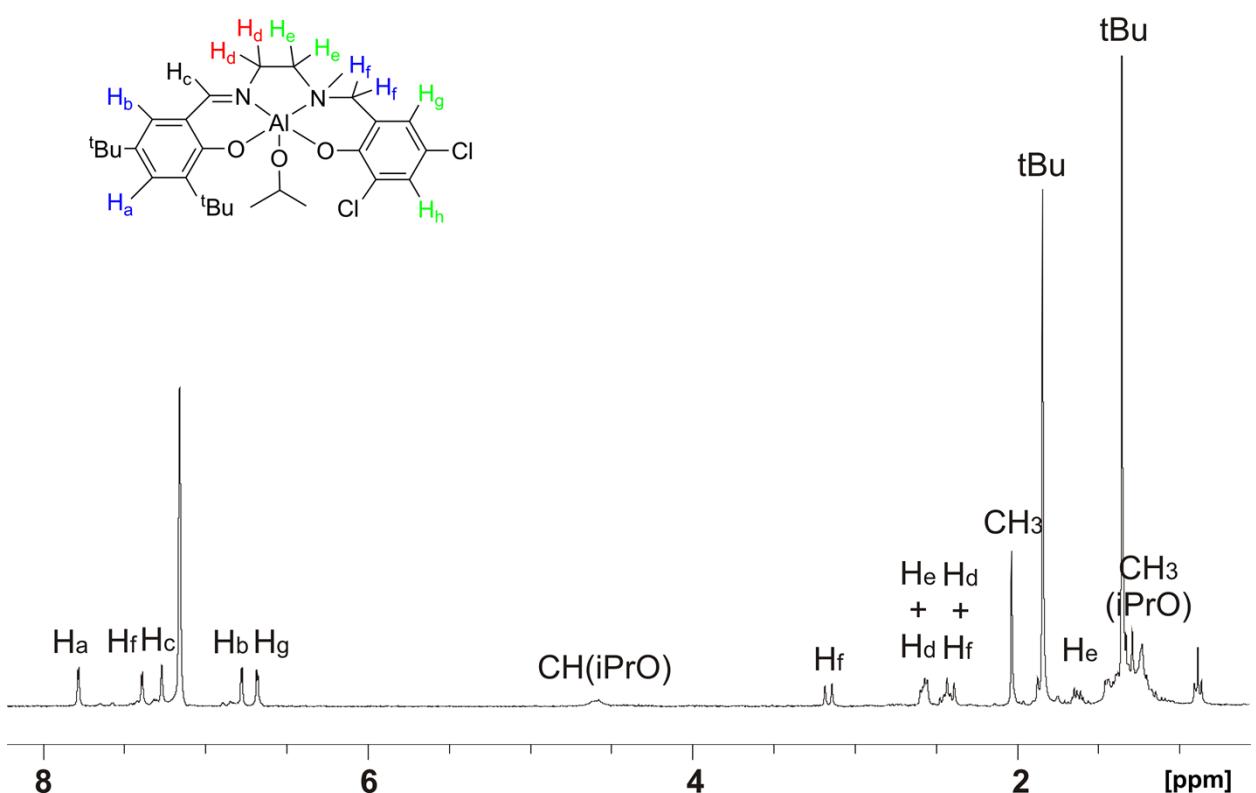
## NMR characterization of complex **2b**.



**Figure S1.** Aliphatic (up) and aromatic (down) area of NMR COSY spectrum of complex **2b** ( $\text{C}_6\text{D}_6$ ).



**Figure S2.** Aliphatic/aromatic cross peaks of NMR COSY spectrum of complex **2b** ( $\text{C}_6\text{D}_6$ ).



**Figure S3.**  $^1\text{H}$  NMR spectrum of complex **2b** ( $\text{C}_6\text{D}_6$ ).

**Table S1.** Tetrad probabilities based on Bernoullian Statistic (Th) and experimental values (Exp) as obtained by NMR analysis (entry 2, Table 1).  $P_m = 0.72$

Tetrad	Formula	Th	Exp
[mmm]	$P_m^2 + P_r P_m / 2$	0.621	0.621
[mmr]	$P_r P_m / 2$	0.101	0.099
[rmm]	$P_r P_m / 2$	0.101	0.101
[rmr]	$P_r^2 / 2$	0.039	0.049
[mrm]	$(P_r^2 + P_r P_m) / 2$	0.140	0.130

**Table S2.** Tetrad probabilities based on Enantiomeric Site Control Statistic (Th) and experimental values (Exp) as obtained by NMR analysis (entry 2, Table 1).  $\alpha = 0.81$

Tetrad	Formula	Th	Exp
[mmm]	$[\alpha^2 + (1-\alpha)^2 + \alpha^3 + (1-\alpha)^3] / 2$	0.621	0.621
[mmr]	$[\alpha^2(1-\alpha) + \alpha(1-\alpha)^2] / 2$	0.077	0.099
[rmm]	$[\alpha^2(1-\alpha) + \alpha(1-\alpha)^2] / 2$	0.077	0.101
[rmr]	$[\alpha^2(1-\alpha) + \alpha(1-\alpha)^2] / 2$	0.077	0.049
[mrm]	$[\alpha(1-\alpha) + \alpha(1-\alpha)] / 2$	0.154	0.130

**Table S3.** Tetrad probabilities based on Bernoullian Statistic (Th) and experimental values (Exp) as obtained by NMR analysis (entry 3, Table 1).  $P_m = 0.69$

Tetrad	Formula	Th	Exp
[mmm]	$P_m^2 + P_r P_m / 2$	0.583	0.583
[mmr]	$P_r P_m / 2$	0.107	0.112
[rmm]	$P_r P_m / 2$	0.107	0.131
[rmr]	$P_r^2 / 2$	0.048	0.060
[mrm]	$(P_r^2 + P_r P_m) / 2$	0.155	0.114

**Table S4.** Ring-Opening Polymerization of L-lactide promoted by **2b** and **S-4b**.

Monomer	Time	Conv	$M_n(\text{th})^c$ (Kg/mol)	$M_n(\text{GPC})^b$ (Kg/mol)	PDI
L-LA	25h	70%	20.2	21.4	1.05

<sup>a</sup> Conditions: initiator: **2b**: 10 µmol + **S-4b**: 10 µmol; toluene: 4 mL; lactide: 4 mmol; temperature: 80 °C.

<sup>b</sup> Experimental  $M_n$  values were determined by GPC analysis in THF using polystyrene standards and corrected by the factor 0.58.

<sup>c</sup>  $144.13 \times [LA]_0/[I]_0 \times \text{conv LA}$ .

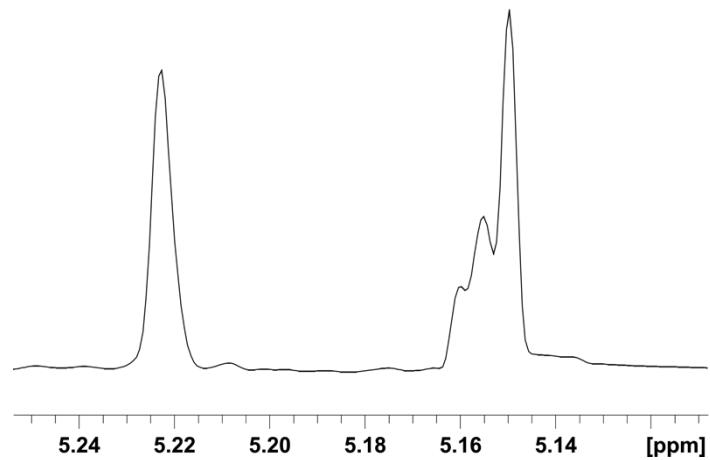
## Meso-Lactide

**Table S5.** Tetrad probabilities based on Bernoullian Statistic (Th) and experimental values (Exp) as obtained by <sup>13</sup>C NMR analysis (entry 6, Table 1).  $P_m = 0.79$

Tetrad	Formula	Th	Exp
[rmr]	$(P_m^2 + P_r P_m) / 2$	0.395	0.382
[rrr]	$P_r^2 + P_r P_m / 2$	0.127	0.136
[rrm]	$P_r P_m / 2$	0.083	0.078
[mrr]	$P_r P_m / 2$	0.083	0.090
[mrm]	$P_m^2 / 2$	0.314	0.314

**Table S6.** Tetrad probabilities based on Enantiomeric Site Control Statistics (Th) and experimental values (Exp) as obtained by  $^{13}\text{C}$  NMR analysis (entry 6, Table 1).  $\alpha = 0.76$

Tetrad	Formula	Th	Exp
[rrr]	$[\alpha(1-\alpha)+\alpha(1-\alpha)]/2$	0.182	0.382
[rrr]	$[\alpha^2+(1-\alpha)^2+\alpha^3+(1-\alpha)^3]/2$	0.544	0.136
[rrm]	$[\alpha^2(1-\alpha)+\alpha(1-\alpha)^2]/2$	0.078	0.078
[mrr]	$[\alpha^2(1-\alpha)+\alpha(1-\alpha)^2]/2$	0.078	0.090
[mrm]	$[\alpha^2(1-\alpha)+\alpha(1-\alpha)^2]/2$	0.078	0.314



**Figure S4.** Homonuclear decoupled  $^1\text{H}$  NMR of the PLA sample obtained by complex **2b** and *meso*-LA in toluene at 80 °C

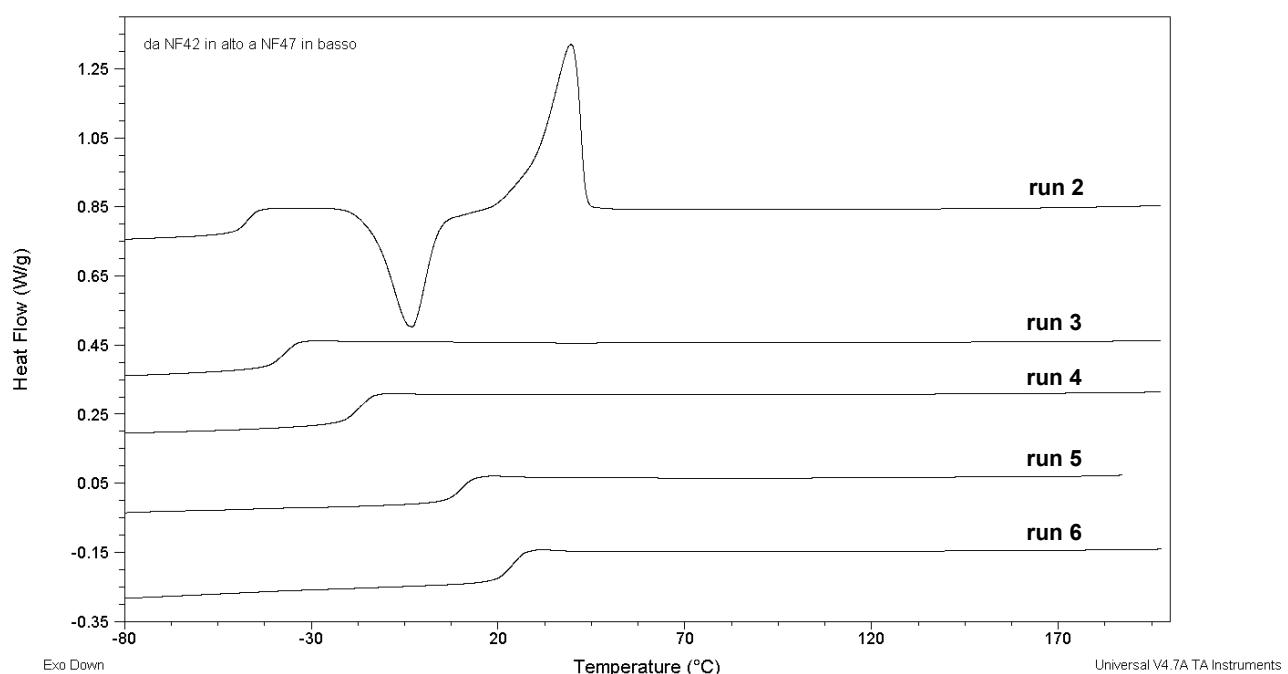
**Determination of Reactivity Ratios:** The reactivity ratios were calculated using the nonlinear least squares (NLLS) method, carrying out the copolymerizations at low conversion with different ratios of the monomers.

**Table S7.** Copolymerization runs at low monomer conversions.

Time	[LA] <sub>0</sub> (f1)	[CL] <sub>0</sub> (f2)	LA conv.,%	CL conv.,%	LA in the copolymer (F1)	CL in the copolymer (F2)
10'	0.1	0.9	11	33	0.01	0.99
30'	0.3	0.7	6	20	0.12	0.88
1h 30'	0.5	0.5	7	12	0.34	0.66
1h 30'	0.7	0.3	7	13	0.58	0.42
1h 30'	0.9	0.1	5	9	0.84	0.16

$$r_{CL} = 2.95 \quad r_{LA} = 0.85$$

### Thermal analysis of the L-LA/ $\epsilon$ -CL copolymers.



**Figure S5.** DSC (second heating run) of random L-LA/ $\epsilon$ -CL copolymers reported in Table 2 of the paper.