

## Electronic Supplementary Information

For

### **First iron and cobalt(II) hexabromoclathrochelates: structural, magnetic, redox and electrocatalytic behavior.**

Alexander V. Dolganov<sup>a</sup>, Alexander S. Belov<sup>a</sup>, Valentin V. Novikov<sup>a</sup>, Anna V. Vologzhanina<sup>a</sup>, Galina V. Romanenko<sup>b</sup>, Yulia G. Budnikova<sup>c</sup>, Genrikh E. Zelinskii<sup>a</sup>, Michail I. Buzin<sup>a</sup>, Yan Z. Voloshin<sup>a</sup>

<sup>a</sup> *Nesmeyanov Institute of Organoelement Compounds of the Russian Academy of Sciences, 119991 Moscow, Russia. E-mail: voloshin@ineos.ac.ru; Fax: +7 (499)135-5085*

<sup>b</sup> *International Tomography Center, the Siberian Branch of the Russian Academy of Sciences, 630090 Novosibirsk, Russia*

<sup>c</sup> *A.E. Arbusov Institute of Organic and Physical Chemistry, Kazan Scientific Center, Russian Academy of Sciences, Kazan, Russian Federation.*

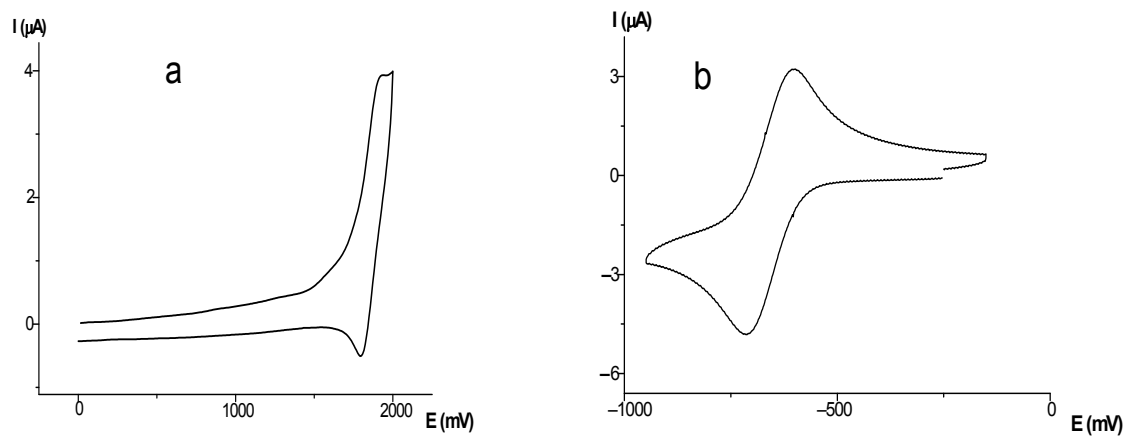


Figure S1. CV for the acetonitrile solution of the clathrochelate  $\text{Fe}(\text{Br}_2\text{Gm})_3(\text{Bn}-\text{C}_4\text{H}_9)_2$  in the anodic (a) and cathodic (b) ranges.

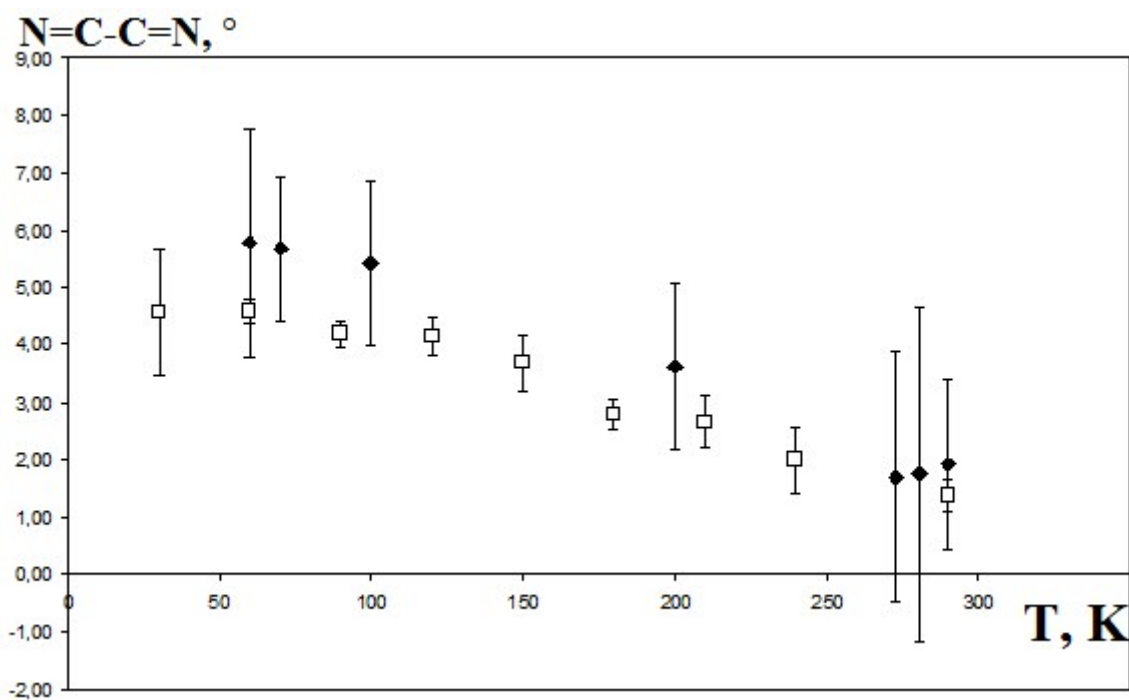


Figure S2. Temperature variation of the average  $\text{N} = \text{C} - \text{C} = \text{N}$  angles for the cobalt(II) hexahalogenoclatrochelates  $\text{Co}(\text{Br}_2\text{Gm})_3(\text{Bn}-\text{C}_4\text{H}_9)_2$  (black diamonds) and  $\text{Co}(\text{Cl}_2\text{Gm})_3(\text{Bn}-\text{C}_4\text{H}_9)_2$  (white squares).

**Table S1.** Maxima of the UV-vis spectra ( $\lambda_{\text{max}}/\text{nm}$ ,  $\varepsilon \cdot 10^{-3} \text{ mol}^{-1} \cdot \text{L} \cdot \text{cm}^3$ ) of the iron and cobalt(II) hexahalogenoclatrochelates.

Complex	$\nu_1$ ( $\varepsilon_1$ )	$\nu_2$ ( $\varepsilon_2$ )	$\nu_3$ ( $\varepsilon_3$ )	$\nu_4$ ( $\varepsilon_4$ )	$\nu_5$ ( $\varepsilon_5$ )	$\nu_6$ ( $\varepsilon_6$ )	$\nu_7$ ( $\varepsilon_7$ )	$\nu_8$ ( $\varepsilon_8$ )
$\text{Fe}(\text{Cl}_2\text{Gm})_3(\text{Bn}-\text{C}_4\text{H}_9)_2$ <sup>3a</sup>	259(7.9)	285(5.4)	313(2.7)			423(4.8)	453 (15)	
$\text{Fe}(\text{Br}_2\text{Gm})_3(\text{Bn}-\text{C}_4\text{H}_9)_2$	262 (11)	264 (1.8)	290 (1.0)	350 (1.7)		428 (2.2)	448 (9.2)	459 (7.7)
$\text{Fe}(\text{I}_2\text{Gm})_3(\text{Bn}-\text{C}_4\text{H}_9)_2$ <sup>4</sup>	264 (6.3)	287 (7.0)			397 (3.6)	430 (5.7)	464 (19)	496 (5.4)
$\text{Co}(\text{Cl}_2\text{Gm})_3(\text{Bn}-\text{C}_4\text{H}_9)_2$ <sup>3d</sup>		267 (20)	287 (8.4)	329 (3.1)	368 (3.2)	421 (0.64)		468 (1.6)
$\text{Co}(\text{Br}_2\text{Gm})_3(\text{Bn}-\text{C}_4\text{H}_9)_2$	249 (7.3)	272 (21)	302 (5.3)	335 (3.7)	371 (4.0)	444 (2.0)		477 (2.2)

**Table S2.** The number of inter- and intramolecular interactions of the hexahalogenoclathrochelates  $\text{Co}(\text{Br}_2\text{Gm})_3(\text{Bn}-\text{C}_4\text{H}_9)_2$  and  $\text{Co}(\text{Cl}_2\text{Gm})_3(\text{Bn}-\text{C}_4\text{H}_9)_2$  at different temperatures

Type of A...Z interaction <sup>a</sup>	$\text{Co}(\text{Br}_2\text{Gm})_3(\text{Bn}-\text{C}_4\text{H}_9)_2$							$\text{Co}(\text{Cl}_2\text{Gm})_3(\text{Bn}-\text{C}_4\text{H}_9)_2$					
	290	281	273	200	100	70	60	290	240	210-120	90	60	30
Intramolecular interactions													
Co...N	6	6	6	6	6	6	6	6	6	6	6	6	6
B...Co	4	4	4	4	4	4	4	4	4	4	4	4	4
O...C	24	24	24	24	24	24	24	24	24	24	24	24	24
O...B	6	6	6	6	6	6	6	6	6	6	6	6	6
O...O	12	12	12	12	12	12	12	12	12	12	12	12	12
O...N	22	22	22	22	22	22	22	22	22	22	22	22	21
N...N	20	20	20	20	20	20	20	20	20	20	20	20	20
N...B	12	12	12	12	12	12	12	12	12	12	12	12	12
N...C	18	14	18	18	18	18	18	18	18	18	18	18	18
C...B	6	6	6	6	6	6	6	6	6	6	6	6	6
B...H	16	16	16	16	16	16	16	16	16	16	16	16	16
Hal...C	14	18	14	14	18	18	18	14	14	18	18	18	18
Hal...Hal	6	6	6	6	6	6	6	6	6	6	6	6	6
Hal...H	8	8	8	8	8	8	8	8	8	8	8	8	8
Hal...O	12	12	12	12	12	12	12	12	12	12	12	12	12
O...H	28	28	28	30	28	28	28	28	28	28	28	28	28
C...C	17	17	17	17	13	17	17	17	17	17	17	17	17
C...H	86	86	86	86	90	86	86	90	90	90	90	90	90
H...H	88	88	88	84	88	88	92	88	88	88	88	88	86
Intermolecular interactions													
Hal...H	92	88	80	87	76	80	84	80	80	80	80	80	78
Hal...O	20	20	20	20	20	20	20	20	20	20	20	20	22
Hal...Hal	22	22	22	22	22	22	22	22	22	22	18	18	20
Hal...N	20	20	20	20	20	20	20	12	12	12	12	12	14
Hal...C	16	16	16	16	16	16	16	16	16	16	16	16	20
H...H	46	46	46	46	42	46	46	42	42	42	42	42	44
H...O	40	40	48	42	56	52	48	52	52	52	52	52	52
C...H	56	56	52	52	56	56	56	52	52	52	52	52	52
N...H	32	32	32	32	32	32	32	32	32	32	32	32	32
Co...H	4	4	4	4	4	4	4	4	4	4	4	4	4
C...C	0	0	0	8	8	4	4	0	4	4	4	4	6

<sup>a</sup> The interacting pairs are denoted as A...Z. The full details of such procedure a comparison of the conformation polymorphs are described in ref. 19.