

Nitrosyl and carbene iron complexes bearing κ^3 -SNS thioamide pincer type ligand

Tatsuya Suzuki^a, Jun Matsumoto^a, Yuji Kajita^b, Tomohiko Inomata^a, Tomohiro Ozawa^a, and Hideki Masuda^{*a}

^aDepartment of Frontier Materials, Graduate School of Engineering, Nagoya Institute of Technology, Gokiso, Showa, Nagoya 466-8555, Japan

^bDepartment of Applied Chemistry, Aichi Institute of Technology, 1247 Yachigusa, Yakusa-cho, Toyota 470-0392, Japan

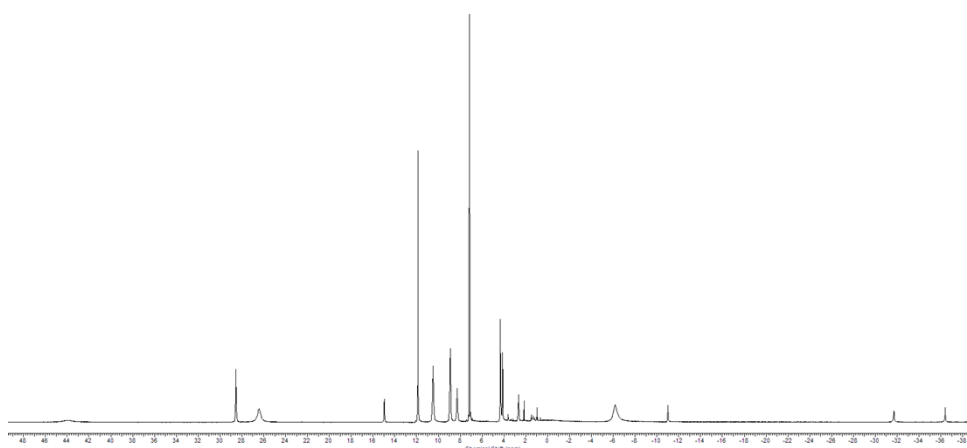


Fig. S1 ¹H-NMR spectrum of [Fe(NHC)(κ^3 -L^{DPM})] in C₆D₆.

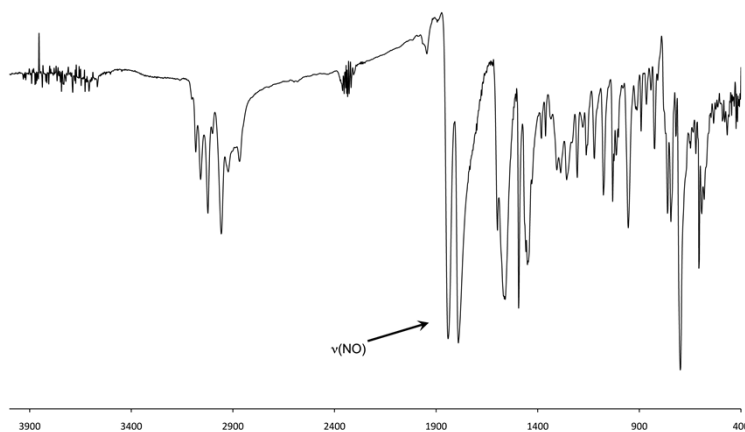


Fig. S2 IR spectrum of [Fe(NO)₂(κ^3 -L^{DPM})].

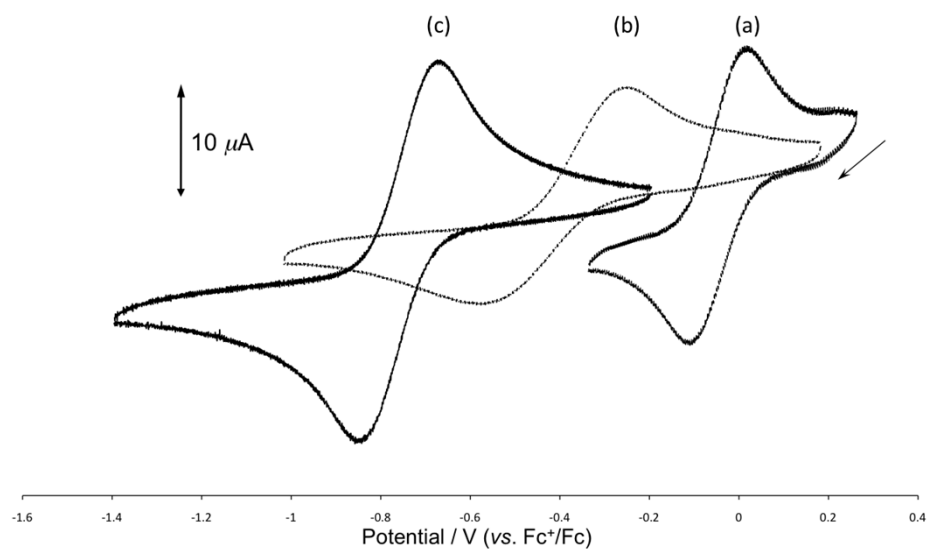


Fig. S3 Cyclic voltammogram of $[\text{Fe}(\text{THF})_2(\kappa^3\text{-LDPM})]$ (1 mM); (a), $[\text{Fe}(\text{NHC})(\kappa^3\text{-LDPM})]$ (1 mM); (b) and $[\text{Fe}(\text{NO})_2(\kappa^3\text{-LDPM})]$ (1 mM); (c) in THF under Ar at sweep rate of 100 mV s^{-1} .

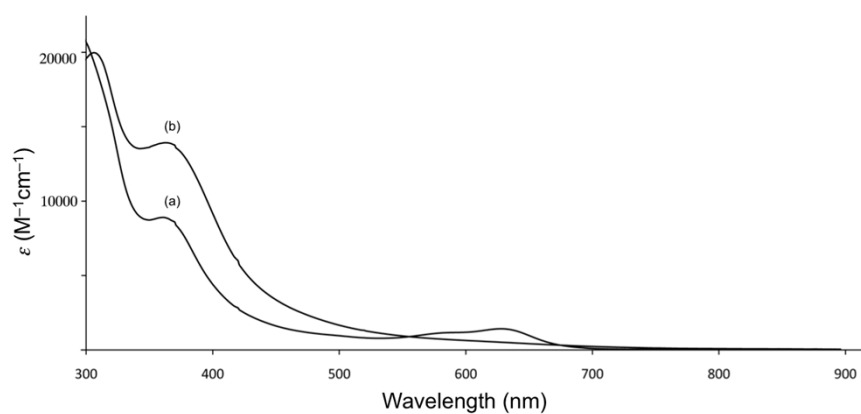


Fig. S4 UV/vis spectra of $[\text{Fe}(\text{NHC})(\kappa^3\text{-LDPM})]$ (a) and $[\text{Fe}(\text{NO})_2(\kappa^3\text{-LDPM})]$ (b) as 0.1 mM in THF.

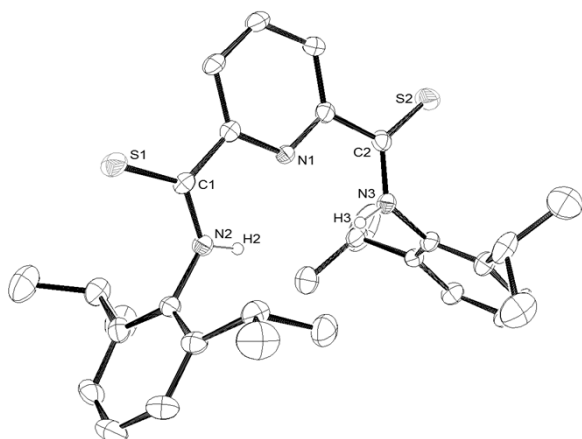


Fig. S5 ORTEP drawing of $\mathbf{H}_2\mathbf{L}^{\text{DIP}}$ ligand with ellipsoids at 30% probability level. The hydrogen atoms and aromatic substituents on N–thioamide group are omitted for clarity except for H(2) and H(3) atoms. Selected bond lengths (Å): C(1)–N(2) 1.337(5), C(2)–N(3) 1.340(5), C(1)–S(1) 1.649(4), C(2)–S(2) 1.642(4).

Table S1. Crystallographic and structure refinement data for compounds $\mathbf{H}_2\mathbf{L}^{\text{DIP}}$, $[\text{Fe}(\kappa^3\text{-H}_2\mathbf{L}^{\text{DIP}})_2](2\cdot\text{Br})$, $[\text{Fe}(\text{NHC})(\kappa^3\text{-L}^{\text{DPM}})]$, and $[\text{Fe}(\text{NO})_2(\kappa^3\text{-L}^{\text{DPM}})]$.

Compound	$\mathbf{H}_2\mathbf{L}^{\text{DIP}}$	$[\text{Fe}(\kappa^3\text{-H}_2\mathbf{L}^{\text{DIP}})_2](2\cdot\text{Br})$	$[\text{Fe}(\text{NHC})(\kappa^3\text{-L}^{\text{DPM}})]$	$[\text{Fe}(\text{NO})_2(\kappa^3\text{-L}^{\text{DPM}})]$
Chemical formula	$\text{C}_{31}\text{H}_{39}\text{N}_3\text{S}_2 \cdot \text{CHCl}_3$	$\text{C}_{62}\text{H}_{78}\text{FeN}_6\text{S}_4 \cdot 2\text{Br} \cdot 3(\text{C}_2\text{H}_6\text{O})$	$\text{C}_{104}\text{H}_{101}\text{FeN}_5\text{S}_2 \cdot 2(\text{C}_6\text{H}_6) \cdot \text{C}_5\text{H}_{12}$	$\text{C}_{77}\text{H}_{65}\text{FeN}_5\text{O}_2\text{S}_2 \cdot \text{CH}_2\text{Cl}_2$
Formula weight	637.17	1389.44	1771.33	1297.29
Temp (°C)	–100	–100	–120	–100
Crystal system	Monoclinic	Triclinic	Monoclinic	Triclinic
Space group	$P2_1/n$ (#14)	$P-1$ (#2)	$P2_1/n$ (#13)	$P-1$ (#2)
$a / \text{Å}$	12.089(3)	12.572(2)	19.615(3)	8.53160(10)
$b / \text{Å}$	15.277(3)	15.106(3)	22.908(3)	17.3689
$c / \text{Å}$	18.575(4)	18.852(3)	20.265(3)	23.9404(2)
$\alpha / ^\circ$		87.160(4)		73.322(4)
$\beta / ^\circ$	97.530(3)	84.309(4)	108.4184(12)	84.257(4)
$\gamma / ^\circ$		87.522(4)		79.592(4)
$V / \text{Å}^3$	3400.9(11)	3805(2)	8639(2)	3338.07(8)
Z	4	2	2	2
$D_{\text{calc}} / \text{g cm}^{-3}$	1.244	1.298	1.142	1.291
$\mu(\text{Mo-K}\alpha) / \text{cm}^{-1}$	4.171	15.036	2.375	4.197
$F(000)$	1344	1460	1892	1356
Reflections collected	26859	28056	40001	26248
Independent reflections	7774	15634	11790	14602
$R(\text{int})$	0.0721	0.0500	0.0359	0.0228
$RI (I > 2\sigma(I))^a$	0.0846	0.0568	0.0552	0.0520
RI (all)	0.1507	0.0768	0.0667	0.0704
$wR2$ (all)	0.1941	0.1459	0.1556	0.1453
GOF	1.074	1.035	1.079	1.052
CCDC number	1028474	1028475	1028476	1028477

$$^a R = \sum \|F_o\| - \|F_c\| / \sum \|F_o\|, wR2 = [\sum (w(F_o^2 - F_c^2)^2) / \sum w(F_o^2)^2]^{1/2}$$