Supporting Information

Color-tunable and white-light emission of one-dimensional L-di-2-thenoyltartaric acid mixed-lanthanide coordination polymers

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Figure S1. IR spectra of the H₂L and complexes 1–7.



Figure S2. UV absorption spectra of the H_2L and complexes 1–7 in methanol solution.

Structural parameter	1	2	3	4
the range of O–Ln bond distance (Å)	2.424(8)-2.679(7)	2.325(8)-2.576(8)	2.303(11)-2.566(9)	2.266(9)-2.538(9)
separation of Ln1…Ln2 (Å)	4.5825(15)	4.4785(15)	4.4568(15)	4.4383(15)
separation of Ln1(2)…Ln1(2) (Å)	9.8870(51)	9.7490(5)	9.7280(51)	9.7170(51)
the range of O–Ln–O bond angles (°)	49.0(2)-146.5(3)	51.2(3)-156.6(5)	50.9(3)-157.6(4)	52.1(3)-158.4(4)
Ln1–O–Ln2 bond angles (°)	122.169(261)	123.209(322)	122.801(351)	123.813(332)

Table S1. Bond lengths (Å) and angles (deg) for complexes 1–4.

Table S2. Partial crystal data for complex 5.

Comp.	$[(Eu_{0.037}Tb_{0.963})_2L_3(CH_3OH)_2(H_2O)_4] \cdot CH_3OH \cdot 2.75H_2O (\textbf{5})$
Shape	rod like
Color	Colorless
Crystal system	Orthorhombic
Space group	$P2_{1}2_{1}2_{1}$
a (Å)	9.732(5)
b (Å)	21.990(5)
c (Å)	31.208(5)
a (deg)	90.00
β (deg)	90.00
γ (deg)	90.00
V (Å ³)	6679(4)

Table S3. Elemental analysis of lanthanide ions by ICP for complexes 5–7.

Comp.	5		6			7		
-	Eu	Tb	Eu	Tb	La	Eu	Tb	Gd
Wt % (Found)	3.5	96.5	7.0	39.2	53.8	4.9	40.9	54.2
Mol %	3.7	96.3	6.8	36.3	56.9	5.1	40.6	54.3







Figure S3. TG-DSC curves of complexes 1–7.







Figure S4. Powder X-ray diffraction patterns and the simulated patterns of complexes 1–7.



Figure S5. Solid-state excitation spectra and emission spectra for H_2L at room temperature.



Figure S6. Solid-state excitation spectra for complexes 1–4 at room temperature.



Figure S7. Decay curves for complexes 2 and 4.



Figure S8. Phosphorescence spectrum for complex 3 in methanol solution at 77K.



Figure S9. Decay curves for complex 5 excited at 326 nm.



Figure S10. Decay curves for complex 6 excited at 322 nm.



Figure S11. Decay curves for complex 7 excited at 323 nm.



Figure S12. Left: Solid-state emission spectra of the complex 6 with excitation wavelengths varying from 310 to 350 nm (scanning interval = 5 nm). Inset: optical photograph of a powder sample of complex 6 excited at corresponding excitation wavelength. Right: The CIE-1931 chromaticity diagram for complex 6 with excitation wavelengths varying from 310 to 390 nm (scanning interval = 5 nm) showing the location of the color-tunable chromaticity of visual emission image.



Figure S13. Left: Solid-state emission spectra of the complex 7 with excitation wavelengths varying from 310 to 350 nm (scanning interval = 5 nm). Inset: optical photograph of a powder sample of complex 7 excited at corresponding excitation wavelength. Right: The CIE-1931 chromaticity diagram showing the emissions for complex 6 with excitation wavelengths varying from 310 to 390 nm (scanning interval = 5 nm) showing the location of the color-tunable chromaticity of visual emission image.

Excitation	5	6	7
λ/nm	CIE (x, y)	CIE (x, y)	CIE (x, y)
310	(0.408, 0.504)	(0.376, 0.441)	(0.395, 0.462)
315	(0.395, 0.467)	(0.370, 0.427)	(0.392, 0.441)
320	(0.360, 0.399)	(0.348, 0.380)	(0.371, 0.396)
325	(0.335, 0.366)	(0.310, 0.294)	(0.329, 0.308)
330	(0.299, 0.292)	(0.277, 0.242)	(0.291, 0.249)
335	(0.277, 0.255)	(0.259, 0.222)	(0.272, 0.228)
340	(0.276, 0.270)	(0.255, 0.233)	(0.271, 0.250)
345	(0.249, 0.219)	(0.238, 0.194)	(0.250, 0.208)
350	(0.265, 0.268)	(0.246, 0.231)	(0.263, 0.259)

Table S4. CIE chromaticity coordinates for the Ln³⁺-doped complexes **5**–7 excited from 310 to 350nm.