

Supporting Information

Solution-processable conjugated polymers containing alternating 1-alkyl-1,2,4-triazole and N=S=N Links

Mingfeng Wang, Fred Wudl*

Department of Chemistry and Biochemistry, Center for Polymers and Organic Solids, Mitsubishi Chemical Center for Advanced Materials and Materials Research Laboratory, University of California, Santa Barbara, CA 93106-6105

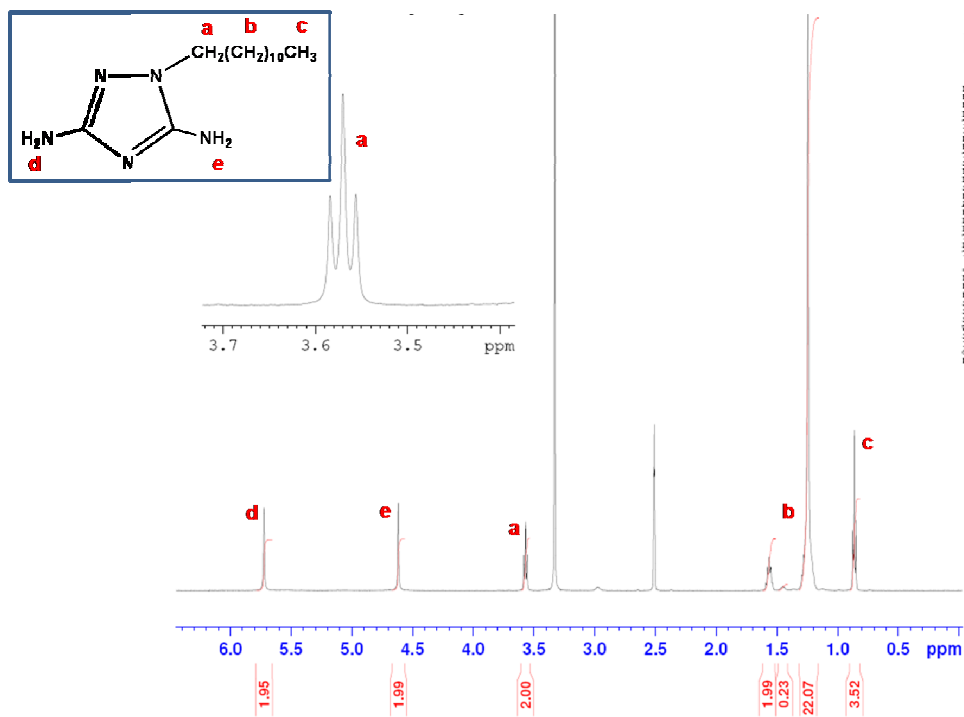


Figure S1. ¹H-NMR (DMSO-d₆) of DDTA.

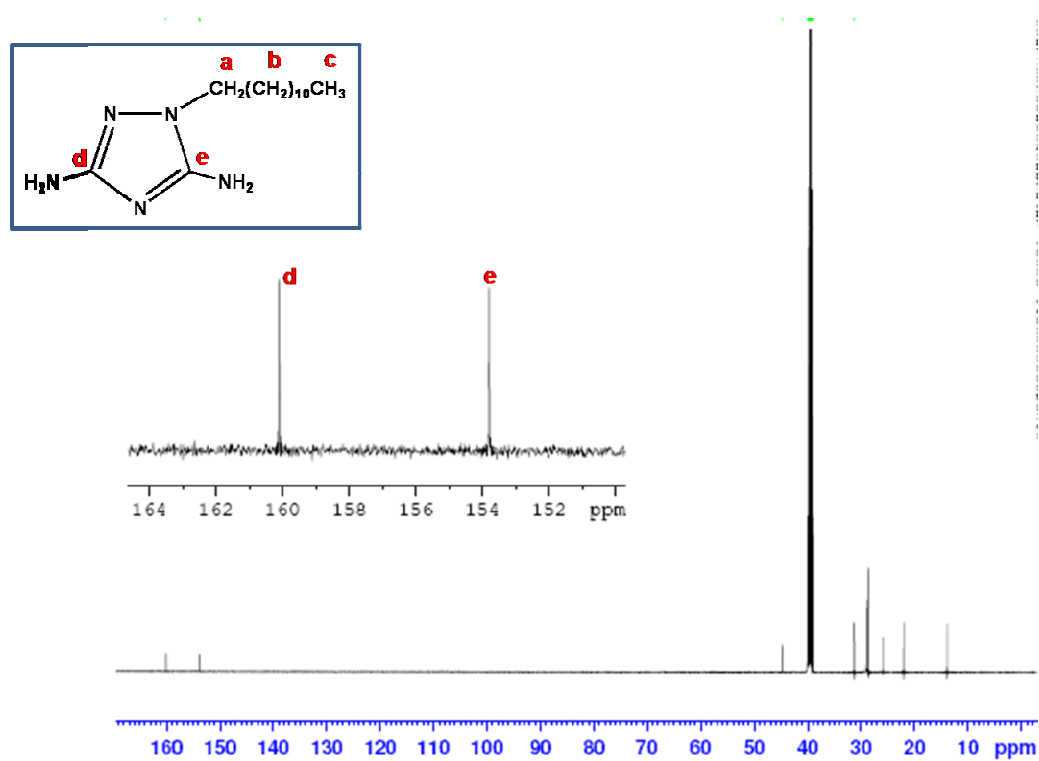


Figure S2. ^{13}C -NMR (DMSO-d_6) of DDTA.

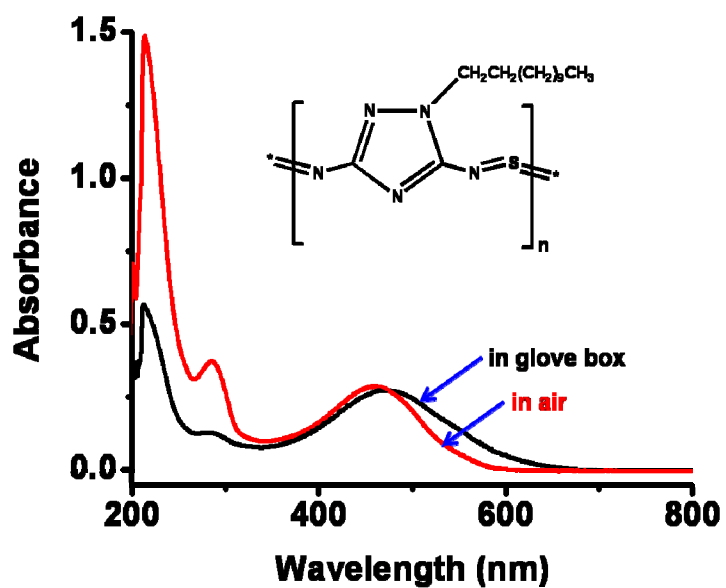


Figure S3. UV-vis absorption spectra of PBSDDT-1 in THF prepared under different conditions. This polymer was synthesized from the polymerization of BSDDT-1 that was prepared by using N-sulfinyl-p-toluenesulfonamide as the sulfinylating agent. DMAP was used as the catalyst for the polymerization.

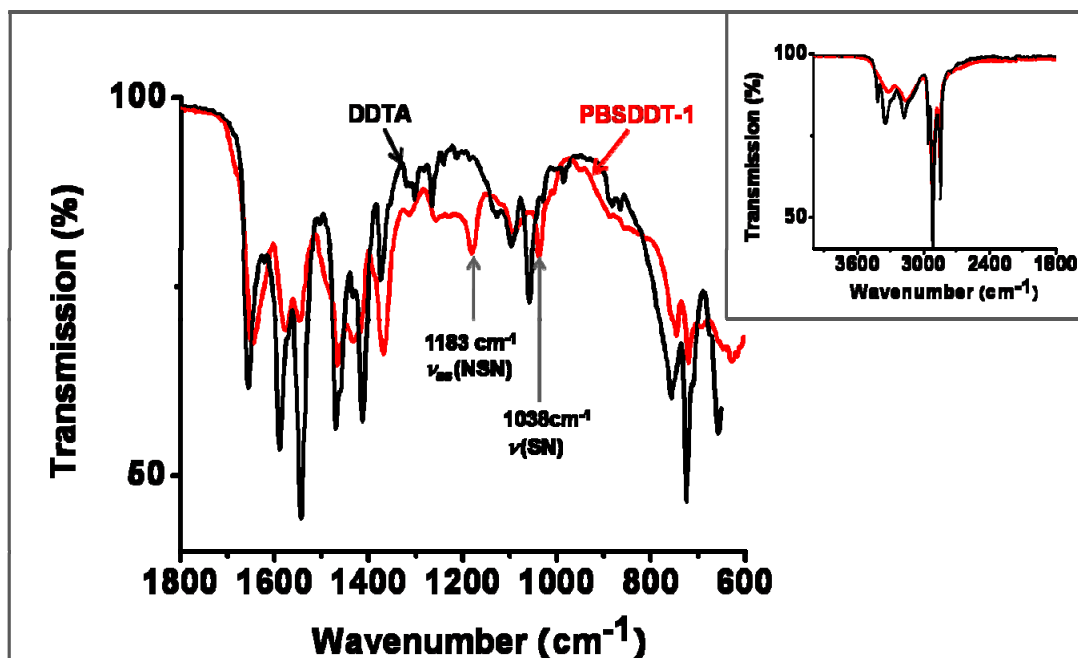


Figure S4. FTIR spectra of DDTA-1 and PBSDDT-1 that was purified twice by precipitation in CH₃CN. This polymer was synthesized from the polymerization of BSDDT that was prepared by using N-sulfinyl-p-toluenesulfonamide as the sulfinylating agent. DMAP was used as the catalyst for the polymerization. The polymer product was purified twice by precipitation in CH₃CN.

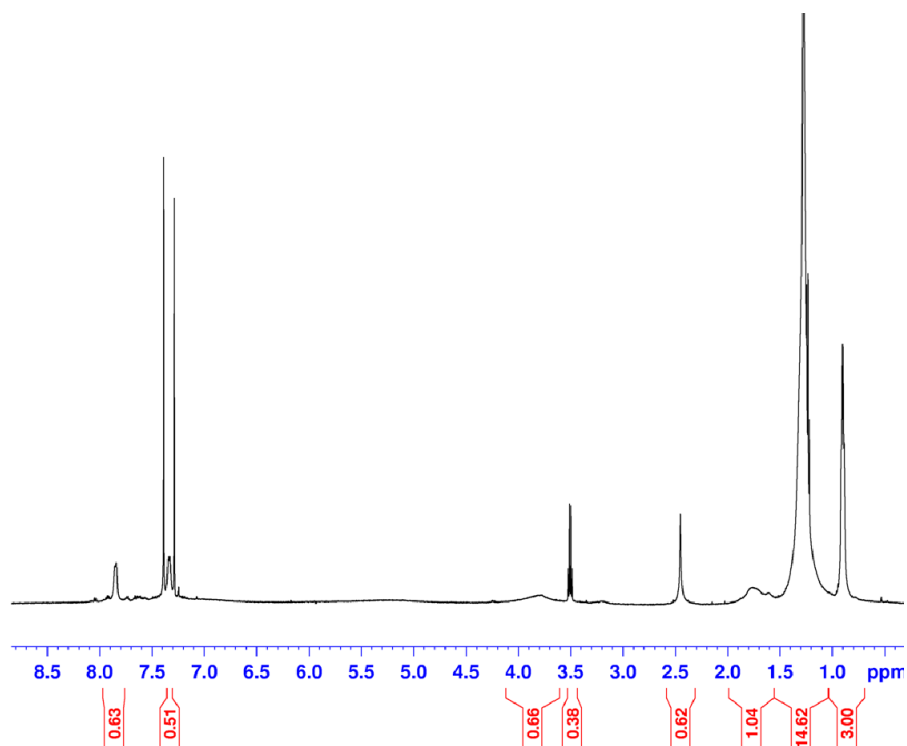


Figure S5. ¹H-NMR (CDCl₃) of PBSDDT-2. This polymer was synthesized from the polymerization of BSDDT-1 that was prepared by using N-sulfinyl-p-toluenesulfonamide as the sulfinylating agent. AlCl₃ was used as the catalyst for the polymerization.

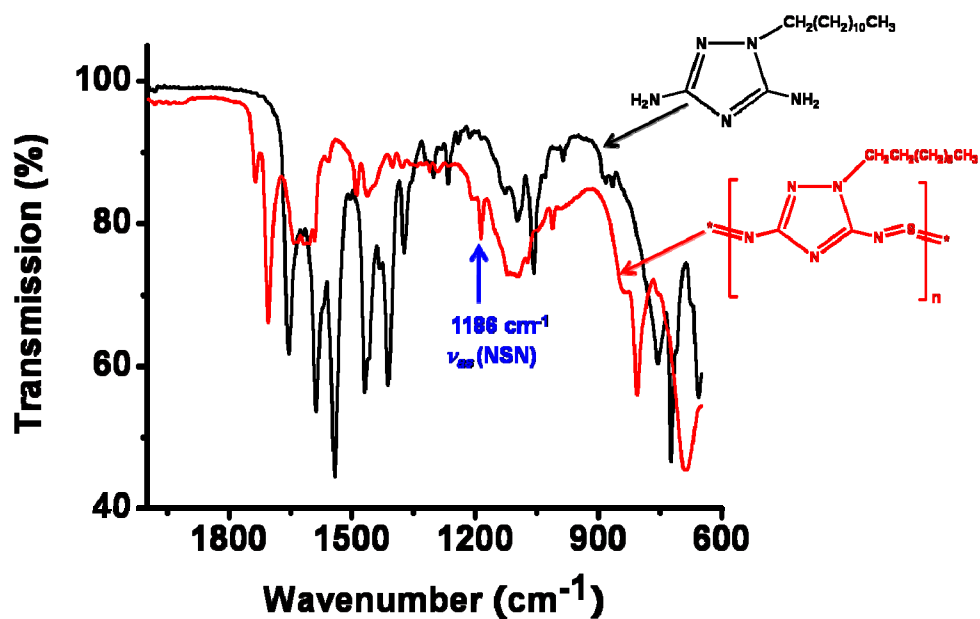


Figure S6. FTIR spectra of DDTA and PBSDDT-2. This polymer was synthesized from the polymerization of BSDDT-1 that was prepared by using N-sulfinyl-p-toluenesulfonamide as the sulfinylating agent. AlCl₃ was used as the catalyst for the polymerization.