

Supporting Information

for

Dual Optical Responses of Phenothiazine Derivatives: Near-IR Chromophore and Water-soluble Fluorescent Organic Nanoparticles

Sheng-Yuan Su,^a Hsin-Hung Lin,^b Cheng-Chung Chang^{*b}

^a Department of chemistry, National Chung Hsing University 250, Kuo Kuang Road, Taichung 402, Taiwan, R.O.C.

^b Graduate Institute of Biomedical Engineering, National Chung Hsing University 250, Kuo Kuang Road, Taichung 402, Taiwan, R.O.C.

Fax: (886)-4-228 52422

Tel: (886)-4-22840734 ext. 24

E-mail: ccchang555@dragon.nchu.edu.tw

1. General procedure for the synthesis of phenothiazine derivatives (Scheme 2).

1. 3-((E)-(4-aminostyryl))-10H-phenothiazine (**PTZ-NH**)
2. 3-((E)-(4-methoxystyryl))-10H-phenothiazine (**PTZ-OMe**)
3. 3-((E)-(4-acetoxystyryl))-10H-phenothiazine (**3**)
4. 3-((E)-2-(phenol-4-yl) vinyl)-10H-phenothiazine (**PTZ-OH**)
5. 3-((E)-2-(phenol-4-yl) vinyl)-10-methyl-phenothiazine (**MPTZ-OH**)

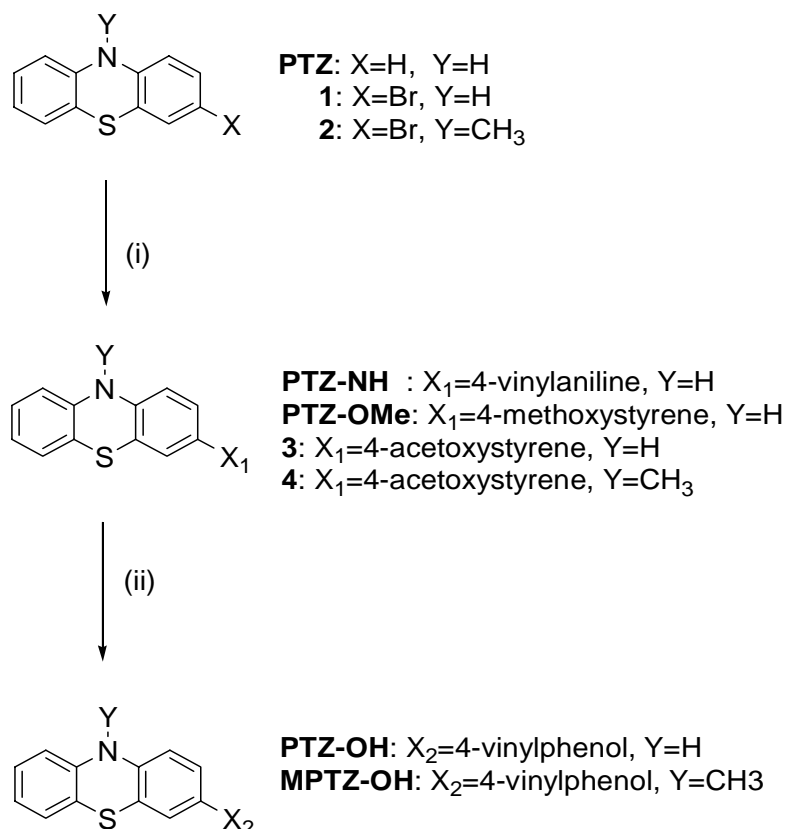
2. Figure S1 Deprotonated titration spectrum of PTZ-OH in Na₃PO₄ containing DMSO solution. Insert shows the first stage for the appearance of phenolate-PTZ, while the second stage shows the unparent appearance of phenothiazinated-PTZ and dominate component hybrid-PTZ.

3. Figure. S2 Absorption (left) and emission (right) spectral illustrations of 25 uM MPTZ-OH in alkaline aqueous solutions (10 mM). Insert photo shows the emission images under the UV light (365 nm)

4. Figure. S3 Absorption (left) and emission (right) spectral illustrations of variable concentration of PTZ-OH in KOH aqueous solutions (10 mM). Insert shows the emission intensities (at 500 nm) plots.

1. General procedure for the synthesis of phenothiazine derivatives (Scheme 2)

Synthesis of the phenothiazine derivatives is shown in Scheme 1. 10*H*-phenothiazine or *N*-methyl-phenothiazine containing solution was subjected to bromination with NBS/THF in an additional funnel. Next, the samples were subjected to the Heck coupling reaction³⁴ with 4-vinylaniline, 4-acetoxystyrene or 4-methoxystyrene under catalyst Pd (OAc)₂. Finally, 4-vinylphenol substituted PTZ derivatives were easily prepared by the addition of KOH in a methanol system.



Scheme 2 Synthesis of PTZ derivatives. Reaction reagents and conditions: (i) Mixture of Pd(OAc)₂/(*o*-tol)₃P complex and 4-vinylaniline (for PTZ-NH) or 4-methoxyly styrene (for PTZ-OMe) or 4-acetoxily styrene (for compound 3 and 4) with Et₃N/MeCN as a solvent pair. Reflux under nitrogen, 48 h. (ii) KOH, methanol, reflux, 4 h

1. 3-((*E*)-(4-aminostyryl))-10*H*-phenothiazine (PTZ-NH)

Compound 1 (5 mmole) was added to a high pressure bottle containing a mixture of palladium (II) acetate (8 mg, Strem) and tri-*o*-tolyl phosphine (80 mg, Aldrich), after which the solvent pair (triethylamine 5 mL/acetonitrile 15 mL) and 4-vinylanile (8 mmole, Acros) was added. The bottle was then sealed after bubbling with nitrogen for 10 min. After keeping the system under ~105 °C for two days, the system was cooled to room temperature and then extracted with CH₂Cl₂/H₂O twice. The organic layer was then dried with MgSO₄ and evaporated in under vacuum, after which the residue was subjected to chromatography on a silica gel using hexane/ethyl acetate (1/2, R_f = 0.4). A

dark-yellow solid compound was then obtained by recrystallization with acetone/hexane (yield: 52%). Data for **PTZ-NH**: ¹HNMR (400 Hz, DMSO-d₆): δ = 8.614 (s, 1H), 7.198 (d, *J* = 8.4 Hz, 2H), 7.102 (d, *J* = 8.0 Hz, 1H), 7.052 (s, 1H), 6.965 (t, *J* = 8.4 Hz, 1H), 6.898 (d, *J* = 8.0 Hz, 1H), 6.840 (d, *J* = 16.0 Hz, 1H), 6.728 (t, *J* = 8.0 Hz, 1H), 6.690 (d, *J* = 16.0 Hz, 1H), 6.650 (d, *J* = 8.0 Hz, 1H), 6.625 (d, *J* = 8.0 Hz, 1H), 6.520 (d, *J* = 8.4 Hz, 2H), 5.285 (b, 2H). Anal. Calcd. %. For C₂₀H₁₆N₂S: C, 75.92; H, 5.10; N, 8.85. C₂₀H₁₈N₂OS·1H₂O: C, 71.83; H, 5.42; N, 8.38. Observation: C, 71.02; H, 5.49; N, 8.29.

2. 3-((E)-(4-methoxystyryl)-10H-phenothiazine (PTZ-OMe)

A procedure similar to that used to prepare the **PTZ-NH** was employed; however, 4-methoxystyrene (10 mmole, Merck) was added. The residue was then chromatographed on silica gel with hexane/acetone (2/1) and recrystallized from acetone/ethyl acetate (yield: 75%) to give a light-yellow solid. Data for **PTZ-OMe**: ¹HNMR (400 Hz, DMSO-d₆): δ = 8.664 (s, 1H), 7.450 (d, *J* = 8.4 Hz, 2H), 7.16 (d, *J* = 8.0 Hz, 1H), 7.124 (s, 1H), 6.980 (d, *J* = 8.0 Hz, 1H), 6.938 (d, *J* = 16.2, 1H), 6.907 (d, *J* = 8 Hz, 2H), 6.879 (d, *J* = 16.2 Hz, 1H), 6.747 (t, *J* = 7.8 Hz, 1H), 6.358 (d, *J* = 7.8 Hz, 1H), 6.637 (d, *J* = 7.8 Hz, 2H), 3.751 (s, 3H). Anal. Calcd. %. For C₂₁H₁₇NOS: C, 76.10; H, 5.17; N, 4.23. Observation: C, 76.01; H, 5.19; N, 4.25.

3. 3-((E)-(4-acetoxystyryl)-10H-phenothiazine (3)

Compound **1** (5 mmole) was added to a high pressure bottle containing a mixture of palladium (II) acetate (8 mg, Strem) and tri-*o*-tolyl phosphine (80 mg, Aldrich), after which the solvent pair (triethylamine 5 mL/acetonitrile 15 mL) and 4-acetoxystyrene (8 mmole, Merck) was added. The bottle was then sealed after bubbling with nitrogen for 10 min. After keeping the system under ~105 °C for two days, the system was cooled to room temperature and then extracted with CH₂Cl₂/H₂O twice. The organic layer was then dried using MgSO₄ and evaporated under vacuum, after which the residue was chromatographed on silica gel with hexane/acetone (2/1). A light yellow solid compound was obtained by recrystallization with acetone/ethyl acetate (yield: 57%). Data for **3**: ¹HNMR (400 Hz, DMSO-d₆): δ = 8.719 (s, 1H), 7.540 (d, *J* = 8.4 Hz, 2H), 7.183 (d, *J* = 8.0 Hz, 1H), 8.163 (s, 1H), 7.090 (d, *J* = 8.4 Hz, 2H), 7.030 (s, 2H, double bond), 6.974 (t, *J* = 8.0 Hz, 1H), 6.907 (d, *J* = 8.0 Hz, 1H), 6.741 (t, *J* = 8.0 Hz, 1H), 6.663 (d, *J* = 8.0 Hz, 1H), 6.649 (d, *J* = 8.0 Hz, 1H), 2.257 (s, 3H).

4. 3-((E)-2-(phenol-4-yl)vinyl)-10H-phenothiazine (PTZ-OH)

After refluxing compound **3** with excess KOH in methanol for four hours, the system was evaporated in vacuum and the sample was then extracted with CH₂Cl₂/1N HCl twice. Next, the organic layer was dried with MgSO₄ and the collected residue was chromatographed on a silica gel using hexane/acetone (1/2). Finally, yellow powders (yield: 92%) were recrystallized with

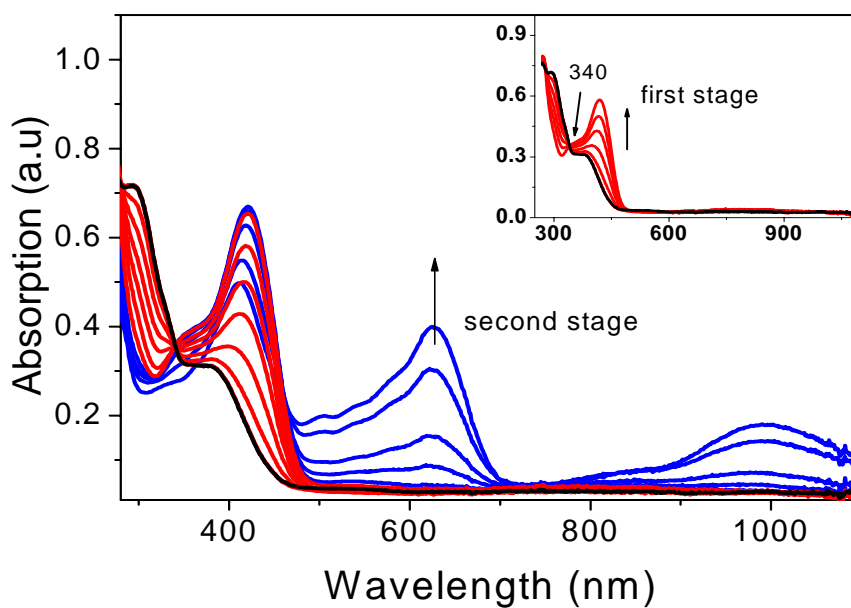
acetone/hexane. Data for **PTZ-OH**: $^1\text{H NMR}$ (DMSO- d_6): δ = 9.497 (s, 1H), 8.650 (s, 1H), 7.332 (d, J = 8.4 Hz, 2H), 7.137 (d, J = 8.0 Hz, 1H), 7.098 (s, 1H), 6.970 (t, J = 8.0 Hz, 1H), 6.925 (d, J = 16.4 Hz, 1H), 6.903 (d, J = 8.0 Hz, 1H), 6.801 (d, J = 16.4 Hz, 1H), 6.735 (t, J = 8.0 Hz, 1H), 6.725 (d, J = 8.4 Hz, 2H), 6.656 (d, J = 8.0 Hz, 1H), 6.628 (d, J = 8.0 Hz, 1H). Anal. Calcd. %. For $\text{C}_{25}\text{H}_{15}\text{NOS}$: C, 75.68; H, 4.76; N, 4.41. Observation: C, 75.58; H, 4.79; N, 4.40.

5. 3-((E)-2-(phenol-4-yl) vinyl)-10-methyl-phenothiazine (MPTZ-OH)

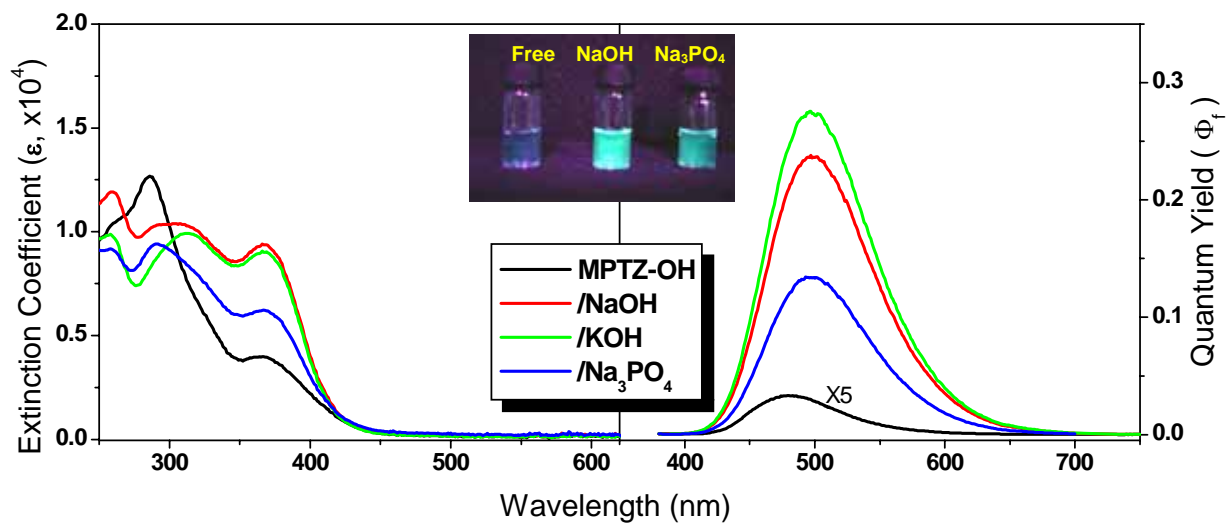
Compound 3-((E)-(4-acetoxystyryl))-10-methyl-phenothiazine (**4**), the precursor of **MPTZ-OH**, was prepared following the same procedure used to prepare compound **3**, except that compound **2** was used as the starting material. The residue was then chromatographed on silica gel by hexane/acetone (3/1, R_f = 0.5). A light-yellow solid was collected by recrystallization with methanol (yield: 65%). Data for **4**: $^1\text{H NMR}$ (DMSO- d_6): δ = 7.570 (d, J = 8.4 Hz, 2H), 7.413 (s, 1H), 7.304 (dd, J = 8.0, 2.4 Hz, 1H), 7.206 (dd, J = 8.0, 1.2 Hz, 1H), 7.166 (dd, J = 8.0, 1.6 Hz, 1H), 7.150 (d, J = 16.2 Hz, 1H), 7.123 (d, J = 16.2 Hz, 1H), 7.098 (d, J = 8.4 Hz, 2H), 6.950 (m, 3H), 3.311 (s, 3H), 2.234 (s, 3H).

Following the preparation of **PTZ-OH**, the final product was chromatographed on silica gel by hexane/acetone (3/1, R_f = 0.4) and a yellow powder was recrystallized from hexane/acetone (yield: 90 %). Data for **MPTZ-OH**: $^1\text{H NMR}$ (400 Hz, DMSO- d_6): δ = 9.559 (s, 1H), 7.358 (d, J = 8.4 Hz, 2H), 7.339 (m, 2H), 7.198 (dd, J = 8.0, 1.6 Hz, 1H), 7.147 (dd, J = 8.0, 1.2 Hz, 1H), 7.018 (d, J = 16.0 Hz, 1H), 6.940 (m, 2H), 6.904 (d, J = 8.0 Hz, 1H), 6.888 (d, J = 16.0 Hz, 1H), 6.734 (d, J = 8.4 Hz, 2H), 3.257 (s, 3H). Anal. Calcd. %. For $\text{C}_{21}\text{H}_{17}\text{NOS}$: C, 76.10; H, 5.17; N, 4.23. Observation: C, 75.69; H, 5.20; N, 4.20.

2. Figure S1 Deprotonated titration spectrum of PTZ-OH in Na_3PO_4 containing DMSO solution. Insert shows the first stage for the appearance of phenolate-PTZ, while the second stage shows the unapparent appearance of phenothiazinated-PTZ and dominate component hybrid-PTZ.



3. Figure. S2 Absorption (left) and emission (right) spectral illustrations of 25 μM MPTZ-OH in alkaline aqueous solutions (10 mM). Insert photo shows the emission images under the UV light (365 nm)



4. Figure. S3 Absorption (left) and emission (right) spectral illustrations of variable concentration of PTZ-OH in KOH aqueous solutions (10 mM). Insert shows the emission intensities (at 500 nm) plots.

