## SUPPLEMENTARY INFORMATION

## Phase Transition Thermochromism Based on C-H Acidity of 4-Alkylflavylium Compounds in Pluronic<sup>®</sup> F-127

João Avó,<sup>a</sup> A. Jorge Parola,<sup>a,\*</sup> João C. Lima,<sup>a</sup> Fernando Pina,<sup>a,\*</sup> Luís Cunha-Silva<sup>b</sup>

<sup>a</sup>REQUIMTE, Departamento de Química, Faculdade de Ciências e Tecnologia, Universidade
Nova de Lisboa, 2829-516 Caparica, Portugal, Fax: +351212948550; Tel: +351212948355;
E-mail:ajp@dq.fct.unl.pt; E-mail: fjp@dq.fct.unl.pt

<sup>b</sup>REQUIMTE, Departamento de Química e Bioquímica, Faculdade de Ciências, Universidade do Porto, 4169-007 Porto, Portugal

<b>1.</b> Crystal packing of the flavylim salt 5b and the C–H…O hydrogen bounds	S2
2. UV-Vis spectra of ethylenic bases	S3
6a (Fig. S2) 6b (Fig. S2)	S3 S3
3. <sup>1</sup> H NMR spectra of 5a and 5b in CDCl <sub>3</sub> and in D <sub>2</sub> O/CD <sub>3</sub> OD (1:1) at pH≈4 and of 6b in CDCl <sub>3</sub> .	S4
5a (Fig. S3 and S4) 5b (Fig. S5 and S6) 6b (Fig. S7)	S4 S5 S6
<b>4.</b> Titration of compounds <b>1</b> and <b>5a</b> in aqueous 20% (w/v) Pluronic <sup>®</sup> F127	S7
1 (Fig. S8) 5a (Fig. S9)	S7 S7



## 1. Crystal packing of the flavylim salt 5b and the C-H…O hydrogen bounds

**Fig. S1** – a) Crystalline packing arrangement of the compound **5b** viewed in the [1 0 0] direction of the unit cell. b) Weak C–H···O hydrogen bound interactions (shown as golden dashed lines) between the perchlorate anion and the neighboring flavylium molecules; for clarity purpose the carbon atoms of distinct organic molecules are drawn in different colors.

Electronic Supplementary Material (ESI) for Journal of Materials Chemistry This journal is The Royal Society of Chemistry 2011

## 2. UV-Vis spectra of ethylenic bases



**Fig. S2** – UV-Vis spectra of ethylenic base solutions in chloroform: A. [**6a**]= $1.3 \times 10^{-5}$  M; B. [**6b**]= $2.0 \times 10^{-5}$  M

3. <sup>1</sup>H NMR spectra of 5a and 5b in CDCl<sub>3</sub> and in D<sub>2</sub>O/CD<sub>3</sub>OD (1:1) at pH $\approx$ 4, and of 6b in CDCl<sub>3</sub>.



**Fig. S3** - <sup>1</sup>H NMR spectrum of **5a** in CDCl<sub>3</sub>.



**Fig. S4** – <sup>1</sup>H NMR spectrum of **5a** in D<sub>2</sub>O/CD<sub>3</sub>OD (1:1) at pH $\approx$ 4.







**Fig. S6** – <sup>1</sup>H NMR spectrum of **5b** in  $D_2O/CD_3OD$  (1:1) at pH≈4.







4. Titration of compounds 1 and 5a in aqueous 20% (w/v) Pluronic® F127

**Fig. S8** – Absorption spectra of compound  $1 \, 1.2 \times 10^{-4}$  M in the presence of 20% (w/v) Pluronic<sup>®</sup> micelles at room temperature. Fitting of the absorbance according to eq. (9).



**Fig. S9** – Absorption spectra of compound **5a**  $2.3 \times 10^{-5}$  M in the presence of 20% (w/v) Pluronic<sup>®</sup> micelles at room temperature. Fitting of the absorbance according to eq. (9).