

Supporting information

A General and Feasible Method for Fabrication of Functional Nanoparticles in Mesoporous Silica Hollow Composite Spheres

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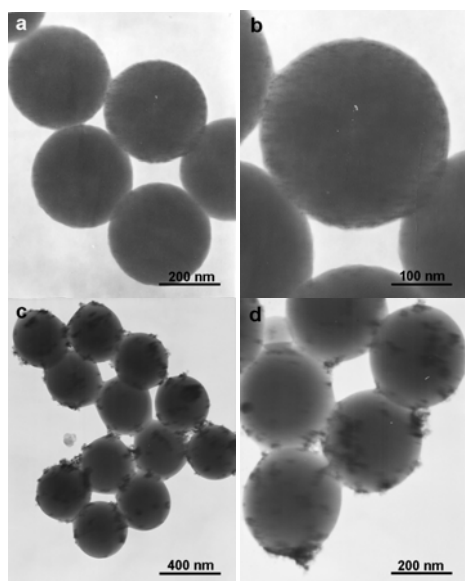


Figure S1. TEM images of (a, b) PS-Pt composite microspheres
and (c, d) PS-Fe₃O₄ composite microspheres.

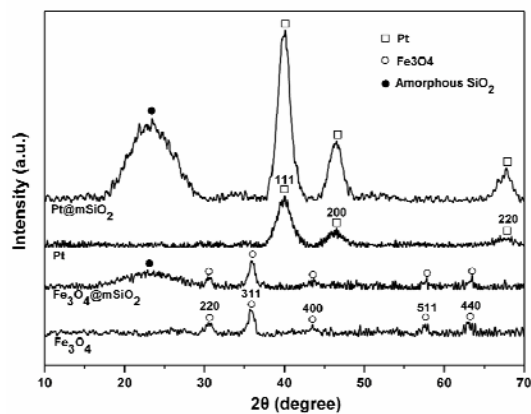


Figure S2. XRD patterns of Pt, Fe₃O₄ nanoparticles and their
HMNSs.

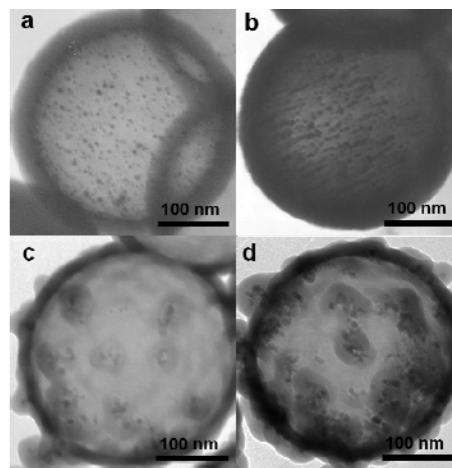


Figure S3. TEM images of Pt@SiO₂ HMNSs containing Pt NPs:
(a) 2.3 wt%, (b) 4.6 wt%, and Fe₃O₄@SiO₂ HMNSs containing
Fe₃O₄ NPs: (c) 1 wt%, (d) 2 wt%.

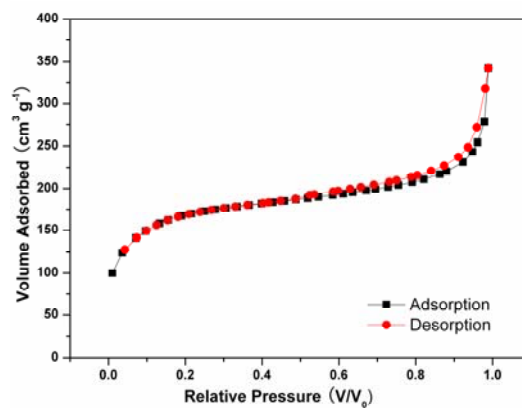


Figure S4. BET of the pure hollow mesoporous silica spheres.

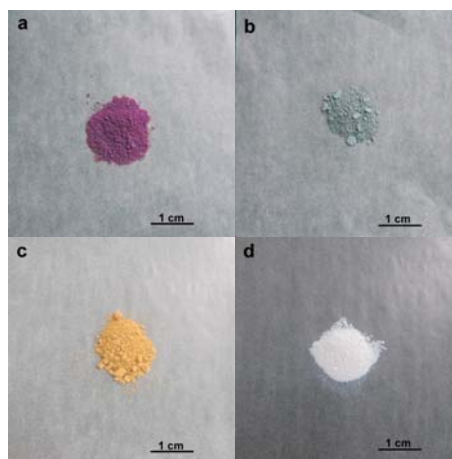


Figure S5. Photographs of (a) Au@SiO₂, (b) Pt@SiO₂, (c)
Fe₃O₄@SiO₂ HMNS powders and (d) pure hollow mesoporous
silica sphere powders.

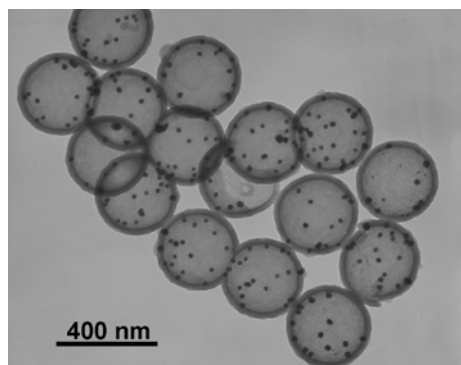


Figure S6. Typical TEM image of the Au@SiO₂ HMNSs after 8 cycles of catalysis.

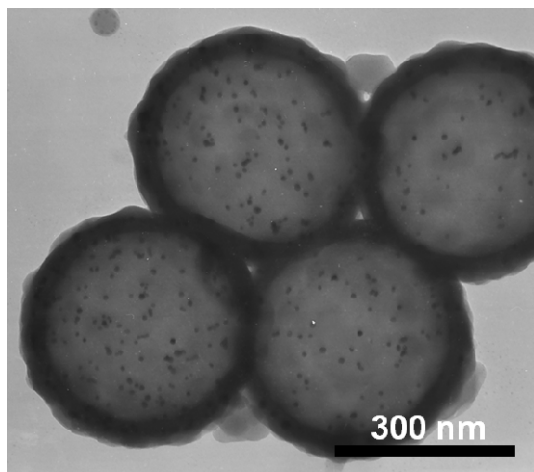
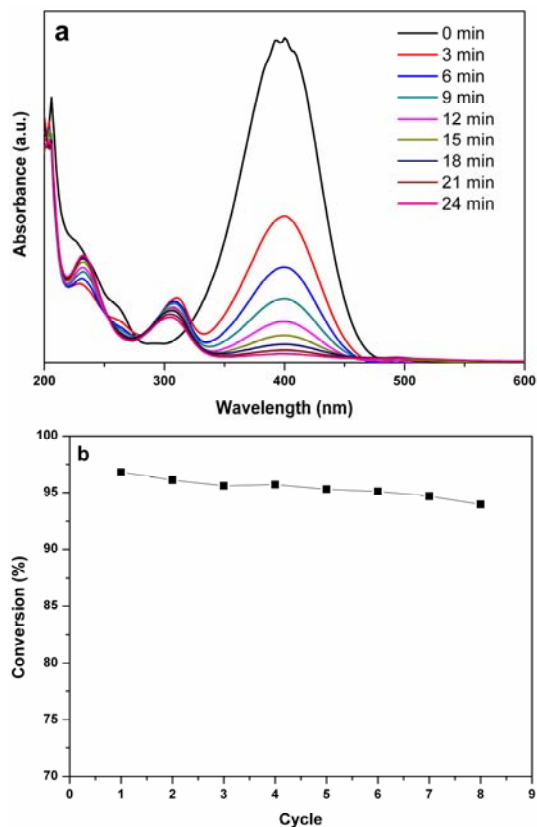
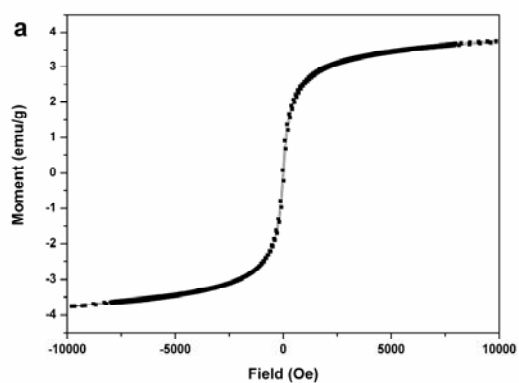


Figure S7. TEM image of Au@SiO₂ HMNSs with Au NPs of 4-5 nm.



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Figure S8. (a) Time-dependent UV-vis spectra of the solution catalyzed by Pt@SiO₂ HMNSs. (b) Reusability of Pt@SiO₂ HMNSs as catalysts for the reduction of 4-nitrophenol with NaBH₄. 1 mg of Pt@SiO₂ composites spheres were used, containing Pt 2.3×10^{-7} mol. Other reaction conditions were the same with the process of Au@SiO₂ as catalysis.



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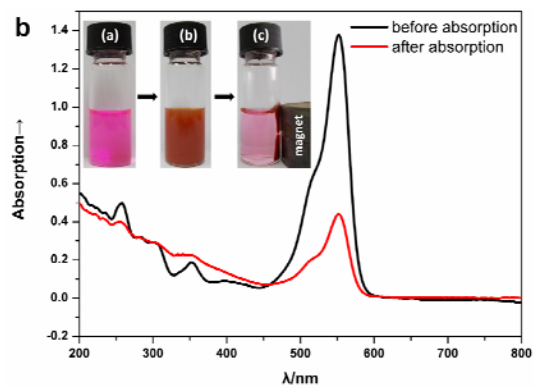


Figure S9. (a) The magnetic hysteresis loops of the $\text{Fe}_3\text{O}_4@\text{SiO}_2$ HMNSs. (b) UV-vis absorption spectra of 10 mg/L Rhodamine solution before and after absorption by the $\text{Fe}_3\text{O}_4@\text{SiO}_2$ HMNSs. Inset is the corresponding digital camera images: (a) 2 mL of 10 mg/L Rhodamine B solution, (b) mixture with 10 mg $\text{Fe}_3\text{O}_4@\text{SiO}_2$ HMNSs, (c) magnetic separation of the $\text{Fe}_3\text{O}_4@\text{SiO}_2$ HMNSs from the Rhodamine B solution under an external magnetic field.

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