Electronic Supplementary Information for

Synthesis of core-shell structured TS-1@mesocarbon material and its application as a tandem catalyst

Honggen Peng, Le Xu, Liyan Zhang, Kun Zhang, Yueming Liu, Peng Wu*

Shanghai Key Laboratory of Green Chemistry and Chemical Processes, Department of

Chemistry, East China Normal University, North Zhongshan Road 3663, Shanghai 200062,

China



Fig. S1 High resolution TEM image of TS-1@MC.



Fig. S2 SEM images of TS-1 (a) and TS-1@MC-TPAOH after calcination in air at 873 K for 6 h. The insert shows the image of uncalcined TS-1@MC-TPAOH.



Fig. S3 XRD patterns of TS-1@MC prepared by etching with TPAOH at different concentrations. Other conditions: 338 K and 4 h.



Fig. S4 N_2 adsorption/desorption isotherms of TS-1@MC prepared by etching with TPAOH at different concentration at 338 K for 4 h. The TPAOH was not removed by calconation in vacuum after TS-1@MS/C etched by TPAOH.

No.	TPAOH (M)	Yield (wt%)	Carbon content (%)	Si/Ti ratio	$\frac{S_{BET}}{(m^2 g^{-1})}$	$\frac{S_{ext}}{(m^2 g^{-1})}$	$\frac{V_{tot}}{(cm^3 g^{-1})}$	D _{meso} (nm)	D _{micro} (nm)
1	0	100	22.0	65	360	32.5	0.259	-	0.56
2	0.1	80.0	27.3	50	533	227	0.463	3.28	0.51
3	0.3	76.0	29.6	45	605	270	0.473	2.94	0.84
4	0.5	71.7	29.1	39	657	347	0.561	2.94	0.51
5	0.8	70.5	30.1	40	661	379	0.547	2.94	0.51

Table S1 Physicochemical properties of TS-1@MC etched with TPAOH at different concentration a

^a Other conditions: 338 K, 4 h. Yield in grams of solid after treatment per gram of starting material. The TPAOH was not removed by calconation in vacuum after TS-1@MS/C etched by TPAOH.



Fig. S5 Pd particle size distribution of Pd/TS-1@MC (A) and Pd/TS-1 (B).