

## Supporting information

**Table S1** Summary of the reported maximum power density of laboratory SOFC-H with hydrogen as fuel in recent years.

Year	Anode	Electrolyte	Cathode	T (°C)	P <sub>max</sub> (mW cm <sup>-2</sup> )	Ref.
2005	Pd	BaCe <sub>0.8</sub> Y <sub>0.2</sub> O <sub>3-δ</sub>	Perovskite	600	1400	[S1]
2006	Ni-BaCe <sub>0.8</sub> Sm <sub>0.2</sub> O <sub>3-δ</sub>	BaCe <sub>0.8</sub> Sm <sub>0.2</sub> O <sub>3-δ</sub>	Ba <sub>0.5</sub> Sr <sub>0.5</sub> Co <sub>0.8</sub> Fe <sub>0.2</sub> O <sub>3-δ</sub>	700	340	[S2]
2006	Ni-BaCe <sub>0.8</sub> Gd <sub>0.2</sub> O <sub>3-δ</sub>	BaCe <sub>0.8</sub> Gd <sub>0.2</sub> O <sub>3-δ</sub>	La <sub>0.5</sub> Sr <sub>0.5</sub> CoO <sub>3-δ</sub>	700	371	[S3]
2006	Ni-BaCe <sub>0.7</sub> Zr <sub>0.1</sub> Y <sub>0.2</sub> O <sub>3-δ</sub>	BaCe <sub>0.7</sub> Zr <sub>0.1</sub> Y <sub>0.2</sub> O <sub>3-δ</sub>	BaCe <sub>0.4</sub> Pr <sub>0.4</sub> Y <sub>0.2</sub> O <sub>3-δ</sub>	700	270	[S4]
2007	Ni-BaCe <sub>0.9</sub> Nd <sub>0.1</sub> O <sub>3-δ</sub>	BaCe <sub>0.9</sub> Nd <sub>0.1</sub> O <sub>3-δ</sub>	La <sub>0.5</sub> Sr <sub>0.5</sub> CoO <sub>3-δ</sub> -BaCe <sub>0.9</sub> Nd <sub>0.1</sub> O <sub>3-δ</sub>	700	335	[S5]
2008	Ni- BaCe <sub>0.7</sub> Zr <sub>0.1</sub> Y <sub>0.2</sub> O <sub>3-δ</sub>	BaCe <sub>0.7</sub> Zr <sub>0.1</sub> Y <sub>0.2</sub> O <sub>3-δ</sub>	GdBaCo <sub>2</sub> O <sub>5+δ</sub>	700	266	[S6]
2008	Ni- BaCe <sub>0.9</sub> Y <sub>0.1</sub> O <sub>3-δ</sub>	BaCe <sub>0.9</sub> Y <sub>0.1</sub> O <sub>3-δ</sub>	Ba <sub>0.5</sub> Sr <sub>0.5</sub> Co <sub>0.8</sub> Fe <sub>0.2</sub> O <sub>3-δ</sub>	700	550	[S7]
2008	Ni- BaCe <sub>0.7</sub> Zr <sub>0.1</sub> Y <sub>0.2</sub> O <sub>3-δ</sub>	BaCe <sub>0.7</sub> Zr <sub>0.1</sub> Y <sub>0.2</sub> O <sub>3-δ</sub>	Sm <sub>0.5</sub> Sr <sub>0.5</sub> CoO <sub>3-δ</sub> -BaCe <sub>0.7</sub> Zr <sub>0.1</sub> Y <sub>0.2</sub> O <sub>3-δ</sub>	700	725	[S8]
2009	Ni- BaCe <sub>0.7</sub> Zr <sub>0.1</sub> Y <sub>0.2</sub> O <sub>3-δ</sub>	BaCe <sub>0.7</sub> Zr <sub>0.1</sub> Y <sub>0.2</sub> O <sub>3-δ</sub>	SmBaCo <sub>2</sub> O <sub>5+δ</sub>	700	382	[S9]
2009	Ni- BaCe <sub>0.7</sub> Zr <sub>0.1</sub> Y <sub>0.2</sub> O <sub>3-δ</sub>	BaCe <sub>0.7</sub> Zr <sub>0.1</sub> Y <sub>0.2</sub> O <sub>3-δ</sub>	PrBaCo <sub>2</sub> O <sub>5+δ</sub>	700	545	[S10]
2009	Ni- BaCe <sub>0.7</sub> Zr <sub>0.1</sub> Y <sub>0.2</sub> O <sub>3-δ</sub>	BaCe <sub>0.7</sub> Zr <sub>0.1</sub> Y <sub>0.2</sub> O <sub>3-δ</sub>	BaCe <sub>0.5</sub> Bi <sub>0.5</sub> O <sub>3-δ</sub>	700	321	[S11]
2009	Ni- BaCe <sub>0.7</sub> Zr <sub>0.1</sub> Y <sub>0.1</sub> Yb <sub>0.1</sub> O <sub>3-δ</sub>	BaCe <sub>0.7</sub> Zr <sub>0.1</sub> Y <sub>0.1</sub> Yb <sub>0.1</sub> O <sub>3-δ</sub>	La <sub>0.8</sub> Sr <sub>0.2</sub> Co <sub>0.2</sub> Fe <sub>0.8</sub> O <sub>3-δ</sub> - BaCe <sub>0.7</sub> Zr <sub>0.1</sub> Y <sub>0.2</sub> O <sub>3-δ</sub>	700	1100	[S12]
2010	Ni- BaCe <sub>0.7</sub> Zr <sub>0.1</sub> Y <sub>0.2</sub> O <sub>3-δ</sub>	BaCe <sub>0.7</sub> Zr <sub>0.1</sub> Y <sub>0.2</sub> O <sub>3-δ</sub>	PrBaCo <sub>2</sub> O <sub>5+δ</sub>	700	520	[S13]
2010	Ni- BaCe <sub>0.7</sub> Zr <sub>0.1</sub> Y <sub>0.2</sub> O <sub>3-δ</sub>	BaCe <sub>0.7</sub> Zr <sub>0.1</sub> Y <sub>0.2</sub> O <sub>3-δ</sub>	Ba <sub>0.5</sub> Sr <sub>0.5</sub> Co <sub>0.8</sub> Fe <sub>0.2</sub> O <sub>3-δ</sub>	700	420	[S14]
2011	Ni- BaCe <sub>0.7</sub> Zr <sub>0.1</sub> Y <sub>0.2</sub> O <sub>3-δ</sub>	BaCe <sub>0.7</sub> Zr <sub>0.1</sub> Y <sub>0.2</sub> O <sub>3-δ</sub>	BaCe <sub>0.7</sub> Zr <sub>0.1</sub> Co <sub>0.2</sub> O <sub>3-δ</sub>	700	370	[S15]
2012	Ni- BaCe <sub>0.7</sub> Zr <sub>0.1</sub> Y <sub>0.2</sub> O <sub>3-δ</sub>	BaCe <sub>0.7</sub> Zr <sub>0.1</sub> Y <sub>0.2</sub> O <sub>3-δ</sub>	BaCo <sub>0.7</sub> Fe <sub>0.2</sub> Nb <sub>0.1</sub> O <sub>3-δ</sub>	700	585	[S16]

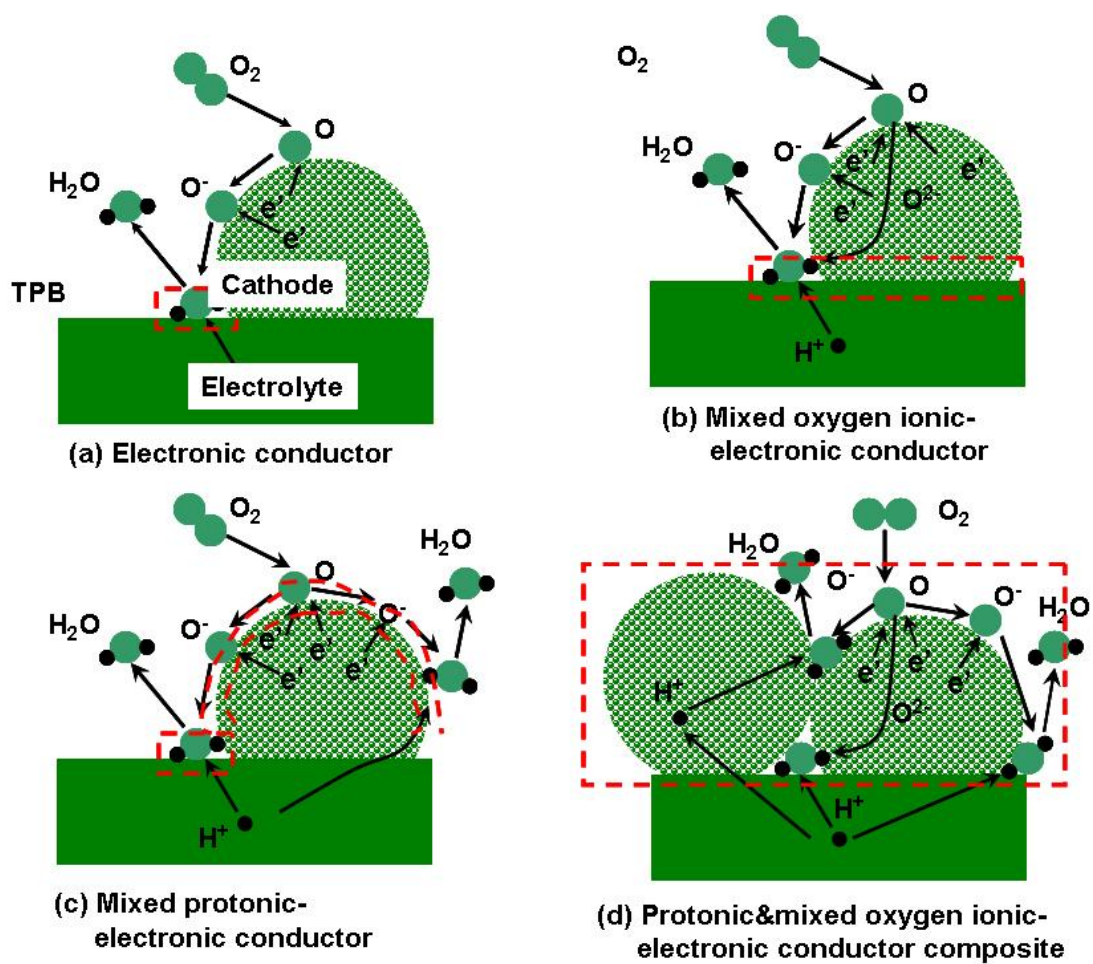
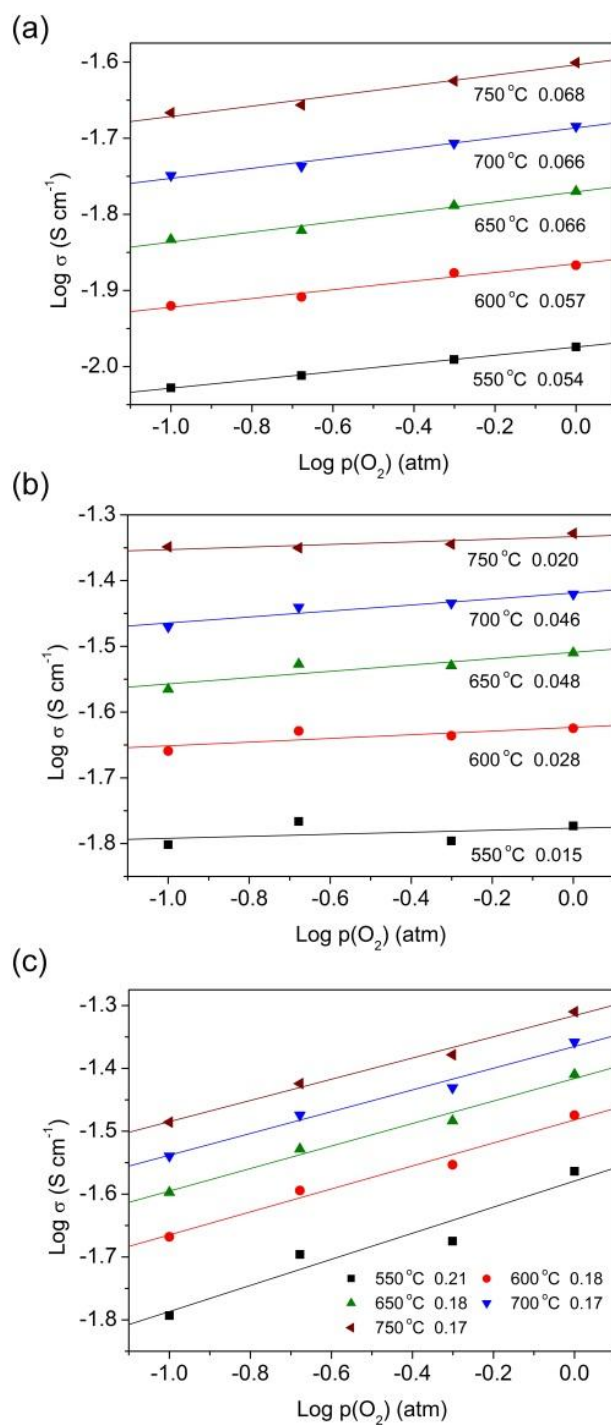
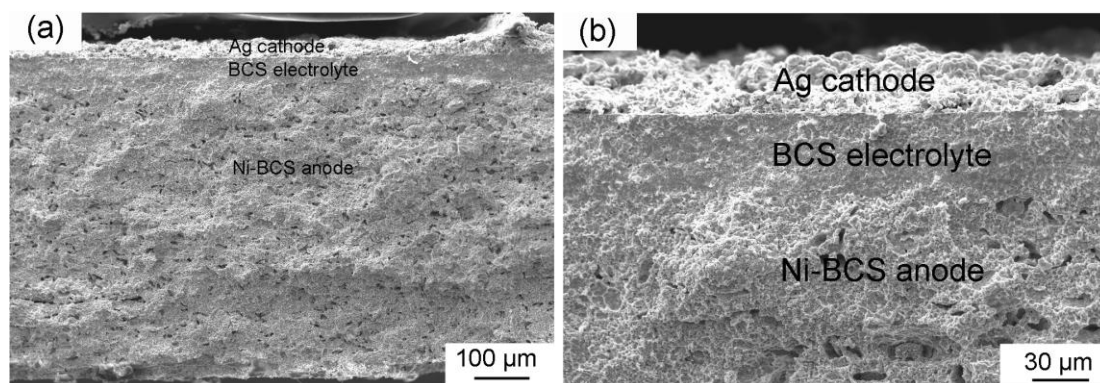


Fig. S1 Reaction mechanism of different cathodes for proton conducting SOFC.

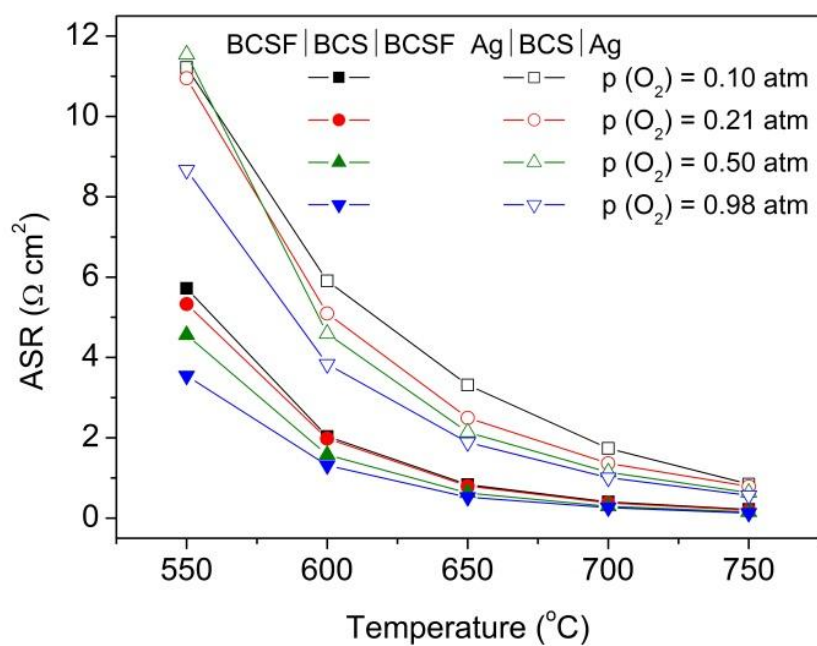


**Fig. S2** Dependence of electrical conductivity of symmetrical cell (a), BCS (b) and BCSF

(c) on  $p(\text{O}_2)$ . The  $n$  values in  $\sigma = k p(\text{O}_2)^n$  are inset.



**Fig. S3** Microstructure of the Ni-BCS|BCS|Ag cell after test.



**Fig. S4** Comparison of ASR of symmetrical cells with BCSF and Ag electrode, respectively.

## Supporting References

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