

Conformational analysis, NMR properties and nitrogen inversion of N-substituted 1,3-oxazines.

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Supporting Information (A total of 4 pages)

Table S1. B3LYP/6-31G** optimized geometries for N-substituted oxazines. Bond lengths are in Å and angles in degrees.

Parameters	Methyl		Ethyl		Propyl		i-Propyl		t-Butyl	
	Axial	Equatorial	Axial	Equatorial	Axial	Equatorial	Axial	Equatorial	Axial	Equatorial
O1-C2	1.433	1.410	1.433	1.411	1.434	1.411	1.438	1.416	1.436	1.417
C2-N3	1.439	1.453	1.441	1.456	1.440	1.456	1.431	1.449	1.435	1.454
N3-C4	1.460	1.456	1.460	1.458	1.459	1.458	1.454	1.455	1.498	1.493
C4-C5	1.550	1.533	1.550	1.533	1.549	1.533	1.548	1.536	1.554	1.536
C5-C6	1.538	1.533	1.536	1.533	1.536	1.532	1.539	1.534	1.536	1.532
N3-R	1.467	1.455	1.478	1.465	1.476	1.465	1.484	1.474	1.498	1.493
C2-H2a	1.102	1.115	1.102	1.115	1.102	1.115	1.103	1.112	1.104	1.112
C2-H2e	1.092	1.093	1.090	1.091	1.090	1.091	1.090	1.093	1.088	1.088
C4-H4a	1.098	1.110	1.099	1.110	1.099	1.109	1.101	1.108	1.101	1.108
C4-H4e	1.091	1.092	1.091	1.092	1.091	1.092	1.090	1.090	1.087	1.087
C6-H6a	1.104	1.103	1.104	1.103	1.104	1.103	1.104	1.104	1.104	1.104
C6-H6e	1.090	1.090	1.090	1.090	1.090	1.090	1.090	1.090	1.090	1.090
<O1C2N3	113.4	111.0	113.4	111.1	113.4	111.1	113.5	110.7	113.9	110.8
<C5C4N3	114.8	110.3	114.9	110.5	114.8	110.4	114.7	110.4	115.6	109.8
<H2aC2H2e	109.5	108.7	109.2	108.7	109.2	108.7	109.1	108.8	108.8	108.8
<H4aC4H4e	107.8	108.0	107.7	107.9	107.7	108.0	107.6	108.0	107.1	108.0
<H6aC6H6e	108.7	109.0	108.7	109.6	108.7	109.0	108.7	109.0	108.6	109.0
<C4C5C6	109.5	109.5	109.4	109.5	109.3	109.5	109.4	109.6	109.6	109.8
<H2aC2N3	108.7	111.6	108.7	111.3	108.7	111.2	109.1	112.7	108.9	112.1
<H4aC4N3	108.0	112.2	108.0	112.4	108.1	112.5	108.5	113.1	107.9	113.1
<C2O1C6C5	-56.6	-56.1	-56.8	-55.8	-56.6	-55.6	-56.4	-55.7	-57.0	-55.2
<C2N3C4C5	49.0	56.3	49.2	56.7	49.5	57.0	48.3	57.8	47.2	58.4
<O1C2N3C7	72.9	171.0	73.9	170.8	74.5	170.5	86.6	160.5	90.3	159.6

Table S2. Total energies (E , hartrees), zero point energy corrections (ZPE, hartrees), enthalpy thermal corrections (H_{corr} , hartrees), Gibb's free energy thermal corrections (G_{corr} , hartrees); entropy (S , cal mol⁻¹ K⁻¹), dipole moments (μ), relative enthalpies (ΔH° , kcal mol⁻¹) and relative Gibb's free energies (ΔG° , kcal mol⁻¹)

Energy ^a	Methyl		Ethyl		Propyl		i-Propyl		t-Butyl	
	Axial	Equatorial	Axial	Equatorial	Axial	Equatorial	Equatorial	Axial	Equatorial	Axial
HF/6-31G(d,p) geometry										
$E_0^{(2)}$ [MP2/6-31G(d,p)]	-569.292365	-569.290142	-608.481374	-608.473577	-647.658985	-647.655581	-647.659508	-647.658793	-686.843460	-686.842988
$E_0^{(3)}$ [MP2/6-311G(d,p)]	-569.525091	-569.523274	-608.725012	-608.719911	-647.917960	-647.914975	-647.918619	-647.917992	-687.116108	-687.115811
$E_0^{(4)}$ [MP2/6-311+G(d,p)]	-569.550367	-569.549218	-608.751020	-608.746301	-647.943959	-647.941837	-647.944753	-647.944921	-687.143164	-687.143571
$E_0^{(5)}$ [MP2/6-311++G(d,p)]	-569.551437	-569.550254	-608.752278	-608.747509	-647.945353	-647.943154	-647.946213	-647.946284	-687.144765	-687.145074
ZPE	0.209094	0.208582	0.237144	0.238833	0.269539	0.269092	0.268836	0.268572	0.298837	0.298501
H_{corr}	0.219900	0.219500	0.247930	0.251118	0.283012	0.282728	0.282263	0.282119	0.313263	0.313155
G_{corr}	0.173775	0.173090	0.202277	0.201030	0.230081	0.229034	0.230308	0.229506	0.259667	0.258010
S^b	97.077	97.677	96.084	105.419	111.403	113.009	109.349	110.734	112.801	116.064
μ (D)	4.160	5.578	4.129	5.483	4.130	5.411	4.149	5.368	4.119	5.272
$E_0^{(4)} + \text{ZPE}$	-569.341277	-569.340636	-608.513876	-608.507468	-647.674420	-647.672745	-647.675917	-647.676349	-686.844327	-686.845070
$E_0^{(4)} + H_{\text{corr}}$	-569.330467	-569.329718	-608.503090	-608.495183	-647.660947	-647.659109	-647.662490	-647.662802	-686.829901	-686.830416
$E_0^{(4)} + G_{\text{corr}}$	-569.376592	-569.376128	-608.548743	-608.545271	-647.713878	-647.712803	-647.714445	-647.715415	-686.883497	-686.885561
$\Delta H^{\circ(4)c}$	-0.470		-4.962		-1.100		0.196		0.324	
$\Delta G^{\circ(4)c}$	-0.291		-2.179		-0.675		0.609		1.296	
B3LYP/6-31G(d,p) geometry										
$E_0^{(6)}$ [B3LYP/6-31G(d,p)]	-570.938542	-570.937188	-610.256524	-610.254212	-649.573022	-649.570567	-649.571062	-649.570629	-688.884268	-688.883853
$E_0^{(7)}$ [B3LYP/6-311G(d,p)]	-571.078165	-571.077212	-610.404188	-610.402319	-649.728232	-649.726405	-649.726520	-649.726317	-689.047357	-689.047142
$E_0^{(8)}$ [B3LYP/6-311+G(d,p)]	-571.091344	-571.091240	-610.417108	-610.416124	-649.741383	-649.740243	-649.73872	-649.739722	-689.059258	-689.060595
$E_0^{(9)}$ [B3LYP/6-311++G(d,p)]	-571.091621	-571.091485	-610.417389	-610.416403	-649.741686	-649.740534	-649.739055	-649.740021	-689.059608	-689.060921
ZPE	0.193964	0.193379	0.222258	0.221826	0.250853	0.250202	0.250116	0.249721	0.278248	0.277814
H_{corr}	0.205482	0.204998	0.235186	0.234867	0.265148	0.264673	0.264435	0.264172	0.293685	0.293471
G_{corr}	0.157929	0.157135	0.184102	0.183239	0.210566	0.209319	0.210523	0.209664	0.238047	0.23656
S^b	100.085	101.12	107.515	108.661	114.876	116.503	113.469	114.72	117.102	119.778
μ (D)	3.528	4.736	3.496	4.654	3.478	4.598	3.498	4.616	3.472	4.527
$E_0^{(8)} + \text{ZPE}$	-570.897380	-570.897861	-610.194850	-610.1942984	-649.490530	-649.490041	-649.488605	-649.490001	-688.781010	-688.782781
$E_0^{(8)} + H_{\text{corr}}$	-570.885862	-570.886242	-610.181922	-610.1812574	-649.476235	-649.475570	-649.474286	-649.475550	-688.765573	-688.767124
$E_0^{(8)} + G_{\text{corr}}$	-570.933415	-570.934105	-610.233006	-610.2328854	-649.530817	-649.530924	-649.528198	-649.530058	-688.821211	-688.824035
$\Delta H^{\circ(8)c}$	0.238454		-0.417168		-0.416854		0.792858		0.973518	
$\Delta G^{\circ(8)c}$	0.432982		-0.075803		0.067583		1.166854		1.772338	

Table S3. Relative free energies (kcal mol⁻¹) of the axial isomer of the methyl and ethyl derivatives obtained at two different levels of theory. The values within parenthesis correspond to the percentage of the axial isomer.

Solvent	Theoretical model	Methyl	Ethyl
Gas	MP2/6-311+G**	-0.29 (62.0)	-0.44 (67.8)
	MP2/6-311+G(3df,2p)	-0.42 (67.2)	-0.58 (72.6)
CH ₂ Cl ₂	MP2/6-311+G**	0.68 (24.1)	0.68 (24.1)
	MP2/6-311+G(3df,2p)	0.40 (33.9)	0.35 (35.5)