

*Electronic Supplementary Information*

## **Homochiral column structure of *rac*- and $\Lambda$ -[M<sup>III</sup>(tn)<sub>3</sub>]P<sub>3</sub>O<sub>9</sub> (M = Co, Cr; tn = 1,3-diaminopropane; P<sub>3</sub>O<sub>9</sub> = cyclotriphosphate(3-)) produced by multiple hydrogen bonds<sup>†</sup>**

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**Table S1** Hydrogen bond distances and angles for *rac*-[Co(tn)<sub>3</sub>]P<sub>3</sub>O<sub>9</sub>·7.3H<sub>2</sub>O

<b>Co1 (N1–N6)</b>				
D–H···A	D···A/Å	D–H/Å	H···A/Å	∠D–H···A/°
N(1)–H(1)···O(5)	2.866(5)	0.920	2.029	151
N(1)–H(2)···O(4)*	2.874(4)	0.920	1.955	178
N(2)–H(9)···O(38)	2.855(4)	0.920	1.942	172
N(2)–H(10)···O(41) <sup>†</sup>	3.251(5)	0.920	2.524	136
N(3)–H(11)···O(5)	3.023(5)	0.920	2.268	139
N(3)–H(12)···O(50) <sup>†</sup>	3.102(5)	0.920	2.290	147
N(4)–H(19)···O(15)	2.904(4)	0.920	2.025	160
N(4)–H(20)···O(36) <sup>‡</sup>	2.950(5)	0.920	2.038	171
N(5)–H(21)···O(5)	2.798(4)	0.920	2.014	142
N(5)–H(22)···O(36) <sup>‡</sup>	2.893(5)	0.920	1.977	174
N(6)–H(29)···O(15)	3.216(4)	0.920	2.423	144
N(6)–H(29)···O(38)	2.987(5)	0.920	2.356	126
N(6)–H(30)···O(4)*	2.959(4)	0.920	2.049	170
<b>Co2 (N7–N12)</b>				
D–H···A	D···A/Å	D–H/Å	H···A/Å	∠D–H···A/°
N(7)–H(31)···O(14)	2.868(4)	0.920	2.022	152
N(7)–H(32)···O(22) <sup>§</sup>	2.964(4)	0.920	2.059	168
N(8)–H(39)···O(48)	2.901(4)	0.920	1.989	171
N(8)–H(40)···O(47)	2.923(5)	0.920	2.100	148
N(9)–H(41)···O(14)	3.157(4)	0.920	2.308	153
N(9)–H(42)···O(43)	2.962(5)	0.920	2.228	136
N(10)–H(49)···O(24)	2.918(5)	0.920	2.023	164
N(10)–H(50)···O(32) <sup>□</sup>	3.031(5)	0.920	2.197	150
N(11)–H(51)···O(14)	2.846(4)	0.920	1.942	167
N(11)–H(52)···O(46) <sup>¶</sup>	2.884(5)	0.920	2.012	158
N(12)–H(59)···O(24)	3.049(4)	0.920	2.158	163
N(12)–H(59)···O(48)	3.038(5)	0.920	2.574	112
N(12)–H(60)···O(22) <sup>§</sup>	2.994(4)	0.920	2.108	162

**Table S1** (continued) Hydrogen bond distances **and angles** for *rac*-[Co(tn)<sub>3</sub>]P<sub>3</sub>O<sub>9</sub>·7.3H<sub>2</sub>O  
**Co1 (N13–N18)**

D–H···A	D···A/Å	D–H/Å	H···A/Å	∠D–H···A/°
N(13)–H(61)···O(23)	2.817(4)	0.920	1.921	164
N(13)–H(62)···O(13) <sup>§</sup>	2.911(4)	0.920	2.003	169
N(14)–H(69)···O(35)	2.916(4)	0.920	2.011	168
N(14)–H(70)···O(34)	3.037(5)	0.920	2.243	144
N(15)–H(71)···O(27)	2.846(5)	0.920	1.942	167
N(15)–H(72)···O(25)	3.034(4)	0.920	2.335	133
N(16)–H(79)···O(6)**	2.854(4)	0.920	2.036	147
N(16)–H(80)···O(33) <sup>□</sup>	2.964(5)	0.920	2.097	157
N(17)–H(81)···O(23)	2.850(4)	0.920	1.944	168
N(17)–H(82)···O(27)	3.039(4)	0.920	2.358	131
N(17)–H(82)···O(33) <sup>□</sup>	3.051(5)	0.920	2.261	144
N(18)–H(89)···O(6)**	3.014(4)	0.920	2.151	156
N(18)–H(89)···O(35)	3.024(5)	0.920	2.545	113
N(18)–H(90)···O(13) <sup>§</sup>	3.059(4)	0.920	2.192	157

Symmetry operators; \*:  $-x -y -z$ ; †:  $x-1 \ 1/2-y \ -1/2+z$ ; ‡:  $x-1 \ y \ z-1$ ; §:  $2-x -y \ 1-z$ ; □:  $x+1 \ y \ z$ ; ¶:  $x \ 1/2-y \ -1/2+z$ ; \*\*:  $x+2 \ y \ z+1$ .

**Table S2** Hydrogen bond distances and angles for *rac*-[Cr(tn)<sub>3</sub>]P<sub>3</sub>O<sub>9</sub>·7.5H<sub>2</sub>O

<b>Cr1 (N1–N6)</b>				
D–H···A	D···A/Å	D–H/Å	H···A/Å	∠D–H···A/°
N(1)–H(1)···O(20)	2.896(4)	0.920	2.030	156
N(1)–H(2)···O(9)	2.892(3)	0.920	1.985	168
N(2)–H(9)···O(8)*	2.903(3)	0.920	1.995	169
N(2)–H(10)···O(6) <sup>†</sup>	3.067(3)	0.920	2.200	157
N(3)–H(11)···O(8)*	3.093(3)	0.920	2.249	152
N(3)–H(11)···O(9)*	3.257(4)	0.920	2.528	136
N(3)–H(12)···O(9)	2.824(3)	0.920	1.915	170
N(4)–H(19)···O(26) <sup>‡</sup>	2.943(3)	0.920	2.187	139
N(4)–H(20)···O(29) <sup>‡</sup>	2.938(4)	0.920	2.026	171
N(5)–H(21)···O(7)	2.928(3)	0.920	2.119	146
N(5)–H(22)···O(20)	3.027(2)	0.920	2.151	159
N(6)–H(29)···O(19)	3.013(3)	0.920	2.156	155
N(6)–H(30)···O(6) <sup>†</sup>	2.899(3)	0.920	2.062	151
<b>Cr2 (N7–N12)</b>				
D–H···A	D···A/Å	D–H/Å	H···A/Å	∠D–H···A/°
N(7)–H(31)···O(32) <sup>§</sup>	2.960(3)	0.920	2.059	166
N(7)–H(32)···O(22)	3.235(5)	0.920	2.431	146
N(8)–H(39)···O(15)	2.800(3)	0.920	1.907	163
N(8)–H(40)···O(16) <sup>□</sup>	2.963(3)	0.920	2.045	175
N(9)–H(41)···O(14) <sup>¶</sup>	3.071(3)	0.920	2.298	141
N(9)–H(41)···O(32) <sup>§</sup>	3.007(2)	0.920	2.350	128
N(9)–H(42)···O(16) <sup>□</sup>	2.946(3)	0.920	2.035	170
N(10)–H(49)···O(15)	2.829(3)	0.920	2.136	131
N(10)–H(50)···O(27)	2.919(3)	0.920	2.004	172
N(11)–H(51)···O(14) <sup>¶</sup>	2.882(3)	0.920	1.996	161
N(11)–H(52)···O(27)	2.930(2)	0.920	2.013	174
N(12)–H(59)···O(17)	2.926(2)	0.920	2.028	165
N(12)–H(60)···O(13)	2.998(2)	0.920	2.243	139

Symmetry operators; \*: 1–x –y 1–z; †: x–1 y z; ‡: x+1 y+1 z+1; §: 2–x 1–y –z; □: 1–x 1–y –z; ¶: x+1 y z.

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**Table S3** Relevant coordination bond lengths (Å) and angles (°) with their estimated standard deviations in parentheses for  $\Lambda$ -[Co(tn)<sub>3</sub>]P<sub>3</sub>O<sub>9</sub>·2.5H<sub>2</sub>O

Bond length / Å			
Co(1)–N(1)	1.977(2)	Co(1)–N(4)	1.998(2)
Co(1)–N(2)	1.9850(19)	Co(1)–N(5)	1.9882(18)
Co(1)–N(3)	2.0090(18)	Co(1)–N(6)	1.9760(18)
Co(2)–N(7)	1.984(2)	Co(2)–N(10)	1.9784(19)
Co(2)–N(8)	1.979(2)	Co(2)–N(11)	1.993(2)
Co(2)–N(9)	1.9842(19)	Co(2)–N(12)	1.9891(19)
	Co–N <sub>av</sub>		1.987 Å
Bond angle / °			
N(1)–Co(1)–N(2)	93.56(8)	N(7)–Co(2)–N(8)	94.06(9)
N(3)–Co(1)–N(4)	88.76(8)	N(9)–Co(2)–N(10)	93.93(8)
N(5)–Co(1)–N(6)	94.43(7)	N(11)–Co(2)–N(12)	89.83(8)

**Table S4** Hydrogen bond distances and angles for  $\Lambda$ -[Co(tn)<sub>3</sub>]P<sub>3</sub>O<sub>9</sub>·2.5H<sub>2</sub>O

<b>Co1 (N1–N6)</b>				
D–H···A	D···A/Å	D–H/Å	H···A/Å	∠D–H···A/°
N(1)–H(1)···O(9)	2.776(2)	0.950	1.851	164
N(1)–H(2)···O(20)	2.924(2)	0.950	2.121	141
N(2)–H(9)···O(4)*	2.821(2)	0.950	1.878	172
N(2)–H(10)···O(15)	2.920(2)	0.950	1.970	178
N(3)–H(11)···O(9)	2.981(2)	0.950	2.176	142
N(3)–H(12)···O(15)	2.910(2)	0.950	1.968	171
N(4)–H(19)···O(4)*	2.870(2)	0.950	2.118	135
N(4)–H(20)···O(17) <sup>†</sup>	2.886(2)	0.950	1.938	175
N(5)–H(21)···O(7)	2.942(2)	0.950	2.043	157
N(5)–H(22)···O(17) <sup>†</sup>	2.847(2)	0.950	1.915	166
N(6)–H(29)···O(6)*	2.986(2)	0.950	2.145	147
N(6)–H(30)···O(23)	3.003(3)	0.950	2.171	146
<b>Co2 (N7–N12)</b>				
D–H···A	D···A/Å	D–H/Å	H···A/Å	∠D–H···A/°
N(7)–H(31)···O(16) <sup>‡</sup>	2.793(2)	0.950	2.021	137
N(7)–H(32)···O(5)	3.193(2)	0.950	2.407	140
N(8)–H(39)···O(13)	2.937(2)	0.950	2.035	158
N(8)–H(40)···O(21) <sup>§</sup>	2.962(2)	0.950	2.129	146
N(9)–H(41)···O(14) <sup>‡</sup>	3.087(2)	0.950	2.321	137
N(9)–H(42)···O(18) <sup>‡</sup>	2.819(3)	0.950	1.947	152
N(10)–H(49)···O(13)	2.858(2)	0.950	1.939	162
N(10)–H(50)···O(8)	2.806(2)	0.950	1.873	167
N(11)–H(51)···O(16) <sup>‡</sup>	2.948(2)	0.950	2.131	143
N(11)–H(52)···O(8) <sup>□</sup>	3.002(2)	0.950	2.080	163
N(12)–H(59)···O(10)	3.261(2)	0.950	2.447	144
N(12)–H(59)···O(13)	3.031(2)	0.950	2.253	139
N(12)–H(60)···O(5)	2.898(2)	0.950	1.973	164

**Table S4** Hydrogen bond distances and angles for  $\Lambda$ -[Co(tn)<sub>3</sub>]P<sub>3</sub>O<sub>9</sub>·2.5H<sub>2</sub>O (continued)

<b>O<sub>H2O</sub></b>				
D–H···A	D···A/Å	D–H/Å	H···A/Å	∠D–H···A/°
O(19)–H(61)···O(5)	2.904(2)	0.857	2.057	170
O(19)–H(62)···O(15)	2.904(2)	0.931	2.035	155
O(20)–H(63)···O(6) <sup>¶</sup>	2.788(2)	0.831	1.960	175
O(20)–H(64)···O(21)	2.898(2)	0.942	1.967	169
O(21)–H(65)···O(14)**	2.761(2)	0.829	1.936	173
O(21)–H(66)···O(22)	2.808(2)	0.869	1.949	170
O(22)–H(67)···O(7) <sup>¶</sup>	2.761(2)	0.920	1.852	169
O(22)–H(68)···O(18) <sup>††</sup>	2.740(2)	0.876	1.868	172
O(23)–H(69)···O(22)	2.835(2)	0.766	2.080	169
O(23)–H(70)···O(8)*	2.805(2)	0.885	1.945	164

Symmetry operators; \*: x–1 y z; †: 1–x 1/2+y 1/2–z; ‡: x+1 y z; §: 3/2–x –y 1/2–z; ¶: 2–x 1/2+y 1/2–z; ¶¶: x–1/2 1/2–y 1–z; \*\*: 1–x –1/2+y 1/2–z; ††: 1/2–x –y 1/2+z.



**Table S5** Relevant coordination bond lengths (Å) and angles (°) with their estimated standard deviations in parentheses for  $\Lambda$ -[Cr(tn)<sub>3</sub>]P<sub>3</sub>O<sub>9</sub>·2.5H<sub>2</sub>O

Bond length / Å			
Cr(1)–N(1)	2.0752(16)	Cr(1)–N(4)	2.0990(17)
Cr(1)–N(2)	2.0793(17)	Cr(1)–N(5)	2.0832(16)
Cr(1)–N(3)	2.1066(18)	Cr(1)–N(6)	2.0839(17)
Cr(2)–N(7)	2.089(2)	Cr(2)–N(10)	2.0712(16)
Cr(2)–N(8)	2.0846(17)	Cr(2)–N(11)	2.0965(18)
Cr(2)–N(9)	2.0994(19)	Cr(2)–N(12)	2.0905(18)
	Cr–N <sub>av</sub>	2.088 Å	
Bond angle / °			
N(1)–Cr(1)–N(2)	92.06(6)	N(7)–Cr(2)–N(8)	89.84(7)
N(3)–Cr(1)–N(4)	86.86(7)	N(9)–Cr(2)–N(10)	90.92(7)
N(5)–Cr(1)–N(6)	92.78(6)	N(11)–Cr(2)–N(12)	86.97(7)

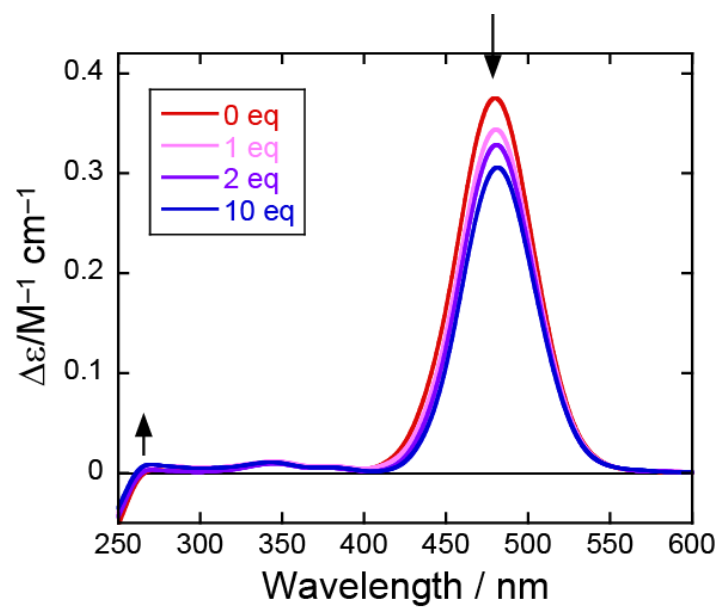
**Table S6** Hydrogen bond distances and angles for  $\Lambda$ -[Cr(tn)<sub>3</sub>]P<sub>3</sub>O<sub>9</sub>·2.5H<sub>2</sub>O

<b>Cr1 (N1–N6)</b>				
D–H···A	D···A/Å	D–H/Å	H···A/Å	∠D–H···A/°
N(1)–H(1)···O(9)	2.775(2)	0.950	1.843	166
N(1)–H(2)···O(20)	2.925(2)	0.950	2.101	144
N(2)–H(9)···O(4)*	2.838(2)	0.950	1.901	168
N(2)–H(10)···O(15)	2.936(2)	0.950	1.986	177
N(3)–H(11)···O(9)	3.014(2)	0.950	2.202	143
N(3)–H(12)···O(15)	2.9115(19)	0.950	1.969	172
N(4)–H(19)···O(4)*	2.878(2)	0.950	2.124	135
N(4)–H(20)···O(17) <sup>†</sup>	2.891(2)	0.950	1.946	173
N(5)–H(21)···O(7)	2.923(2)	0.950	2.019	158
N(5)–H(22)···O(17) <sup>†</sup>	2.866(2)	0.950	1.933	167
N(6)–H(29)···O(6)*	2.933(2)	0.950	2.079	149
N(6)–H(30)···O(23)	2.972(2)	0.950	2.125	148
<b>Cr2 (N7–N12)</b>				
D–H···A	D···A/Å	D–H/Å	H···A/Å	∠D–H···A/°
N(7)–H(31)···O(18) <sup>‡</sup>	3.122(2)	0.950	2.186	168
N(7)–H(32)···O(16) <sup>‡</sup>	2.778(2)	0.950	2.228	116
N(8)–H(39)···O(13)	2.926(2)	0.950	2.058	151
N(8)–H(40)···O(21) <sup>§</sup>	2.937(2)	0.950	2.069	151
N(9)–H(41)···O(14) <sup>‡</sup>	3.007(2)	0.950	2.198	142
N(9)–H(42)···O(18) <sup>‡</sup>	2.843(2)	0.950	1.943	157
N(10)–H(49)···O(13)	2.848(2)	0.950	1.936	160
N(10)–H(50)···O(8) <sup>□</sup>	2.8057(19)	0.950	1.866	170
N(11)–H(51)···O(16) <sup>‡</sup>	2.912(2)	0.950	2.135	138
N(11)–H(52)···O(8) <sup>□</sup>	3.086(2)	0.950	2.176	160
N(12)–H(59)···O(10)	3.319(2)	0.950	2.477	148
N(12)–H(59)···O(13)	3.117(2)	0.950	2.381	134
N(12)–H(60)···O(5)	2.8822(18)	0.950	1.969	161

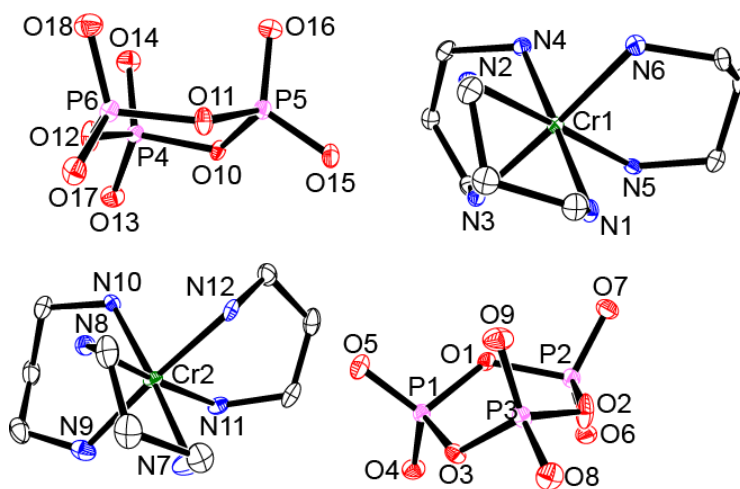
**Table S6** Hydrogen bond distances and angles for  $\Lambda$ -[Cr(tn)<sub>3</sub>]P<sub>3</sub>O<sub>9</sub>·2.5H<sub>2</sub>O (continued)

<b>O<sub>H2O</sub></b>				
D–H···A	D···A/Å	D–H/Å	H···A/Å	∠D–H···A/°
O(19)–H(61)···O(5)	2.925(2)	0.910	2.111	149
O(19)–H(62)···O(15)	2.884(2)	1.004	1.890	170
O(20)–H(63)···O(6) <sup>¶</sup>	2.796(2)	0.921	1.883	171
O(20)–H(64)···O(21)	2.8928(19)	0.857	2.051	167
O(21)–H(65)···O(14)**	2.7628(19)	0.980	1.788	173
O(21)–H(66)···O(22)	2.785(2)	0.918	1.895	163
O(22)–H(67)···O(7) <sup>¶</sup>	2.745(2)	0.873	1.877	173
O(22)–H(68)···O(18) <sup>††</sup>	2.771(2)	0.919	1.856	173
O(23)–H(69)···O(22)	2.865(2)	0.892	1.977	174
O(23)–H(70)···O(8)*	2.794(2)	0.982	1.833	165

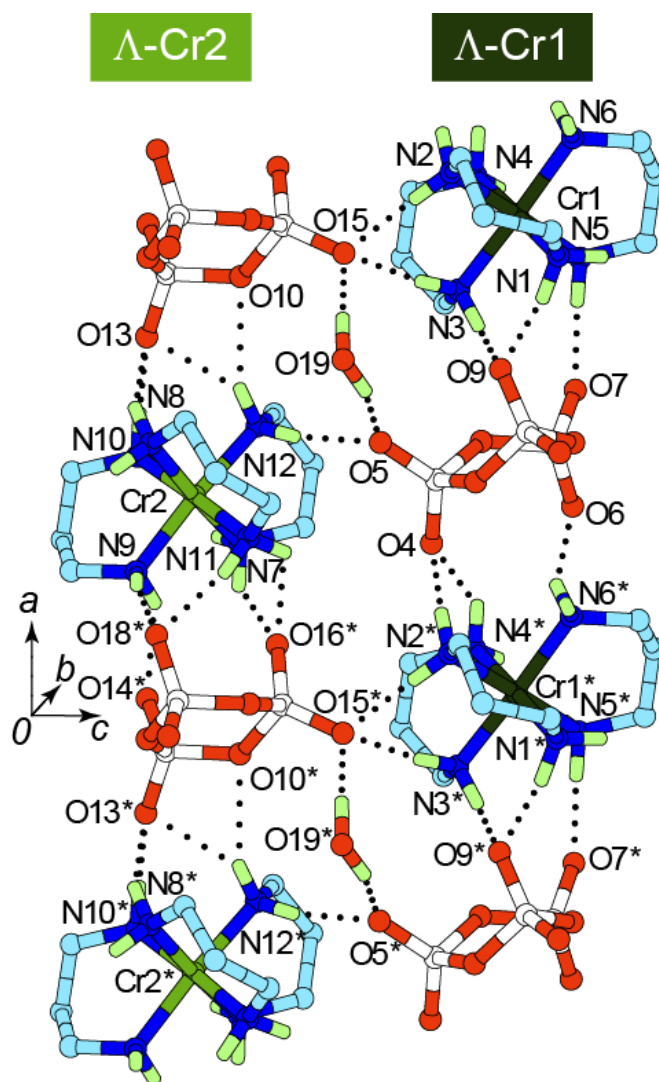
Symmetry operators; \*: x–1 y z; †: 1–x 1/2+y 1/2–z; ‡: x+1 y z; §: 3/2–x –y 1/2–z; □: 2–x 1/2+y 1/2–z; ¶: x–1/2 1/2–y 1–z; \*\*: 1–x –1/2+y 1/2–z; ††: 1/2–x –y 1/2+z.



**Fig. S1** CD spectral changes of  $\Lambda$ -[Cr(tn)<sub>3</sub>]Br<sub>3</sub> (2.5 mM) as a function of Na<sub>3</sub>P<sub>3</sub>O<sub>9</sub> concentration (0–25 mM).



**Fig. S2** ORTEP view of  $\Lambda$ -[Cr(tn)<sub>3</sub>]P<sub>3</sub>O<sub>9</sub>·2.5H<sub>2</sub>O with selected atom numbering schemes. Water molecules and hydrogen atoms are omitted for clarity.



**Fig. S3** Homochiral column-pair structure of  $\Lambda$ -[Cr(tn)<sub>3</sub>]P<sub>3</sub>O<sub>9</sub>·2.5H<sub>2</sub>O, which was constructed by multiple hydrogen bonds between the alternately stacked [Cr(tn)<sub>3</sub>]<sup>3+</sup> cations and P<sub>3</sub>O<sub>9</sub><sup>3-</sup> anions, and the neighboring columns are further linked by hydrogen bonds.

Symmetry operator \*:  $x-1, y, z$ .