

Electronic Supplementary Information (ESI)

Fluorescence enhancement of a tetrazole-based pyridine coordination polymer hydrogel

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Fig. S1 Photograph of hydrogel **2** (20 mM) with Mg^{2+} (4 equiv) at pH= 12.

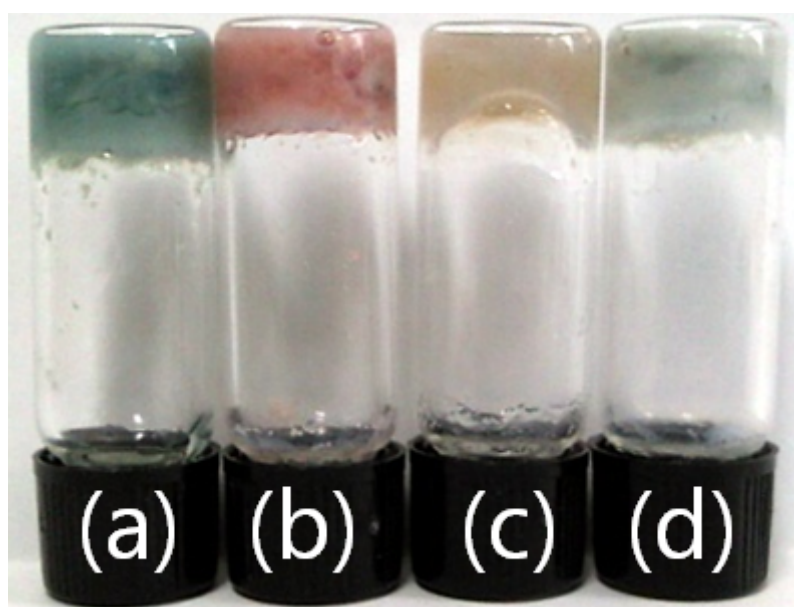


Fig. S2 Photograph of hydrogels **1** (20 mM) with (a) Cu^{2+} , (b) Co^{2+} , (c) Zn^{2+} and (d) Ni^{2+} ions at pH=12.

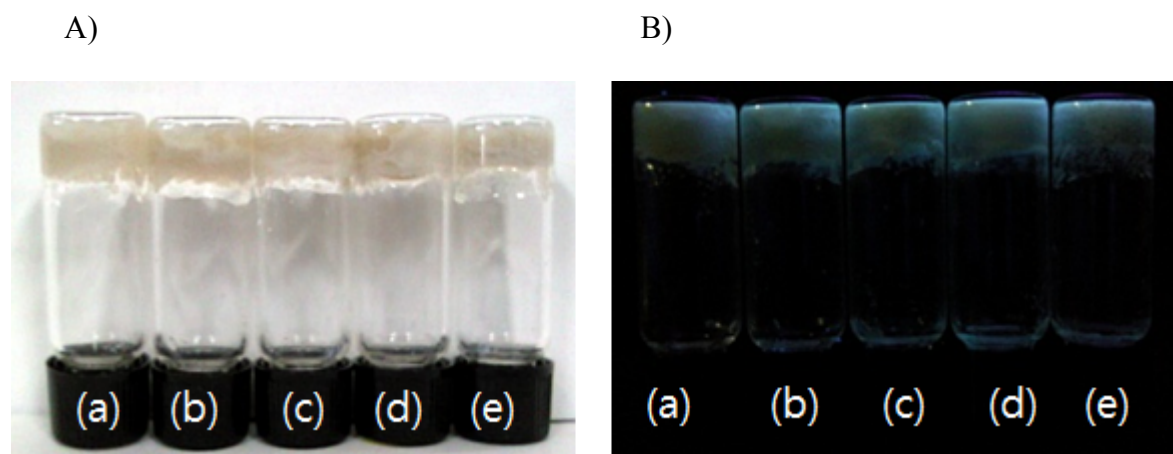


Fig. S3 Photographs of hydrogels **1** (20 mM) in the presence of magnesium anions (4 equiv) (a) NO_3^- , (b) Br^- , (c) SO_4^{2-} , (d) Cl^- and (e) I^- (A) without and (B) with irradiation of UV light.

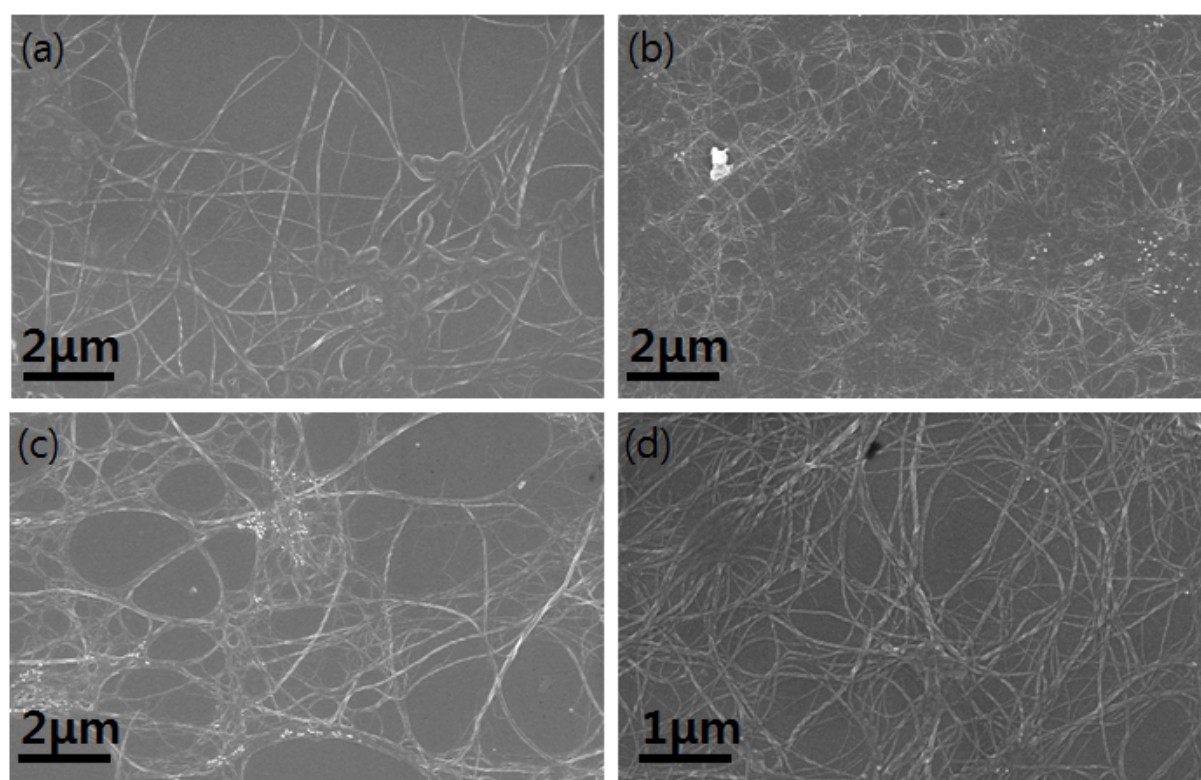


Fig. S4 SEM images of Mg^{2+} coordination polymeric gel **1** with different anions; (a) MgSO_4 , (b) MgBr_2 , (c) MgI_2 and (d) $\text{Mg}(\text{NO}_3)_2$.

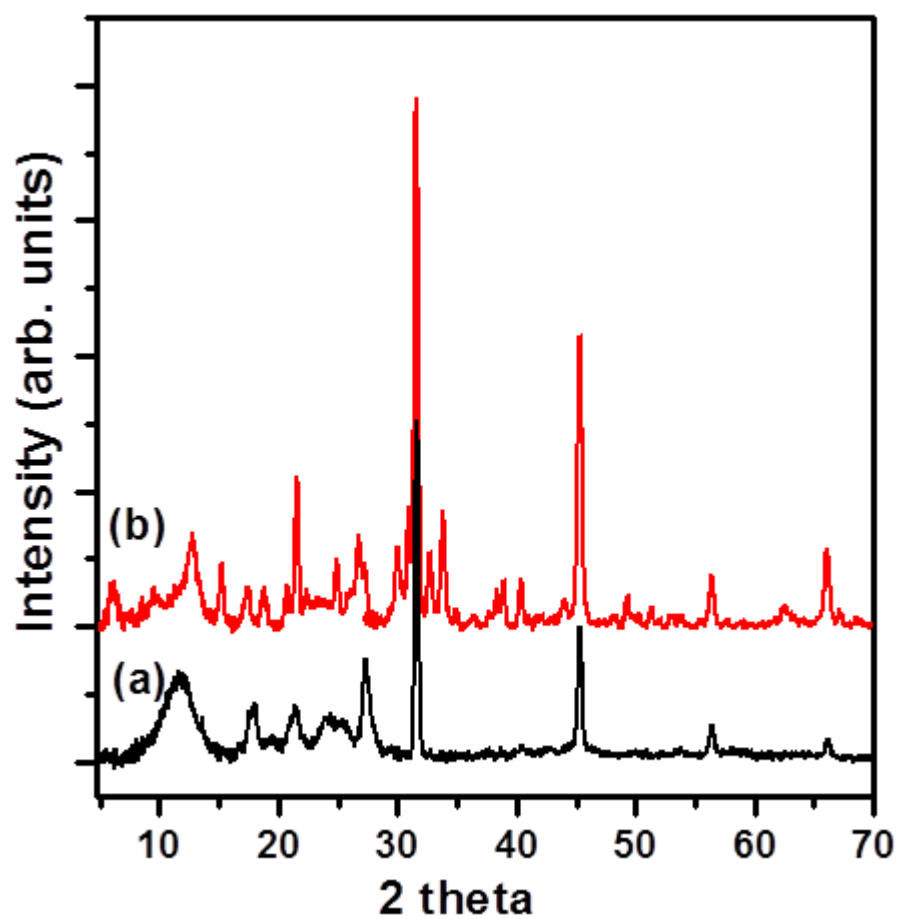


Fig. S5 X-ray powder diffraction pattern of (a) ligand **1** and (b) Mg²⁺ coordination polymer gel **1**.

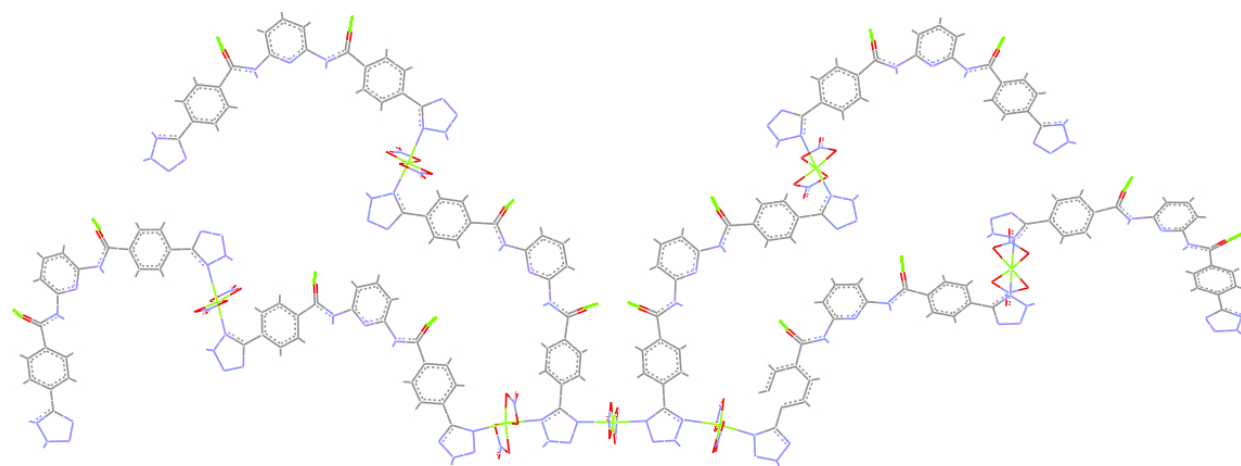


Fig. S6 The 2D expanded structure of Mg^{2+} coordination polymer gel **1** from B3LYP/3-21G* optimized structure.

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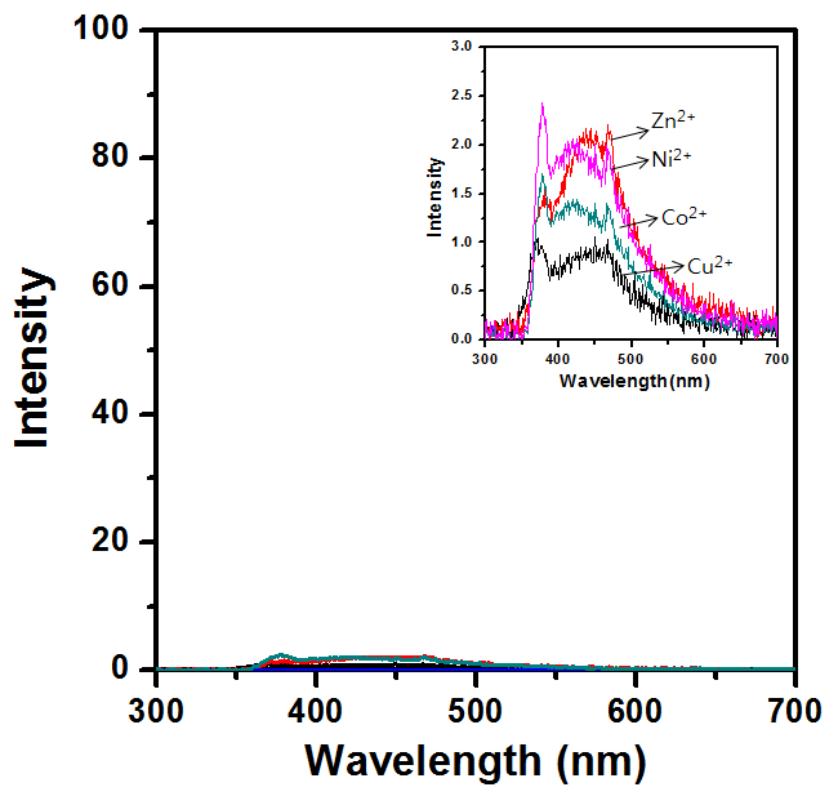


Fig. S7 Fluorescence spectra of **1** (20 mM) in the presence of Co²⁺, Zn²⁺, Ni²⁺ and Cu²⁺ (4 equiv) at pH= 12.

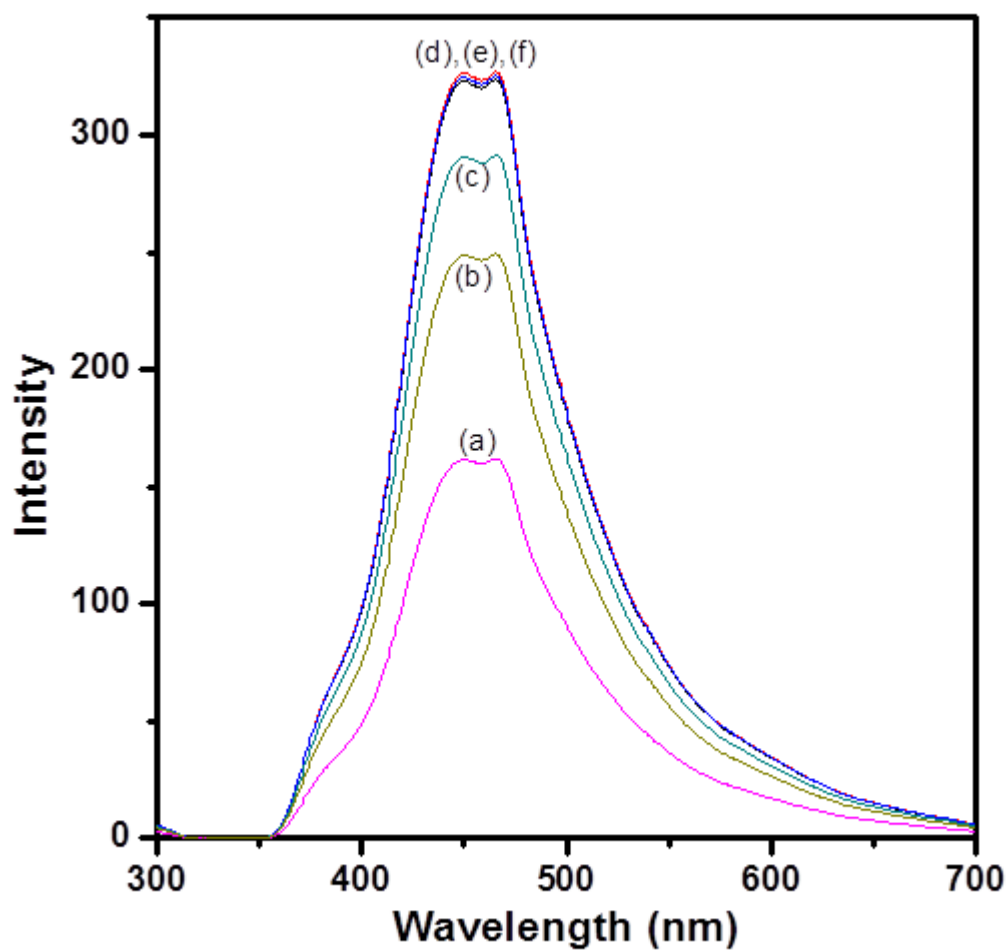


Fig. S8 Fluorescence spectra of hydrogel **1** (20 mM) by the addition of Mg^{2+} ion; (a) 1.0 equiv, (b) 2.0 equiv, (c) 3.0 equiv, (d) 4.0 equiv, (e) 5.0 equiv and (f) 6 equiv.

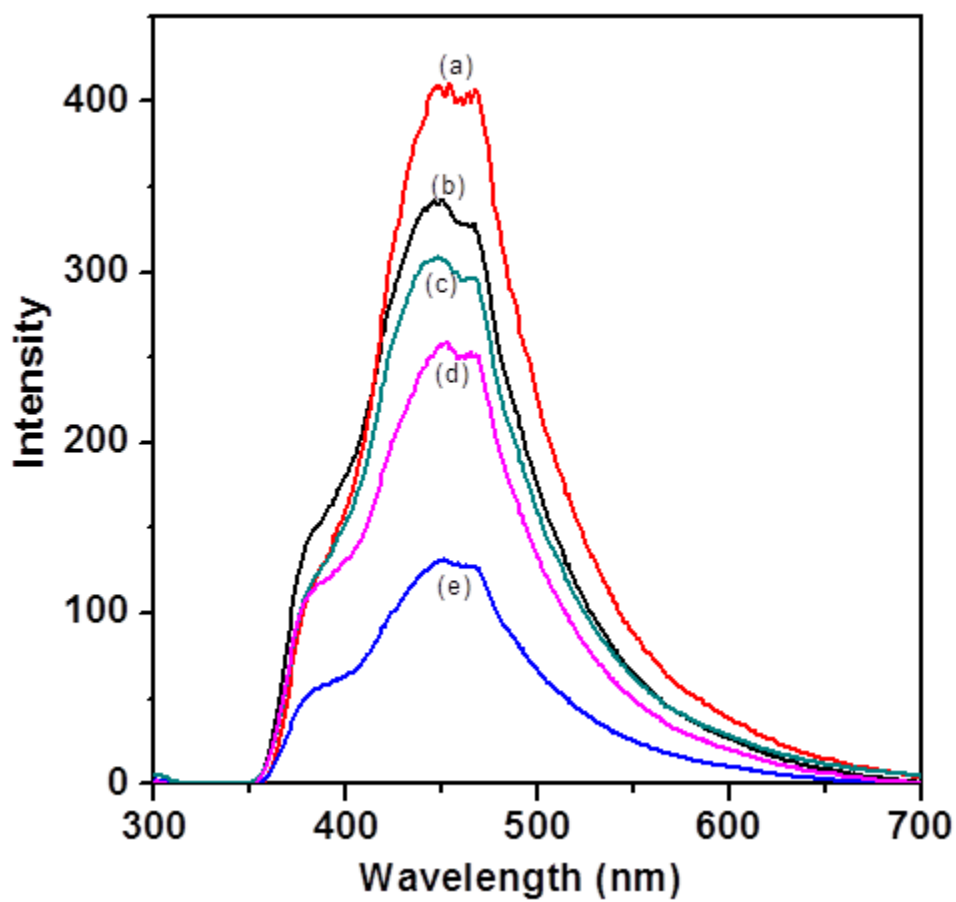


Fig. S9 Fluorescence spectra of Mg²⁺ (4 equiv) coordination polymer gel **1** (20 mM) with different anions; (a) NO₃⁻, (b) Cl⁻, (c) SO₄²⁻, (d) Br⁻, (e) I⁻.

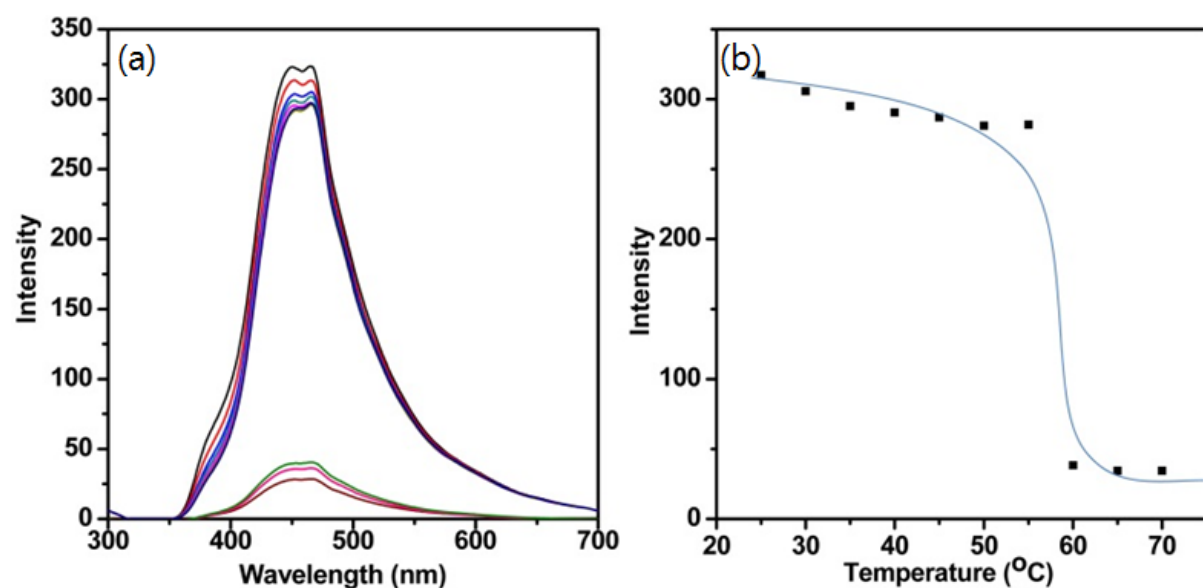


Fig. S10 (a) Fluorescence spectra of Mg^{2+} (4 equiv) coordination polymeric gel **1** (20 mM) at different temperatures (20 °C, 25 °C, 30 °C, 35 °C, 40 °C, 45 °C, 50 °C, 55 °C, 60 °C, 65 °C, 70 °C). (b) Plot of fluorescence intensity of Mg^{2+} coordination polymeric gel **1** function of temperature.

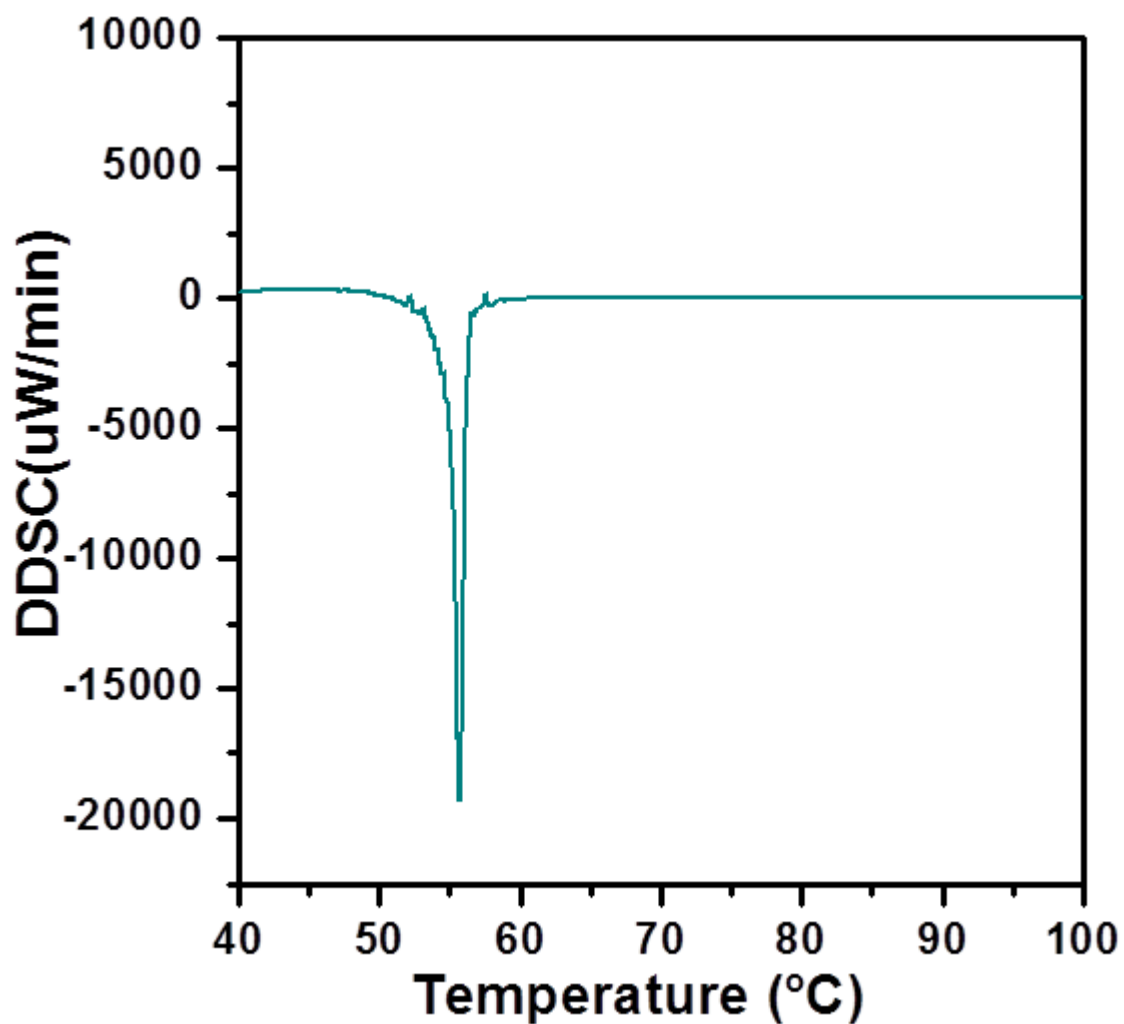


Fig. S11 Differential scanning calorimetric thermogram (DSC) of Mg^{2+} (4 equiv) coordination polymer gel **1** (20 mM).

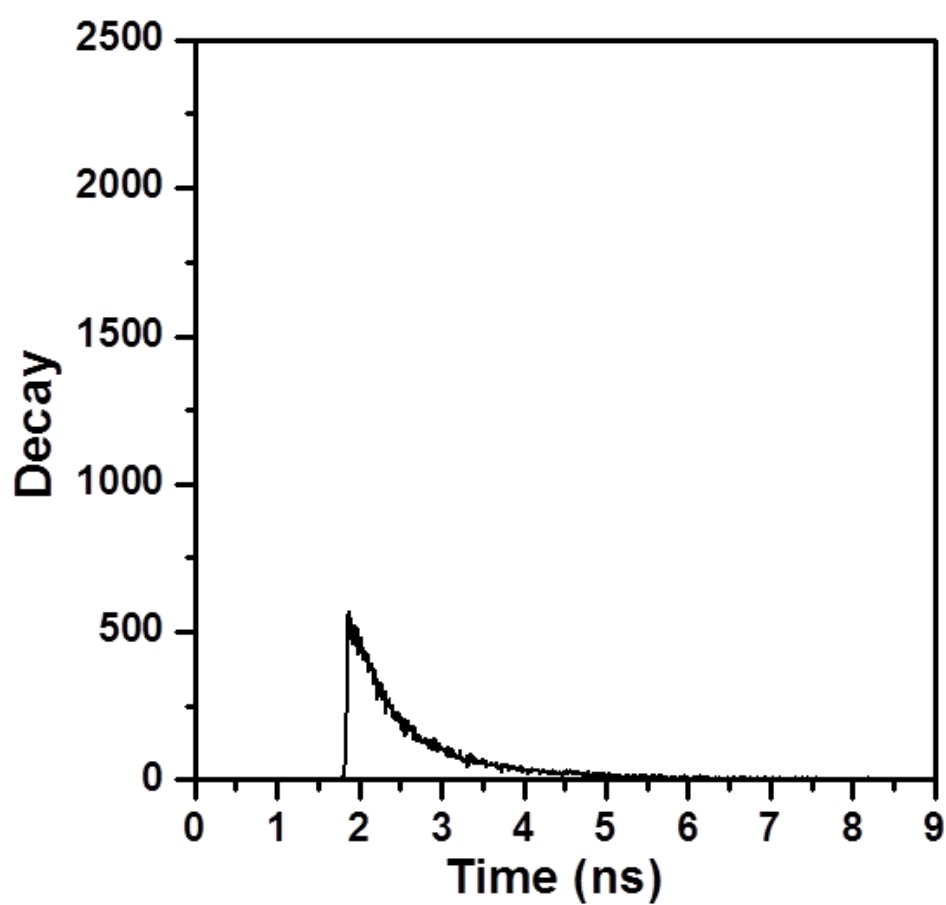


Fig. S12 Fluorescence decay of Mg²⁺ Coordination polymer gel **1** obtained by time-resolved fluorescence confocal microscopy.

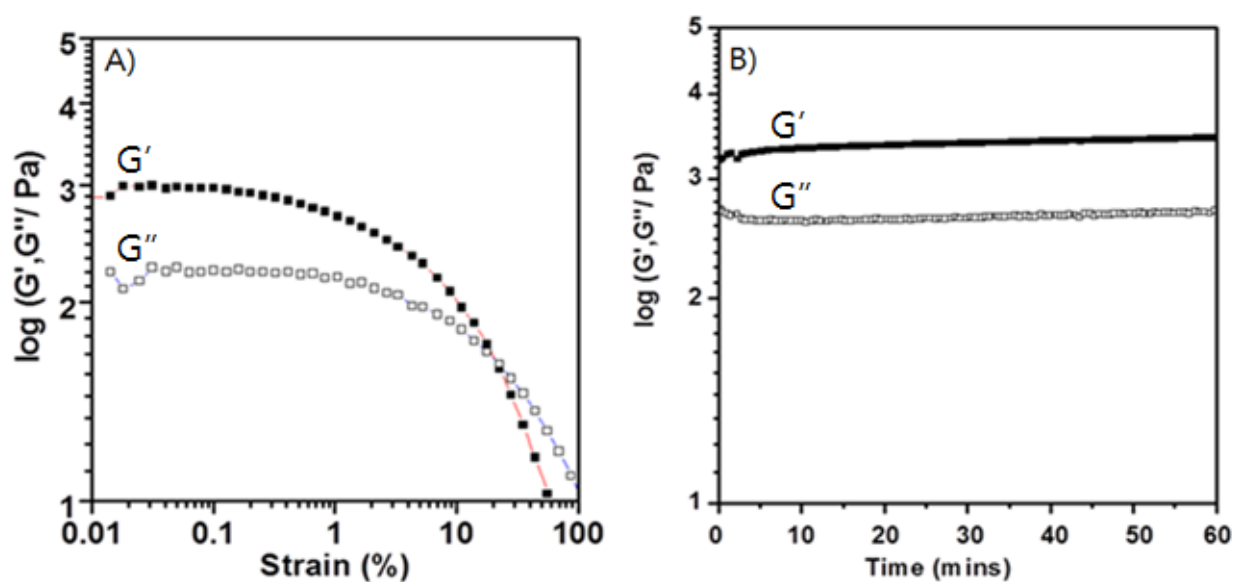


Fig. S13 (A) Strain sweep at a frequency of 1 rad s^{-1} of Mg^{2+} coordination polymer gel **1**. (B) Time sweep for Mg^{2+} coordination polymer gel **1** at strain of 0.1%.

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