

Stability trends and tautomerization of chalcocyclopentadienes. The role of aromaticity

Marcela Hurtado, Al-Mokhtar Lamsabhi, Otilia Mó, Manuel Yáñez and Jean-Claude Guillemin

Contribution from Departamento de Química, C-9. Universidad Autónoma de Madrid. Campus de Excelencia UAM-CSIC Cantoblanco. 28049-Madrid. Spain, and École Nationale Supérieure de Chimie de Rennes, CNRS, UMR 6226, Avenue du Général Leclerc, CS 50837, 35708 Rennes Cedex 7, France

Supporting Information

(Total number of pages 4)

Table S1. Calculated Gibbs free energies (hartrees)

		ΔG^a	ΔG^b
O	iso1	-269.26087	-268.578126
	iso2-2	-269.276363	-268.590375
	iso3-2	-269.277279	-268.591346
S	iso1	-592.240462	-591.19169
	iso2-1	-592.247604	-591.196924
	iso3-1	-592.247502	-591.196636
Se	iso1	-2593.46115	-2593.44468
	iso2-1	-2593.46706	-2593.44699
	iso3-1	-2593.46725	-2593.44665

^a at B3LYP/6-31+G(d,p)

^b CCSD/6-311+G(d,p)

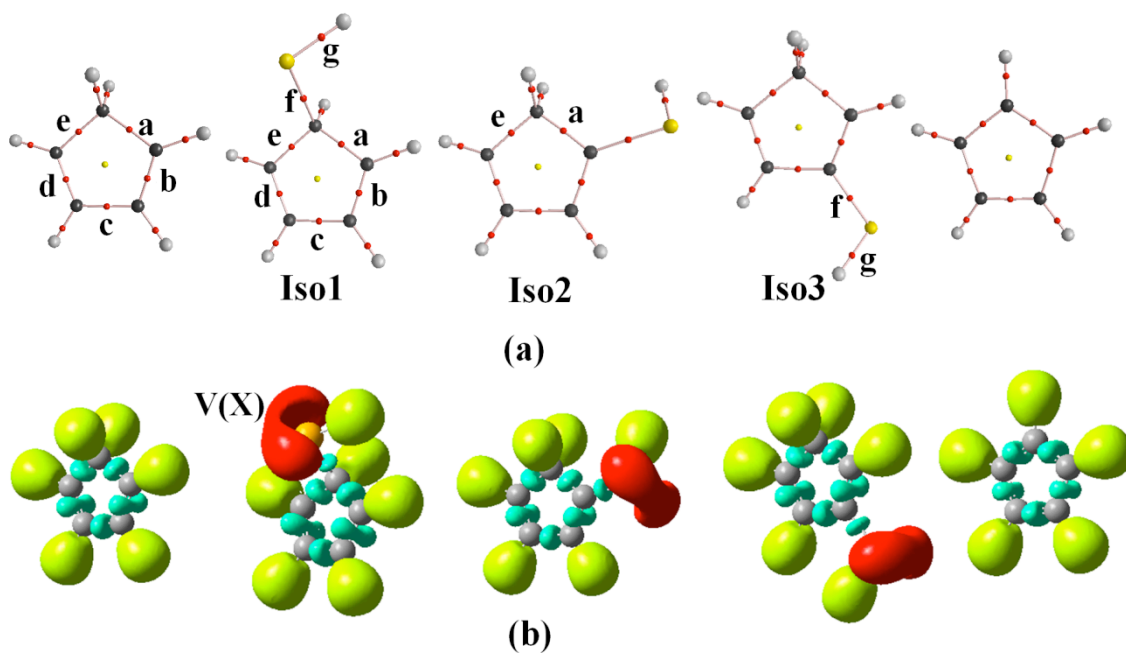


Table S2. Electron density (a.u.) at the BCP; **basin population** (e^-); **bond length** (\AA) and **bond order**, evaluated at the C-C bonds of the ring (**a**, **b**, **c**, **d** and **e**), at the C-X bond (**f**) and at the X-H bond (**g**). **V(X)** is the electron-pair basin on the heteroatom.

	a	b	c	d	e	f	g	V(X)
Cyclopentadiene	0.256	0.337	0.273					
	1.97	3.44	2.20					
	1.507	1.352	1.469					
	1.04	1.81	1.13					

Cyclopentadienyl anion	0.295 2.96 1.418 1.40							
O-ISO1	0.256 2.04 1.519 1.00	0.339 3.43 1.347 1.84	0.269 2.160 1.479 1.10	0.340 3.46 1.346 1.84	0.260 2.04 1.511 1.01	0.254 1.26 1.427 0.93	0.365 1.71 0.966 0.77	4.72
O-ISO2	0.262 2.06 1.504 1.02	0.335 3.68 1.355 1.70	0.271 2.18 1.470 1.12	0.337 3.50 1.352 1.82	0.254 1.97 1.510 1.03	0.286 1.47 1.362 1.03	0.363 1.73 0.967 0.75	4.48
O-ISO3	0.253 1.95 1.508 1.03	0.335 3.70 1.354 1.70	0.279 2.28 1.466 1.10	0.339 3.42 1.349 1.82	0.256 1.95 1.509 1.04	0.285 1.44 1.365 1.02	0.364 1.75 0.966 0.76	4.53
S-ISO1	0.258 2.00 1.507 1.03	0.338 3.43 1.350 1.81	0.272 2.180 1.472 1.12	0.339 3.43 1.50 1.81	0.256 2.02 1.509 1.03	0.164 1.48 1.859 0.97	0.212 1.87 1.350 0.97	4.54
S-ISO2	0.253 2.00 1.513 1.02	0.334 3.53 1.356 1.73	0.274 2.20 1.466 1.13	0.337 3.43 1.352 1.81	0.256 1.95 1.508 1.04	0.194 1.82 1.771 1.09	0.212 1.85 1.351 0.96	4.39
S-ISO3	0.256 1.95 1.506 1.04	0.335 3.54 1.354 1.73	0.270 2.20 1.474 1.10	0.339 3.43 1.349 1.82	0.256 1.97 1.506 1.04	0.192 1.78 1.778 1.07	0.212 1.86 1.349 0.96	4.36
Se-ISO1	0.262 2.01 1.498 1.05	0.337 3.42 1.352 1.79	0.271 2.180 1.468 1.13	0.337 3.4 1.352 1.80	0.259 2.04 1.503 1.04	0.135 1.39 1.989 0.92	0.170 1.86 1.468 0.98	4.91
Se-ISO2	0.254 1.99 1.510 1.03	0.334 3.49 1.355 1.74	0.274 2.22 1.466 1.12	0.337 3.42 1.352 1.81	0.256 1.98 1.507 1.03	0.160 1.79 1.890 1.06	0.171 1.85 1.468 0.97	4.78
Se-ISO3	0.255 1.94 1.506 1.03	0.334 3.55 1.353 1.73	0.271 2.21 1.471 1.11	0.339 3.46 1.349 1.81	0.257 1.97 1.505 1.04	0.160 1.77 1.894 1.05	0.172 1.86 1.463 0.98	4.72
Te-ISO1	0.264 2.04 1.492 1.07	0.334 3.41 1.356 1.76	0.275 2.22 1.465 1.14	0.335 3.38 1.356 1.77	0.260 2.05 1.497 1.06	0.093 1.34 2.237 0.86	0.125 1.86 1.679 0.98	5.29
Te-ISO2	0.253 1.94 1.512 1.03	0.334 3.38 1.356 1.77	0.274 2.19 1.468 1.12	0.337 3.41 1.353 1.81	0.256 1.97 1.506 1.03	0.111 1.95 2.128 0.98	0.123 1.93 1.680 0.97	5.18
Te-ISO3	0.256 1.94 1.507 1.03	0.335 3.43 1.354 1.78	0.269 2.16 1.474 1.12	0.338 3.44 1.351 1.82	0.256 1.96 1.506 1.04	0.110 1.86 2.134 0.97	0.124 1.84 1.677 0.98	5.36

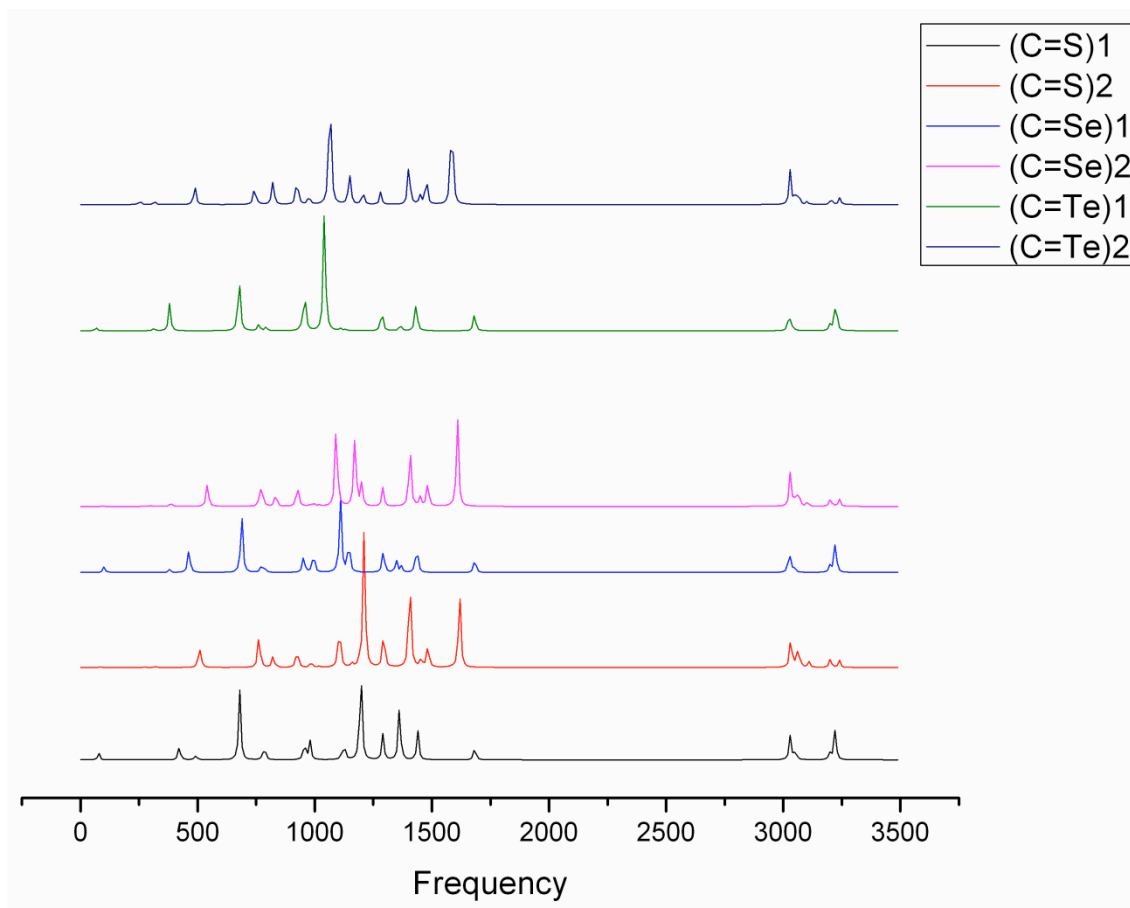


Figure S1. IR spectrum for the keto forms of S, Se and Te containing derivatives.
Frequencies in cm^{-1} .