

## **Supporting information for**

### **Tripod facial surfactants with benzene as central core: design, synthesis and self-assembly study**

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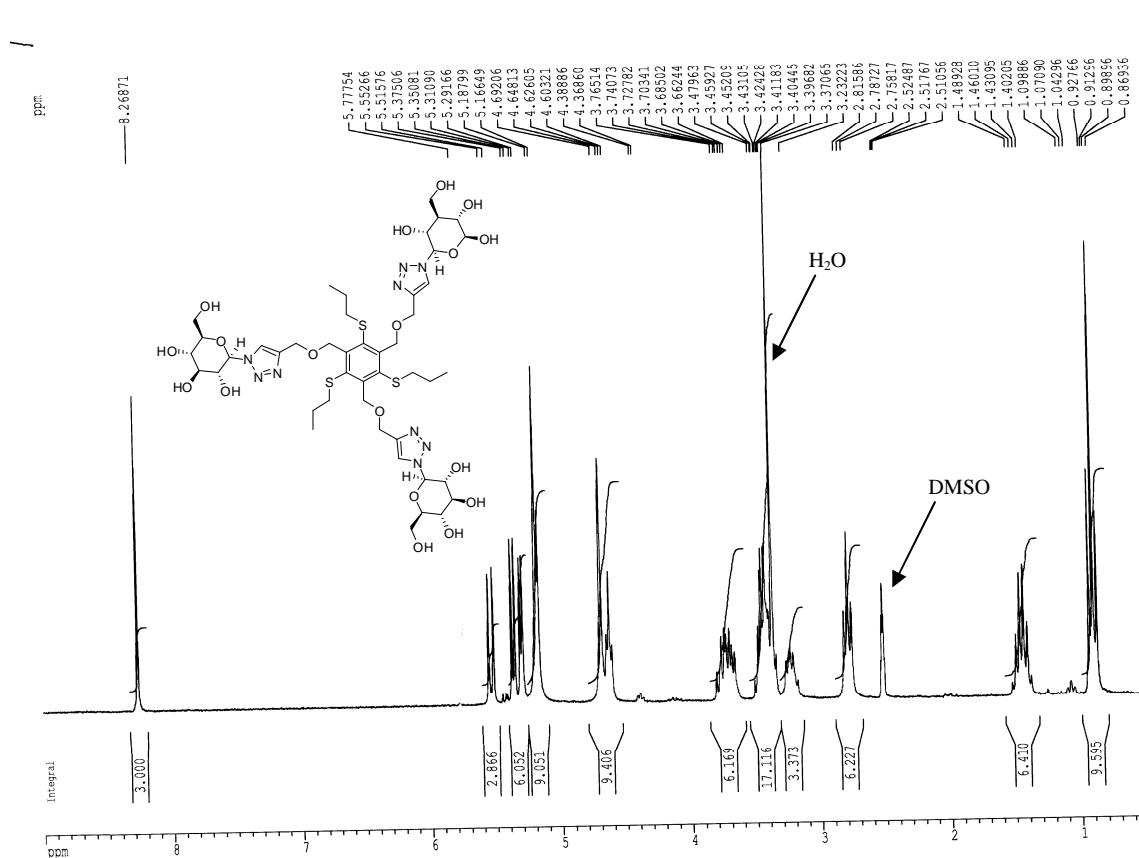
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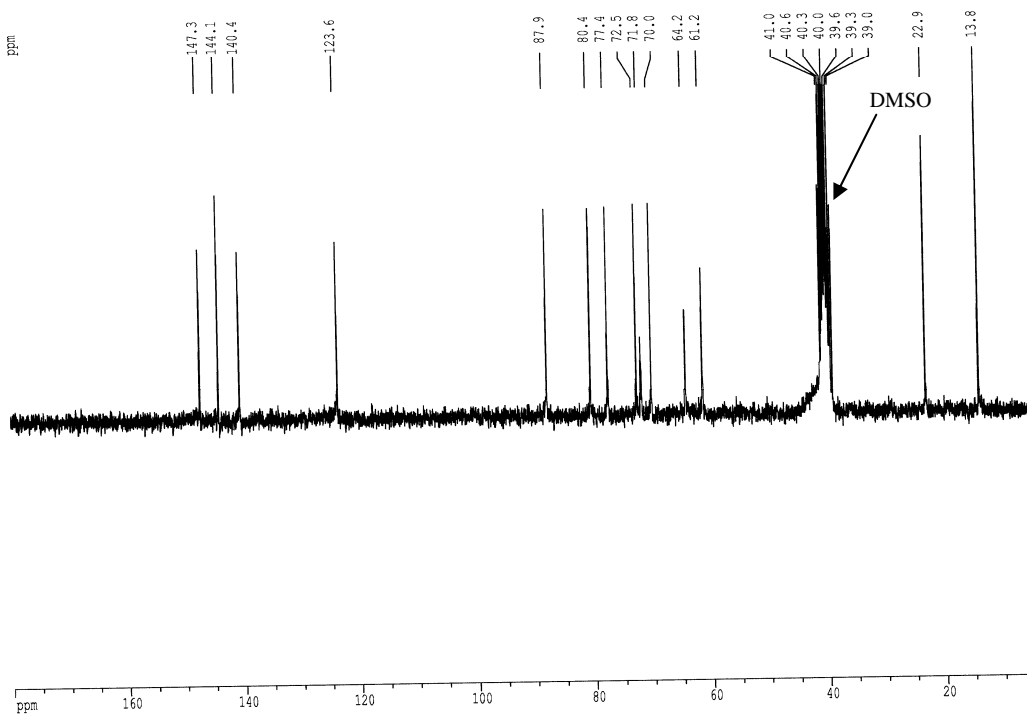
Paul Scherrer Institut, 5232 Villigen PSI, Switzerland

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### <sup>1</sup>H NMR and <sup>13</sup>C NMR spectra scanned

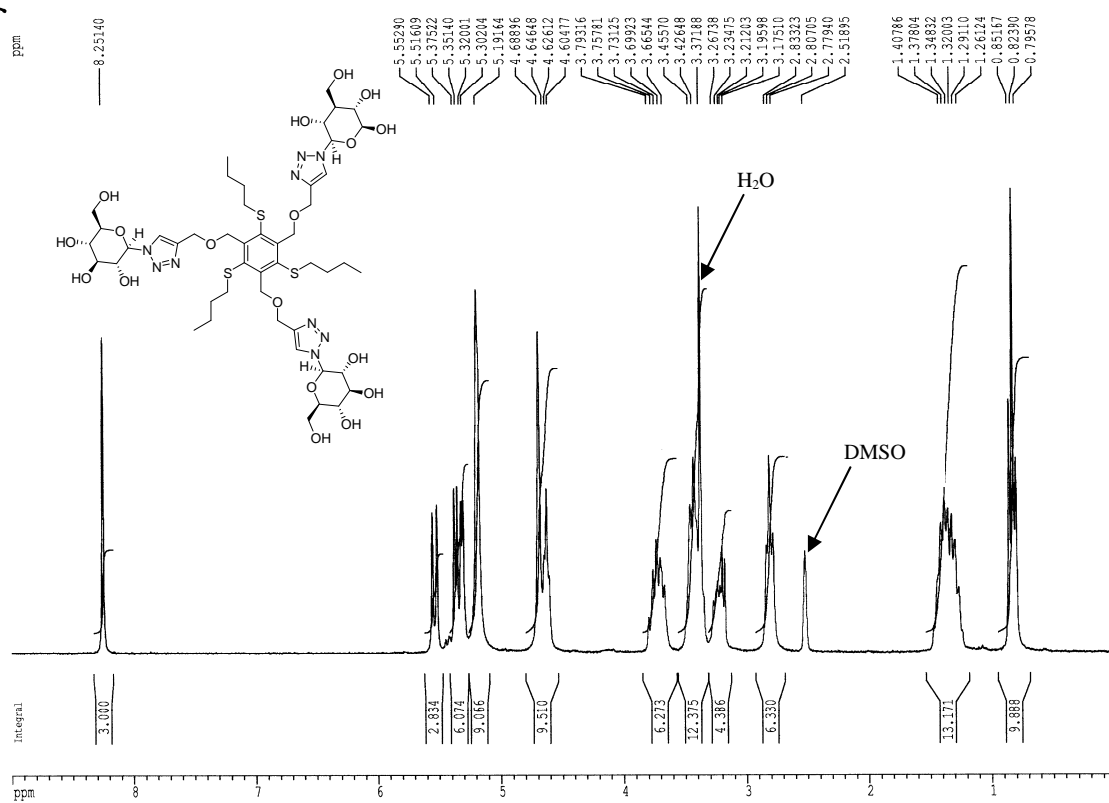


<sup>1</sup>H NMR (DMSO) for compound C3Glu3.

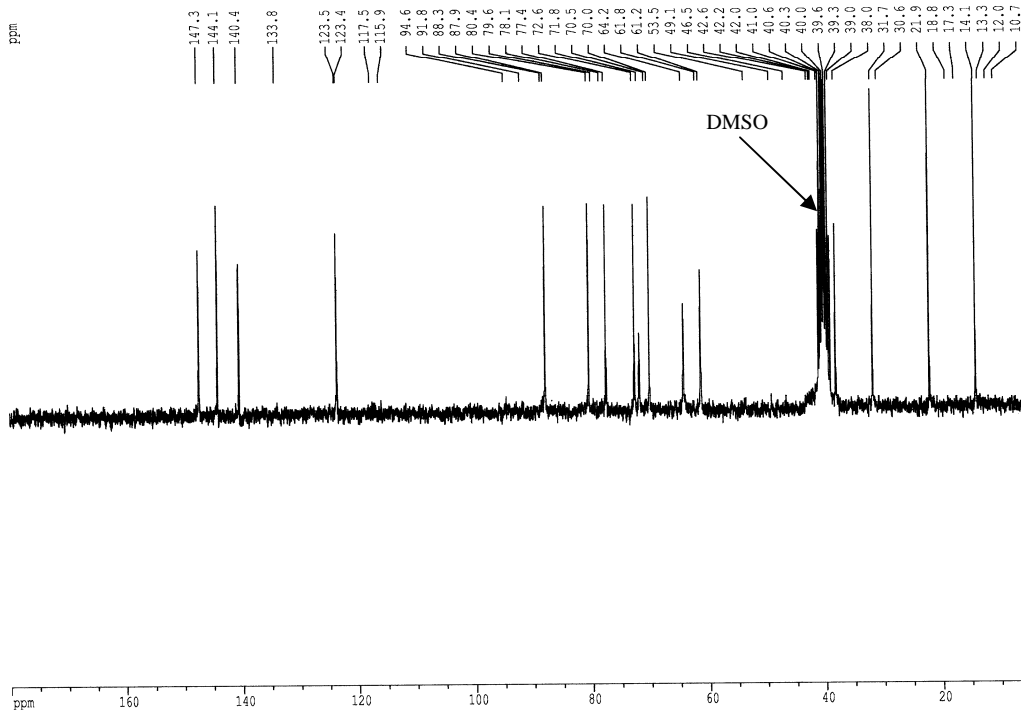


<sup>13</sup>C NMR (DMSO) for compound C3Glu3.

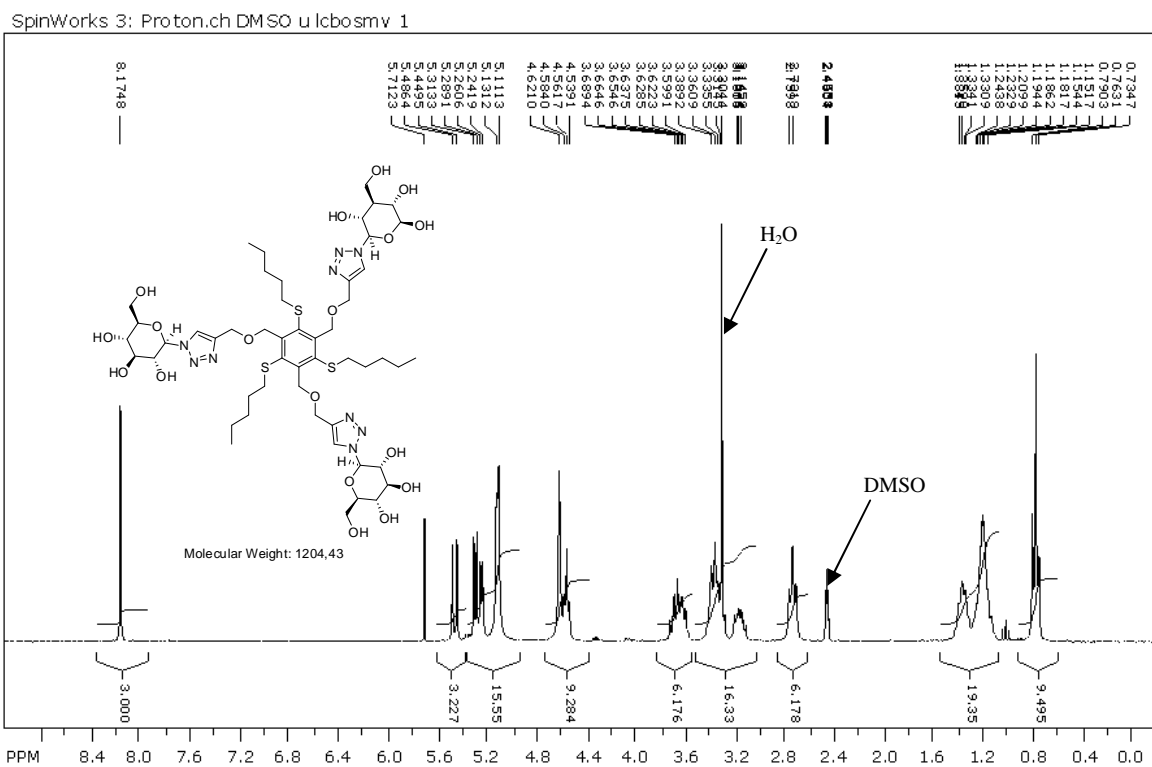
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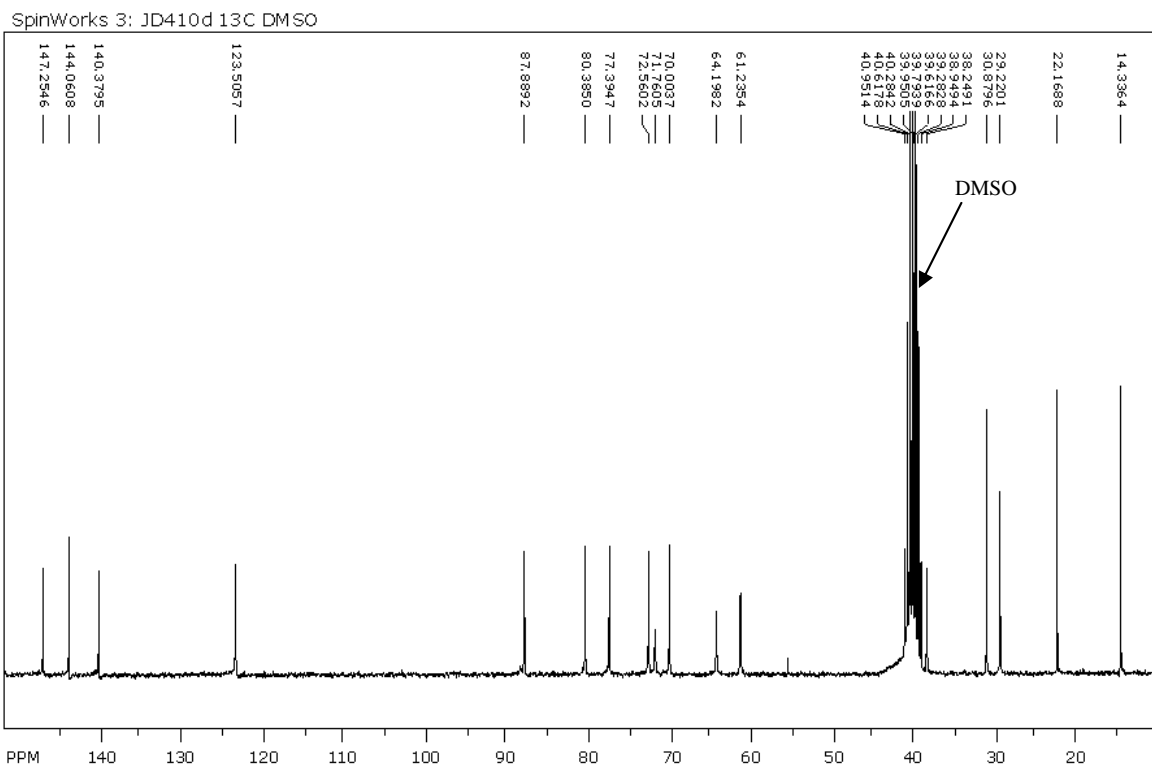
<sup>1</sup>H NMR (DMSO) for compound C4Glu3.



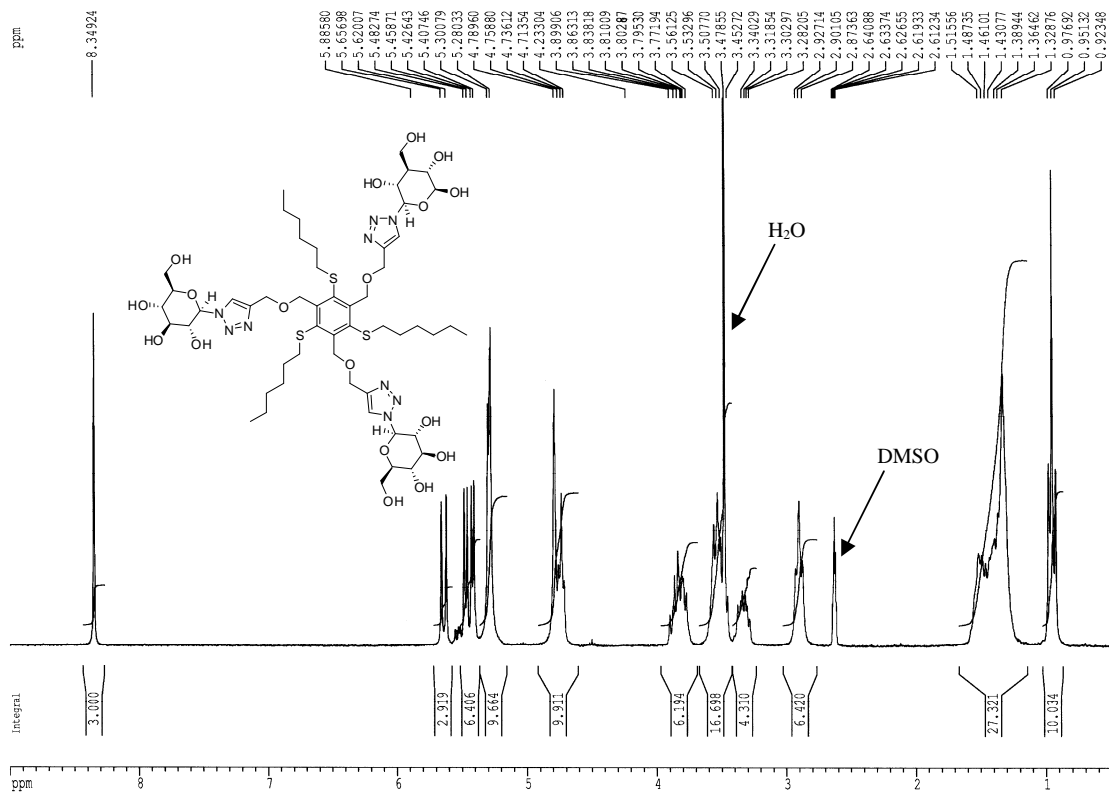
<sup>13</sup>C NMR (DMSO) for compound C4Glu3.



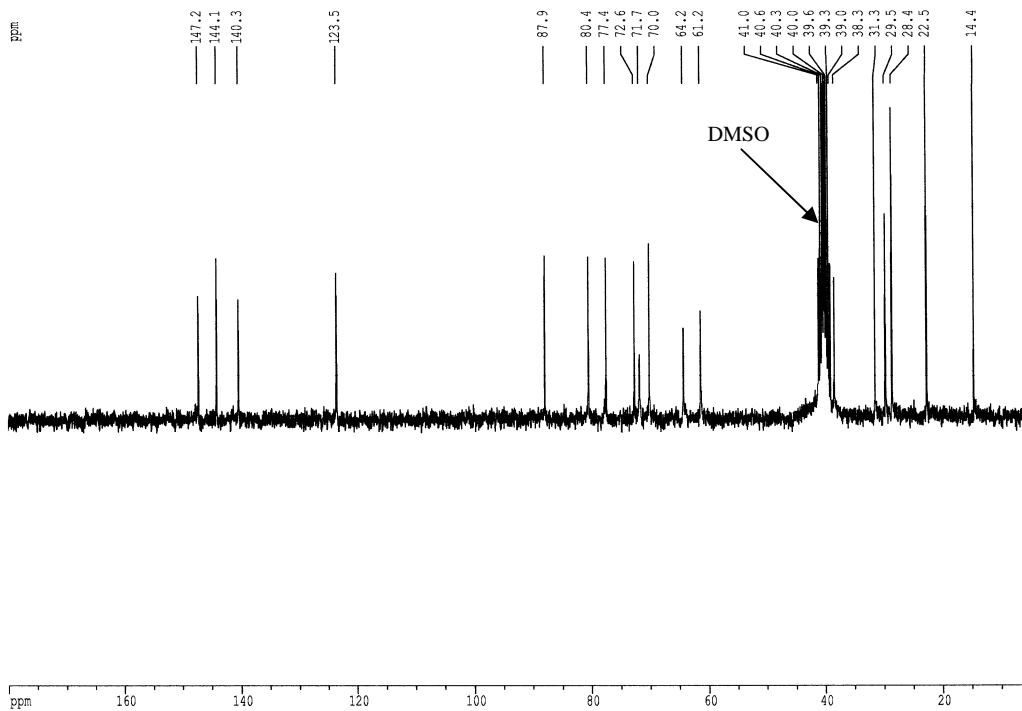
<sup>1</sup>H NMR (DMSO) for compound C5Glu3.



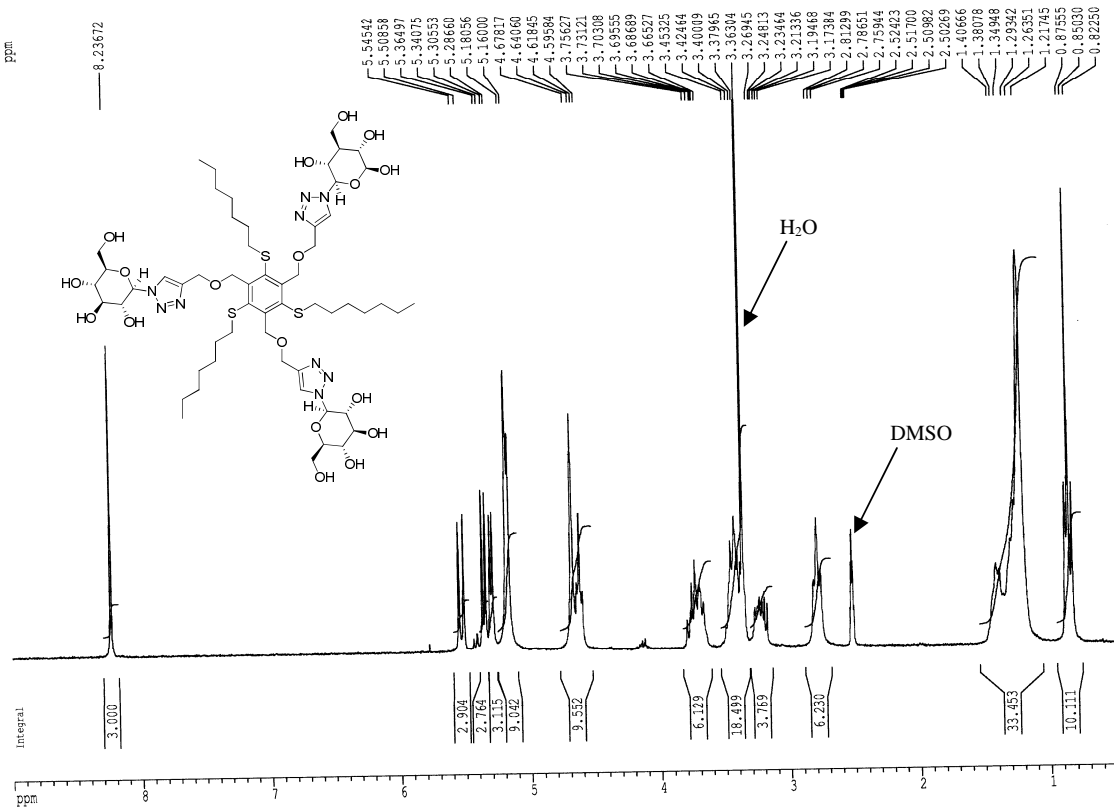
<sup>13</sup>C NMR (DMSO) for compound C5Glu3.



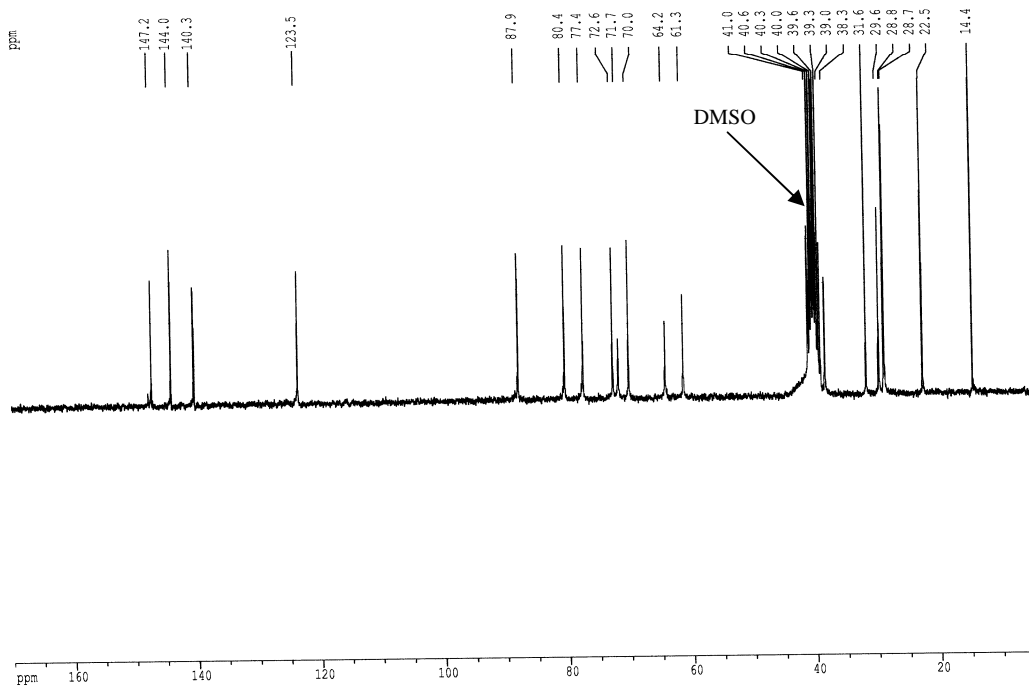
**<sup>1</sup>H NMR (DMSO) for compound C6Glu3.**



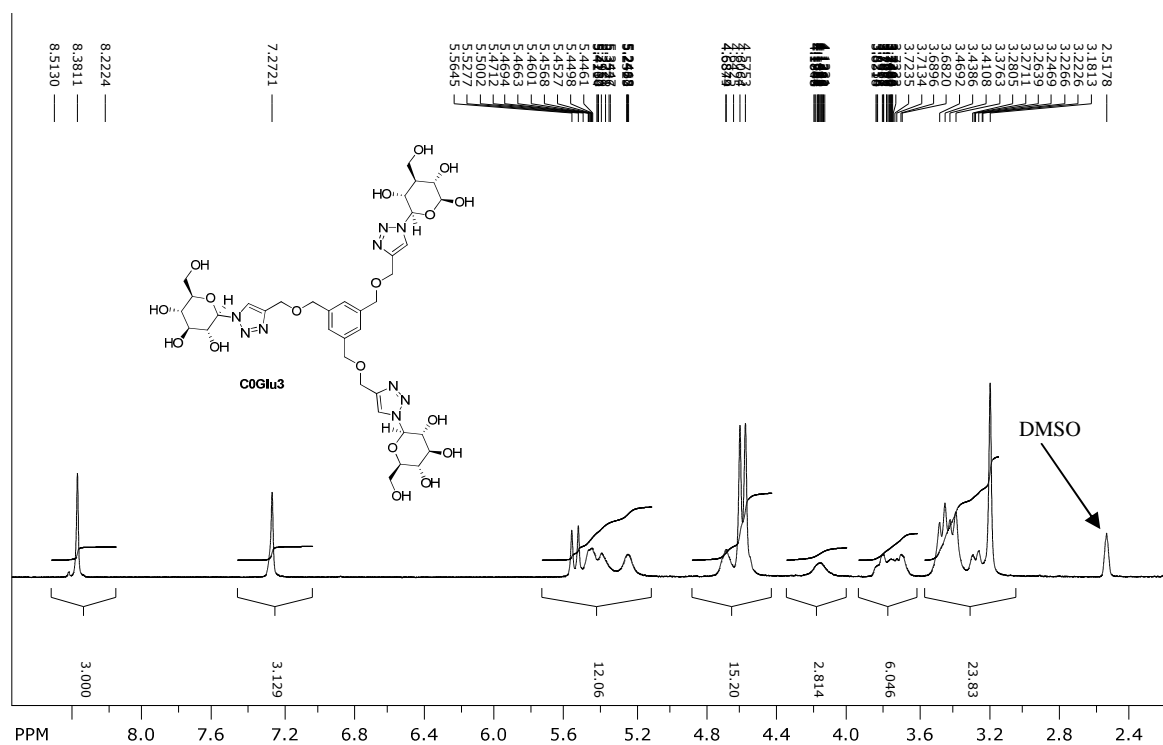
**<sup>13</sup>C NMR (DMSO) for compound C6Glu3.**



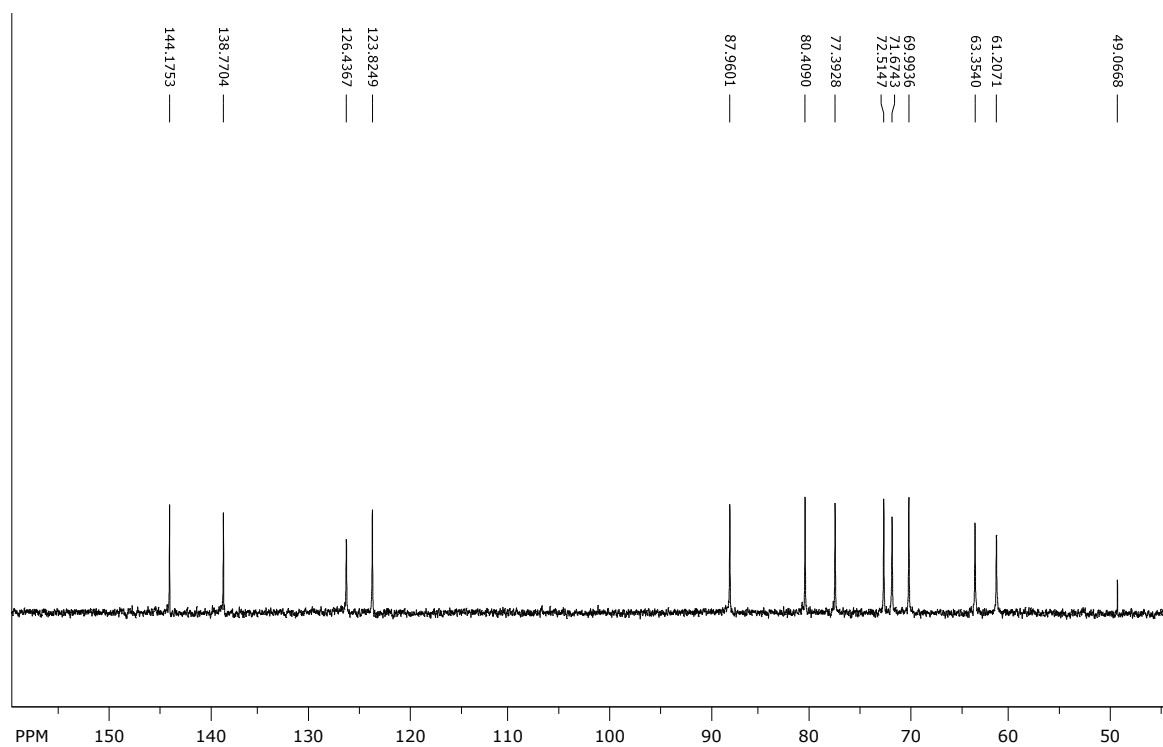
<sup>1</sup>H NMR (DMSO) for compound C7Glu3.



<sup>13</sup>C NMR (DMSO) for compound C7Glu3.



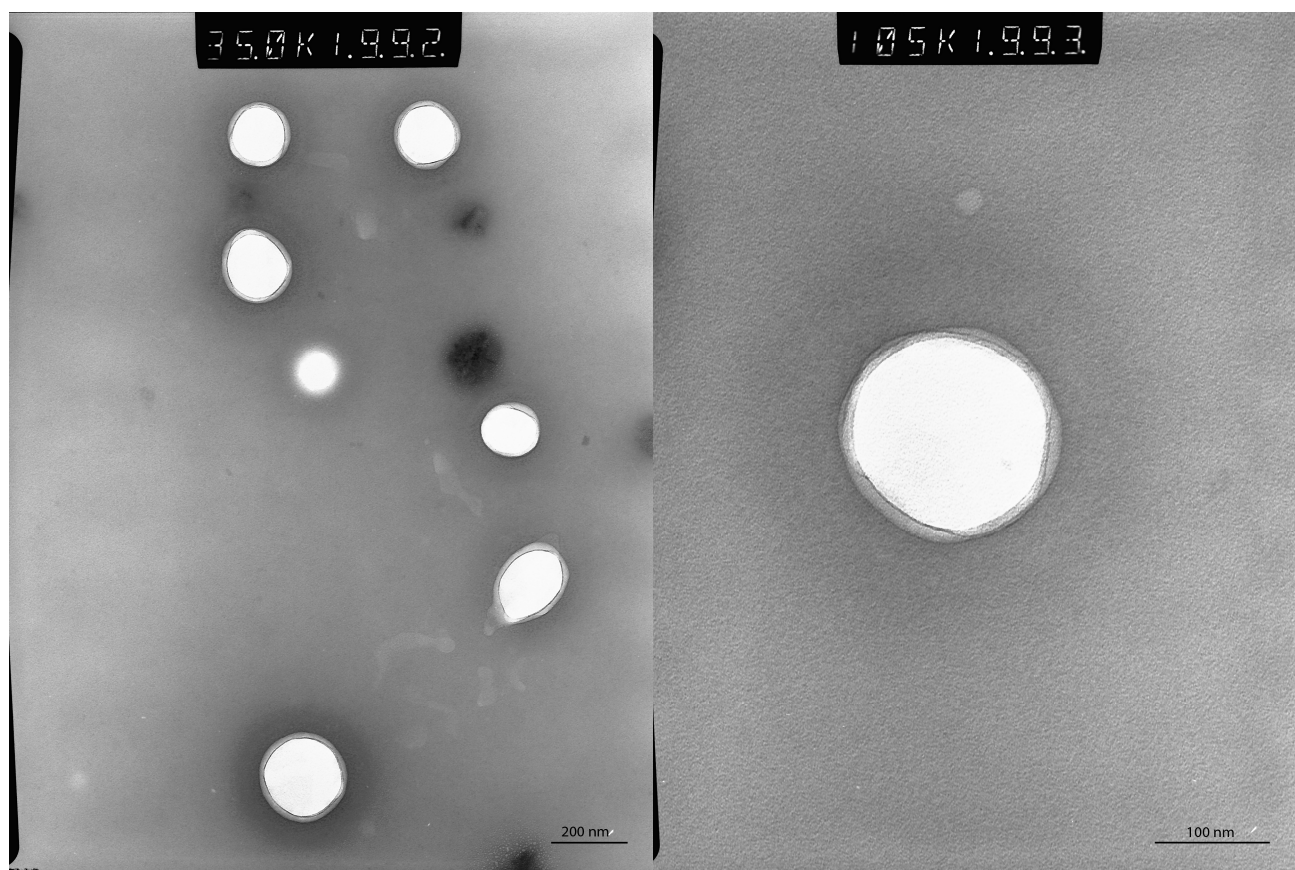
**<sup>1</sup>H NMR (DMSO) for compound C0Glu3.**



**<sup>13</sup>C NMR (DMSO) for compound C0Glu3.**

### Electron microscopy visualisation by negative staining

A solution of the compound **C7Glu3** at 10 mg/mL was sonicated with a microprobe during 15 min and/or 1 h in a bath at 50°C to allow the formation of vesicles. The solutions were passed through a 0.45 µm filter. A drop of the sample was layed down on a metallic grid coated with a film of carbon and the excess of liquid was imbibed with a blotting paper. Then, a drop of a 3% ammonium molybdate was put on the grid. We waited 30 s to optimise the adsorption and the excess of liquid was removed again. The visualisation by electron microscopy with negative coloration was monitored on a Philips CM/2 transmission electron microscope.



**Figure 1.** Pictures of compound **C7Glu3**.

The same sonicated samples was analysed by DLS. These results were collected in the table below and compared to particle sizes found by TEM.

**Table 1.** Data of vesicles for compound **C7Glu3** measured by DLS.

Compound	d (nm) DLS	% (volume)	d <sup>1/2</sup> (nm)	Count rate (kcps)	d (nm) MET <sup>a</sup>
<b>C7Glu3</b>	538.0	52%	149.7	269.7	167-233

<sup>a</sup> Values correspond to the interval between the smaller and the bigger particles observed on the pictures.



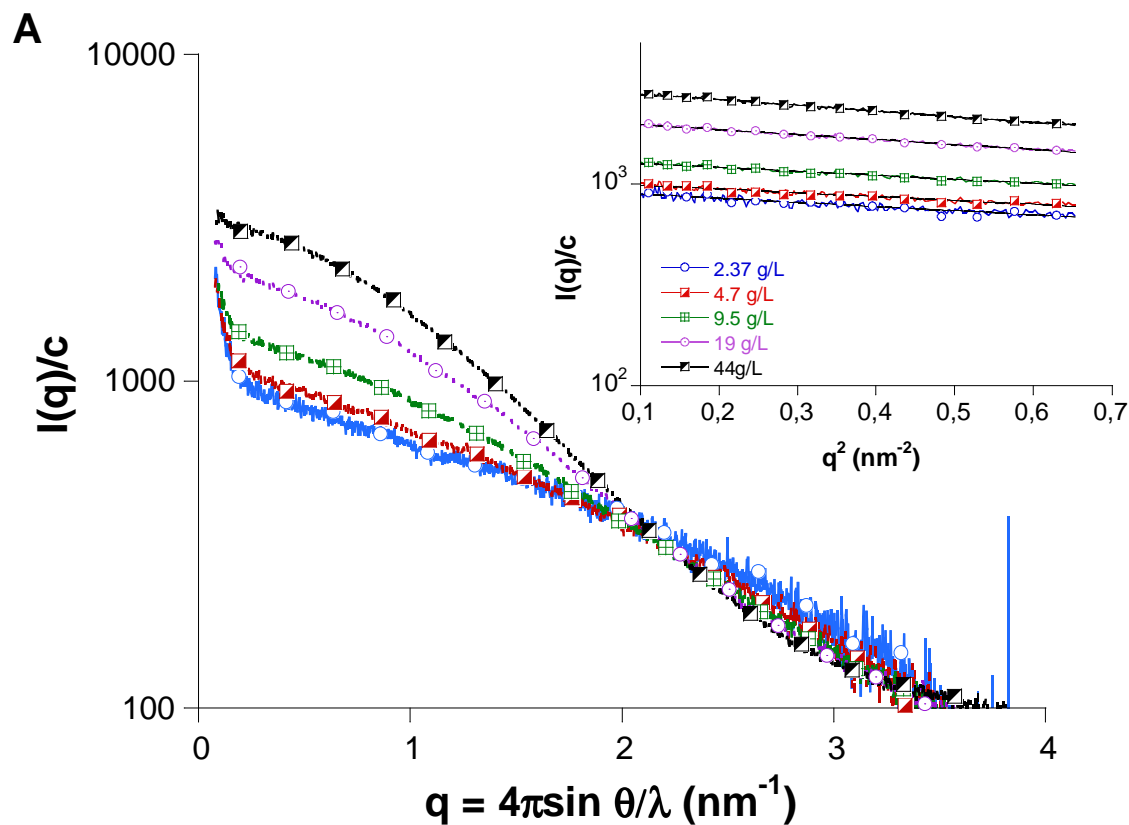
## Small Angle X-Ray Scattering

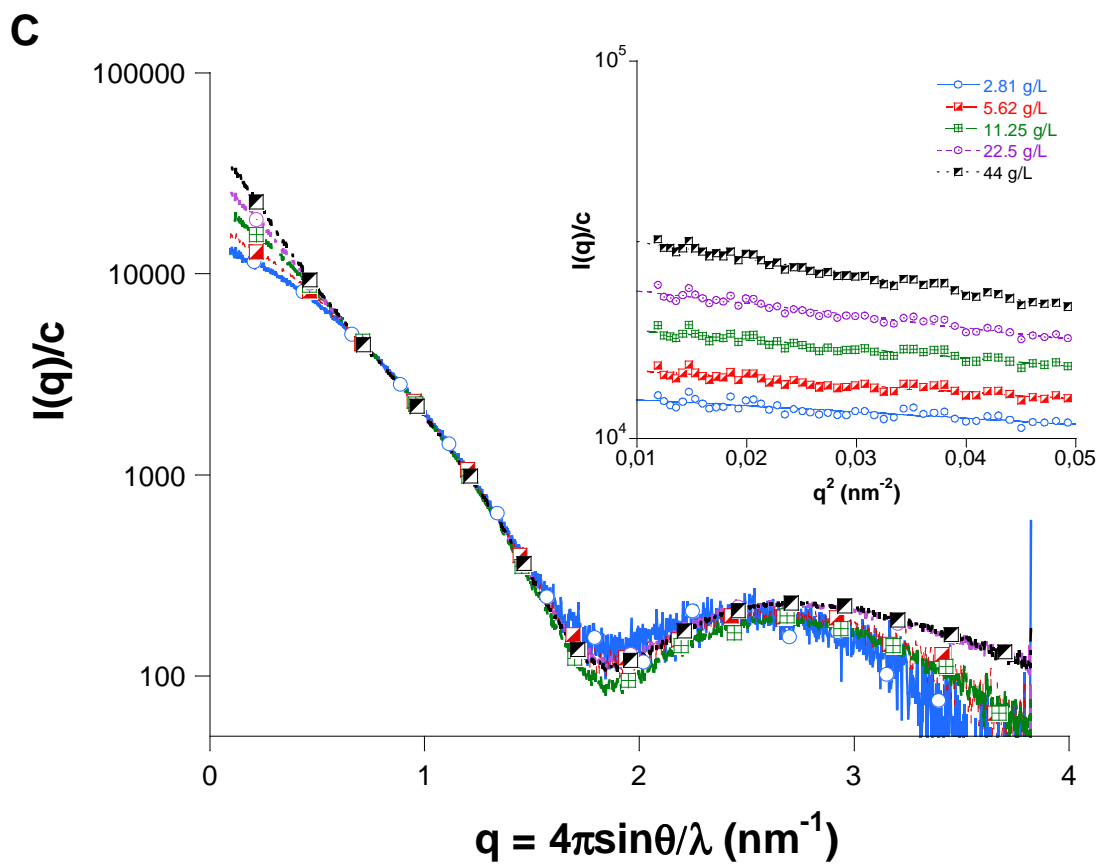
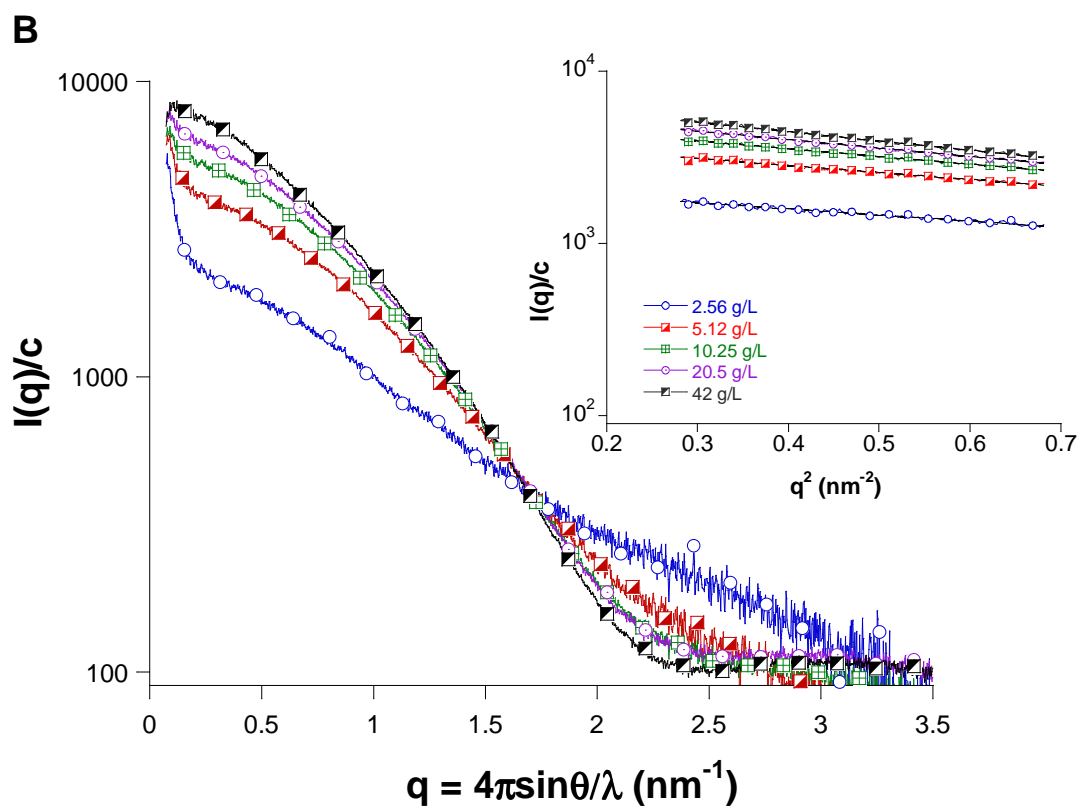
**Table 2** SAXS data of **CnGlu3** (n= 3, 4, 5) vs. concentration (g.L<sup>-1</sup>):

Entry	c (g/L)	MW <sup>#</sup> (g/mol)	Rg <sup>§</sup> (nm)	H <sup>¶</sup> (nm)	Nagg
<b>C3Glu3</b>	2.4	941.2	1.18	n a	0.8
	44	3003.4	1.40	n a	2.7
<b>C4Glu3</b>	2.6*	2480.8	1.56	n a	2.1*
	42	8280.0	1.93	5.8	7.1
<b>C5Glu3</b>	2.8	16244	3.61	6.4	13
	5.6	19355	3.76	7.9	16
	11.2	24995	4.19	10.5	20
	22.5	32452	4.85	14	27
	42	45513	5.67	19.8	38

<sup>#</sup> Molecular mass calculated from forward intensity; <sup>§</sup> Radius of gyration obtained from Guinier approximation; <sup>¶</sup> Height of rod obtained from SASfit models

\* This value is an intermediate state between the monomer, which could be characterized at CMC or below, and the final aggregate. Measurement below CMC could not be done because of a poor SAXS signal below the CMC or CAC.





**Figure 2.** SAXS patterns of C3Glu3 (A) C4Glu3 (B) and C5Glu3 (C) as a function of concentration. Guinier plots are shown in insert.

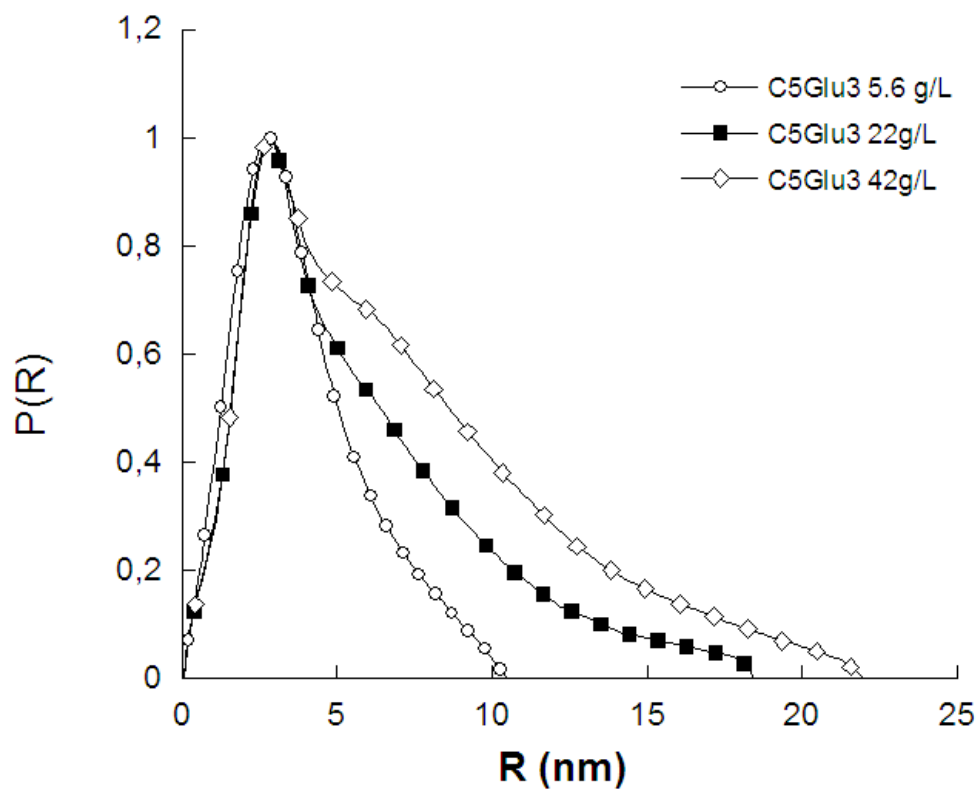


Figure 3. Pair distribution function of C5Glu3 at different concentrations