

Electronic Supplementary Information (ESI) for the paper

Self-assembly and decay of Mn(II) pivalate-phosphonate complexes

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Table S1. Crystal data and structure refinement statistics for manganese(II) compounds.

Parameter	1	2·2C₇H₁₆·2MeOH	3	3a
CSD number	951019	951020	975925	982652
Empirical formula	C ₉₆ H ₁₈₈ Cl ₄ Mn ₁₀ O ₄₄ P ₄	C ₁₆₄ H ₃₅₄ Cl ₂ Mn ₂₀ O ₈₈ P ₁₂	C ₇₆ H ₁₅₆ Mn ₄ O ₃₆ P ₄	C ₇₆ H ₁₅₆ Mn ₄ O ₃₆ P ₄
Molecular mass	2861.53	5275.79	1989.64	1989.64
<i>T</i> /K	150(2)	173(2)	184(2)	120(2)
Space group	Tetragonal, <i>I</i> -42 <i>m</i>	Triclinic, <i>P</i> -1	Monoclinic, <i>C</i> 2/ <i>c</i>	Trigonal, <i>P</i> -3
<i>Z</i>	2	1	4	2
<i>a</i> /Å	22.4667(10)	20.016(1)	27.413(17)	16.2519(15)
<i>b</i> /Å	22.4667(10)	20.421(1)	17.073(11)	16.2519(15)
<i>c</i> /Å	14.6274(7)	20.656(1)	26.290(16)	25.431(5)
α /°	90.00	86.268(7)	90.00	90
β /°	90.00	68.581(7)	108.213(10)	90
γ /°	90.00	62.388(7)	90.00	120
<i>V</i> /Å ³	7383.2(7)	6909.1(6)	11688(13)	5817.0(15)
<i>d</i> _{calc} /g·cm ⁻³	1.287	1.268	1.131	1.160
μ /mm ⁻¹	1.008	1.037	0.543	0.548
θ /° scanning range	1.28–31.18	2.14–28.70	1.43–26.02	0.80–27.48
Reflections measured	47010	73931	41298	47336
Independent reflections	6132	35351	11482	8921
Reflections with <i>I</i> >2 σ _{<i>I</i>}	5609	21024	4132	3948
<i>R</i> _{int}	0.0441	0.0412	0.1125	0.1169
Number of refined parameters	199	1434	1012	687
<i>T</i> _{min} / <i>T</i> _{max}	0.707/0.924	0.713/0.886	0.909/0.958	0.875/0.942
<i>Goof</i>	1.086	1.011	1.401	1.321
<i>R</i> ₁ (<i>I</i> >2 σ _{<i>I</i>})	0.0264	0.0619	0.1148	0.0842
<i>wR</i> ₂ (<i>I</i> >2 σ _{<i>I</i>})	0.0712	0.1631	0.2744	0.2364
<i>R</i> ₁ (all data)	0.0350	0.1210	0.2036	0.1827
<i>wR</i> ₂ (all data)	0.0851	0.2075	0.2985	0.2927
Flack parameter	0.005(5)			

Table S2. Selected bond lengths [Å], bond angles [deg] and H-bonds parameters in manganese compounds.

Parameter	1*	2·2C₇H₁₆·2MeOH**	3***	3a****
Mn–O(μ -OH)	–	2.210(3), 2.135(3)	–	
Mn–Cl(μ -Cl)	2.4721(6)	2.5301(12), 2.5404(12)	–	
Mn–O(μ -Piv)	2.133(3)–2.224(2)	2.055(3)–2.281(3)	–	
Mn–O(O ₃ PBu ^t)	2.110(2)–2.292(2)	2.047(3)–2.500(3)	2.109(11)– 2.169(9)	2.077(10)– 2.165(5)
Mn–O(HPiv)	2.292(2)	2.208(5)–2.327(7)	2.272(12)– 2.356(13)	2.221(7)– 2.386(15)
Mn–O(H ₂ O)	–	2.094(3)–2.375(2)	–	
P–O	1.534(2)–1.539(2)	1.506(3)–1.551(3)	1.472(11)– 1.550(11)	1.404(11)– 1.620(11)
O–P–O	104.28(12)– 112.74(8)	103.5(2)–114.6(2)	108.3(6)– 113.6(7)	105.0(6)– 116.5(7)
C–O(Piv)	1.240(4)–1.286(4)	1.239(6)–1.278(5)	–	
C–O(HPiv)	1.213(4), 1.319(5)	1.209(6)–1.323(8)	1.153(9)– 1.261(9)	1.162(7)– 1.285(8)
O–C–O	122.6(3)–123.9(3)	122.8(4)–126.6(4)	125.0(11)– 133.7(12)	122.5(13)– 128.8(12)
H-bonds				
O...O	2.603	2.552–2.613	2.505–2.621	2.536–2.591
H...O	1.78	1.72–1.78	1.67–1.80	1.72–1.75
O–H–O	166.0	170.2–173.0	167.0–173.8	163.0–174.6

* Symmetry codes: (i) $x, -y+1, -z+1$; (ii) $-x+1, -y+1, z$; (iii) $-x+1, -y+1, z$; (iv) y, x, z ; (v) $x, -y+1, -z+1$; (vi) $-x+1, y, -z+1$; (vii) $-x+1, y, -z+1$;

** Symmetry code: (i) $-x+2, -y+1, -z+1$.

*** Symmetry code: (i) $-x, y, -z+1/2$;

**** Symmetry code: (i) $-y+1, x-y, z$; (ii) $-x+y+1, -x+1, z$

Table S3. Mass spectrum of 1·2H₂O under TGA experiment conditions.

<i>m/z</i>	Ion	<i>I</i> /%	<i>m/z</i>	Ion	<i>I</i> /%
12	[C] ⁺	3.9	39	[C ₃ H ₃] ⁺	37.4
15	[CH ₃] ⁺	0.4	41	[C ₃ H ₅] ⁺ , [C ₂ HO] ⁺	11.3
17	[OH] ⁺	15.8	43	[C ₃ H ₇] ⁺	3.3
18	[H ₂ O] ⁺	59.1	44	[CO ₂] ⁺ , [C ₃ H ₈] ⁺	33.5
27	[C ₂ H ₃] ⁺	5.9	45	[COOH] ⁺ , [C ₂ H ₅ O] ⁺ [¹³ CO ₂] ⁺	0.7
28	[C ₂ H ₄] ⁺ , [CO] ⁺	28.6	56	[C ₄ H ₈] ⁺	3.1
29	[C ₂ H ₅] ⁺	8.4	57	[CCOOH] ⁺ [C ₄ H ₉] ⁺	0.5
31	[C ₂ H ₇] ⁺	0.4	58	[HCCOOH] ⁺	2.7
35	[³⁵ Cl] ⁺	0.2	59	[H ₂ CCOOH] ⁺	3.2
36	[H ³⁵ Cl] ⁺	100	87	[C ₃ H ₇ COOH] ⁺	0.5
38	[H ³⁷ Cl] ⁺	22.2	-	-	-

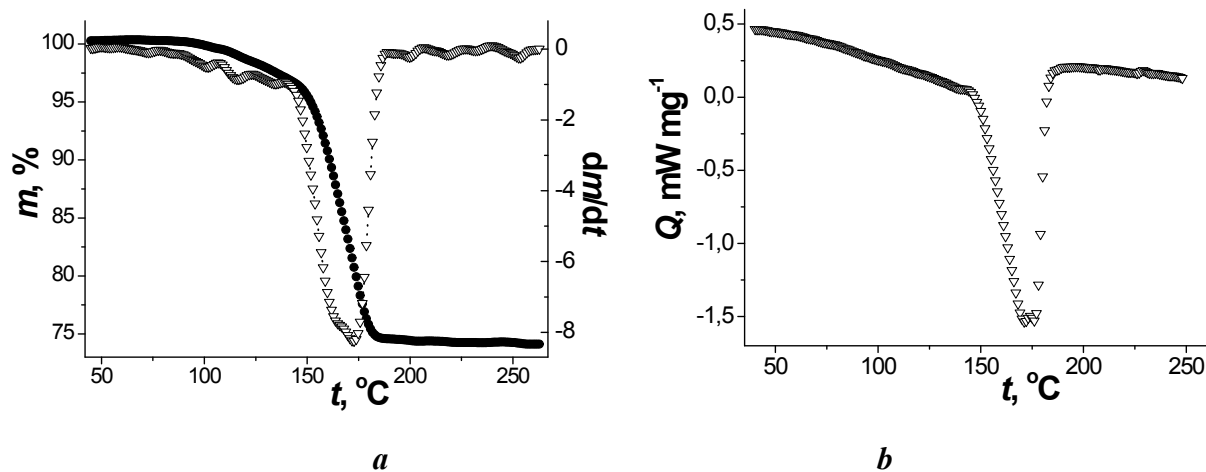


Fig. S1. Plots of mass change, first derivative of mass (*a*), and thermal flux *versus* temperature (*b*) for **1**.

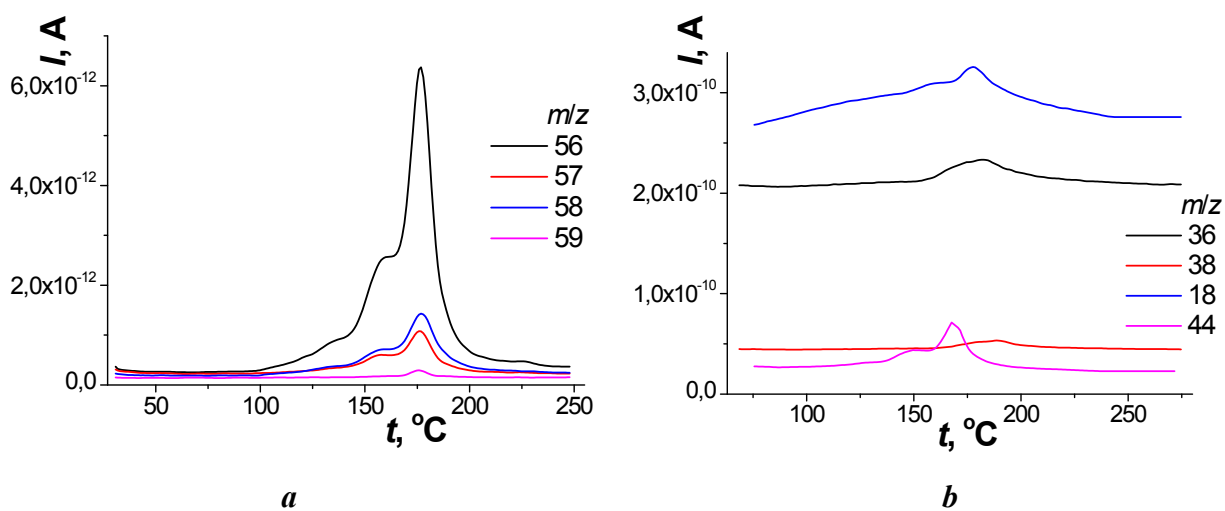


Fig. S2. Fragments of the mass spectrum (*a* and *b*) of **1** under the conditions of the TGA experiment.

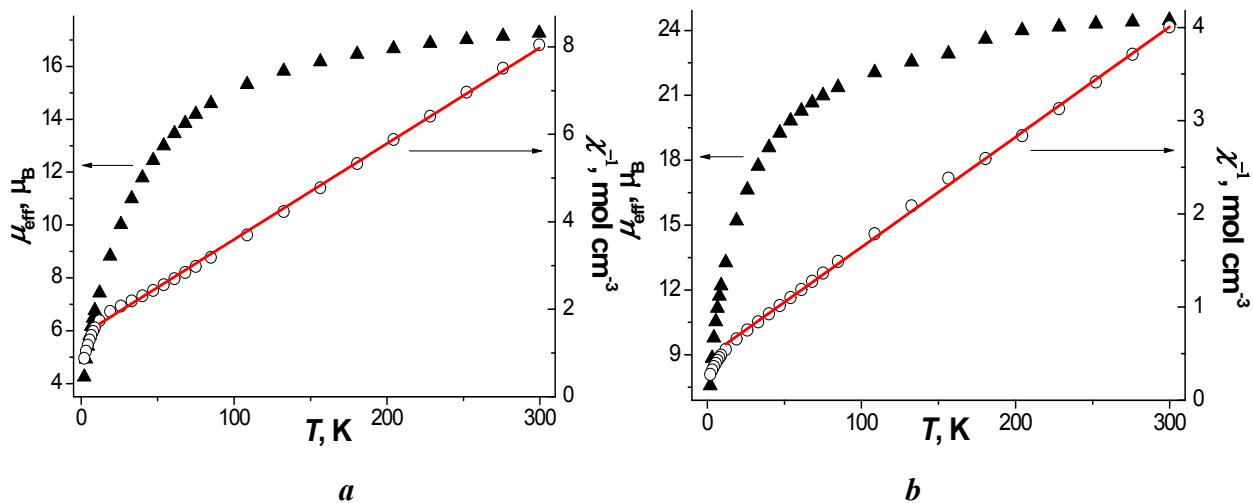


Fig. S3. Magnetic properties of **1** (a) and **2** (b): $\mu_{\text{eff}}(T)$ (\blacktriangle) and $1/\chi(T)$ (\circ) plots; calculated data (see main text) are shown as lines.

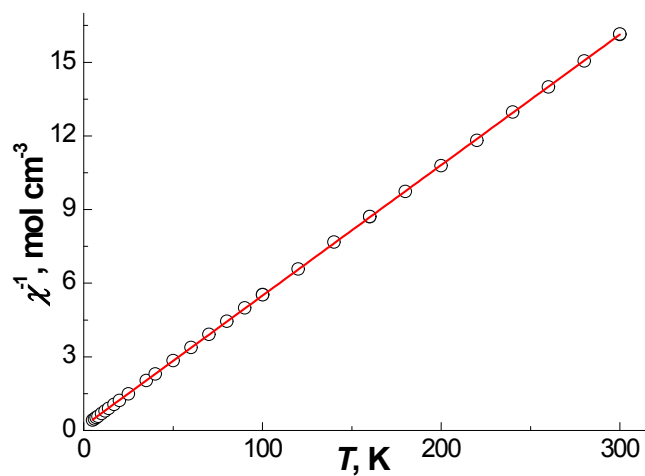


Fig. S4. $1/\chi(T)$ (\circ) plot and calculated data (line, see main text) for **3**·8H₂O.

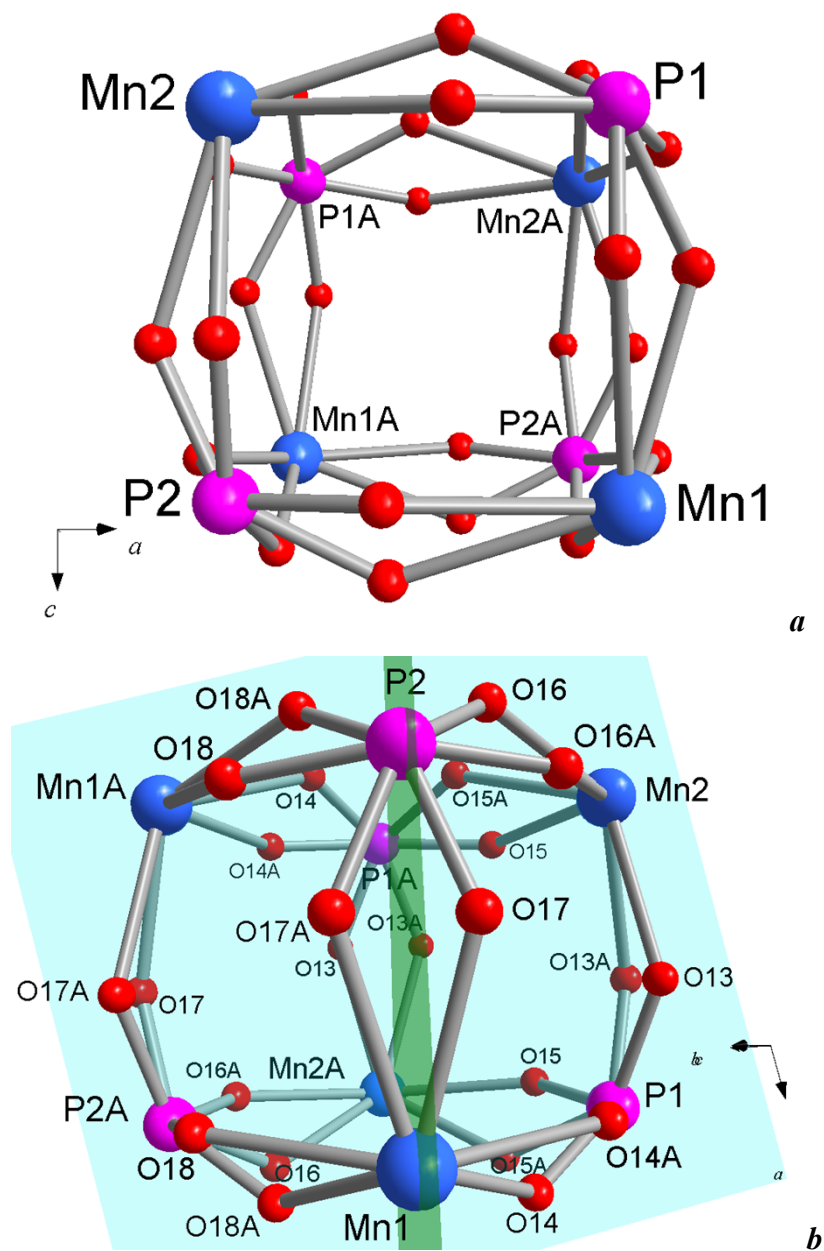


Fig. S5. Disorder of the O_3P fragments in the structure **3** (Piv anions and t Bu-groups at P atoms are omitted). *b* - Two groups of O atoms (without index and with index “A”) correspond to different isomers. Blue and green planes correspond two “non-crystallography” *m*-planes passing through Mn1, Mn2A, P1A, P2 and Mn1A, Mn2, P1, P2A atoms.