

Electronic Supplementary Information (ESI)

Hydrothermal synthesis of porous α -Fe₂O₃ nanostructures for highly efficient Cr(VI) removal

*Er-tao Liu,^a Huiping Zhao,^a Hui Li,^a Guangfang Li,^a Yunling Liu^b and Rong Chen^{*a}*

^a School of Chemistry and Environmental Engineering, Wuhan Institute of Technology, Xiongchu Street, Wuhan, 430073, PR China

^b State Key Laboratory of Inorganic Synthesis and Preparative Chemistry, College of Chemistry, Jilin University, Changchun, 130012, PR China

* Corresponding author. Tel.: +86 13659815698; Fax: +86 2787194560.

E-mail: rchenhku@hotmail.com

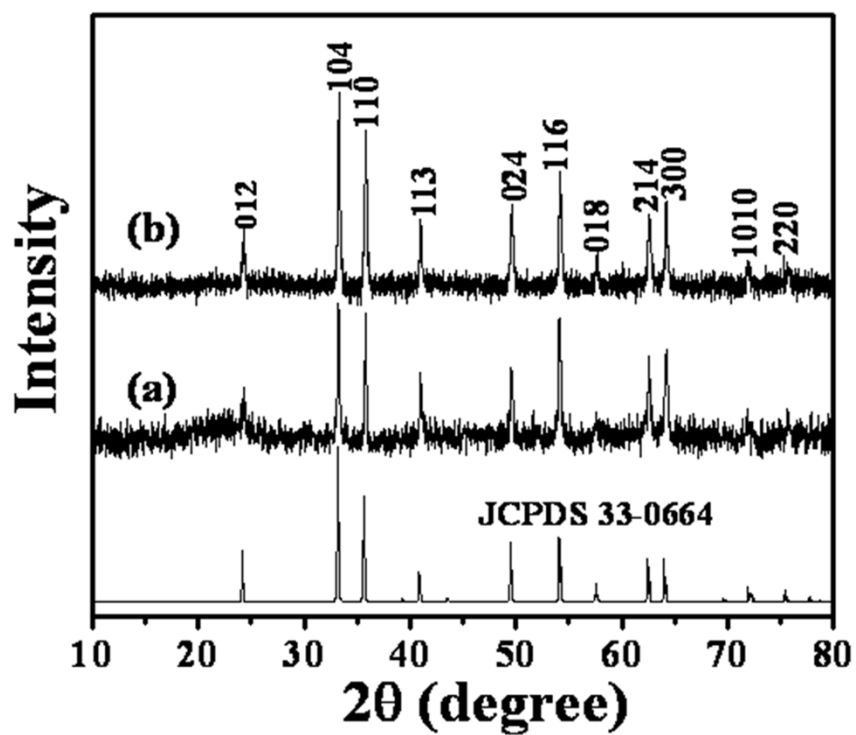


Fig. S1 XRD patterns of α - Fe_2O_3 : (a) S2; (b) S3.

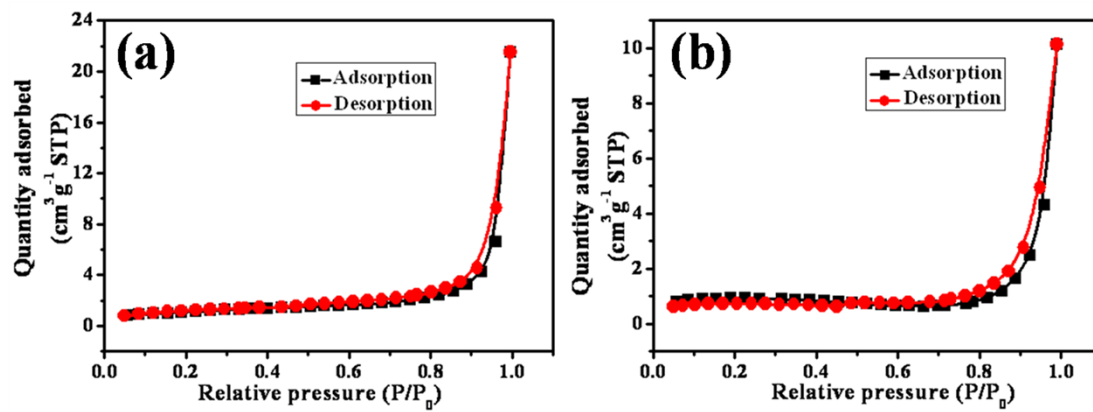


Fig. S2 Nitrogen adsorption–desorption isotherms of α -Fe₂O₃ nanostructures: (a) S2; (b) S3.

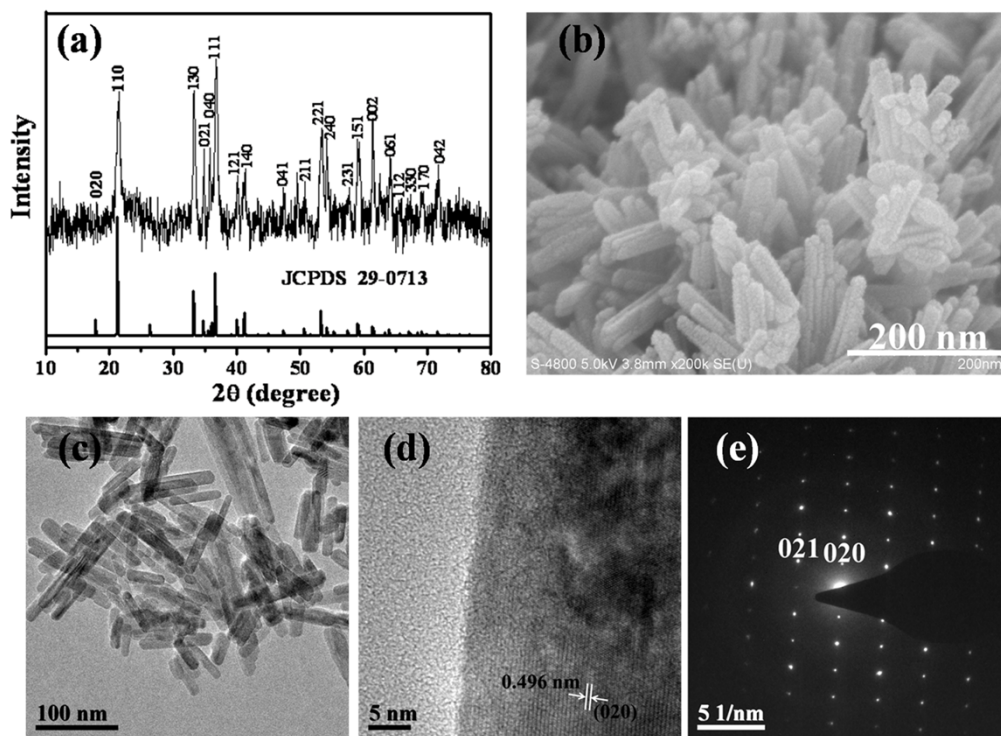


Fig. S3 (a) XRD pattern, (b) SEM image, (c) TEM image (d) HR-TEM image and (e) SAED pattern of α -FeOOH sample (S1 before calcinations).

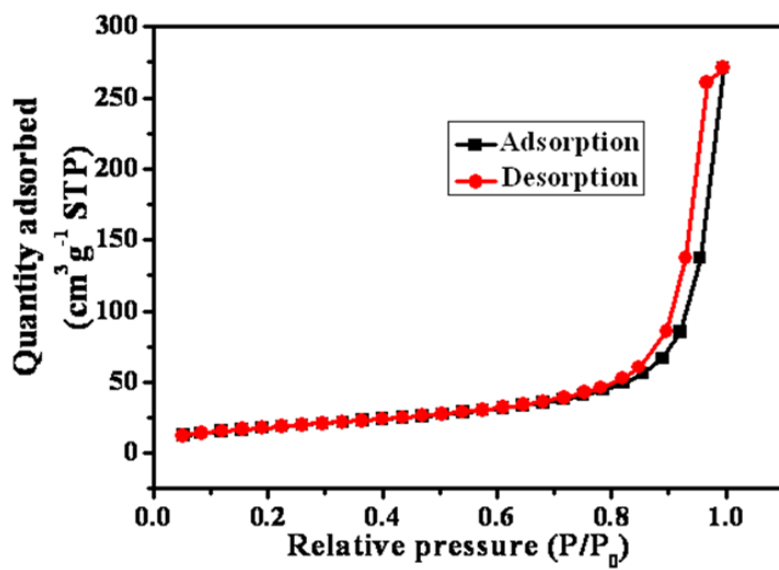


Fig. S4 Nitrogen adsorption–desorption isotherm of α -FeOOH nanorods.

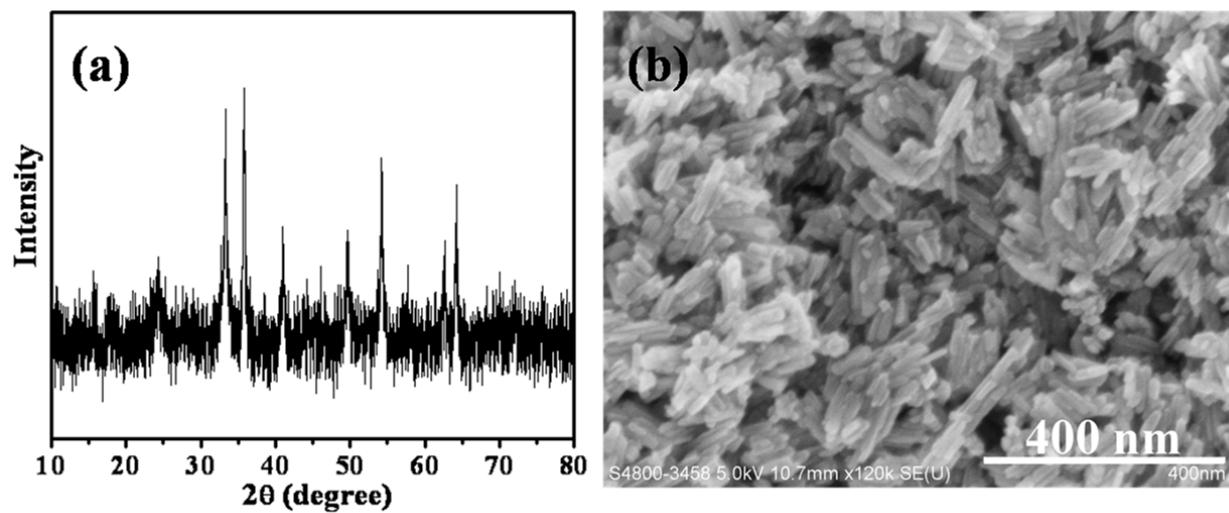


Fig. S5 XRD pattern (a) and SEM image (b) of porous α -Fe₂O₃ nanorods after four cycles.

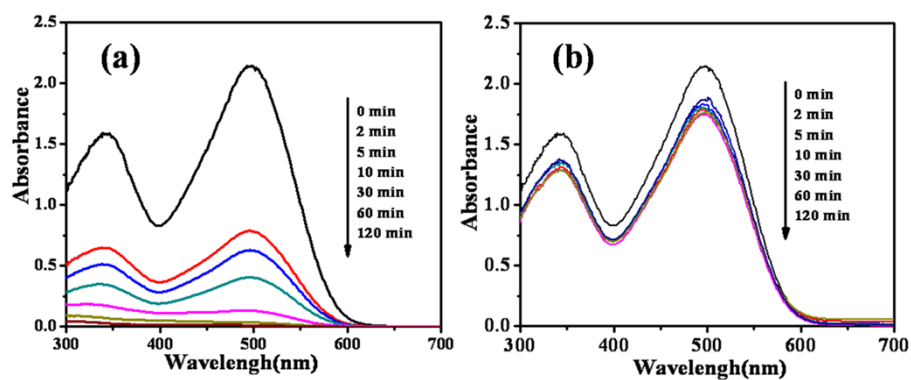


Fig. S6 Time-dependent optical absorption spectra of Congo red solution with an initial concentration of 100 mg L^{-1} in the presence of 30 mg of porous $\alpha\text{-Fe}_2\text{O}_3$ nanorod (a), and commercial $\alpha\text{-Fe}_2\text{O}_3$ (b), respectively.