

Steric effects of the polar substituents evaluated in terms of energy by means of isodesmic reactions

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Supplementary information

Table S1 DFT-calculated energies of sterically crowded cyclopropane derivatives **1** to **3**
(in a. u.)

Substituent	1 ΔE	2 ΔE	3 ΔE
H	-117.9309084	-275.2290857	-157.2579373
C \equiv CH	-194.0948265	-351.3899348	-233.4216306
CF ₃	-455.0862038	-612.3762048	-494.4105586
CN	-210.1977365	-367.4941312	-249.5251671
Si(CH ₃) ₃	-526.6697711	-683.9591712	-565.9935877
NH ₂	-173.2968432	-330.5892632	-212.6225809
NO ₂	-322.4977465	-479.7883340	-361.8227074
OH	-193.1677423	-350.4662822	-232.4971684
OCH ₃	-232.4818905	-389.7770680	-271.8074088
SO ₂ CH ₃	-705.9051243	-863.1953047	-745.2297445
F	-217.1957774	-374.4917291	-256.5221648
Cl	-577.5531000	-734.8471926	-616.8794002
C \equiv C ⁻	-193.4788957	-350.7782931	-232.8062814
CO ₂ ⁻	-306.0096953	-463.3044223	-345.3349930
NH ₃ ⁺	-173.6553086	-330.9564492	-212.9840522
N(CH ₃) ₃ ⁺	-291.6233002	-448.9093220	-231.8953720
O ⁻	-192.5670923	-349.8696834	-231.8953723

Table S2 DFT-calculated bond angles in sterically crowded cyclopropane derivatives **2** and **3**

Substituent	2	2	3	3
	$\vartheta_1 = \angle X-C1-C2$	$\vartheta_2 = \angle C1-C2-C$	$\vartheta_1 = \angle X-C1-C2$	$\vartheta_2 = \angle C1-C2-C$
H		124.5		120.4
C≡CH	124.8	126.9	121.3	121.4
CF ₃	128.0	131.2	123.5	124.2
CN	123.6	126.5	120.3	121.3
Si(CH ₃) ₃	136.1	129.7	128.7	123.3
NH ₂	124.1	128.9	124.6	122.1
NO ₂	123.0	129.4	119.9	123.7
OH	118.8	126.4	120.2	121.4
OCH ₃	119.2	126.3	120.2	121.6
SO ₂ CH ₃	127.9	131.1	124.7	124.7
F	119.6	126.1	117.8	121.0
Cl	124.3	128.9	121.2	122.9
C≡C ⁻	124.1	126.3	121.9	120.4
CO ₂ ⁻	124.9	131.1	121.9	122.3
NH ₃ ⁺	121.5	126.1	119.1	123.0
N(CH ₃) ₃ ⁺	133.1	128.9	128.5	128.7
O ⁻	123.4	124.5	122.4	117.9