

Electronic supplementary information

An unexpected epoxidation of benzil derivatives in their reaction with a germene

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General Procedures.

All manipulations were carried out under N₂ or Ar using standard Schlenk techniques with solvents freshly distilled from sodium benzophenone. ¹H and ¹³C NMR spectra were recorded in CDCl₃ on Bruker AC 200, AC 250 and Avance 300 instruments at 200.13, 250.13 and 300.13 MHz (¹H), and 50.10, 62.89 and 75.47 MHz (¹³C) respectively. Mass spectra were measured on a Hewlett-Packard 5989 A spectrometer by EI at 70 eV and on a Nermag R10-10 spectrometer by CI (NH₃). Melting points were determined on a Leitz microscope heating stage 250. IR spectra were recorded on a Perkin-Elmer 1600 FT-IR spectrometer. Elemental analyses were performed by the "Service de Microanalyse de l'Ecole de Chimie de Toulouse".

General procedure for the reaction of 1 with substituted benzils

A solution of substituted benzil (2.1 mmol) in Et₂O (10 mL) was added by syringe to the crude solution of **1** (prepared from 1.00 g of fluorogerme Mes₂Ge(F)CHR₂ (2.02 mmol) and one equivalent of *tert*-butyllithium). The initially orange reaction mixture turned light yellow by warming to room temperature. After removal of LiF by filtration, Et₂O was evaporated under vacuum and replaced by pentane ; a fractional crystallization at – 20°C afforded crystals of adducts **2**.

2b 1.11 g (76%), mp = 220 °C.

δ_{H} (300.18 MHz): 2.24 and 2.34 (2s, 2 x 3H, *p*-Me of Mes), 0.50 – 3.40 (extremely broad s ($\Delta\nu_{1/2}$ = 260 Hz) 12H, *o*-Me of Mes), [6.29 – 6.33 (m, 4H), 6.70 – 7.10 (very broad s, 4H),

6.90 (t, $^3J_{HH} = 7.5$ Hz, 2H), 7.19 – 7.44 (m, 7H), 7.62 (d, $^3J_{HH} = 7.5$ Hz, 1H), 7.68 (d, $^3J_{HH} = 7.5$ Hz, 2H)] (arom H).

δ_C (75.47 MHz): 20.89 and 21.03 (*p*-Me of Mes), 22.78 and 23.61 (*o*-Me of Mes), 60.84 (CR₂), 77.90 (R₂CCO), 91.62 (GeOCO), 113.26 and 114.69 (2d, $^2J_{CF} = 21.1$ Hz, *m*-CH of C₆H₄F), 120.08, 120.14, 125.25, 125.40, 126.95, 126.97, 128.77, 128.81, 128.92, 129.09, 130.14 and 130.25 (arom CH), 128.43, 128.65, 132.13, 132.17, 132.42, 133.30, 139.35, 139.44, 140.99, 141.51, 143.44, and 144.96 (arom C), 160.48 (d, $^2J_{CF} = 98.1$ Hz, CF), 164.75 (d, $^2J_{CF} = 98.9$ Hz, CF).

MS (60 eV, m/z) : 722 (M, 10), 627 (M – C₆H₄F, 10), 599 (M – C₆H₄F – CO, 8), 476 (Mes₂Ge=CR₂, 25), 451 (M – CR₂ – CC₆H₄F, 30), 394 (M – Mes₂GeO, 45), 311 (Mes₂Ge – 1, 38), 270 (R₂CCC₆H₄F – 1, 45), 123 (FC₆H₄CO, 100).

Anal. Calcd for C₄₅H₃₈F₂GeO₂ (722.205) C, 74.92; H, 5.31; F, 5.27% . Found : C, 75.02; H, 5.10; F, 5.30%.

2c 1.01 g (67%), mp = 158 °C.

δ_H (300.18 MHz): 2.27 and 2.38 (2s, 2 x 3H, *p*-Me of Mes), 2.70 (extremely broad s ($\Delta\nu_{1/2} = 150$ Hz) 12H, *o*-Me of Mes), 3.26 and 3.60 (2s, 2 x 3H, OMe), 2.40 – 3.10 (extremely broad s due to coalescence, *o*-Me of Mes), 6.12 (d, $^3J_{HH} = 7.6$ Hz, 2H, H of Ph), [6.01 – 6.03 (m, 1H), 6.06 (d, $^3J_{HH} = 7.6$ Hz, 1H), 6.35 – 6.39 (m, 1H), 6.55 (t, $^3J_{HH} = 7.6$ Hz, 1H), 6.78 – 6.94 (m, 6H), 7.19 – 7.42 (m, 6H), 7.52, 7.63, 7.69 and 7.78 (4d, $^3J_{HH} = 7.6$ Hz, 4 x 1H)] (arom H).

δ_C (75.47 MHz): 20.91 and 21.04 (*p*-Me of Mes), 23.29 and 23.66 (*o*-Me of Mes), 54.84 and 55.09 (OMe), 61.02 (CR₂), 78.46 (R₂CCO), 91.90 (GeOCO), 112.41, 112.87, 113.95 and 114.32 (arom CH in ortho position in relation to COMe), 119.97, 120.06, 121.38, 125.46, 125.49, 126.76, 126.85, 126.95, 127.00 and 127.13 (other arom CH), 132.59, 133.52, 134.31, 137.98, 139.24, 139.33, 141.09, 141.56, 143.79 and 145.18 (arom C), 157.56 and 158.93 (COMe).

MS (60 eV, m/z) : 746 (M, 5), 639 (M – C₆H₄OMe, 20), 491 (OGeMes₂CR₂, 17), 476 (Mes₂Ge=CR₂, 15), 418 (M – Mes₂GeO, 100), 311 (Mes₂Ge – 1, 45), 135 (MeOC₆H₄CO, 51). Anal. Calcd for C₄₇H₄₄GeO₄ (746.245) C, 75.73; H, 5.95%. Found : C, 75.92; H, 6.10%.

2d 1.12 g (74%), mp = 135 °C.

δ_H (300.18 MHz): 2.10 and 2.20 (2s, 2 x 3H, *p*-Me of Mes), the *o*-Me of the Mes groups could not be seen due to coalescence, 3.29 and 3.66 (2s, 2 x 3H, OMe), [6.04 and 6.17 (2d, $^3J_{HH} =$

8.1 Hz, 2 x 2H), 6.51 – 6.80 (m, 4H), 6.64 (d, $^3J_{HH}$ = 7.5 Hz, 2H), 7.07 (td, $^3J_{HH}$ = 7.5 Hz, $^4J_{HH}$ = 1.1 Hz, 1H), 7.08 (td, $^3J_{HH}$ = 7.5 Hz, $^4J_{HH}$ = 1.1 Hz, 1H), 7.15 – 7.27 (m, 4H), 7.30 (d, $^3J_{HH}$ = 7.5 Hz, 1H), 7.48 (d, $^3J_{HH}$ = 7.5 Hz, 1H), 7.56 (d, $^3J_{HH}$ = 7.5 Hz, 2H)] (arom H).

δ_C (75.47 MHz): 20.89 and 21.04 (*p*-Me of Mes), 23.23 and 23.61 (*o*-Me of Mes), 54.67 and 55.15 (OMe), 61.17 (CR₂), 77.97 (R₂CCO), 91.90 (GeOCO), 111.55, 113.00, 119.95, 119.98, 125.50, 125.70, 126.81, 126.88, 126.99, 128.77, 129.78, 132.43 and 132.70 (arom CH), 125.13, 128.72, 132.79, 133.59, 139.12, 139.23, 141.04, 141.50, 143.92 and 145.35 (arom C), 157.91 and 159.40 (COMe).

MS (60 eV, m/z) : 639 (M – C₆H₄OMe, 2), 491 (OGeMes₂CR₂, 5), 418 (M – Mes₂GeO, 30), 311 (Mes₂Ge – 1, 15), 271 (M – Mes₂GeCR₂, 25), 135 (MeOC₆H₄CO, 100).

Anal. Calcd for C₄₇H₄₄GeO₄ (746.245) C, 75.73; H, 5.95%. Found : C, 75.62; H, 6.01%.

2e 1.01 g (70%), mp = 176 °C.

δ_H (300.18 MHz): 1.86, 2.11, 2.19 and 2.20 (4s, 4 x 3H, *p*-Me of Mes and Me of C₆H₄Me), 2.10 – 3.10 (extremely broad s ($\Delta\nu_{1/2}$ = 150 Hz) 12H, *o*-Me of Mes), [6.15 and 6.30 (2d, $^3J_{HH}$ = 8.1 Hz, 2 x 2H), 6.50 – 6.85 (m, 4H), 6.92 (d, $^3J_{HH}$ = 8.4 Hz, 2H), 7.03 – 7.28 (m, 6H), 7.30, 7.48, 7.54 and 7.59 (4dt, $^3J_{HH}$ = 6.9 Hz, $^4J_{HH}$ = 0.9 Hz, 4 x 1H)] (arom H).

δ_C (75.47 MHz): 20.89, 21.02 and 21.25 (*p*-Me of Mes and *p*-Me of C₆H₄Me), 23.23 and 23.60 (*o*-Me of Mes), 61.19 (CR₂), 78.25 (R₂CCO), 91.93 (GeOCO), 119.88, 119.91, 125.49, 125.56, 126.73, 126.78, 126.82, 126.95, 127.33, 128.28128.32, 128.41, 128.82 and 129.08 (arom CH), 130.01, 132.82, 133.60, 133.65, 135.96, 137.62, 139.10, 139.21, 140.99, 141.51, 143.90 and 145.28 (arom C).

MS (60 eV, m/z) : 714 (M, 2), 623 (M – PhMe, 5), 475 (Mes₂Ge=CR₂ – 1, 10), 386 (M – Mes₂GeO, 35), 311 (Mes₂Ge – 1, 52), 119 (Mes, 100).

Anal. Calcd for C₄₇H₄₄GeO₂ (714.255) C, 79.12; H, 6.22%. Found : C, 79.02; H, 6.10%.