

## **Supplementary Data**

**Diagnostic fragmentations of adducts formed between  
carbanions and carbon disulfide in the gas phase. A joint  
experimental and theoretical study.**

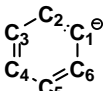
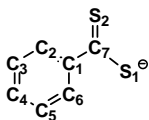
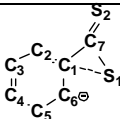
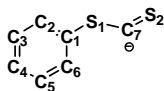
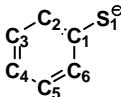
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Australia, 5005.

**Table 1**  
**Reaction Pathway for the CS<sub>2</sub> Addition to the Singlet Phenyl Anion. Anion Geometries and Energies. Carbon-Sulfur Skeleton Only.**

Level of theory used - UCCSD(T)/6-31+G(d,p)//B3LYP/6-31+G(d,p).

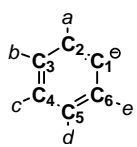
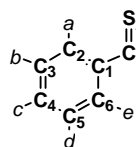
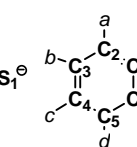
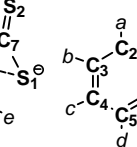
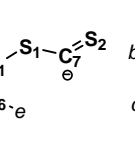
Relative energies in kJ mol<sup>-1</sup> with respect to a+CS<sub>2</sub> (0 kJ mol<sup>-1</sup>).

					
	<b>A</b>	<b>B</b>	<b>TSB/C</b>	<b>C</b>	<b>D</b>
<b>State</b>	<sup>1</sup> A <sub>1</sub>	<sup>1</sup> A	<sup>1</sup> A	<sup>1</sup> A'	<sup>1</sup> A'
<b>Symmetry</b>	C <sub>2v</sub>	C <sub>1</sub>	C <sub>1</sub>	C <sub>s</sub>	C <sub>s</sub>
<b>Energy (Hartrees)</b>	-230.85327 [-1064.1525]	-1064.26773	-1064.14790	-1064.19111	-628.60034 [-1064.17814]
<b>Energy Relative to a+CS<sub>2</sub></b>	0 [A+CS <sub>2</sub> ]	-303	+12	-101	-68 [D+CS]
<b>Gibbs Free Energy</b>	-230.8801 [-1064.19975]	-1064.30188	-1064.181691	-1064.22630	-628.63016 [-1064.22854]
<b>Free Energy Relative to a+CS<sub>2</sub></b>	0 [A+CS <sub>2</sub> ]	-268	+47	-70	-76 [D+CS]
<b>Bond Lengths (Å)</b>					
C <sub>1</sub> S <sub>1</sub>	-	-	3.019	1.791	1.750
C <sub>1</sub> C <sub>2</sub>	1.421	1.407	1.440	1.405	1.424
C <sub>1</sub> C <sub>7</sub>	-	1.512	1.491	-	-
C <sub>2</sub> C <sub>3</sub>	1.407	1.396	1.388	1.396	1.395
C <sub>3</sub> C <sub>4</sub>	1.402	1.400	1.405	1.401	1.403
C <sub>4</sub> C <sub>5</sub>	1.401	1.398	1.405	1.399	1.403
C <sub>5</sub> C <sub>6</sub>	1.407	1.397	1.387	1.397	1.394
C <sub>6</sub> C <sub>1</sub>	1.422	1.408	1.441	1.407	1.424
C <sub>7</sub> S <sub>1</sub>	-	1.706	1.937	1.823	-
C <sub>7</sub> S <sub>2</sub>	-	1.705	1.654	1.682	-
<b>Bond Angles (°)</b>					
S <sub>1</sub> C <sub>7</sub> S <sub>2</sub>	-	125.5	104.4	112.4	-
S <sub>1</sub> C <sub>1</sub> C <sub>2</sub>	-	-	-	123.0	122.4
S <sub>1</sub> C <sub>1</sub> C <sub>6</sub>	-	-	-	118.4	122.3
S <sub>1</sub> C <sub>7</sub> C <sub>1</sub>	-	117.2	75.6	-	-
S <sub>2</sub> C <sub>7</sub> C <sub>1</sub>	-	117.2	180.0	-	-
C <sub>1</sub> C <sub>2</sub> C <sub>3</sub>	125.0	121.3	120.9	120.3	122.3
C <sub>2</sub> C <sub>3</sub> C <sub>4</sub>	119.9	120.4	121.7	121.0	121.0
C <sub>3</sub> C <sub>4</sub> C <sub>5</sub>	118.2	119.0	117.5	118.9	118.0
C <sub>4</sub> C <sub>5</sub> C <sub>6</sub>	119.9	120.5	121.8	120.5	121.0
C <sub>5</sub> C <sub>6</sub> C <sub>1</sub>	124.9	121.2	120.8	120.8	122.3
C <sub>6</sub> C <sub>1</sub> C <sub>2</sub>	112.0	117.6	115.1	118.6	115.3
C <sub>7</sub> C <sub>1</sub> C <sub>2</sub>	-	121.2	120.0	-	-
C <sub>7</sub> S <sub>1</sub> C <sub>1</sub>	-	-	56.7	103.4	-
<b>Dihedral Angle (°)</b>					
C <sub>1</sub> C <sub>2</sub> C <sub>3</sub> C <sub>4</sub>	0.0	1.5	-7.2	0.0	0.0
S <sub>1</sub> C <sub>1</sub> C <sub>2</sub> C <sub>3</sub>	-	-	-118.6	180.0	180.0
C <sub>2</sub> C <sub>3</sub> C <sub>4</sub> C <sub>5</sub>	0.0	-0.8	-2.6	0.0	-0.316
S <sub>2</sub> C <sub>7</sub> S <sub>1</sub> C <sub>1</sub>	-	-	180.0	-	-

**Table 1(a)**  
**Reaction Pathway for the CS<sub>2</sub> Addition to the Singlet Phenyl Anion. Anion Geometries and Energies. C-H Only.**

Level of theory used - UCCSD(T)/6-31+G(d,p)//B3LYP/6-31+G(d,p).

Relative energies in kJ mol<sup>-1</sup> with respect to a+CS<sub>2</sub> (0 kJ mol<sup>-1</sup>).

					
	<b>A</b>	<b>B</b>	<b>TSB/C</b>	<b>C</b>	<b>D</b>
State	<sup>1</sup> A <sub>1</sub>	<sup>1</sup> A	<sup>1</sup> A	<sup>1</sup> A'	<sup>1</sup> A'
Symmetry	C <sub>2v</sub>	C <sub>1</sub>	C <sub>1</sub>	C <sub>s</sub>	C <sub>s</sub>
Energy (Hartrees)	-230.85327 [-1064.1525]	-1064.26773	-1064.14790	-1064.19111	-628.60034 [-1064.17814]
Energy Relative to a+CS <sub>2</sub>	0 [A+CS <sub>2</sub> ]	-303	+12	-101	-68 [D+CS]
Gibbs Free Energy	-230.8801 [-1064.19975]	-1064.30188	-1064.181691	-1064.22630	-628.63016 [-1064.22854]
Free Energy Relative to a+CS <sub>2</sub>	0 [A+CS <sub>2</sub> ]	-268	+47	-70	-76 [D+CS]
<b>Bond Lengths (Å)</b>					
C <sub>2</sub> a*	1.097	1.085	1.087	1.086	1.087
C <sub>3</sub> b	1.094	1.088	1.089	1.088	1.090
C <sub>4</sub> c	1.090	1.087	1.087	1.087	1.088
C <sub>5</sub> d	1.094	1.089	1.090	1.089	1.090
C <sub>6</sub> e	1.097	1.084	1.086	1.088	1.087
<b>Bond Angles (°)</b>					
aC <sub>2</sub> C <sub>1</sub>	118.9	118.1	118.4	117.5	117.8
aC <sub>2</sub> C <sub>3</sub>	116.1	120.6	120.3	122.2	119.9
bC <sub>3</sub> C <sub>4</sub>	119.6	120.0	119.7	119.7	119.8
cC <sub>4</sub> C <sub>5</sub>	120.9	120.5	121.2	120.5	121.1
dC <sub>5</sub> C <sub>6</sub>	120.4	119.5	118.5	119.4	119.8
eC <sub>6</sub> C <sub>1</sub>	119.0	118.1	118.4	119.2	117.8
<b>Dihedral Angle (°)</b>					
aC <sub>2</sub> C <sub>3</sub> C <sub>4</sub>	180.0	-178.1	-179.0	180.0	180.0
cC <sub>4</sub> C <sub>3</sub> C <sub>2</sub>	180.0	179.2	-179.6	180.0	180.0
cC <sub>4</sub> C <sub>5</sub> C <sub>6</sub>	180.0	179.3	179.6	180.0	180.0
eC <sub>6</sub> C <sub>5</sub> C <sub>4</sub>	180.0	-178.1	180.0	-179.0	-

\* Individual Hydrogens have been designated letters a - e

S<sub>1</sub>=C=S<sub>2</sub>; State <sup>1</sup>A<sub>1</sub>, Symmetry C<sub>2v</sub>, Energy (Hartrees) -833.29923, Gibbs Free Energy -833.31965, S<sub>1</sub>=C 1.563Å, C=S<sub>2</sub> 1.563Å, S<sub>1</sub>=C=S<sub>2</sub> 179.7°

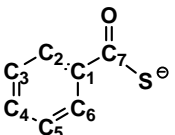
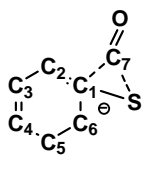
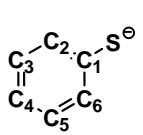
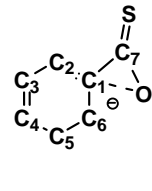
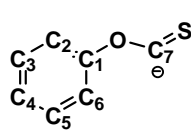
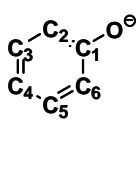
C=S; State <sup>1</sup>SG, Symmetry C<sub>sv</sub>, Energy (Hartrees) -435.5778, Gibbs Free Energy -435.59838, C=S 1.544Å

**Table 2**

**Reaction Pathway for the Collision Induced Dissociation of the Singlet  
PhCOS<sup>-</sup> Anion. Anion Geometries and Energies.  
Carbon-Sulfur-Oxygen Skeleton Only.**

Level of theory used - UCCSD(T)/6-31+G(d,p)//B3LYP/6-31+G(d,p).

Relative energies in kJ mol<sup>-1</sup> with respect to A (0 kJ mol<sup>-1</sup>).

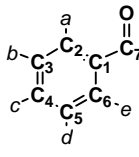
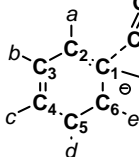
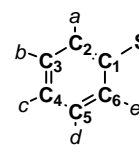
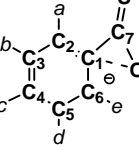
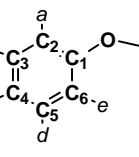
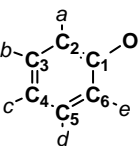
						
	A	TS A/B	B	TS A/C	C	D
<b>State</b>	<sup>1</sup> A	<sup>1</sup> A	<sup>1</sup> A'	<sup>1</sup> A	<sup>1</sup> A'	<sup>1</sup> A'
<b>Symmetry</b>	C <sub>1</sub>	C <sub>1</sub>	C <sub>s</sub>	C <sub>1</sub>	C <sub>s</sub>	C <sub>s</sub>
<b>Energy (Hartrees)</b>	-741.66644	-741.55247	-628.60034 [-741.64111]	-741.55040	-741.57485	-305.97915 [-741.55695]
<b>Energy Relative to A</b>	0	+299	+67 [B+CO]	+305	+241	+288 [E+CS]
<b>Gibbs Free Energy</b>	-741.69969	-741.58543	-628.63016 [-741.69007]	-741.58300	-741.60937	-306.00803 [-741.60641]
<b>Free Energy Relative to A</b>	0	+300	+25 [B+CO]	+306	+237	+245 [E+CS]
<b>Bond Lengths (Å)</b>						
C <sub>1</sub> S	-	1.865	1.750	-	-	-
C <sub>1</sub> C <sub>2</sub>	1.403	1.427	1.424	1.443	1.407	1.448
C <sub>1</sub> C <sub>7</sub>	1.537	1.686	-	1.537	-	-
C <sub>2</sub> C <sub>3</sub>	1.398	1.401	1.395	1.389	1.395	1.392
C <sub>3</sub> C <sub>4</sub>	1.400	1.394	1.403	1.406	1.401	1.407
C <sub>4</sub> C <sub>5</sub>	1.399	1.415	1.403	1.406	1.400	1.407
C <sub>5</sub> C <sub>6</sub>	1.399	1.383	1.394	1.389	1.396	1.392
C <sub>6</sub> C <sub>1</sub>	1.405	1.437	1.424	1.442	1.408	1.449
C <sub>7</sub> S	1.738	2.033	-	1.686	1.682	-
C <sub>7</sub> O	1.244	1.187	-	1.342	1.434	-
C <sub>1</sub> O	-	-	-	1.633	1.362	1.275
<b>Bond Angles (°)</b>						
SC <sub>7</sub> O	125.7	127.0	-	111.3	109.0	-
SC <sub>1</sub> C <sub>2</sub>	-	118.5	122.4	-	-	-
SC <sub>1</sub> C <sub>6</sub>	-	119.8	122.3	-	-	-
SC <sub>7</sub> C <sub>1</sub>	118.4	-	-	180.0	-	-
OC <sub>7</sub> C <sub>1</sub>	115.9	143.1	-	68.7	-	-
C <sub>1</sub> C <sub>2</sub> C <sub>3</sub>	120.9	121.9	122.3	120.6	120.5	122.6
C <sub>2</sub> C <sub>3</sub> C <sub>4</sub>	120.4	121.2	121.0	122.2	120.6	121.7
C <sub>3</sub> C <sub>4</sub> C <sub>5</sub>	119.3	118.0	118.0	117.4	118.7	117.5
C <sub>4</sub> C <sub>5</sub> C <sub>6</sub>	120.2	121.3	121.0	122.2	121.5	121.7
C <sub>5</sub> C <sub>6</sub> C <sub>1</sub>	121.1	122.0	122.3	120.6	119.6	122.6
C <sub>6</sub> C <sub>1</sub> C <sub>2</sub>	118.2	115.3	115.3	116.0	119.1	113.9
C <sub>7</sub> C <sub>1</sub> C <sub>2</sub>	123.4	-	-	120.9	-	-
C <sub>7</sub> SC <sub>1</sub>	-	-	-	-	-	-
OC <sub>1</sub> C <sub>2</sub>	-	-	-	115.4	115.9	123.1
C <sub>7</sub> OC <sub>1</sub>	-	-	-	61.3	118.5	-
<b>Dihedral Angle (°)</b>						
C <sub>1</sub> C <sub>2</sub> C <sub>3</sub> C <sub>4</sub>	0.0	2.8	0.0	5.2	0.0	0.0
C <sub>2</sub> C <sub>3</sub> C <sub>4</sub> C <sub>5</sub>	0.0	0.8	-0.316	1.8	0.0	0.0
OC <sub>1</sub> C <sub>2</sub> C <sub>3</sub>	-	-	-	127.8	180.0	-179.0
SC <sub>7</sub> OC <sub>1</sub>	-	-	-	180.0	-179.0	-

**Table 2(a)**

**Reaction Pathway for the Collision Induced Dissociation of the Singlet  
PhCOS<sup>-</sup> Anion. Anion Geometries and Energies.  
C-H Only.**

Level of theory used - UCCSD(T)/6-31+G(d,p)//B3LYP/6-31+G(d,p).

Relative energies in kJ mol<sup>-1</sup> with respect to A (0 kJ mol<sup>-1</sup>).

						
	A	TS A/B	B	TS A/D	D	E
State	<sup>1</sup> A	<sup>1</sup> A	<sup>1</sup> A'	<sup>1</sup> A	<sup>1</sup> A'	<sup>1</sup> A'
Symmetry	C <sub>1</sub>	C <sub>1</sub>	C <sub>s</sub>	C <sub>1</sub>	C <sub>s</sub>	C <sub>s</sub>
Energy (Hartrees)	-741.66644	-741.55247	-628.60034 [-741.64111]	-741.55040	-741.57485	-305.97915 [-741.55695]
Energy Relative to A	0	+299	+67 [B+CO]	+305	+241	+288 [E+CS]
Gibbs Free Energy	-741.69969	-741.58543	-628.63016 [-741.69007]	-741.58300	-741.60937	-306.00803 [-741.60641]
Free Energy Relative to A	0	+300	+25 [B+CO]	+306	+237	+245 [E+CS]
<b>Bond Lengths (Å)</b>						
C <sub>2</sub> a*	1.086	1.087	1.087	1.085	1.085	1.089
C <sub>3</sub> b	1.088	1.090	1.090	1.091	1.089	1.092
C <sub>4</sub> c	1.088	1.088	1.088	1.086	1.087	1.087
C <sub>5</sub> d	1.089	1.090	1.090	1.090	1.089	1.092
C <sub>6</sub> e	1.085	1.086	1.087	1.086	1.082	1.089
<b>Bond Angles (°)</b>						
aC <sub>2</sub> C <sub>1</sub>	117.6	118.1	117.8	118.7	118.1	116.9
aC <sub>2</sub> C <sub>3</sub>	121.5	119.9	119.9	120.6	121.4	120.5
bC <sub>3</sub> C <sub>4</sub>	119.9	119.9	119.8	119.5	120.0	119.2
cC <sub>4</sub> C <sub>5</sub>	120.3	120.8	121.1	121.3	120.7	121.3
dC <sub>5</sub> C <sub>6</sub>	119.9	119.0	119.8	118.3	118.9	119.0
eC <sub>6</sub> C <sub>1</sub>	117.5	117.9	117.8	118.7	117.9	116.9
<b>Dihedral Angle (°)</b>						
aC <sub>2</sub> C <sub>3</sub> C <sub>4</sub>	-179.0	-178.0	180.0	-178.4	180.0	180.0
cC <sub>4</sub> C <sub>3</sub> C <sub>2</sub>	180.0	178.3	180.0	179.1	180.0	-179.0
eC <sub>6</sub> C <sub>5</sub> C <sub>4</sub>	180.0	-178.9	180.0	-179.0	180.0	180.0
eC <sub>6</sub> C <sub>5</sub> C <sub>4</sub>	180.0	178.5	-	178.3	-179.0	180.0

\* Individual Hydrogens have been designated letters a - e

C=O; State <sup>1</sup>SG, Symmetry C<sub>∞v</sub>, Energy (Hartrees) -113.04077, Gibbs Free Energy -113.05991, C=O 1.138Å

C=S; State <sup>1</sup>SG, Symmetry C<sub>∞v</sub>, Energy (Hartrees) -435.57780, Gibbs Free Energy -435.59838, C=S 1.544Å

**Table 3**

**Possible Reaction Pathway for the CS<sub>2</sub> Addition to the Singlet Benzyl Anion. Anion Geometries and Energies. Carbon and Sulfur Skeleton Only.**

Level of theory used - UCCSD(T)/6-31+G(d,p)//B3LYP/6-31+G(d,p).

Relative energies in kJ mol<sup>-1</sup> with respect to A (0 kJ mol<sup>-1</sup>).

	A	B
State	<sup>1</sup> A	<sup>1</sup> A'
Symmetry	C <sub>1</sub>	C <sub>1</sub>
Energy (Hartrees)	-1103.43726	-704.55370 [-1103.37578]
Energy Relative to A	0	+161 [B+H <sub>2</sub> S]
Gibbs Free Energy	-1103.47418	-704.58712 [-1103.42847]
Free Energy Relative to A	0	+120 [ B+H <sub>2</sub> S ]
<b>Bond Lengths (Å)</b>		
C <sub>1</sub> S <sub>1</sub>	1.699	1.656
C <sub>1</sub> S <sub>2</sub>	1.700	-
C <sub>1</sub> C <sub>2</sub>	1.549	1.240
C <sub>2</sub> C <sub>3</sub>	1.516	1.409
C <sub>3</sub> C <sub>4</sub>	1.404	1.425
C <sub>4</sub> C <sub>5</sub>	1.397	1.393
C <sub>5</sub> C <sub>6</sub>	1.399	1.402
C <sub>6</sub> C <sub>7</sub>	1.399	1.403
C <sub>7</sub> C <sub>8</sub>	1.398	1.393
C <sub>8</sub> C <sub>3</sub>	1.404	1.425
<b>Bond Angles (°)</b>		
S <sub>1</sub> C <sub>1</sub> S <sub>2</sub>	127.3	-
S <sub>1</sub> C <sub>1</sub> C <sub>2</sub>	116.4	180.0
S <sub>2</sub> C <sub>1</sub> C <sub>2</sub>	116.3	-
C <sub>1</sub> C <sub>2</sub> C <sub>3</sub>	112.9	180.0
C <sub>2</sub> C <sub>3</sub> C <sub>4</sub>	121.0	121.9
C <sub>3</sub> C <sub>4</sub> C <sub>5</sub>	121.0	121.7
C <sub>4</sub> C <sub>5</sub> C <sub>6</sub>	120.3	121.0
C <sub>5</sub> C <sub>6</sub> C <sub>7</sub>	119.3	118.4
C <sub>6</sub> C <sub>7</sub> C <sub>8</sub>	120.3	121.0
C <sub>7</sub> C <sub>8</sub> C <sub>3</sub>	121.0	121.7
C <sub>8</sub> C <sub>3</sub> C <sub>2</sub>	120.9	121.9
C <sub>8</sub> C <sub>3</sub> C <sub>4</sub>	118.2	116.2
<b>Dihedral Angle (°)</b>		
C <sub>1</sub> C <sub>2</sub> C <sub>3</sub> C <sub>4</sub>	89.7	0.0
S <sub>1</sub> C <sub>1</sub> C <sub>2</sub> C <sub>3</sub>	-89.6	180.0
C <sub>1</sub> C <sub>2</sub> C <sub>3</sub> C <sub>8</sub>	-89.8	0.0
C <sub>2</sub> C <sub>3</sub> C <sub>4</sub> C <sub>5</sub>	-179.7	-179.0

**Table 3(a)**

**Possible Reaction Pathway for the CS<sub>2</sub> Addition to the Singlet Benzyl Anion. Anion Geometries and Energies.**

**C-H Only.**

Level of theory used - UCCSD(T)/6-31+G(d,p)//B3LYP/6-31+G(d,p).

Relative energies in kJ mol<sup>-1</sup> with respect to A (0 kJ mol<sup>-1</sup>).

	A	B
<b>State</b>	<sup>1</sup> A	<sup>1</sup> A'
<b>Symmetry</b>	C <sub>1</sub>	C <sub>1</sub>
<b>Energy (Hartrees)</b>	-1103.43726	-704.55370 [-1103.37578]
<b>Energy Relative to A</b>	0	+161 [ <b>B</b> +H <sub>2</sub> S ]
<b>Gibbs Free Energy</b>	-1103.47418	-704.58712 [-1103.42847]
<b>Free Energy Relative to A</b>	0	+120 [ <b>B</b> +H <sub>2</sub> S ]
<b>Bond Lengths (Å)</b>		
C <sub>2</sub> a	1.095	-
C <sub>2</sub> b	1.094	-
C <sub>4</sub> a	-	1.087
C <sub>4</sub> c	1.087	-
C <sub>8</sub> b	-	1.086
C <sub>8</sub> d	1.086	-
<b>Bond Angles (°)</b>		
aC <sub>2</sub> b	108.9	-
aC <sub>2</sub> C <sub>1</sub>	108.4	-
aC <sub>2</sub> C <sub>3</sub>	109.0	-
aC <sub>4</sub> C <sub>3</sub>	-	118.1
bC <sub>2</sub> C <sub>1</sub>	108.5	-
bC <sub>2</sub> C <sub>3</sub>	109.0	-
bC <sub>8</sub> C <sub>3</sub>	-	118.2
cC <sub>4</sub> C <sub>3</sub>	118.7	-
dC <sub>8</sub> C <sub>3</sub>	118.7	-
<b>Dihedral Angle (°)</b>		
aC <sub>4</sub> C <sub>3</sub> C <sub>8</sub>	-	180.0
bC <sub>2</sub> C <sub>3</sub> C <sub>8</sub>	30.9	-
dC <sub>8</sub> C <sub>3</sub> C <sub>4</sub>	-178.4	-

H<sub>1</sub>-S-H<sub>2</sub>;

State <sup>1</sup>A', Symmetry C<sub>s</sub>, Energy (Hartrees) -398.82208, Gibbs Free Energy -398.84135, H<sub>1</sub>-S 1.348 Å, H<sub>2</sub>-S 1.348 Å, Angle H<sub>1</sub>-S-H<sub>2</sub> 92.7°

**Table 4**

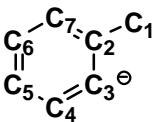
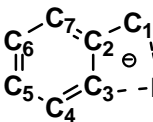
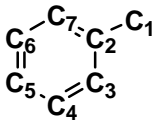
**Reaction Pathway for the Interconversion of the Singlet  
 Anion  $\sigma\text{-CH}_3(\text{C}_6\text{H}_4)^-$  to Singlet  $\text{PhCH}_2^-$ .**

**Anion Geometries and Energies.**

**Carbon Skeleton Only.**

Level of theory used - UCCSD(T)/6-31+G(d,p)//B3LYP/6-31+G(d,p).

Relative energies in  $\text{kJ mol}^{-1}$  with respect to A ( $0 \text{ kJ mol}^{-1}$ ).

			
	A	TS A/B	B
State	$^1A'$	$^1A$	$^1A$
Symmetry	$C_s$	$C_1$	$C_1$
Energy (Hartrees)	-270.03120	-269.96265	-270.05125
Energy Relative to A	0	+180	-53
Gibbs Free Energy	-270.06021	-269.99218	-270.08051
Free Energy Relative to A	0	+179	-53
<b>Bond Lengths (Å)</b>			
$C_1C_2$	1.526	1.509	1.395
$C_2C_3$	1.422	1.423	1.451
$C_3C_4$	1.420	1.404	1.386
$C_4C_5$	1.405	1.412	1.412
$C_5C_6$	1.401	1.399	1.413
$C_6C_7$	1.401	1.408	1.386
$C_7C_2$	1.409	1.403	1.450
<b>Bond Angles (°)</b>			
$C_1C_2C_3$	119.3	105.3	123.2
$C_2C_3C_4$	113.3	118.5	122.8
$C_3C_4C_5$	124.9	120.6	122.1
$C_4C_5C_6$	119.4	120.0	116.7
$C_5C_6C_7$	118.5	120.4	122.1
$C_6C_7C_2$	120.8	119.3	122.8
$C_7C_2C_1$	117.5	133.5	123.2
$C_7C_2C_3$	123.2	121.1	113.6
<b>Dihedral Angle (°)</b>			
$C_1C_2C_3C_4$	-179.0	178.7	-179.0
$C_1C_2C_7C_6$	180.0	-175.4	180.0
$C_2C_3C_4C_5$	0.0	-3.0	0.0
$C_2C_7C_6C_5$	0.0	-1.7	0.0



## Table 4(a)

### Reaction Pathway for the Interconversion of the Singlet

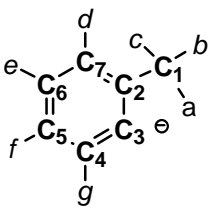
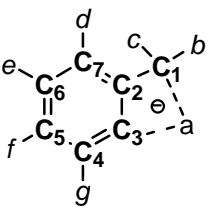
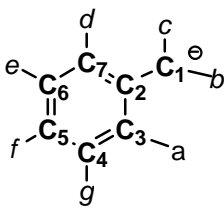
### Anion $\sigma\text{-CH}_3(\text{C}_6\text{H}_4)^-$ to Singlet $\text{PhCH}_2^-$ .

### Anion Geometries and Energies.

### C-H Only.

Level of theory used - UCCSD(T)/6-31+G(d,p)//B3LYP/6-31+G(d,p).

Relative energies in  $\text{kJ mol}^{-1}$  with respect to A ( $0 \text{ kJ mol}^{-1}$ ).

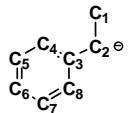
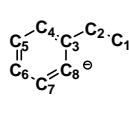
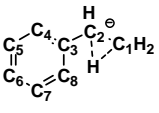
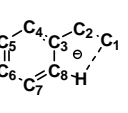
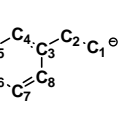
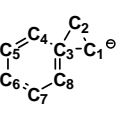
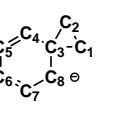
			
	A	TS A/B	B
State	$^1A'$	$^1A$	$^1A$
Symmetry	$C_s$	$C_1$	$C_1$
Energy (Hartrees)	-270.03120	-269.96265	-270.05125
Energy Relative to A	0	+180	-53
Gibbs Free Energy	-270.06021	-269.99218	-270.08051
Free Energy Relative to A	0	+179	-53
<b>Bond Lengths (Å)</b>			
$C_1a$	1.102	1.537	-
$C_1b$	1.102	1.105	1.087
$C_1c$	1.093	1.104	1.087
$C_3a$	-	1.421	1.090
$C_4g$	1.097	1.093	1.091
$C_5f$	1.094	1.092	1.087
$C_6e$	1.090	1.091	1.092
$C_7d$	1.095	1.091	1.090
<b>Bond Angles (°)</b>			
$aC_1C_2$	112.2	72.2	-
$aC_1b$	106.7	132.2	-
$aC_1c$	108.3	110.4	-
$bC_1C_2$	112.2	115.5	121.3
$bC_1c$	108.3	108.6	117.5
$cC_1C_2$	108.9	113.0	121.2
$dC_7C_6$	118.8	119.5	119.2
$eC_6C_5$	121.0	119.7	119.3
$fC_5C_4$	120.8	120.4	121.6
$gC_4C_3$	118.8	120.9	118.6
<b>Dihedral Angle (°)</b>			
$aC_1C_2C_3$	-119.9	-3.2	-
$bC_1C_2C_3$	119.9	-132.2	-179.0

\* Individual Hydrogens have been designated letters a - g

**Table 5**  
**Reaction Pathway for the Interconversion of the Singlet Anion**  
**PhCH<sub>2</sub>CH<sub>2</sub><sup>-</sup>. Anion Geometries and Energies.**  
**Carbon Skeleton Only.**

Level of theory used - UCCSD(T)/6-31+G(d,p)//B3LYP/6-31+G(d,p).

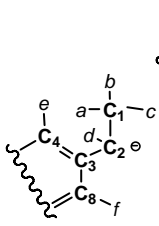
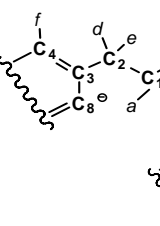
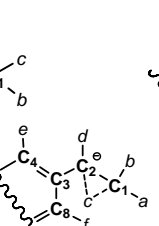
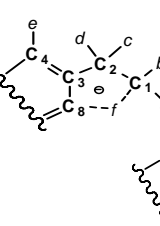
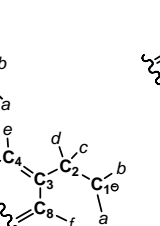
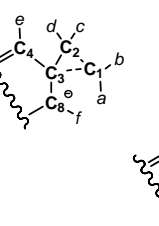
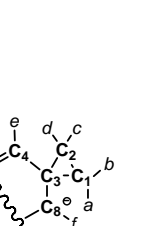
Relative energies in kJ mol<sup>-1</sup> with respect to A (0 kJ mol<sup>-1</sup>).

							
	D	C	TS A/D	TS A/C	A	TS A/B	B
State	<sup>1</sup> A	<sup>1</sup> A	<sup>1</sup> A	<sup>1</sup> A	<sup>1</sup> A	<sup>1</sup> A	<sup>1</sup> A
Symmetry	C <sub>1</sub>	C <sub>1</sub>	C <sub>1</sub>	C <sub>1</sub>	C <sub>1</sub>	C <sub>1</sub>	C <sub>1</sub>
Energy (Hartrees)	-309.22336	-309.20572	-309.12837	-309.16471	-309.18370	-309.17008	-309.17799
Energy Relative to A	-104	-58	+144	+50	0	+36	+13
Gibbs Free Energy	-309.25480	-309.23759	-309.15998	-309.19645	-309.21513	-309.20086	-309.20928
Free Energy Relative to A	-104	-59	+145	+49	0	+38	+15
<b>Bond Lengths (Å)</b>							
C <sub>1</sub> C <sub>2</sub>	1.504	1.539	1.564	1.534	1.514	1.503	1.519
C <sub>2</sub> C <sub>3</sub>	1.392	1.525	1.430	1.555	1.529	1.475	1.517
C <sub>1</sub> C <sub>3</sub>	-	-	-	-	-	1.902	1.555
C <sub>3</sub> C <sub>4</sub>	1.453	1.408	1.433	1.399	1.404	1.458	1.499
C <sub>4</sub> C <sub>5</sub>	1.390	1.403	1.386	1.403	1.399	1.384	1.379
C <sub>5</sub> C <sub>6</sub>	1.407	1.398	1.414	1.402	1.402	1.415	1.420
C <sub>6</sub> C <sub>7</sub>	1.417	1.408	1.402	1.406	1.399	1.415	1.420
C <sub>7</sub> C <sub>8</sub>	1.383	1.416	1.396	1.402	1.402	1.383	1.378
C <sub>8</sub> C <sub>3</sub>	1.452	1.424	1.435	1.402	1.402	1.458	1.499
<b>Bond Angles (°)</b>							
C <sub>1</sub> C <sub>2</sub> C <sub>3</sub>	123.3	112.6	123.7	109.3	116.2	79.4	61.6
C <sub>2</sub> C <sub>3</sub> C <sub>4</sub>	123.4	118.2	121.8	124.4	121.0	121.7	119.2
C <sub>3</sub> C <sub>4</sub> C <sub>5</sub>	122.3	120.6	122.4	119.8	121.6	121.0	121.3
C <sub>4</sub> C <sub>5</sub> C <sub>6</sub>	122.3	118.4	121.1	119.6	119.9	122.0	123.1
C <sub>5</sub> C <sub>6</sub> C <sub>7</sub>	116.9	119.6	117.7	119.6	119.1	117.1	116.6
C <sub>6</sub> C <sub>7</sub> C <sub>8</sub>	121.9	124.7	121.9	121.6	120.5	122.0	123.1
C <sub>7</sub> C <sub>8</sub> C <sub>3</sub>	122.8	113.4	121.4	117.8	121.0	121.0	121.3
C <sub>8</sub> C <sub>3</sub> C <sub>2</sub>	122.8	118.5	122.7	114.0	121.2	121.7	119.2
C <sub>8</sub> C <sub>3</sub> C <sub>4</sub>	113.8	123.3	115.5	121.6	117.8	114.2	112.3
<b>Dihedral Angle (°)</b>							
C <sub>1</sub> C <sub>2</sub> C <sub>3</sub> C <sub>4</sub>	0.0	-105.2	-173.7	-179.0	177.4	-99.5	-108.0
C <sub>1</sub> C <sub>2</sub> C <sub>3</sub> C <sub>8</sub>	180.0	73.9	7.6	0.0	-3.364	99.5	108.1
C <sub>2</sub> C <sub>3</sub> C <sub>4</sub> C <sub>5</sub>	-179.0	178.8	-177.3	180.0	-179.2	-179.0	-162.2
C <sub>2</sub> C <sub>3</sub> C <sub>8</sub> C <sub>7</sub>	180.0	-178.6	176.6	-179.0	-179.6	179.7	162.2

**Table 5(a)**  
**Reaction Pathway for the Interconversion of the Singlet Anion**  
**PhCH<sup>-</sup>CH<sub>3</sub>. Anion Geometries and Energies.**  
**C-H Only.**

Level of theory used - UCCSD(T)/6-31+G(d,p)//B3LYP/6-31+G(d,p).

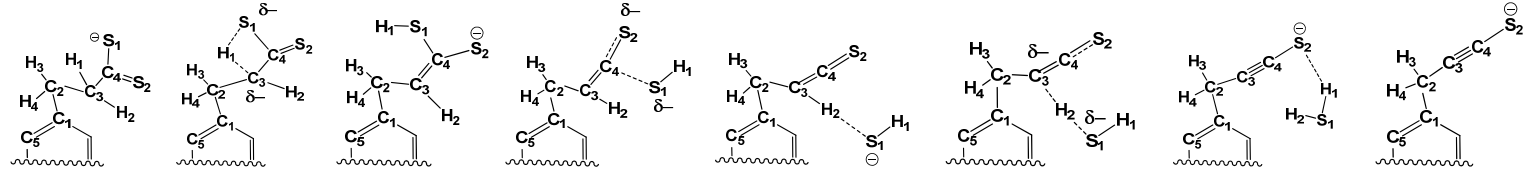
Relative energies in kJ mol<sup>-1</sup> with respect to A (0 kJ mol<sup>-1</sup>).

							
	<b>D</b>	<b>C</b>	<b>TS A/D</b>	<b>TS A/C</b>	<b>A</b>	<b>TS A/B</b>	<b>B</b>
State	<sup>1</sup> A	<sup>1</sup> A	<sup>1</sup> A	<sup>1</sup> A	<sup>1</sup> A	<sup>1</sup> A	<sup>1</sup> A
Symmetry	C <sub>1</sub>	C <sub>1</sub>	C <sub>1</sub>	C <sub>1</sub>	C <sub>1</sub>	C <sub>1</sub>	C <sub>1</sub>
Energy (Hartrees)	-309.22336	-309.20572	-309.12837	-309.16471	-309.18370	-309.17008	-309.17799
Energy Relative to A	-104	-58	+144	+50	0	+36	+13
Gibbs Free Energy	-309.25480	-309.23759	-309.15998	-309.19645	-309.21513	-309.20086	-309.20928
Free Energy Relative to A	-104	-59	+145	+49	0	+38	+15
<b>Bond Lengths (Å)</b>							
C <sub>1</sub> a	1.110	1.098	1.089	1.100	1.100	1.086	1.092
C <sub>1</sub> b	1.109	1.100	1.095	1.099	1.103	1.085	1.092
C <sub>1</sub> c	1.098	1.095	1.401	-	-	-	-
C <sub>1</sub> f	-	-	-	1.528	-	-	-
C <sub>2</sub> c	-	-	1.220	1.104	1.104	1.098	1.091
C <sub>2</sub> d	1.089	1.103	1.092	1.104	1.126	1.098	1.091
C <sub>2</sub> e	-	1.097	-	-	-	-	-
C <sub>4</sub> e	1.088	-	1.090	1.094	1.091	1.090	1.090
C <sub>4</sub> f	-	1.095	-	-	-	-	-
C <sub>8</sub> f	1.090	-	1.085	1.361	1.088	1.090	1.090
<b>Bond Angles (°)</b>							
aC <sub>1</sub> b	105.0	108.1	115.2	109.1	109.2	118.3	114.8
aC <sub>1</sub> c	106.4	107.9	112.5	-	-	-	-
aC <sub>1</sub> C <sub>2</sub>	113.8	110.9	112.1	113.1	111.5	120.1	118.3
bC <sub>1</sub> c	106.4	108.9	131.5	-	-	-	-
bC <sub>1</sub> C <sub>2</sub>	113.9	111.9	116.7	113.1	112.9	120.1	118.3
cC <sub>1</sub> C <sub>2</sub>	110.8	108.9	48.2	-	-	-	-
cC <sub>2</sub> C <sub>1</sub>	-	-	58.9	113.1	-	120.1	-
cC <sub>2</sub> C <sub>3</sub>	-	-	112.2	108.0	-	113.9	-
cC <sub>2</sub> d	-	-	115.4	105.1	103.4	109.8	113.5
dC <sub>2</sub> C <sub>1</sub>	117.8	109.2	116.6	113.2	117.4	116.8	118.4
dC <sub>2</sub> C <sub>3</sub>	118.8	110.8	116.1	107.9	103.4	115.7	117.8
eC <sub>2</sub> C <sub>3</sub>	-	108.3	-	-	-	-	-
eC <sub>4</sub> C <sub>3</sub>	118.8	-	118.3	120.5	118.8	118.7	118.1
fC <sub>8</sub> C <sub>3</sub>	117.9	-	119.0	95.3	116.6	118.7	118.1
fC <sub>4</sub> C <sub>3</sub>	-	120.4	-	-	-	-	-
<b>Dihedral Angle (°)</b>							
aC <sub>1</sub> C <sub>2</sub> C <sub>3</sub>	60.2	63.0	161.6	-117.7	-166.0	-97.0	106.9
dC <sub>2</sub> C <sub>3</sub> C <sub>8</sub>	0.0	-163.5	165.4	123.5	126.7	-145.7	-1.0
eC <sub>4</sub> C <sub>3</sub> C <sub>8</sub>	180.0	-	-178.6	180.0	-179.0	166.8	166.9
fC <sub>8</sub> C <sub>3</sub> C <sub>4</sub>	-179.0	-	178.6	180.0	-178.0	-166.8	-166.9

**Table 6**

**Reaction Pathway for the Dissociation of the Singlet PhCH<sub>2</sub>CH<sub>2</sub>CS<sub>2</sub><sup>-</sup> Anion. Energies and Selected Geometries of Anion Minima and Transition States.**

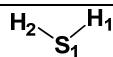
Level of theory used – UCCSD(T)/6-31+G(d,p)//B3LYP/6-31+G(d,p). Relative energies in kJ mol<sup>-1</sup> with respect to A (0 kJ mol<sup>-1</sup>)



	A	TS A/B	B	TS B/C	C	TS C/D	D	E
State	<sup>1</sup> A	<sup>1</sup> A	<sup>1</sup> A	<sup>1</sup> A	<sup>1</sup> A	<sup>1</sup> A	<sup>1</sup> A	<sup>1</sup> A
Symmetry	C <sub>1</sub>	C <sub>1</sub>	C <sub>1</sub>	C <sub>1</sub>	C <sub>1</sub>	C <sub>1</sub>	C <sub>1</sub>	C <sub>1</sub>
Energy (Hartree)	-1142.61346	-1142.52437	-1142.58933	-1142.54184	-1142.55548	-1142.54932	-1142.55202	-743.71570 [-1142.53778]
Energy Relative to A	0	234	63	188	152	168	161	199 [E+F]
Gibbs Free Energy (Hartree)	-1142.64983	-1142.56038	-1142.62546	-1142.58123	-1142.59489	-1142.58866	-1142.59341	-1142.591365
Gibbs Free Energy Relative to A	0	235	64	180	144	161	148	154 [E+F]
<b>Bond Length (Å)</b>								
C <sub>1</sub> C <sub>2</sub>	1.511	1.519	1.520	1.526	1.521	1.524	1.532	1.532
C <sub>2</sub> C <sub>3</sub>	1.550	1.533	1.508	1.516	1.523	1.488	1.459	1.457
C <sub>3</sub> C <sub>4</sub>	1.535	1.452	1.363	1.317	1.297	1.258	1.230	1.233
C <sub>4</sub> S <sub>1</sub>	1.702	1.809	1.828	3.382	—	—	—	—
C <sub>4</sub> S <sub>2</sub>	1.703	1.679	1.739	1.586	1.596	1.639	1.678	1.674
C <sub>3</sub> H <sub>1</sub>	1.095	1.605	—	—	—	—	—	—
C <sub>3</sub> H <sub>2</sub>	1.095	1.097	1.090	1.091	1.137	1.626	—	—
S <sub>1</sub> H <sub>1</sub>	2.787	1.578	1.359	1.354	1.353	1.348	1.347	—
S <sub>1</sub> H <sub>2</sub>	—	—	—	2.643	2.246	1.498	1.397	—
C <sub>2</sub> H <sub>3</sub>	1.096	1.105	1.101	1.100	1.097	1.102	1.104	—
C <sub>2</sub> H <sub>4</sub>	1.096	1.098	1.099	1.098	1.095	1.100	1.104	—

**Table 6 Continued.**

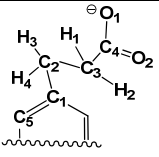
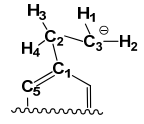
<b>Bond Angle (°)</b>								
<b>C<sub>1</sub>C<sub>2</sub>C<sub>3</sub></b>	113.0	113.9	114.2	115.3	113.8	113.6	116.4	116.4
<b>C<sub>2</sub>C<sub>3</sub>C<sub>4</sub></b>	111.1	119.4	123.2	122.1	126.5	146.6	178.3	179.3
<b>C<sub>3</sub>C<sub>4</sub>S<sub>1</sub></b>	116.4	105.6	118.6	79.5	—	—	—	—
<b>C<sub>3</sub>C<sub>4</sub>S<sub>2</sub></b>	116.4	127.9	129.0	168.9	177.8	173.5	179.4	179.9
<b>H<sub>1</sub>C<sub>3</sub>C<sub>4</sub></b>	109.6	73.8	—	—	—	—	—	—
<b>H<sub>2</sub>C<sub>3</sub>C<sub>4</sub></b>	109.6	113.6	120.2	116.9	120.1	109.4	66.2	—
<b>H<sub>3</sub>C<sub>2</sub>C<sub>1</sub></b>	110.0	107.7	108.3	108.0	109.0	108.0	107.0	106.6
<b>H<sub>4</sub>C<sub>2</sub>C<sub>1</sub></b>	110.0	108.9	108.8	108.5	109.2	108.7	107.0	106.6
<b>Dihedral Angle (°)</b>								
<b>C<sub>3</sub>C<sub>2</sub>C<sub>1</sub>C<sub>5</sub></b>	-89.4	-127.1	-124.6	-162.8	114.9	109.8	179.6	179.9
<b>C<sub>4</sub>C<sub>3</sub>C<sub>2</sub>C<sub>1</sub></b>	180.0	-165.2	-107.0	-118.0	120.1	-79.5	—	-141.9
<b>S<sub>1</sub>C<sub>4</sub>C<sub>3</sub>C<sub>2</sub></b>	88.9	71.2	-179.5	175.7	-177.0	-179.6	—	—
<b>S<sub>2</sub>C<sub>4</sub>C<sub>3</sub>C<sub>2</sub></b>	-88.8	-117.8	0.9	-3.4	175.1	-177.4	—	136.9
<b>H<sub>1</sub>C<sub>3</sub>C<sub>4</sub>S<sub>1</sub></b>	-31.3	-16.9	-0.6	-19.2	13.5	5.1	-0.4	—
<b>H<sub>2</sub>C<sub>3</sub>C<sub>4</sub>S<sub>1</sub></b>	-150.8	-155.8	1.6	-4.6	1.5	-0.4	3.5	—
<b>H<sub>3</sub>C<sub>2</sub>C<sub>1</sub>C<sub>5</sub></b>	31.4	-3.7	-0.7	-40.5	-126.2	-128.3	-56.2	-55.4
<b>H<sub>4</sub>C<sub>2</sub>C<sub>1</sub>C<sub>5</sub></b>	149.8	110.7	115.7	73.9	-8.8	-13.6	55.4	55.7



[F]; State <sup>1</sup>A', Symmetry C<sub>s</sub>, Energy (Hartree) -398.82208, Gibbs Free Energy (Hartree) -398.84201, H<sub>1</sub>S<sub>1</sub> 1.348 Å, H<sub>2</sub>S<sub>1</sub> 1.348 Å, ∠H<sub>1</sub>S<sub>1</sub>H<sub>2</sub> 92.7°.

**Table 7**  
**Energies and Selected Geometries of Stationary Points on the**  
**Potential Energy Surface of Dissociation of the Singlet**  
**PhCH<sub>2</sub>CH<sub>2</sub>CO<sub>2</sub><sup>-</sup> Anion.**

Level of theory used – UCCSD(T)/6-31+G(d,p)//B3LYP/6-31+G(d,p). Relative energies in kJ mol<sup>-1</sup> with respect to A (0 kJ mol<sup>-1</sup>)

		
	<b>A</b>	<b>B</b>
State	<sup>1</sup> A	<sup>1</sup> A
Symmetry	C <sub>1</sub>	C <sub>1</sub>
Energy (Hartree)	-497.40907	-309.18830 [ -497.31026 ]
Energy Relative to A	0	+259 [ B+C ]
GibbsFree Energy (Hartree)	-497.44402	-309.21764 [ -497.35621 ]
GibbsFree Energy Relative to A	0	+231 [ B+C ]
<b>Bond Length (Å)</b>		
C <sub>1</sub> C <sub>2</sub>	1.510	1.517
C <sub>2</sub> C <sub>3</sub>	1.541	1.530
C <sub>3</sub> C <sub>4</sub>	1.571	—
C <sub>4</sub> O <sub>1</sub>	1.261	—
C <sub>4</sub> O <sub>2</sub>	1.259	—
C <sub>3</sub> H <sub>1</sub>	1.096	1.103
C <sub>3</sub> H <sub>2</sub>	1.100	1.102
C <sub>2</sub> H <sub>3</sub>	1.097	1.103
C <sub>2</sub> H <sub>4</sub>	1.095	1.103
<b>Bond Angle (°)</b>		
C <sub>1</sub> C <sub>2</sub> C <sub>3</sub>	113.7	113.3
C <sub>2</sub> C <sub>3</sub> C <sub>4</sub>	113.8	—
C <sub>3</sub> C <sub>4</sub> O <sub>1</sub>	116.1	—
C <sub>3</sub> C <sub>4</sub> O <sub>2</sub>	114.8	—
H <sub>1</sub> C <sub>3</sub> C <sub>2</sub>	111.1	111.2
H <sub>2</sub> C <sub>3</sub> C <sub>4</sub>	107.0	—
H <sub>3</sub> C <sub>2</sub> C <sub>1</sub>	109.5	108.1
H <sub>4</sub> C <sub>2</sub> C <sub>1</sub>	110.9	108.1
<b>Dihedral Angle (°)</b>		
C <sub>3</sub> C <sub>2</sub> C <sub>1</sub> C <sub>5</sub>	-93.3	-88.7
C <sub>4</sub> C <sub>3</sub> C <sub>2</sub> C <sub>1</sub>	-171.8	—
O <sub>1</sub> C <sub>4</sub> C <sub>3</sub> C <sub>2</sub>	27.9	—
O <sub>2</sub> C <sub>4</sub> C <sub>3</sub> C <sub>2</sub>	-153.8	—
H <sub>1</sub> C <sub>3</sub> C <sub>2</sub> O <sub>1</sub>	64.8	60.6
H <sub>2</sub> C <sub>3</sub> C <sub>2</sub> O <sub>1</sub>	-52.3	-60.5
H <sub>3</sub> C <sub>2</sub> C <sub>1</sub> C <sub>5</sub>	28.7	34.0
H <sub>4</sub> C <sub>2</sub> C <sub>1</sub> C <sub>5</sub>	145.8	148.7
<b>O<sub>1</sub>=C<sub>4</sub>=O<sub>2</sub> [C]; State <sup>1</sup>A', Symmetry C<sub>s</sub>, Energy (Hartree) -188.12196, Gibbs Free Energy (Hartree) -188.13857, O<sub>1</sub>C<sub>4</sub> 1.563 Å, O<sub>2</sub>C<sub>4</sub> 1.563 Å, ∠H<sub>1</sub>S<sub>1</sub>H<sub>2</sub> 180.0°.</b>		