**Electronic Supporting Information** 

# Thiophene-functionalized isoindigo dyes bearing electron donor substituents with absorptions approaching the near infrared region

David Bialas, Sabin-Lucian Suraru, Ralf Schmidt and Frank Würthner\*

Universität Würzburg, Institut für Organische Chemie and Röntgen Research Center for Complex Material Systems, Am Hubland, 97074 Würzburg, Germany.

E-mail: wuerthner@chemie.uni-wuerzburg.de; Fax: +49 (0)931 31 84756; Tel: +49 (0)931 31 85340

## **Table of contents**

1.	Computational Details	<b>S2</b>
2.	Cyclic voltammetry	<b>S4</b>
3.	NMR spectra	<b>S</b> 5
4.	References	S20

## 1. Computational Details

The isoindigo derivatives were calculated with the DFT method in combination with the B3-LYP<sup>S1</sup> functional. All calculations were performed with the TURBOMOLE program (version 5.10).<sup>S2</sup> The TZVP<sup>S3</sup> basis for C, N, O, and S atoms was combined with the TZV<sup>S2</sup> basis for the H atom. This basis is designated as TZV(P) and was found to be an excellent compromise between accuracy and computational effort for systems of similar size.<sup>S4</sup> All dyes were fully geometry-optimized, emanating from structures computed on semi-empirical AM1 level. Long alkyl chains were replaced by methyl groups.

Transition densities were calculated on CC2 level with the TZV(P) basis based on the previously geometry-optimized structures using the TURBOMOLE program (version 5.9). In all CC2 computations, the resolution-of-the-identity (RI) approximation was employed.<sup>S5</sup> Corresponding auxiliary basis sets were applied for fitting the charge density.<sup>S6</sup> The frozen core approximation was used to freeze orbitals with eigenvalues smaller than -3.0 a. u. in the calculation of the correlation and excitation energies.



**Fig. S 1** HOMO, LUMO according to DFT (B3-LYP) calculations and transition density (CC2 level) of **3a-f**.

Electronic Supplementary Material (ESI) for Organic and Biomolecular Chemistry This journal is C The Royal Society of Chemistry 2011

## 2. Cyclic voltammetry



**Fig. S 2** Cyclic voltammograms of **3a-f** in dry dichloromethane  $(10^{-4} \text{ M})$ ; scan rate 100 mV s<sup>-1</sup>; supporting electrolyte: tetrabutylammonium hexafluorophosphate (NBu<sub>4</sub>PF<sub>6</sub>, 0.1 M).

## 3. NMR-spectra



Fig. S 3 400 MHz <sup>1</sup>H NMR spectrum of 2d in CDCl<sub>3</sub>.



**Fig. S 4** 400 MHz <sup>1</sup>H NMR spectrum of 2e in CD<sub>2</sub>Cl<sub>2</sub>.



Fig. S 5 400 MHz <sup>1</sup>H NMR spectrum of 2f in CDCl<sub>3</sub>.



Fig. S 6 400 MHz  $^{1}$ H NMR spectrum of 3a in CD<sub>2</sub>Cl<sub>2</sub>.



Fig. S 7 400 MHz  $^{1}$ H NMR spectrum of 3b in CD<sub>2</sub>Cl<sub>2</sub>.



**Fig. S 8** 400 MHz <sup>1</sup>H NMR spectrum of 3c in CD<sub>2</sub>Cl<sub>2</sub>.



Fig. S 9 400 MHz <sup>1</sup>H NMR spectrum of 3d in CD<sub>2</sub>Cl<sub>2</sub>.



Fig. S 10 400 MHz <sup>1</sup>H NMR spectrum of 3e in CD<sub>2</sub>Cl<sub>2</sub>.



Fig. S 11 400 MHz <sup>1</sup>H NMR spectrum of 3f in CD<sub>2</sub>Cl<sub>2</sub>.



Fig. S 12 101 MHz  $^{13}$ C NMR spectrum of 3a in CD<sub>2</sub>Cl<sub>2</sub>.



Fig. S 13 101 MHz  $^{13}$ C NMR spectrum of 3b in CD<sub>2</sub>Cl<sub>2</sub>.



Fig. S 14 101 MHz 13C NMR spectrum of 3c in CD<sub>2</sub>Cl<sub>2</sub>.



Fig. S 15 101 MHz  $^{13}$ C NMR spectrum of 3d in CD<sub>2</sub>Cl<sub>2</sub>.



Fig. S 16 101 MHz <sup>13</sup>C NMR spectrum of 3e in CD<sub>2</sub>Cl<sub>2</sub>.



Fig. S 17 151 MHz  $^{13}$ C NMR spectrum of 3f in CD<sub>2</sub>Cl<sub>2</sub>.

## 4. References

- A. D. Becke, *Phys. Rev. A*, 1988, **38**, 3098; C. Lee, W. Yang and R. G. Parr; *Phys. Rev. B*, 1988, **37**, 785; A. D. Becke, *J. Chem. Phys.* 1993, **98**, 5648.
- R. Ahlrichs, M. Bär, M. Häser, H. Horn and C. Köhmel, *Chem. Phys. Lett.*, 1989, 162, 165; R.
  Ahlrichs and M. von Arnim, *Methods and Techniques in Computational Chemistry: METECC-*95; Clementi, E.; Corongiu, G., Eds.; Club Européen MOTECC, 1995.
- 3 A. Schäfer, C. Huber and R. Ahlrichs, *J. Chem. Phys.*, 1994, **100**, 5829.
- 4 R. F. Fink, J. Seibt, V. Engel, M. Renz, M. Kaupp, S. Lochbrunner, H.-M. Zhao, J. Pfister, F. Würthner and B. Engels, *J. Am. Chem. Soc.*, 2008, **130**, 12858.
- 5 O. Treutler and R. Ahlrichs, J. Chem. Phys., 1995, **102**, 346.
- 6 K. Eichkorn, F. Weigend, O. Treutler and R. Ahlrichs, *Theor. Chim. Acta*, 1997, **97**, 119.