

Cholesterol appended pyridinium ureas: A case of gel making and breaking for selective visual readout of F⁻

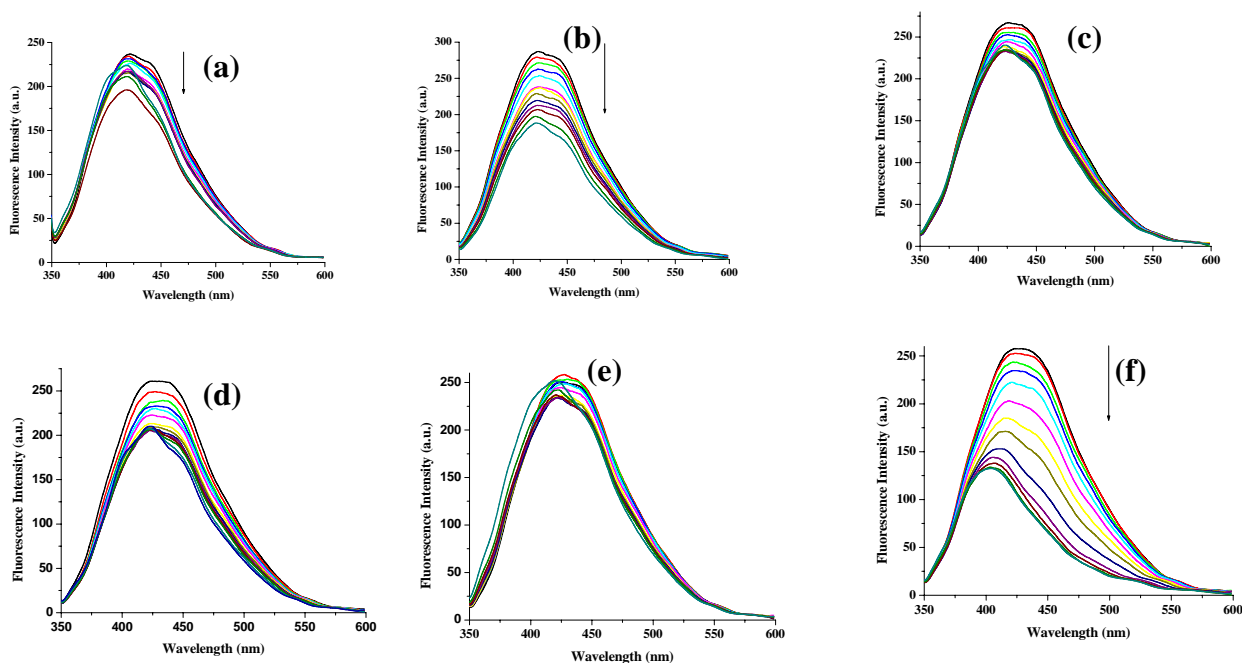
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1. Change in emission of receptor **1** with HSO₄⁻, Cl⁻, Br⁻, I⁻, ClO₄⁻, AcO⁻, H₂PO₄⁻ in CH₃CN.
2. Change in of absorbance receptor **1** with HSO₄⁻, Cl⁻, Br⁻, I⁻, ClO₄⁻, AcO⁻, H₂PO₄⁻ in CH₃CN.
3. Change of emission spectra of receptor **1** upon addition of 2 equivalent of different anions in CH₃CN.
4. Change in emission of receptor **1** with HSO₄⁻, Cl⁻, Br⁻, I⁻, ClO₄⁻, AcO⁻, H₂PO₄⁻ in DMSO.
5. Change in absorbance of receptor **1** with HSO₄⁻, Cl⁻, Br⁻, I⁻, ClO₄⁻, AcO⁻, H₂PO₄⁻ in DMSO.
6. Change of emission spectra of receptor **1** upon addition of 15 equivalent of different anion in DMSO.
7. Benesi–Hilderband plot for **1** with fluoride in CH₃CN.
8. Benesi–Hilderband plot for **1** with F⁻ in DMSO.
9. Photograph of formation of gel for receptor **1** with 5 equiv. amounts of Fluoride in DMSO.
10. Job Plot of **1** with TBAF in DMSO.
11. Comparision of IR Septra.
12. Characterization spectra
13. Comparative ¹H NMR and FTIR spectra
14. UV-vis titration spectra for 2 with the anions.

1. Change in emission of receptor **1** with various anions in CH₃CN.



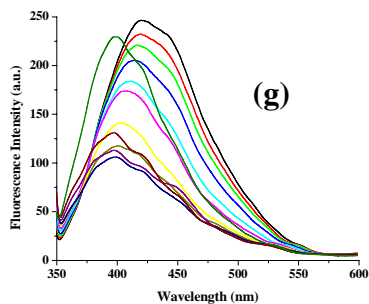


Figure 1S. Fluorescence titration spectra for **1** ($c = 6.32 \times 10^{-5}$ M) with (a) HSO₄⁻, (b) Cl⁻, (c) Br⁻, (d) I⁻, (e) ClO₄⁻, (f) AcO⁻ (g) H₂PO₄⁻ in CH₃CN (in all cases [anion] = 1 x 10⁻³ M).

2. Change in absorbance of receptor 1 with various anions in CH₃CN.

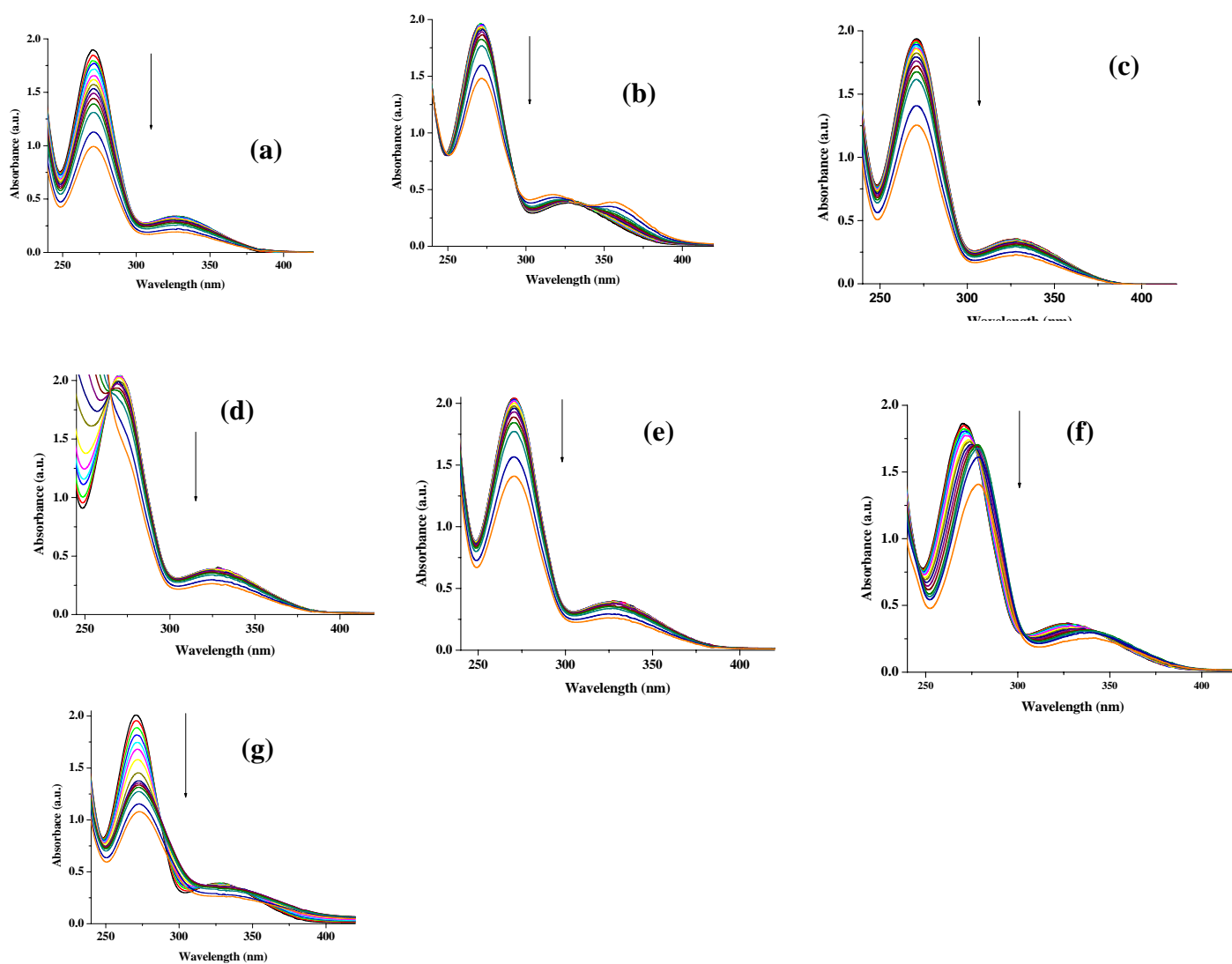


Figure 2S. Change in absorbance of receptor **1** ($c = 6.32 \times 10^{-5}$ M) with (a) HSO₄⁻, (b) Cl⁻, (c) Br⁻, (d) I⁻, (e) ClO₄⁻ (f) AcO⁻ (g) H₂PO₄⁻ in CH₃CN (in all cases [anion] = 1 x 10⁻³ M).

3. Change of emission spectra of receptor 1 upon addition of 2 equivalent of different anion in CH₃CN.

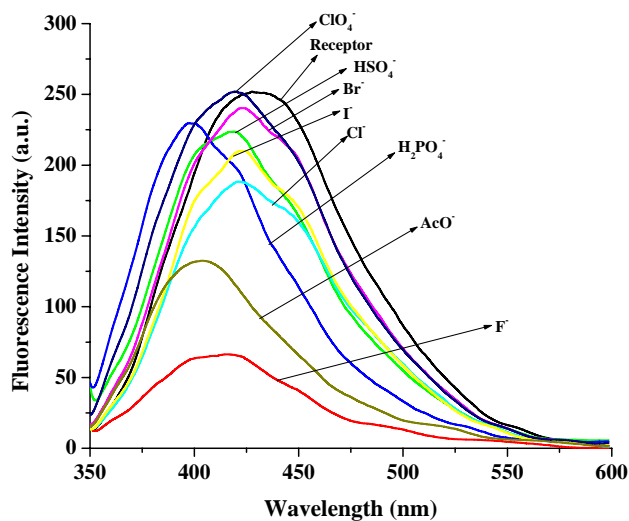
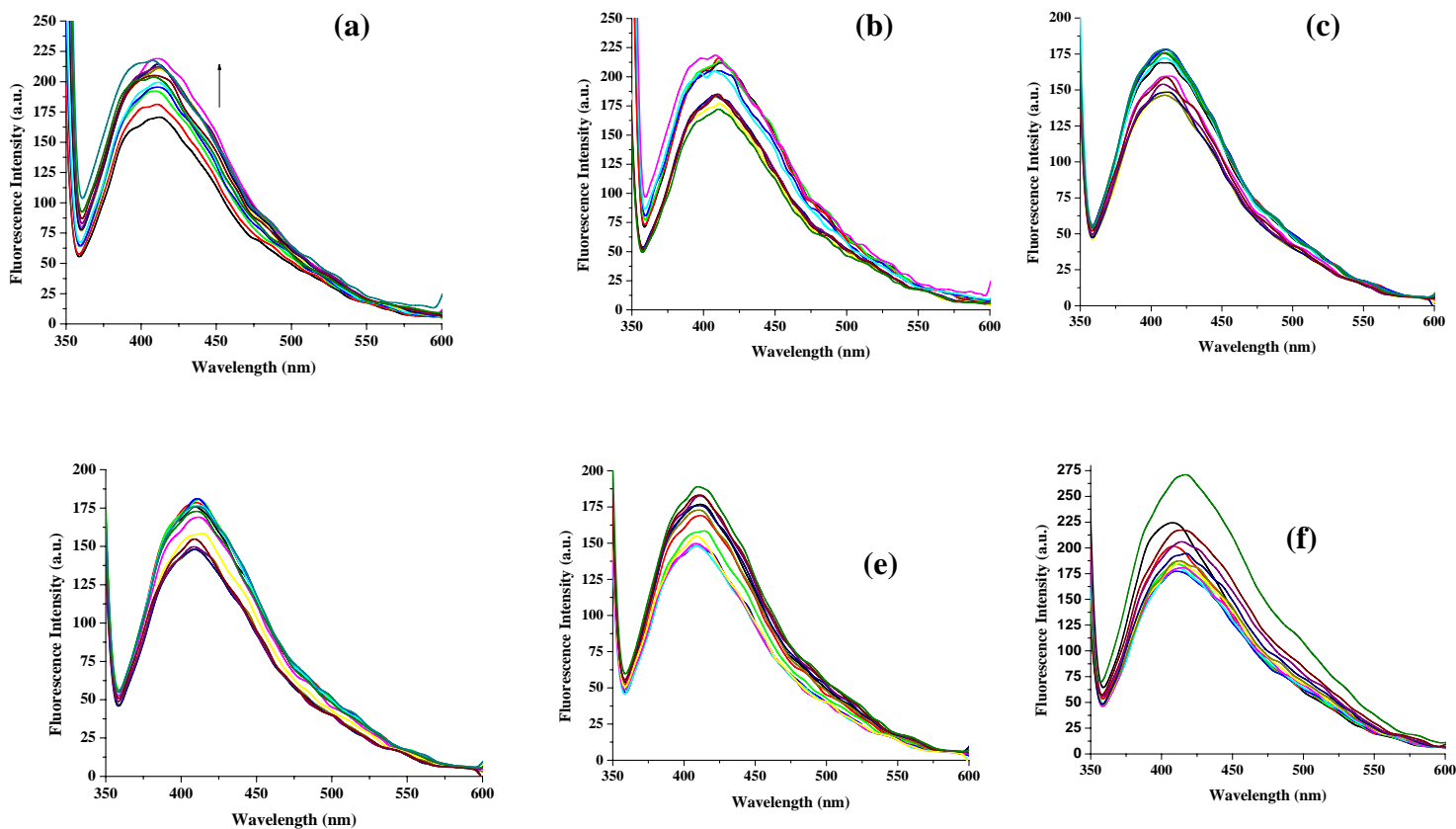


Figure 3S: Change in emission of receptor 1 ($c = 6.32 \times 10^{-5}$ M) upon addition of 2 eqv. of different Guests ($c = 1 \times 10^{-3}$ M) in CH₃CN.

4. Change in emission of receptor 1 with various anions in DMSO.



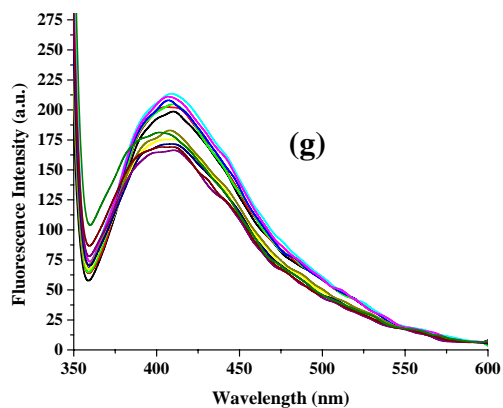
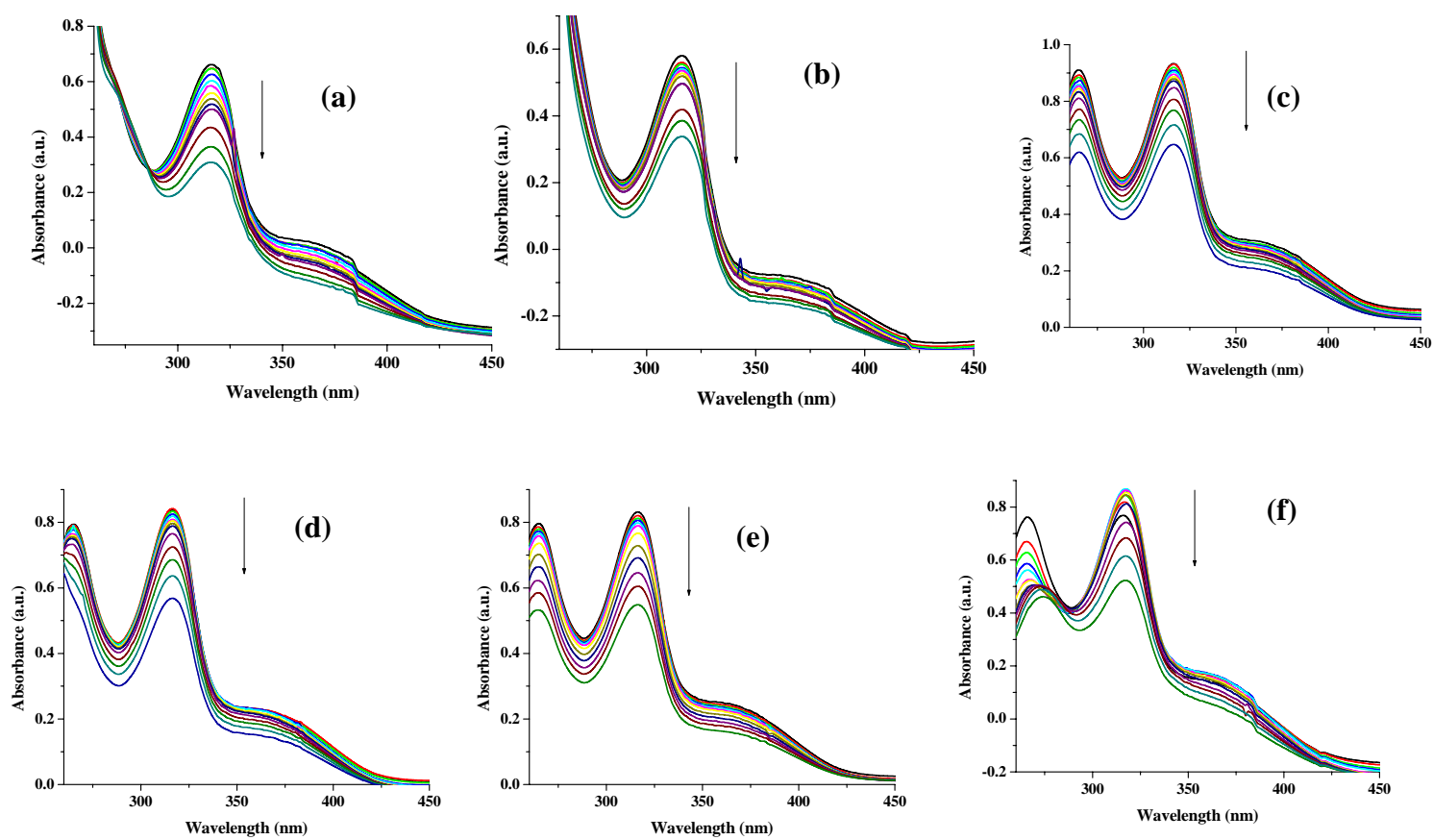


Figure 4S. Fluorescence titration spectra for **1** (c = 6.01 x 10⁻⁵ M) with (a) HSO₄⁻, (b) Cl⁻, (c) Br⁻, (d) I⁻, (e) ClO₄⁻, (f) AcO⁻ (g) H₂P₂O₇⁴⁻ in DMSO (in all cases [anion] = 1 x 10⁻³ M)

5. Change in absorbance of receptor 1 with various anions in DMSO.



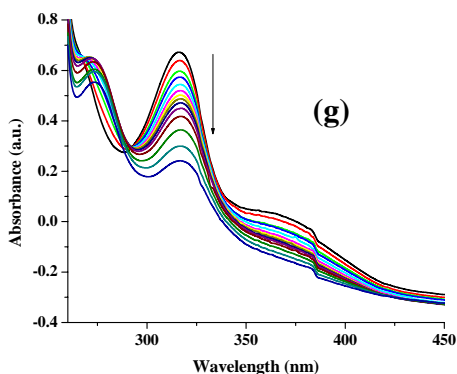


Figure 5S. Change in absorbance of receptor **1** ($c = 6.01 \times 10^{-5}$ M) with (a) HSO_4^- , (b) Cl^- , (c) Br^- , (d) I^- , (e) ClO_4^- (f) AcO^- (g) H_2PO_4^- in DMSO (in all cases $[\text{anion}] = 1 \times 10^{-3}$ M).

6. Change of emission spectra of receptor 1 upon addition of 15 equivalent of different anion in DMSO.

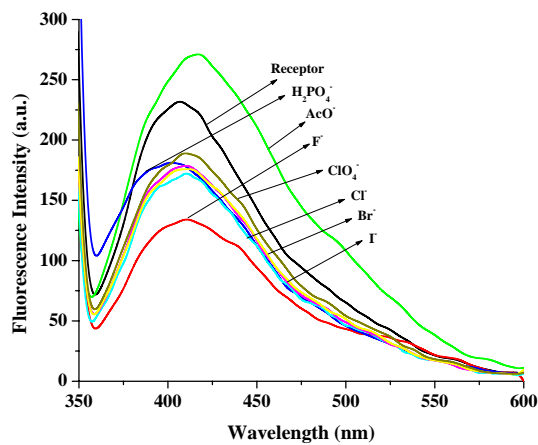


Figure 6S: Change in emission of receptor **1** ($c = 6.01 \times 10^{-5}$ M) upon addition of 15 eqv. of Different Guest ($c = 1 \times 10^{-3}$ M) in DMSO

7. Benesi–Hilderband plot for **1** with F⁻ in CH₃CN.

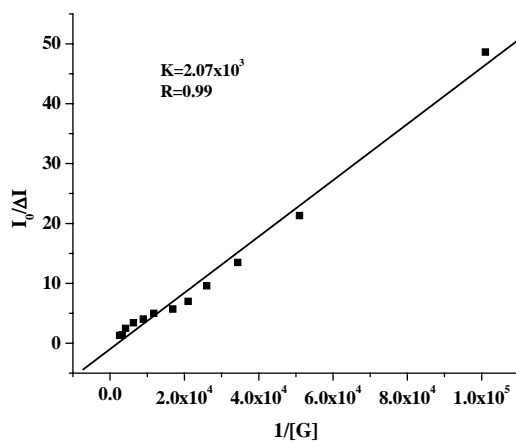


Figure 7S: Benesi–Hilderband plot for receptor **1** ($c = 6.32 \times 10^{-5}$ M) with fluoride ($c = 1 \times 10^{-3}$ M) at 430 nm in CH₃CN

8. Benesi–Hilderband plot for **1** with F⁻ in DMSO.

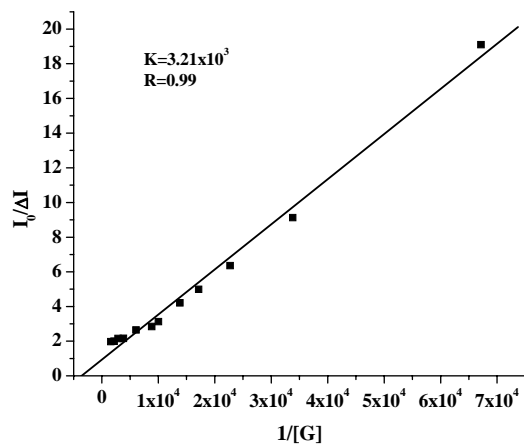


Figure 8S: Benesi–Hilderband plot for receptor **1** ($c = 6.01 \times 10^{-5}$ M) with fluoride ($c = 1 \times 10^{-3}$ M) at 410 nm in DMSO.

9. Photograph of formation of gel for receptor 1 with 5 equiv. amounts of Fluoride in DMSO.



Fig 9S. Photograph of Receptor 1 ($c = 3.58 \times 10^{-3}$ M) with 5 equiv. amounts of Fluoride ($c = 1.5 \times 10^{-2}$ M) in DMSO.

10. Job plot of 1 with F^- in DMSO.

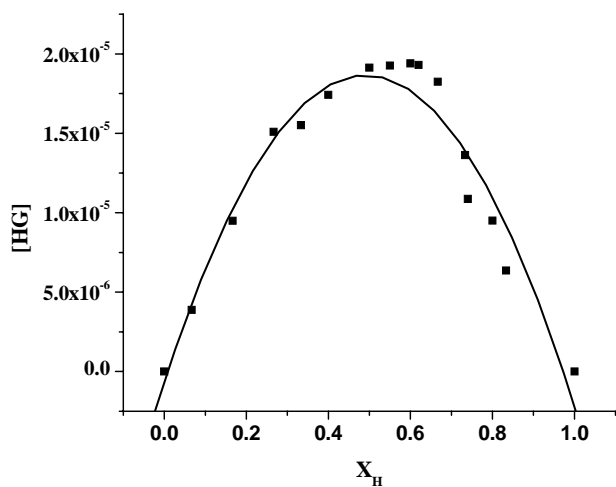


Fig 10S. Job plot for 1 ($c = 5.96 \times 10^{-5}$ M) with TBAF in DMSO.

11. Comparison of IR- Spectra.

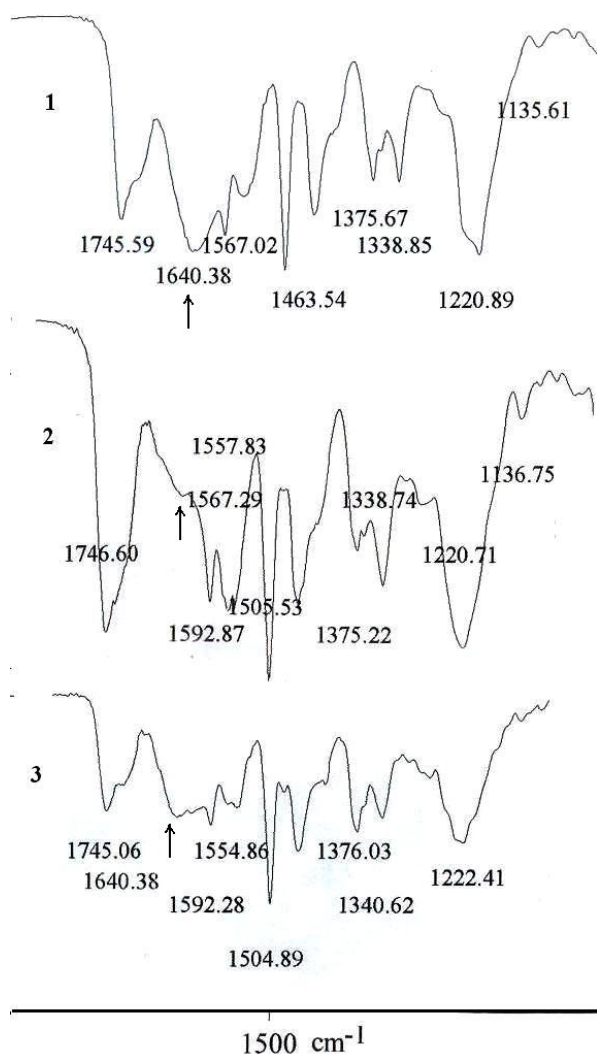
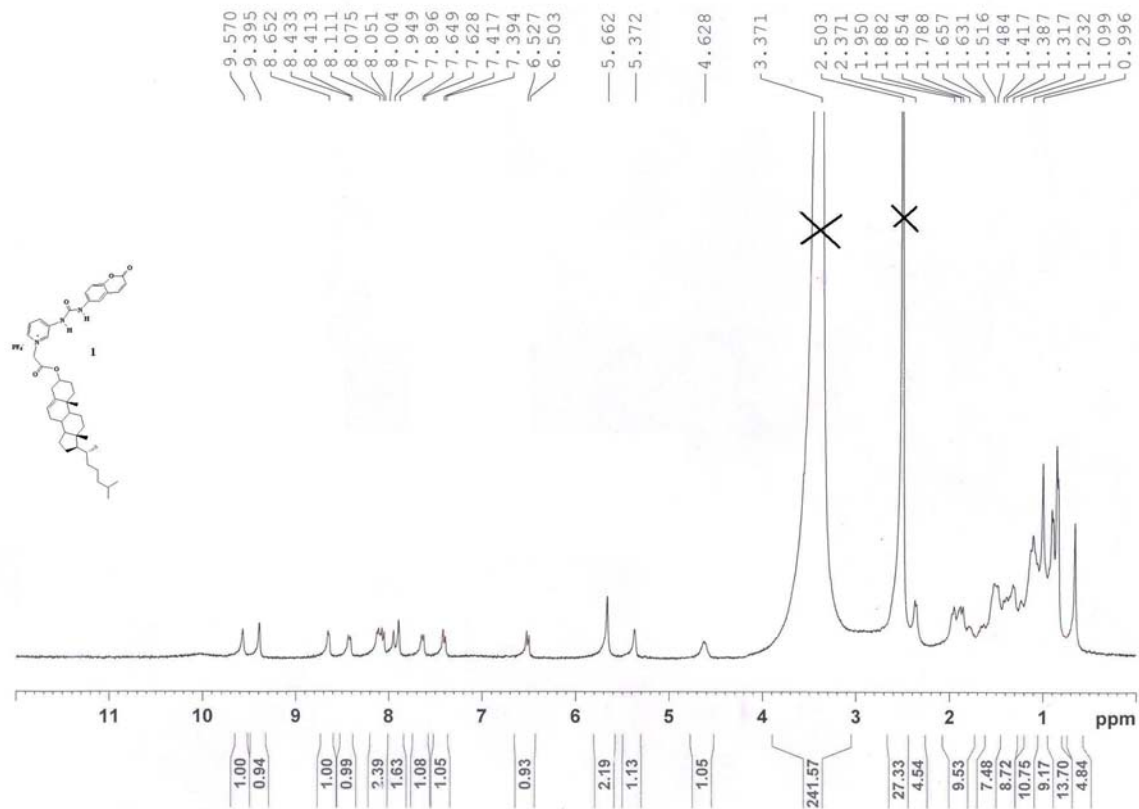


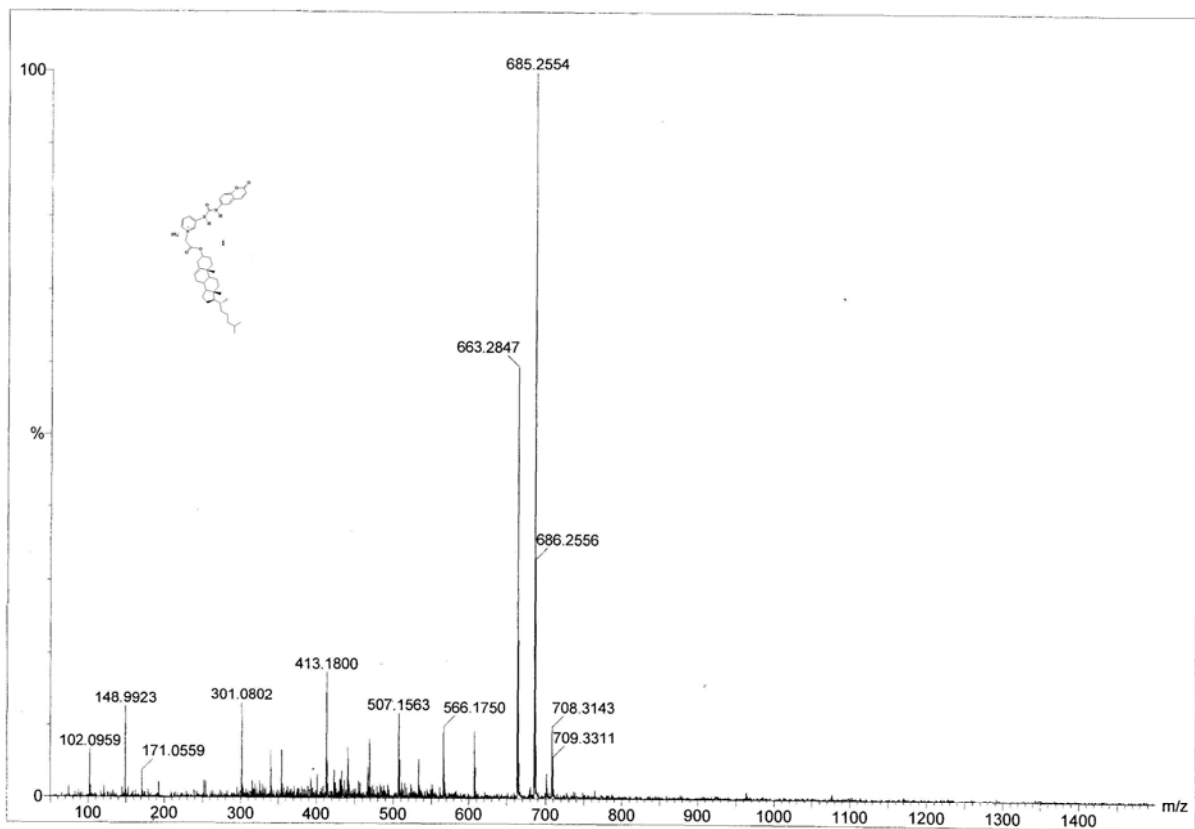
Fig 11S. 1 Partial IR spectrum of **2a**. 2 partial IR spectrum of **2a** in gel state. 3 Partial IR spectrum of **2a** after breaking the gel with fluoride.

12. Characterization spectra

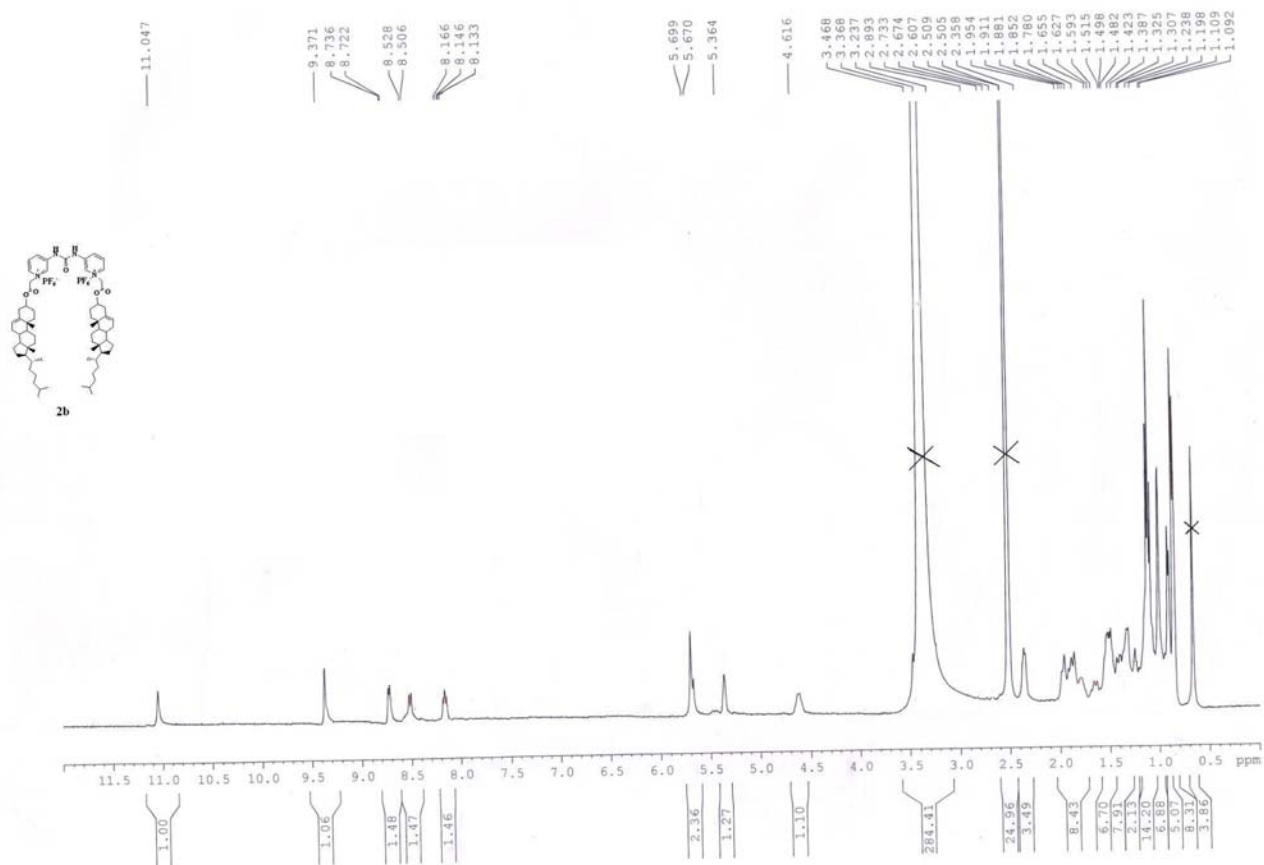
¹H NMR (400 MHz, d₆-DMSO)



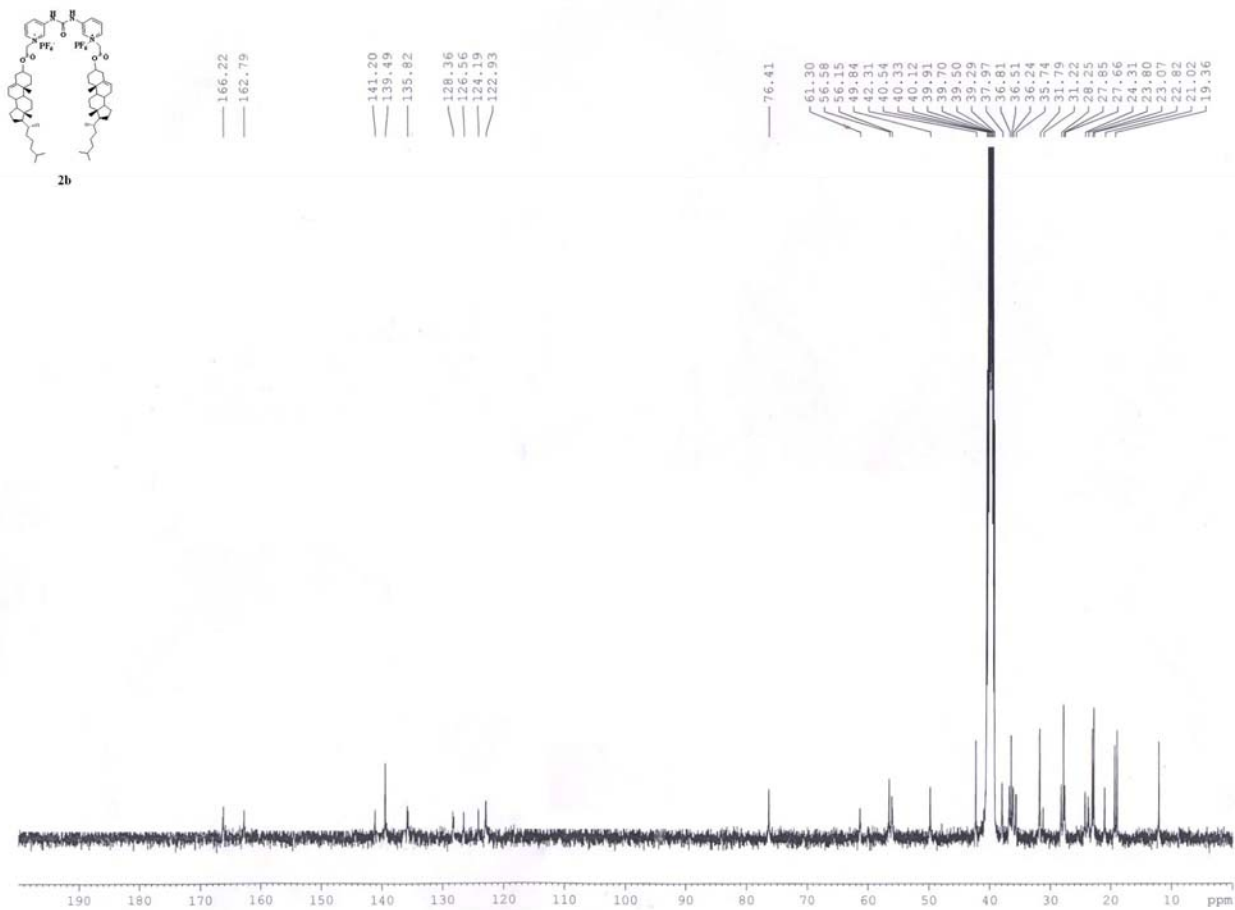
Mass



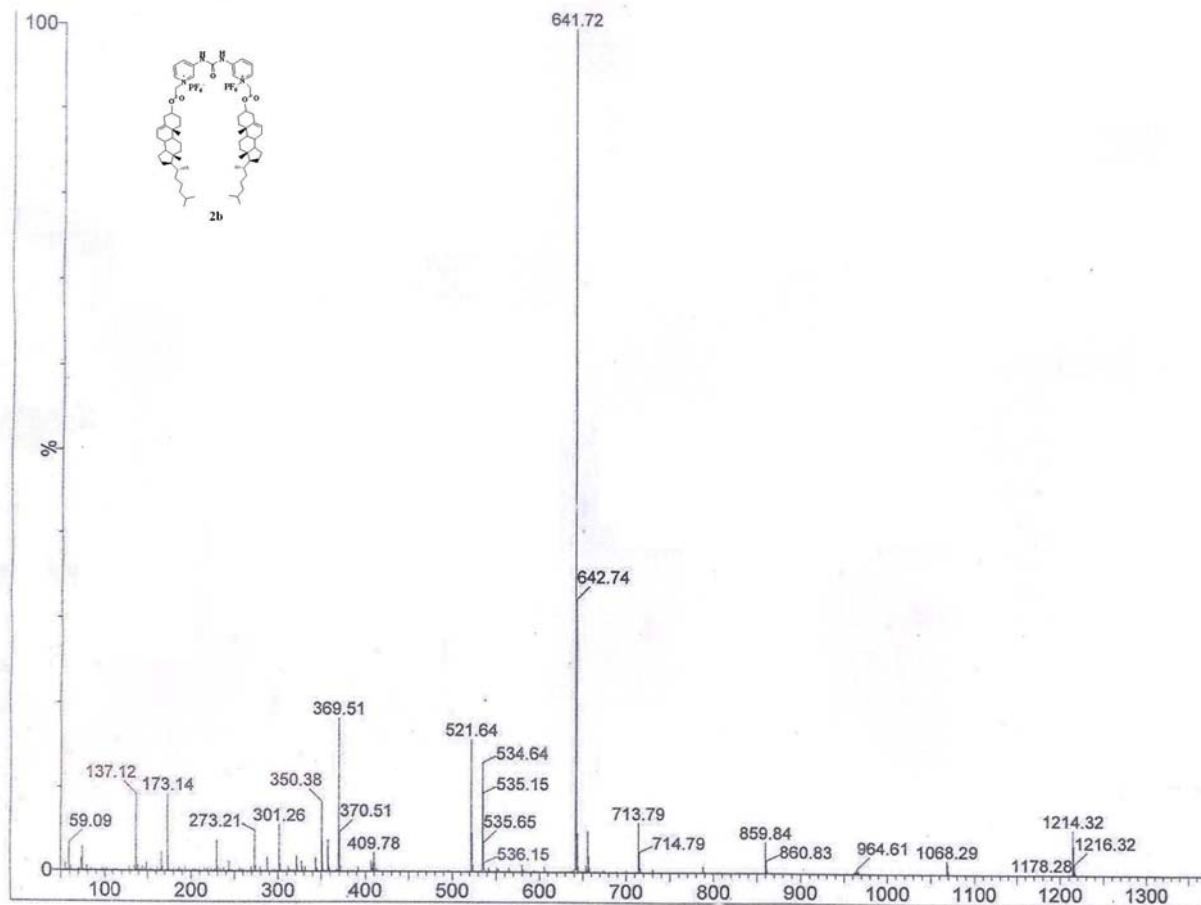
¹H NMR (400 MHz, d₆-DMSO)



^{13}C NMR (100 MHz, $\text{d}_6\text{-DMSO}$)



Mass



13. Comparative ^1H NMR and FTIR spectra

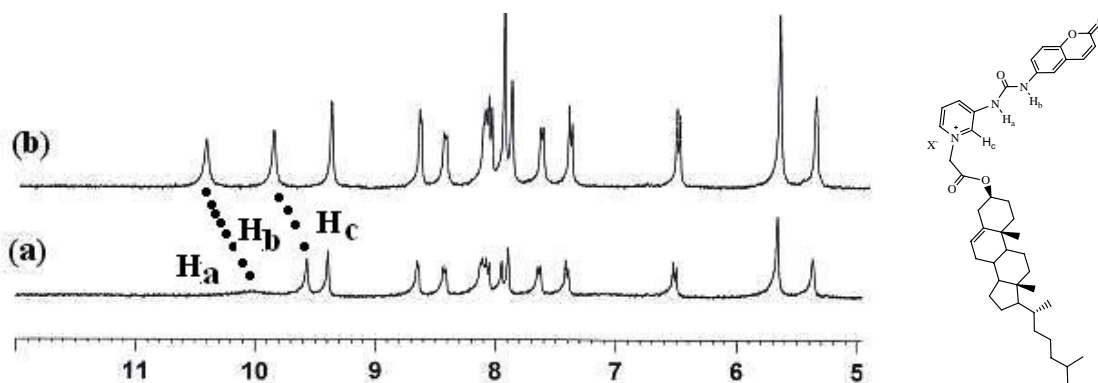


Fig 12S. Partial ^1H NMR (400 MHz, d_6DMSO) of (a) **1** ($\text{X} = \text{PF}_6^-$, $c = 4.8 \times 10^{-3} \text{ M}$) and (b) $\text{X} = \text{Cl}^-$.

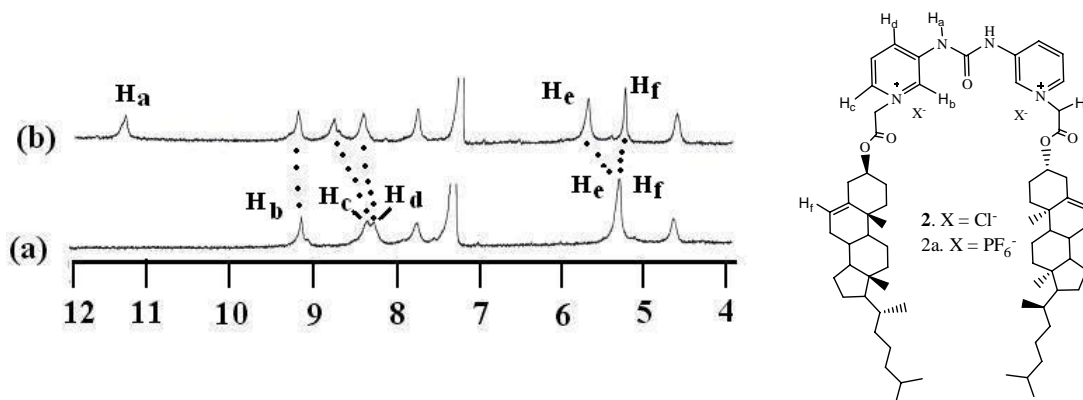


Fig 13S. Partial ^1H NMR (400 MHz, CDCl_3 containing 5% d_6DMSO) of (a) **2a** and (b) **2**.

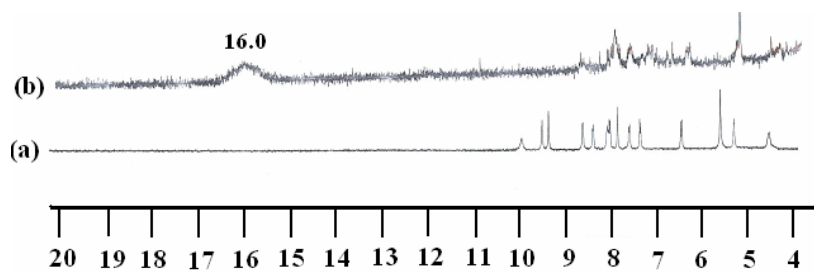


Fig 14S. Partial ^1H NMR (400 MHz, d_6DMSO) of (a) **1** and (b) after addition of 3 equivalent amounts of tetrabutylammonium fluoride.

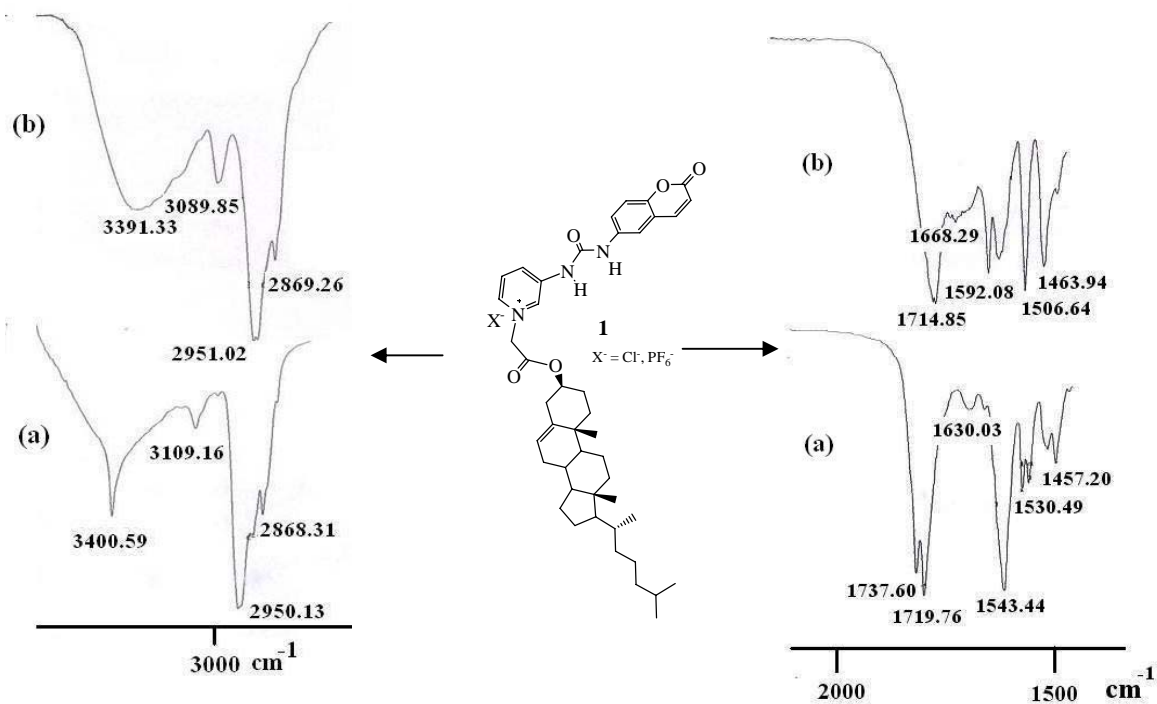


Fig 15S. Partial FT-IR (in KBr) of (a) **1** and (b) chloride salt of **1**.

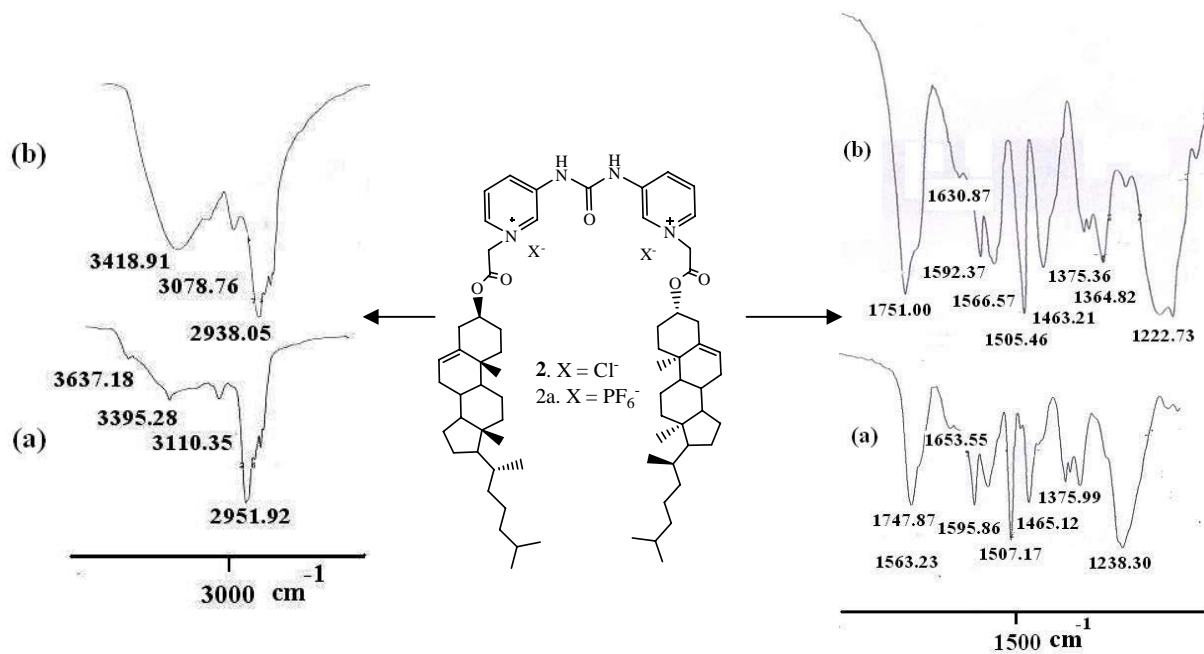


Fig 16S. Partial FT-IR (in KBr) of (a) **2a** and (b) **2**

14. UV-vis titration spectra for **2** with the anions.

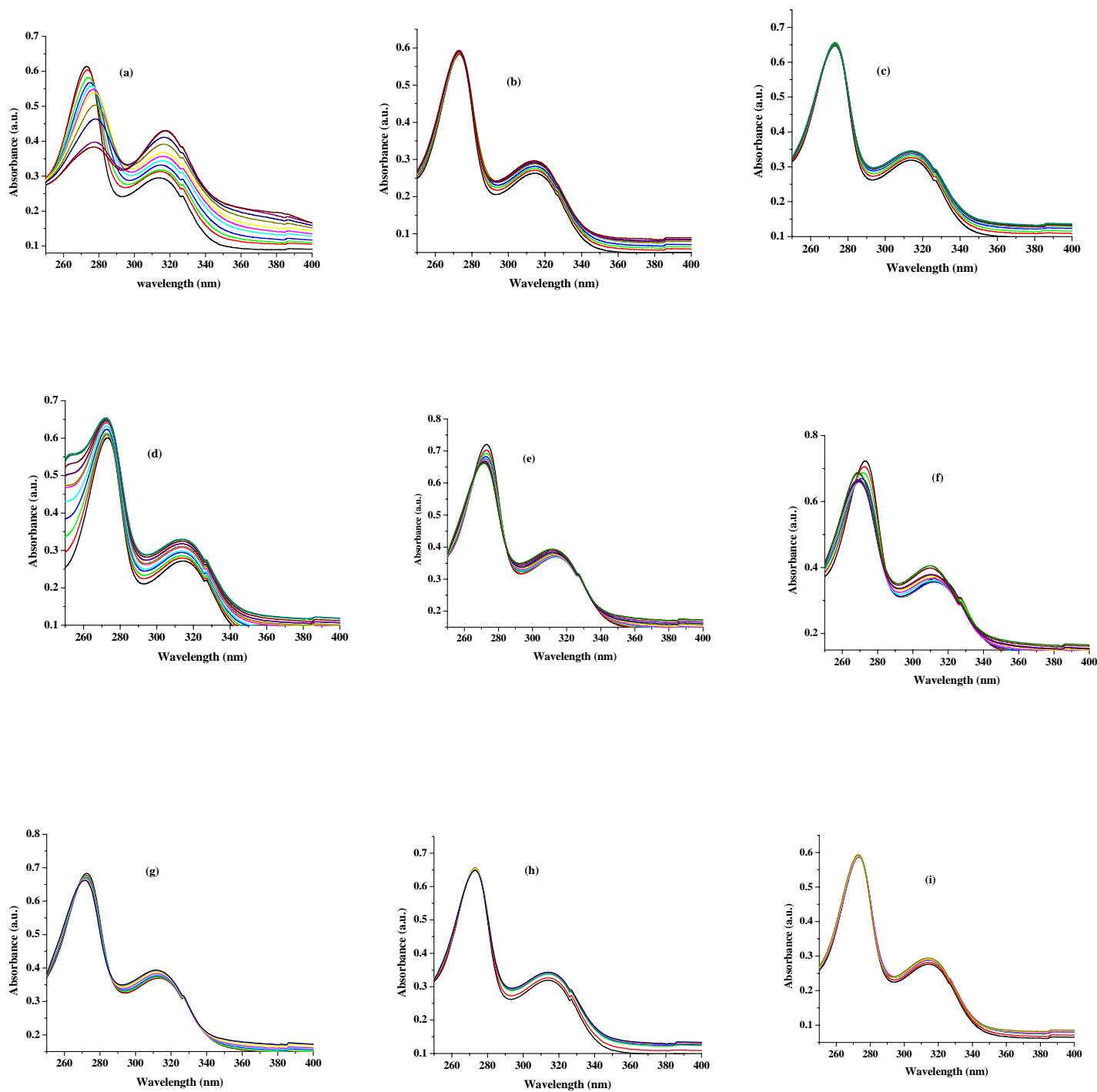


Fig 17S. Absorption titration spectra for **2** ($c = 4.05 \times 10^{-5}$ M) with (a) F^- , (b) Cl^- , (c) Br^- , (d) I^- , (e) $H_2PO_4^-$, (f) HSO_4^- , (g) AcO^- , (h) NO_3^- , (i) ClO_4^- in $CHCl_3$ (in all cases $[anion] = 1 \times 10^{-3}$ M).