Palladium-Catalyzed ortho-Functionalization of Azoarenes with Aryl

Acylperoxides

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1. General experimental information

Unless otherwise noted, reactions were conducted in dry solvents. Purifications of reaction products were carried out by flash chromatography using silica gel (40-63 mm). Infrared spectra (IR) were recorded on a spectrophotometer and are reported as wavelength numbers (cm⁻¹). Infrared spectra were recorded by preparing a KBr pellet containing the title compound. ¹H and ¹³C NMR spectra were recorded with tetramethylsilane (TMS) as internal standard at ambient temperature unless otherwise indicated on a spectrometer operating at 400 MHz for ¹H NMR and 100 MHz for ¹³CNMR Chemical shifts are reported in parts per million (ppm) and coupling constants are reported as Hertz (Hz). Splitting patterns are designated as singlet (s), broad singlet (bs), doublet (d), triplet (t). Splitting patterns that could not be interpreted or easily visualized are designated as multiple (m). Low resolution mass spectra were recorded on an IF-TOF spectrometer (Micromass).Gas chromatograph mass spectra were obtained with a model spectrometer.

2. General procedure for reaction condition screening

Table S1. General procedure for ortho-acyloxylationofazobenzenes:

A 10 mL of reaction tube was charged with azobenzene (**1a**) (0.15 mmol), benzoylperoxide (**2a**) (0.30 mmol), Pd catalyst(10 mol %), and CH₃CN (2.0 mL). After the reaction was carried out at 60 $^{\circ}$ C for 24 h, it was cooled to room temperature and concentrated in vacuum, followed by flash chromatography on SiO₂ to provide the corresponding desired product **3a**.^{*a*}

	N _N + 0 1a	$\frac{Pd \text{ catalyst}}{Solvent, heat}$	$ \begin{array}{c} $
Entry	Pd catalyst	Solvent	Yield (%) ^b
1	PdCl ₂	CH ₃ CN	43
2	PdCl ₂	DMSO	NR
3	PdCl ₂	DMF	NR
4	PdCl ₂	THF	14
5	PdCl ₂	Toluene	34
6	PdCl ₂	CH_2Cl_2	26

7	PdCl ₂	PhCF ₃	37
8	Pd(TFA) ₂	CH ₃ CN	60
9	$Pd(OAc)_2$	CH ₃ CN	82
10	Pd(MeCN) ₂ Cl ₂	CH ₃ CN	33
11	$Pd(OAc)_2$	CH ₃ CN	59 ^c
12	Pd(OAc) ₂	CH ₃ CN	67 ^d

^{*a*} Unless otherwise noted, the reactions were carried out using azobenzene (**1a**) (0.15 mmol) and benzoyl peroxide (**2a**) (0.30 mmol) with palladium catalyst (10 mol %) in solvent (2.0 mL) at 60 °C for 24 h. ^{*b*} Isolated yield. ^{*c*} Reaction temperature:80 °C. ^{*d*} Reaction temperature:40 °C.

Table S2. General procedure for *ortho*-arylationofazobenzenes:

A 10 mL of reaction tube was charged with azobenzene (1a) (0.15 mmol), benzoylperoxide (2a) (0.30 mmol), Pd catalyst (10 mol %), and PhCl (1.0 mL). After the reaction was carried out at 100 $^{\circ}$ C for 24 h, it was cooled to room temperature and concentrated in vacuum. The residue was purified by flash chromatography to afford 4a.^{*a*}

NSN	+	Pd catalyst Solvent, heat	N _N	
1 a	2a		4a	
Entry	PdX ₂	Solvent	Yield (%) ^b	
1	Pd(OAc) ₂	HOAc	NR	
2	Pd(OAc) ₂	DMSO	Trace	
3	$Pd(OAc)_2$	DMF	Trace	
4	$Pd(OAc)_2$	CHCl ₂ CHCl ₂	Trace	
5	$Pd(OAc)_2$	PhCl	56	
6	$Pd(OAc)_2$	PhCF ₃	43	
7	$Pd(OAc)_2$	CH ₂ ClCH ₂ Cl	Trace	
8	$Pd(OAc)_2$	PhNO ₂	NR	
9	PdCl ₂	PhCl	47	
10	Pd(TFA) ₂	PhCl	63	
11	Pd(MeCN) ₂ Cl ₂	PhCl	55	
12	$Pd(OAc)_2$	PhCl	76°	
13	$Pd(OAc)_2$	PhCl	49 ^d	
13	$Pd(OAc)_2$	PhCl	58 ^e	
^a Unless otherwise noted, the reactions were carried out using azobenzene (1a) (0.15				

sis otherwise noted, the reactions were carried out using azobenzene (14) (0.15

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mmol) and benzoyl peroxide (**2a**) (0.30 mmol) with palladium catalyst (10 mol %) in solvent (2.0 mL) at 100 °C for 24 h. ^{*b*}Isolated yield. ^{*c*}Reaction temperature:130 °C. ^{*d*}Reaction temperature: 90 °C. ^{*e*}1 equiv of **2a**.

3. ¹H NMR and ¹³C NMR spectrum for all the isolated products (CDCl₃ as solvent)



3.1 ¹H NMR and ¹³C NMR spectrum for **1b** (CDCl₃ as solvent)







3.3 1 H NMR and 13 C NMR spectrum for **1d** (CDCl₃ as solvent)





3.5 ¹H NMR and ¹³C NMR spectrum for **1g** (CDCl₃ as solvent)





3.6 ¹H NMR and ¹³C NMR spectrum for **1f** (CDCl₃ as solvent)





3.8 1 H NMR and 13 C NMR spectrum for **1i** (CDCl₃ as solvent)









3.10 1 H NMR and 13 C NMR spectrum for 1k (CDCl₃ as solvent)



3.11 ¹H NMR and ¹³C NMR spectrum for 2g (CDCl₃ as solvent)



3.12 ¹H NMR and ¹³C NMR spectrum for **2h** (CDCl₃ as solvent)

3.13 ¹H NMR and ¹³C NMR spectrum for **2i** (CDCl₃ as solvent)







3.15 ¹H NMR and ¹³C NMR spectrum for **3b** (CDCl₃ as solvent)



3.16 1 H NMR and 13 C NMR spectrum for **3c** (CDCl₃ as solvent)







3.18 1 H NMR and 13 C NMR spectrum for **3e** (CDCl₃ as solvent)









3.20 1 H NMR and 13 C NMR spectrum for **3g** (CDCl₃ as solvent)



3.22 ¹H NMR and ¹³C NMR spectrum for 3j (CDCl₃ as solvent)









3.24 1 H NMR and 13 C NMR spectrum for **3m** (CDCl₃ as solvent)



3.25 1 H NMR and 13 C NMR spectrum for **3n** (CDCl₃ as solvent)

3.26 1 H NMR and 13 C NMR spectrum for **3p** (CDCl₃ as solvent)





3.27 ¹H NMR and ¹³C NMR spectrum for 3q (CDCl₃ as solvent)











3.30 1 H NMR and 13 C NMR spectrum for **4a** (CDCl₃ as solvent)





3.32 $^1\!H$ NMR and $^{13}\!C$ NMR spectrum for 4c (CDCl3 as solvent)











3.35 1 H NMR and 13 C NMR spectrum for **4f** (CDCl₃ as solvent)









3.38 1 H NMR and 13 C NMR spectrum for **4i** (CDCl₃ as solvent)

3.39 ¹H NMR and ¹³C NMR spectrum for 4j (CDCl₃ as solvent)







3.41 1 H NMR and 13 C NMR spectrum for **4m** (CDCl₃ as solvent)



3.42 1 H NMR and 13 C NMR spectrum for **4n** (CDCl₃ as solvent)





3.43 1 H NMR and 13 C NMR spectrum for 4q (CDCl₃ as solvent)



3.44 ¹H NMR and ¹³C NMR spectrum for 4r (CDCl₃ as solvent)

3.45 1 H NMR and 13 C NMR spectrum for **5** (CDCl₃ as solvent)









3.47 1 H NMR and 13 C NMR spectrum for 7 (CDCl₃ as solvent)

3.48 1 H NMR and 13 C NMR spectrum for **8** (DMSO-D6 as solvent)

