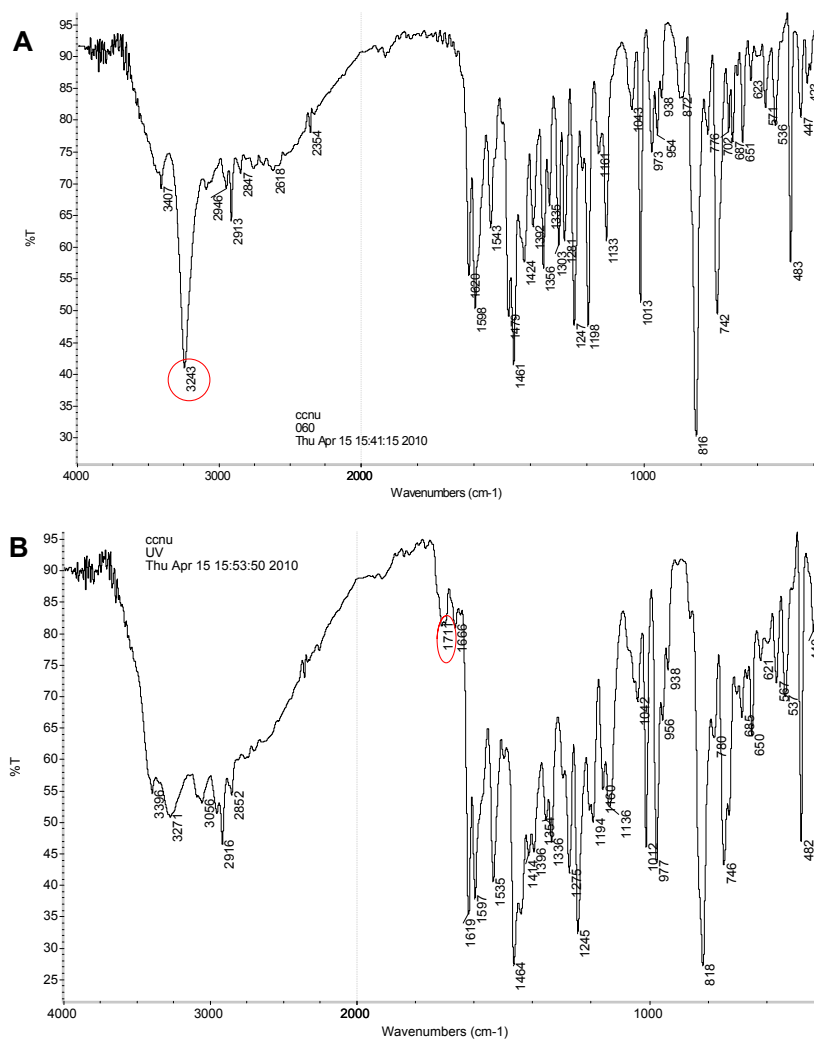


## Supporting Information

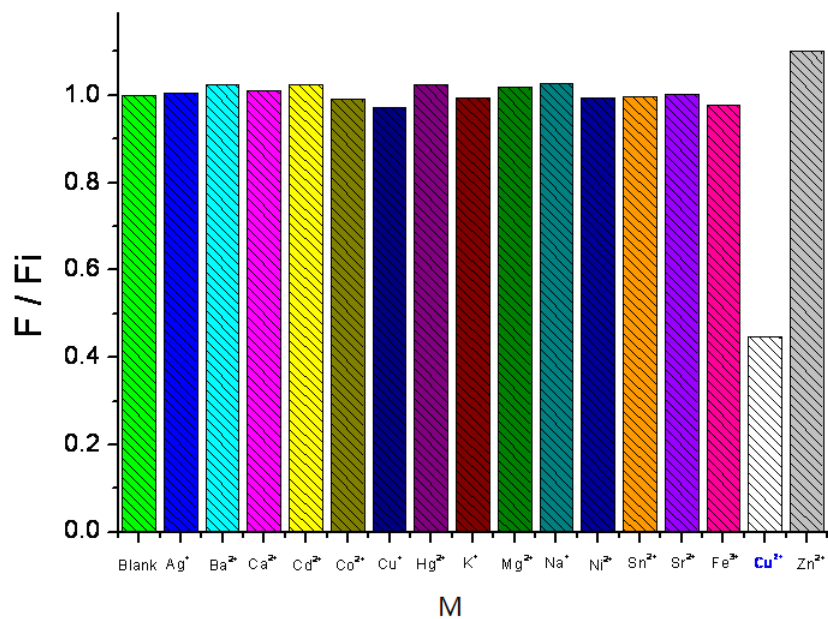
# Chemical control of photochromism and a multiresponsive molecular switch based on a diarylethene derivative containing naphthol

Wenju Liu, Ziyong Li, Fang Hu, Jun Yin \*, Guang-Ao Yu , Sheng Hua Liu\*

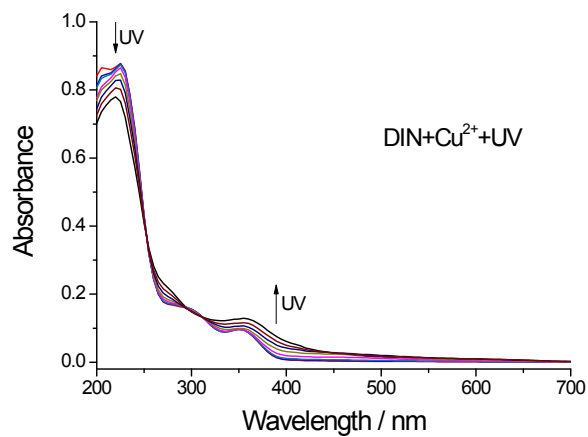
Key Laboratory of Pesticide and Chemical Biology, Ministry of Education, College of Chemistry,  
Central China Normal University, Wuhan 430079, PR China



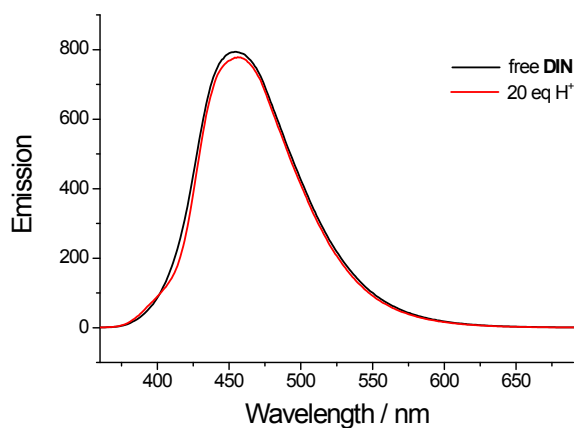
**Figure S1:** IR spectrum of diarylethene DIN before and after UV irradiation.(A: IR spectrum before UV irradiation; B: IR spectrum after UV irradiation).



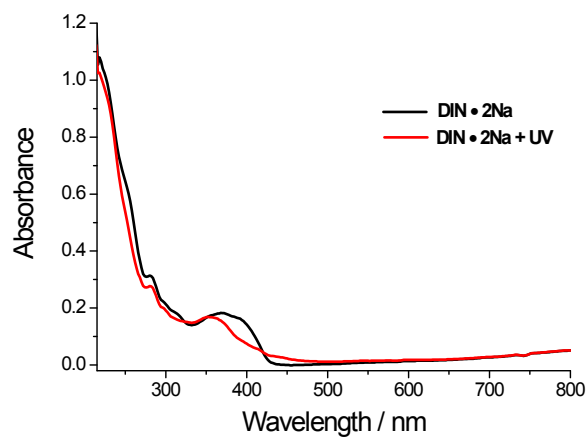
**Figure S2:** Fluorescence responses of **DIN** to various ions in acetonitrile ( $c = 2 \times 10^{-5}$  mol/L).  $F/F_i$  represents the final integrated emission ( $F$ ) over initial integrated fluorescence response ( $F_i$ ).



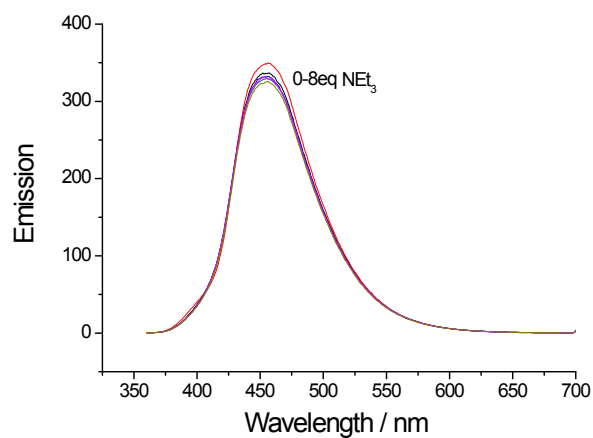
**Figure S3:** Optical response of diarylethene **DIN** in acetonitrile upon the addition of  $Cu^{2+}$  after irradiation with 302 nm.



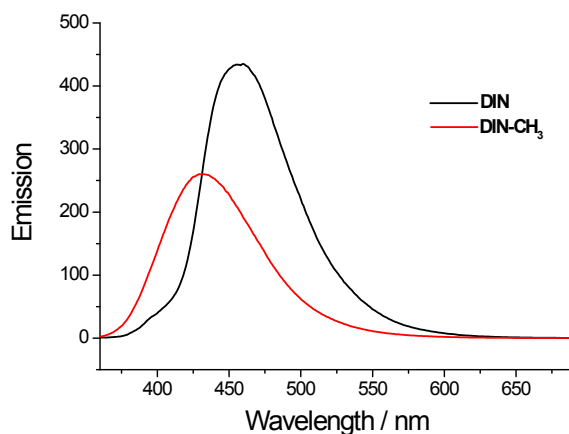
**Figure S4:** Fluorescence responses of diarylethene **DIN** in acetonitrile ( $c = 2 \times 10^{-5}$  mol/L) upon protonation with HCl (20 eq).



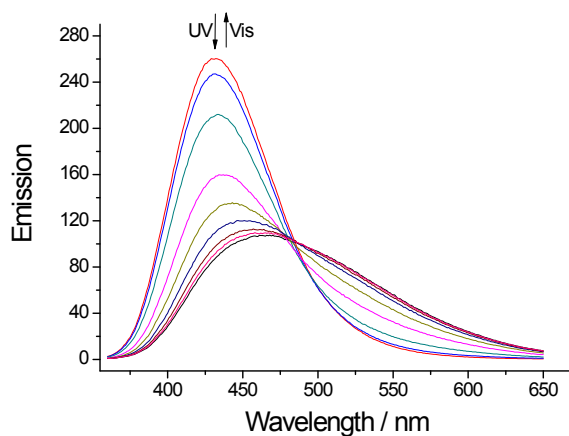
**Figure S5:** Absorbance responses of diarylethene **DIN•2Na** in acetonitrile ( $c = 2 \times 10^{-5}$  mol/L) upon the addition of NaOH (10 eq) upon irradiation with UV light.



**Figure S6:** Fluorescence responses of diarylethene **DIN** in acetonitrile ( $c = 2 \times 10^{-5}$  mol/L) upon the addition of NEt<sub>3</sub> (0-8 eq).



**Figure S7:** Fluorescence changes of diarylethenes **DIN** and **DIN•CH<sub>3</sub>** in acetonitrile ( $c = 2 \times 10^{-5}$  mol/L) before irradiation with UV light.



**Figure S8:** Fluorescence responses of diarylethene **DIN•CH<sub>3</sub>** in acetonitrile ( $c = 2 \times 10^{-5}$  mol/L) upon irradiation with UV and visible light.

**Table S1:** Crystal data and structure refinement parameters for **DIN**.

<b>DIN</b>	
Empirical formula	$C_{23}H_{16}Cl_2N_2O S_2$
Formula weight	471.40
Temperature	298(2) K
Wavelength	0.71073 Å
Crystal system	Orthorhombic
Space group	P2(1)2(1)2(1)
a (Å)	7.366(4)
b (Å)	8.121(4)

c (Å)	36.109(17)
$\alpha$ (deg)	90
$\beta$ (deg)	90
$\gamma$ (deg)	90
Volume(Å <sup>3</sup> )	2160.0(18)
Z	4
Density (calculated)	1.450 Mg/m <sup>3</sup>
Absorption coefficient	0.512 mm <sup>-1</sup>
F(000)	968
Crystal size	0.20 × 0.10 × 0.10 mm <sup>3</sup>
Theta range for data collection	2.57 to 25.99°.
Index ranges	-9<=h<=8, -9<=k<=9, -44<=l<=19
Reflections collected	9666
Independent reflections	4173 [R(int) = 0.1408]
Completeness to theta = 26.00	99.6 %
Absorption correction	None
Max. and min. transmission	0.9506 and 0.9045
Refinement method	Full-matrix least-squares on F <sup>2</sup>
Data / restraints / parameters	4173 / 0 / 274
Goodness-of-fit on F <sup>2</sup>	1.072
Final R indices [I>2sigma(I)]	R1 = 0.0721, wR2 = 0.1689
R indices (all data)	R1 = 0.0999, wR2 = 0.1774
Largest diff. peak and hole	0.294 and -0.309 e <sup>-3</sup>

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