

## Electronic Supplementary Information

# Oligostilbenes from *Vatica mangachapoi* with Xanthine Oxidase and Acetylcholinesterase Inhibitory Activities

Yan Hua Qin, Jie Zhang, Jiang Tao Cui, Zhi Kai Guo, Nan Jiang, Ren Xiang Tan and Hui Ming Ge

### Table of Contents

Figure S1  $^1\text{H}$ -NMR spectrum of compound **1** (DMSO- $d_6$ )

Figure S2  $^{13}\text{C}$ -NMR spectrum of compound **1** (DMSO- $d_6$ )

Figure S3 HMQC spectrum of compound **1** (DMSO- $d_6$ )

Figure S4 HMBC spectrum of compound **1** (DMSO- $d_6$ )

Figure S5  $^1\text{H}$ - $^1\text{H}$  COSY spectrum of compound **1** (DMSO- $d_6$ )

Figure S6 NOESY spectrum of compound **1** ((DMSO- $d_6$ )

Figure S7 HREIMS spectrum of compound **1** (positive mode)

Figure S8 FT-IR spectrum of compound **1** (KBr)

Figure S9 The theoretically predicted UV spectrum for **1**

Figure S10 UV spectrum of compound **1** (CH<sub>3</sub>OH)

Figure S11  $^1\text{H}$ -NMR spectrum of compound **2** (DMSO- $d_6$ )

Figure S12  $^{13}\text{C}$ -NMR spectrum of compound **2** (DMSO- $d_6$ )

Figure S13 HMQC spectrum of compound **2** (DMSO- $d_6$ )

Figure S14 HMBC spectrum of compound **2** (DMSO- $d_6$ )

Figure S15  $^1\text{H}$ - $^1\text{H}$  COSY spectrum of compound **2** (DMSO- $d_6$ )

Figure S16 NOESY spectrum of compound **2** ((DMSO- $d_6$ )

Figure S17 HREIMS spectrum of compound **2** (positive mode)

Figure S18 FT-IR spectrum of compound **2** (KBr)

Figure S19 UV spectrum of compound **2** (CH<sub>3</sub>OH)

Figure S20  $^1\text{H}$ -NMR spectrum of compound **3** (acetone- $d_6$ )

Figure S21  $^{13}\text{C}$ -NMR spectrum of compound **3** (acetone - $d_6$ )

Figure S22 HSQC spectrum of compound **3** (acetone - $d_6$ )

Figure S23 HMBC spectrum of compound **3** (acetone - $d_6$ )

Figure S24  $^1\text{H}$ - $^1\text{H}$  COSY spectrum of compound **3** (acetone - $d_6$ )

Figure S25 ROESY spectrum of compound **3** (acetone - $d_6$ )

Figure S26 HREIMS spectrum of compound **3** (positive mode)

Figure S27 FT-IR spectrum of compound **3** (KBr)

Figure S28 UV spectrum of compound **3** (CH<sub>3</sub>OH)

Table S1 The optimized structural information for **1** within polarizable continuum model

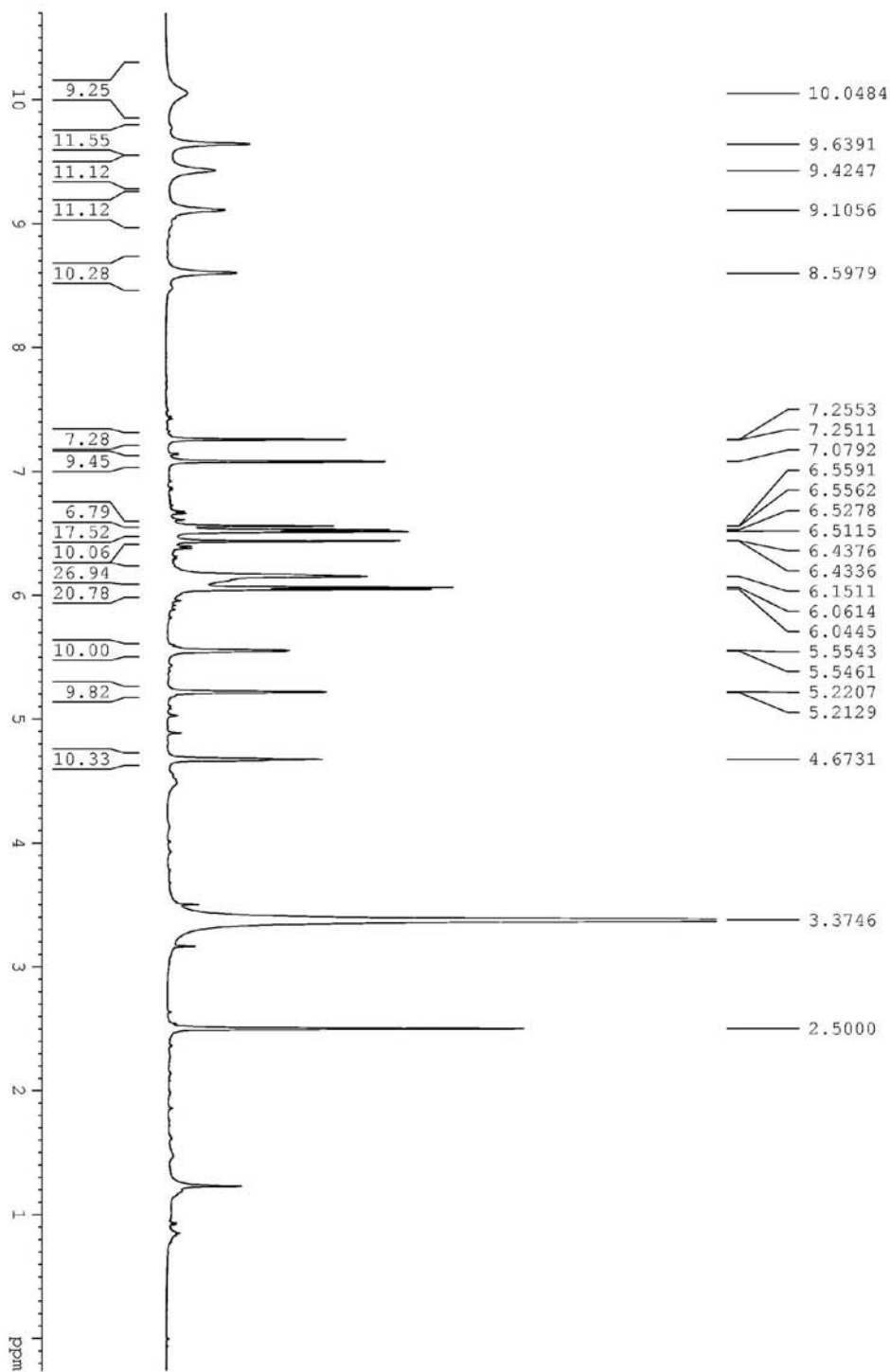
Table S2 TDDFT results for the optimized conformer of (7*bS*)-**1** (200nm< $\lambda$ <400nm)

Table S3 The optimized structural information for **2** within polarizable continuum model

Table S4 TDDFT results for the optimized conformer of (7*aR*, 8*aR*, 7*bR*)-**2** (200nm< $\lambda$ <400nm)

Table S5 The optimized structural information for **3** within polarizable continuum model

Table S6 TDDFT results for the optimized conformer of (7*aR*, 8*aR*)-**3** (200nm< $\lambda$ <400nm)



qyh-1 DMSO-d6 BRUKER DRX500 08-12-22

Figure S1  $^1\text{H-NMR}$  spectrum of compound **1** ( $\text{DMSO-}d_6$ )

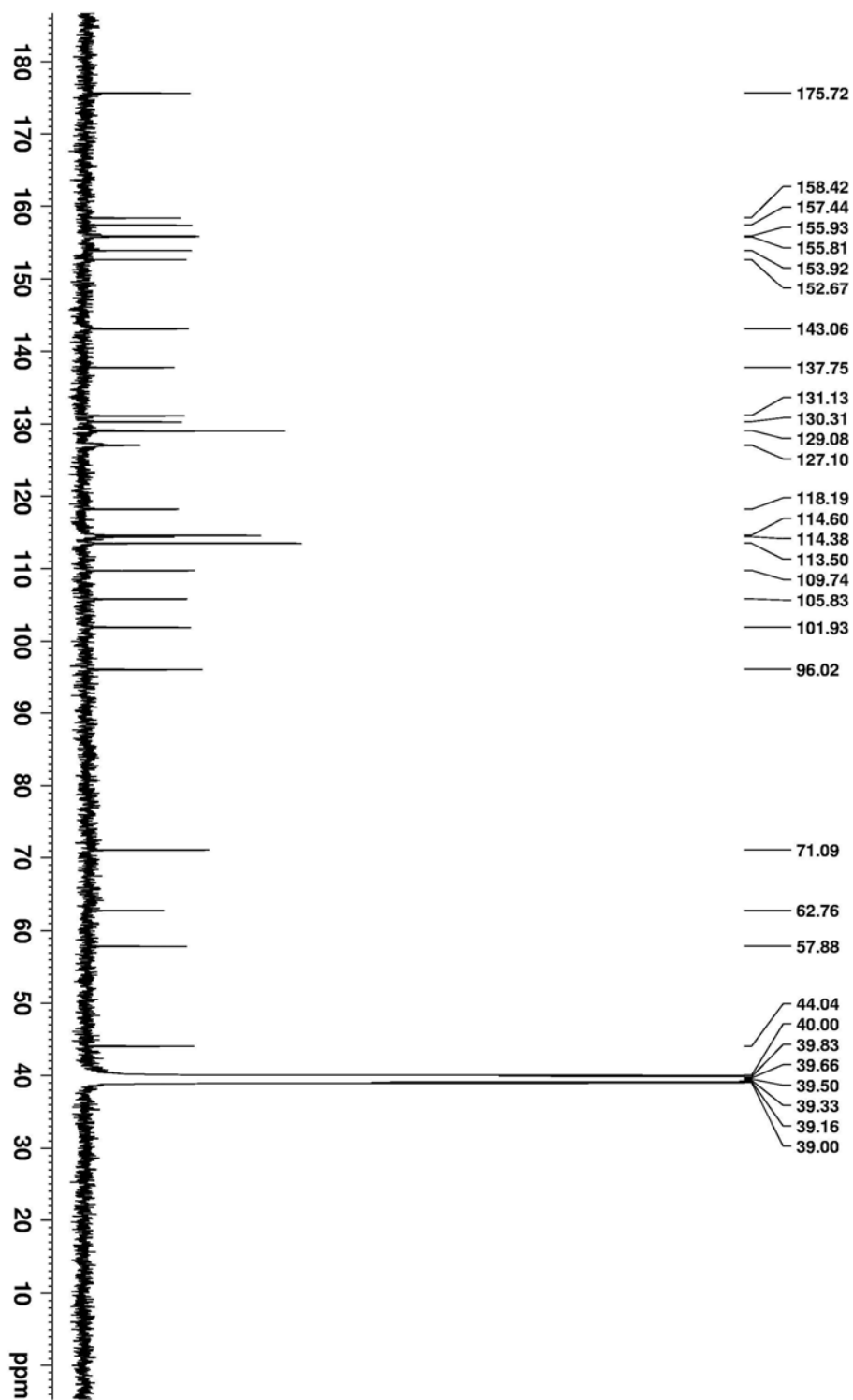


Figure S2  $^{13}\text{C}$ -NMR spectrum of compound **1** (DMSO- $d_6$ )

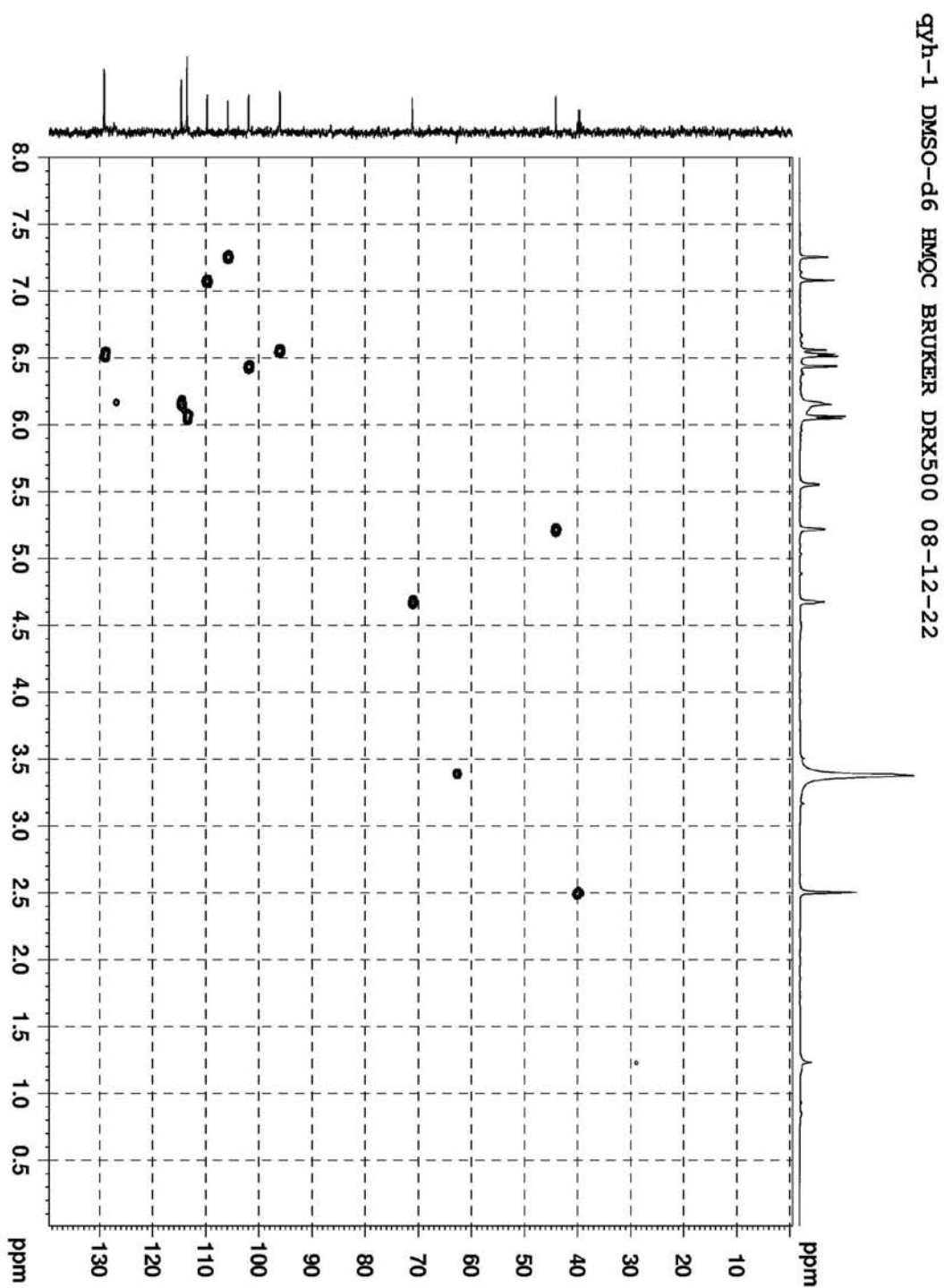


Figure S3 HMQC spectrum of compound **1** (DMSO-*d*<sub>6</sub>)

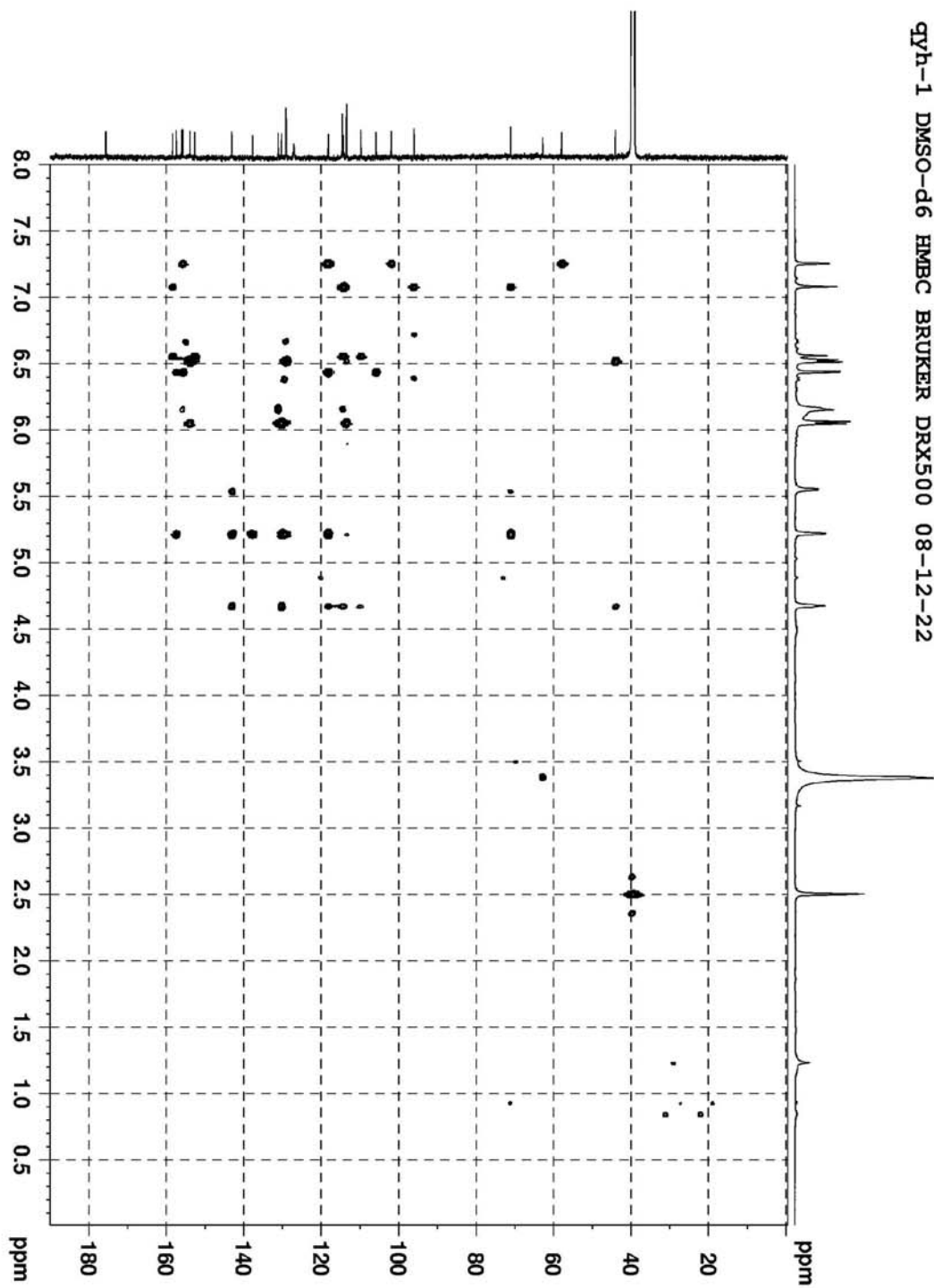


Figure S4 HMBC spectrum of compound **1** (DMSO-*d*<sub>6</sub>)

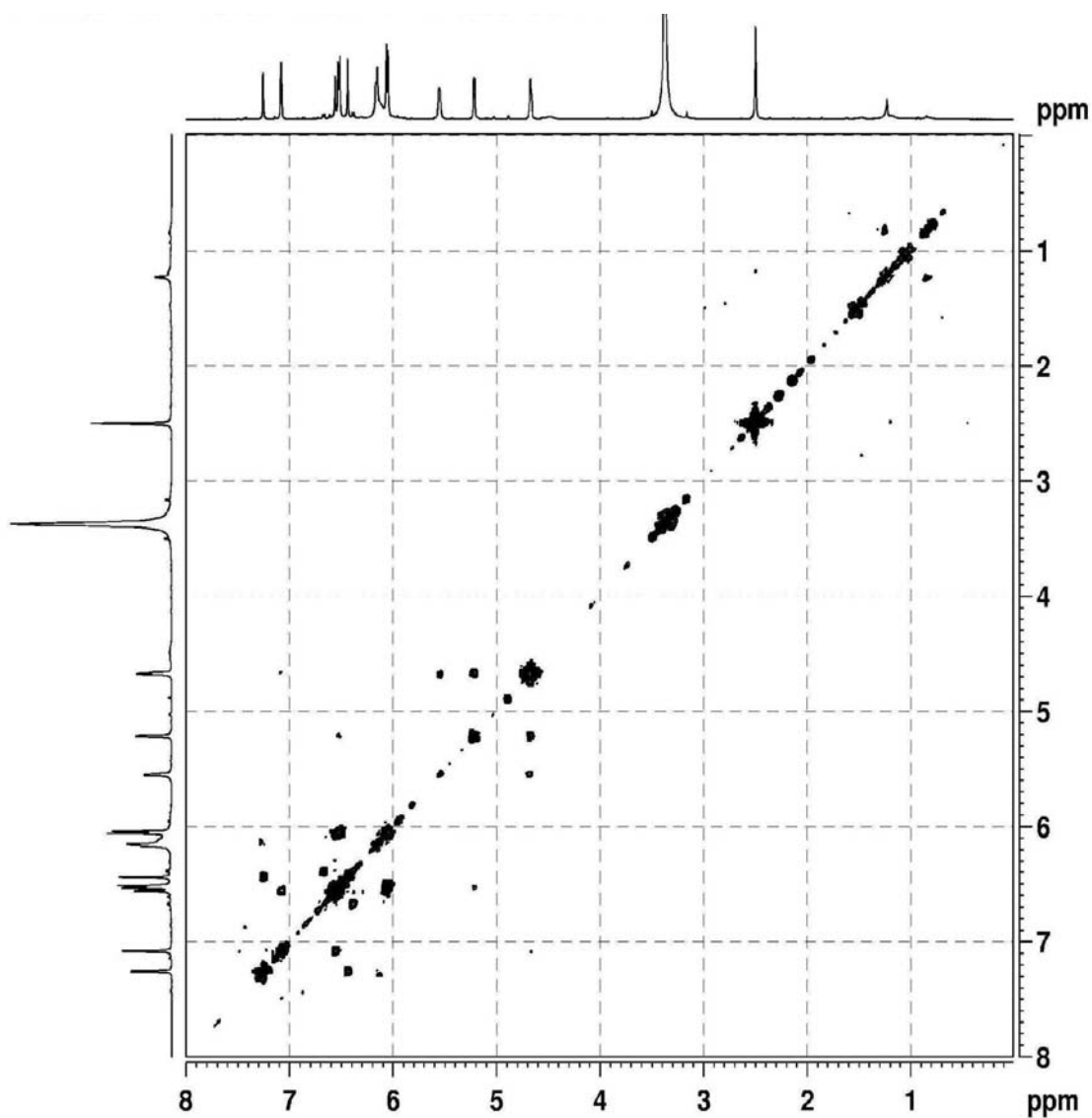


Figure S5  $^1\text{H}$ - $^1\text{H}$  COSY spectrum of compound **1** ( $\text{DMSO}-d_6$ )

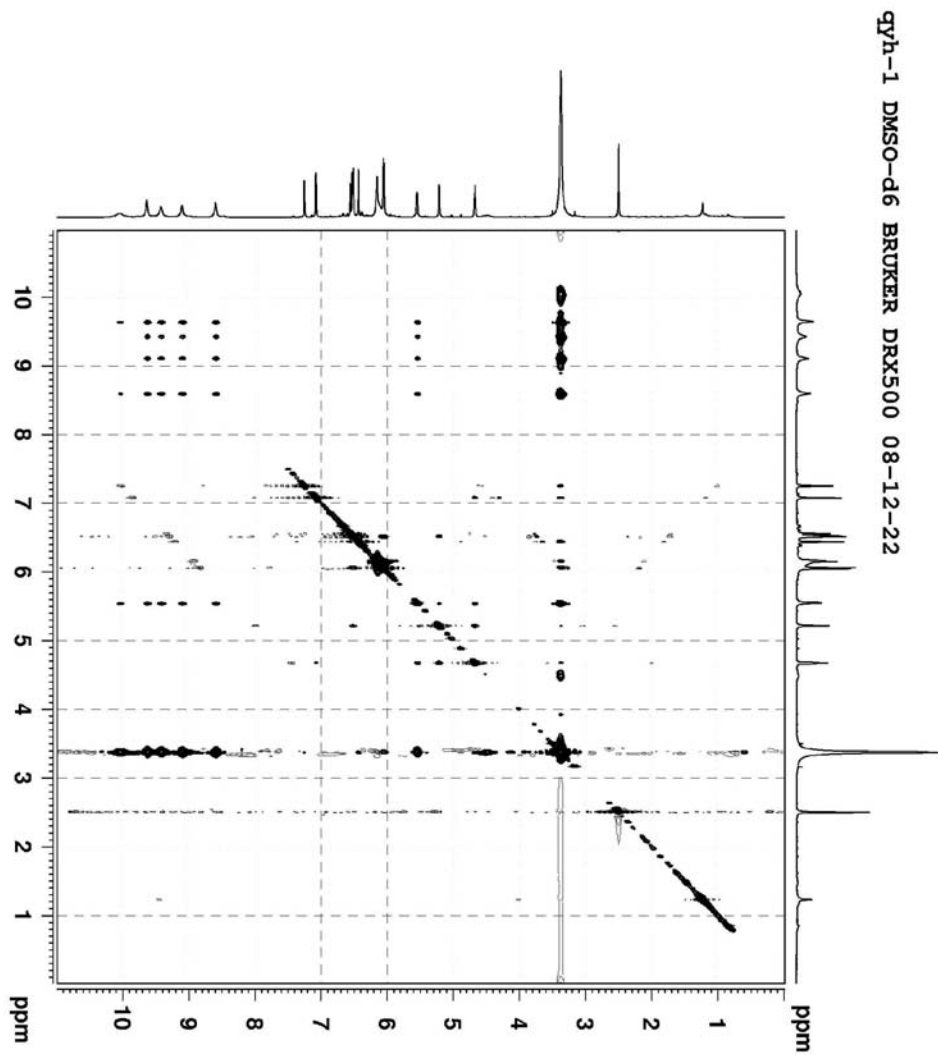


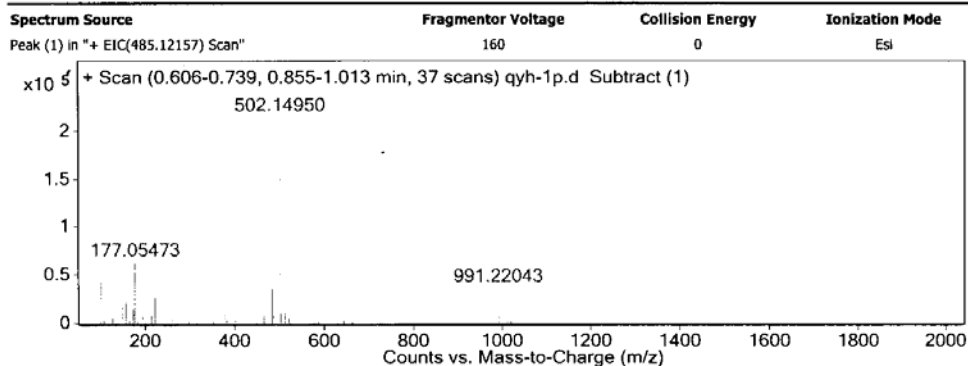
Figure S6 NOESY spectrum of compound **1** (DMSO- $d_6$ )



## Qualitative Analysis Report

<b>Data Filename</b>	qyh-1p.d	<b>Sample Name</b>	qyh-1
<b>Sample Type</b>	Sample	<b>Position</b>	P1-D1
<b>Instrument Name</b>	Instrument 1	<b>User Name</b>	
<b>Acq Method</b>	positive_2.m	<b>IRM Calibration Status</b>	XXXXXXXXXX
<b>DA Method</b>	Default.m	<b>Comment</b>	

### User Spectra



#### Peak List

m/z	z	Abund.	Formula	Ion
102.12795		42352		
149.02351		43355		
177.05473		64307		
223.09286		26539		
245.0785		23818		
485.12299		35487	C <sub>28</sub> H <sub>21</sub> O <sub>8</sub>	(M+H) <sup>+</sup>
502.1495	1	226394		
503.15258	1	62350		
507.10472	1	55020		
991.22043	1	36900		

--- End Of Report ---

Figure S7 HRESIMS spectrum of compound **1** (positive mode)

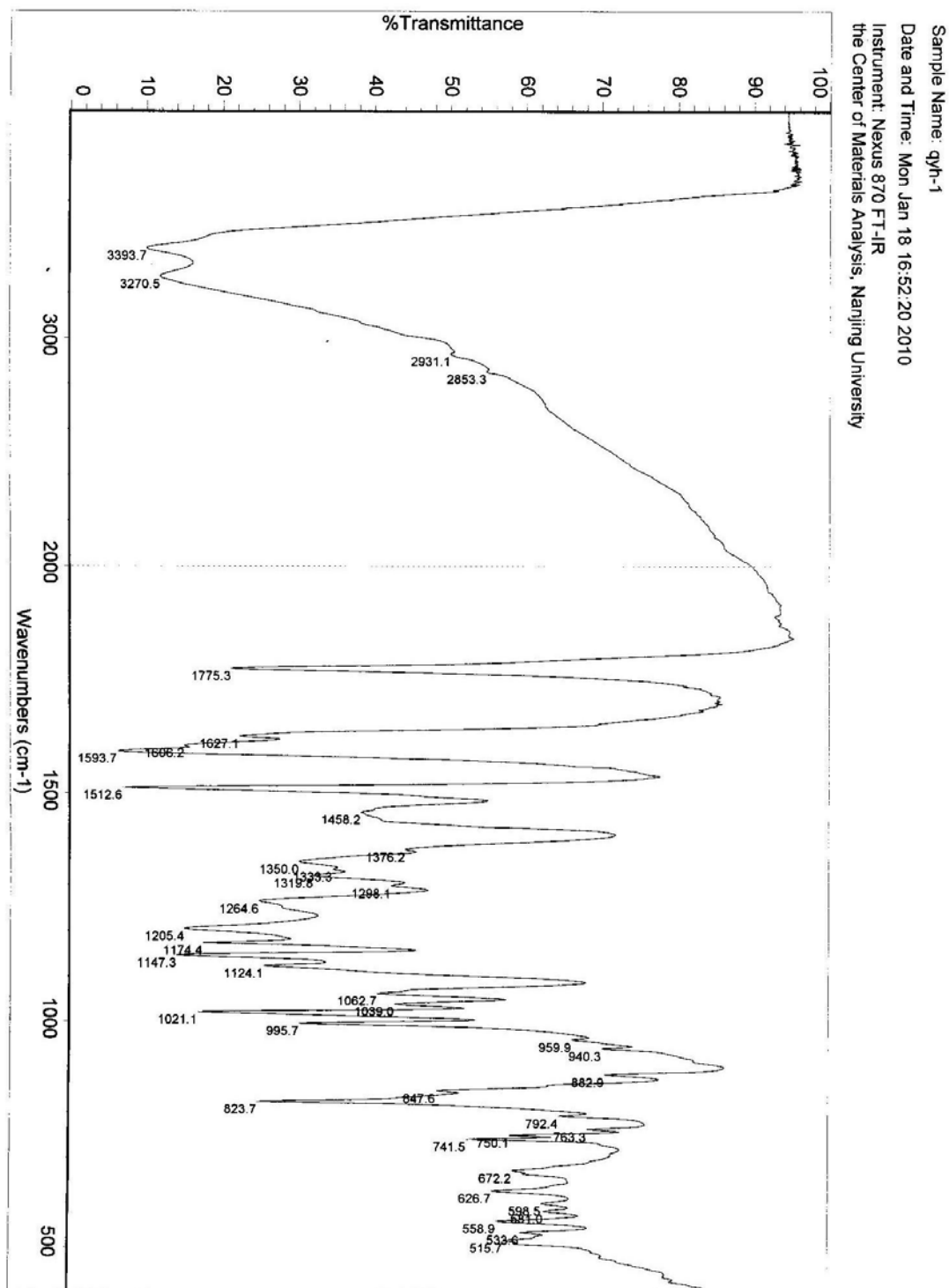


Figure S8 FT-IR spectrum of compound **1** (KBr)

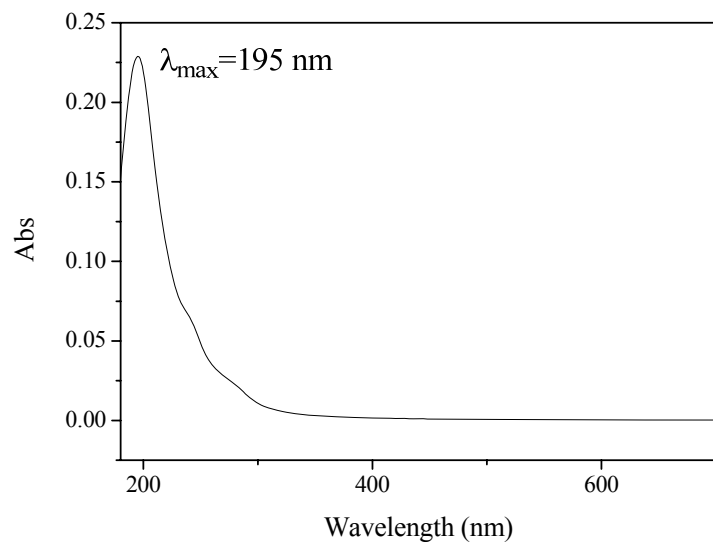
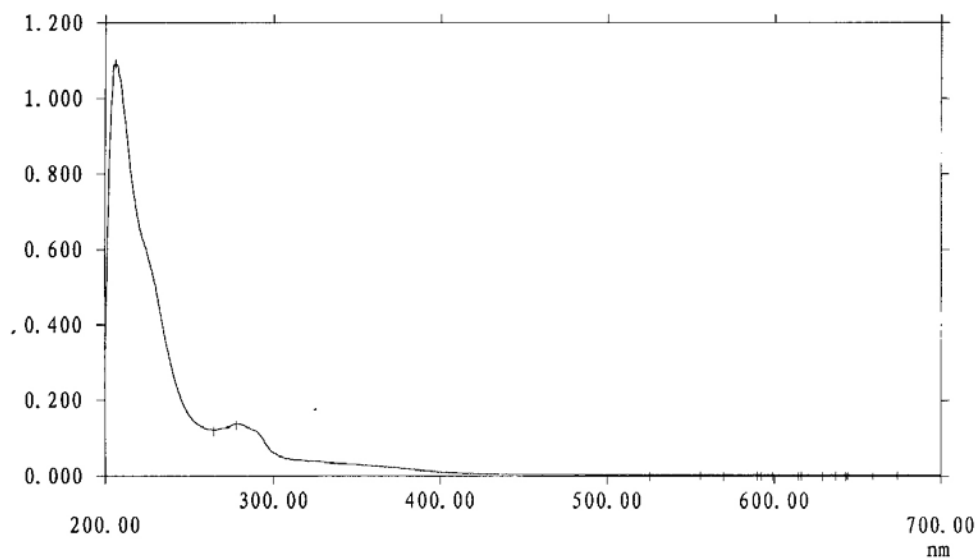


Figure S9 The theoretically predicted UV spectrum for **1**. The calculations are carried out at B3LYP/6-31G(d) level within the polarizable continuum model (CH<sub>3</sub>OH solvent: dielectric constant  $\epsilon = 32.63$ ).



No.	WL (nm)	Peak (Abs)	No.	WL (nm)	Valley (Abs)
1	673.50	0.004	1	658.50	0.003
2	643.50	0.004	2	642.50	0.003
3	636.50	0.004	3	628.50	0.003
4	616.00	0.004	4	614.00	0.003
5	598.50	0.004	5	592.00	0.003
6	589.50	0.004	6	569.50	0.003
7	556.00	0.004	7	525.50	0.003
8	277.50	0.137	8	264.00	0.121
9	206.00	1.096			

Figure S10 UV spectrum of compound **1** (CH<sub>3</sub>OH)

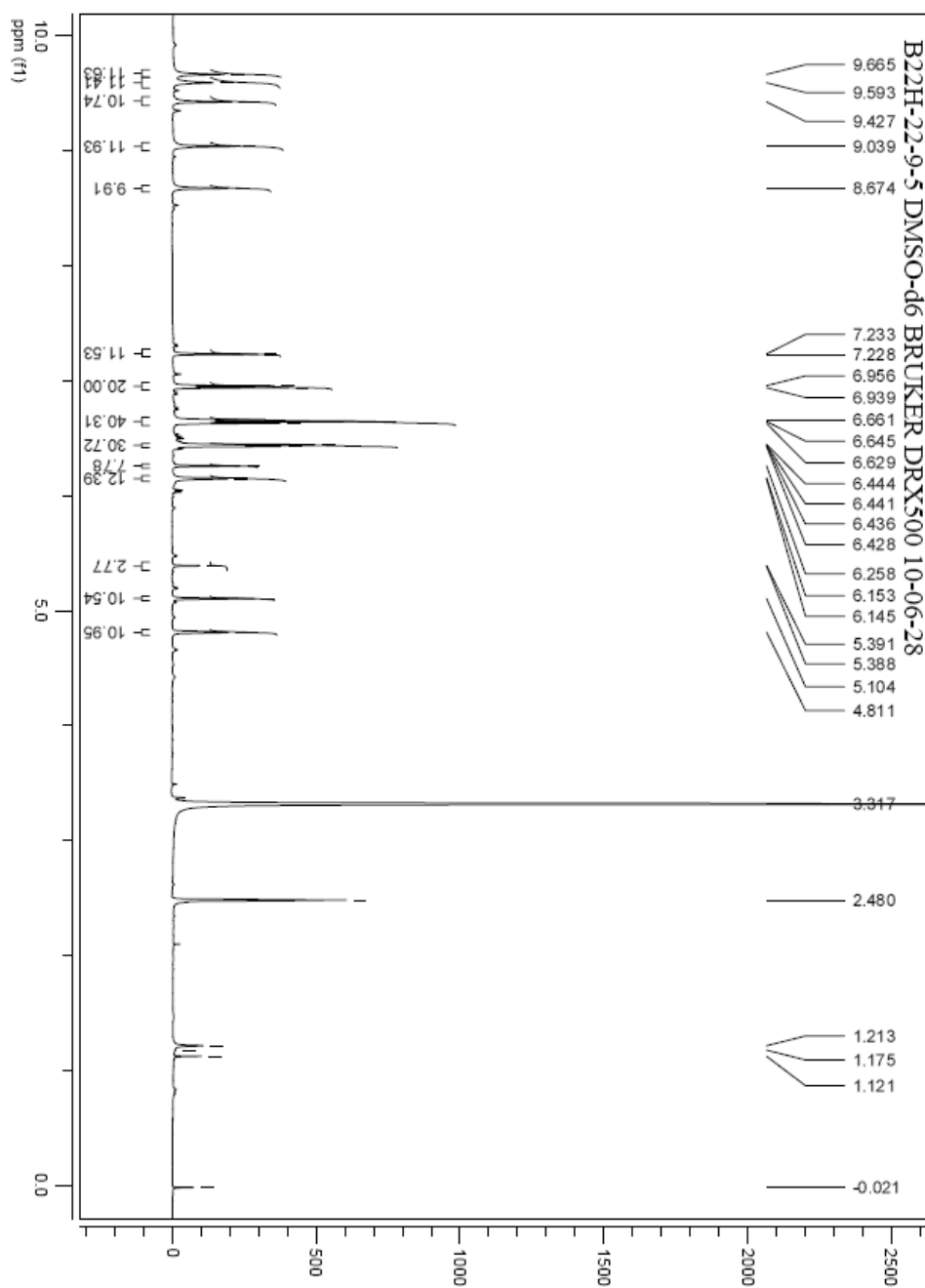


Figure S11  $^1\text{H}$ -NMR spectrum of compound **2** ( $\text{DMSO-}d_6$ )

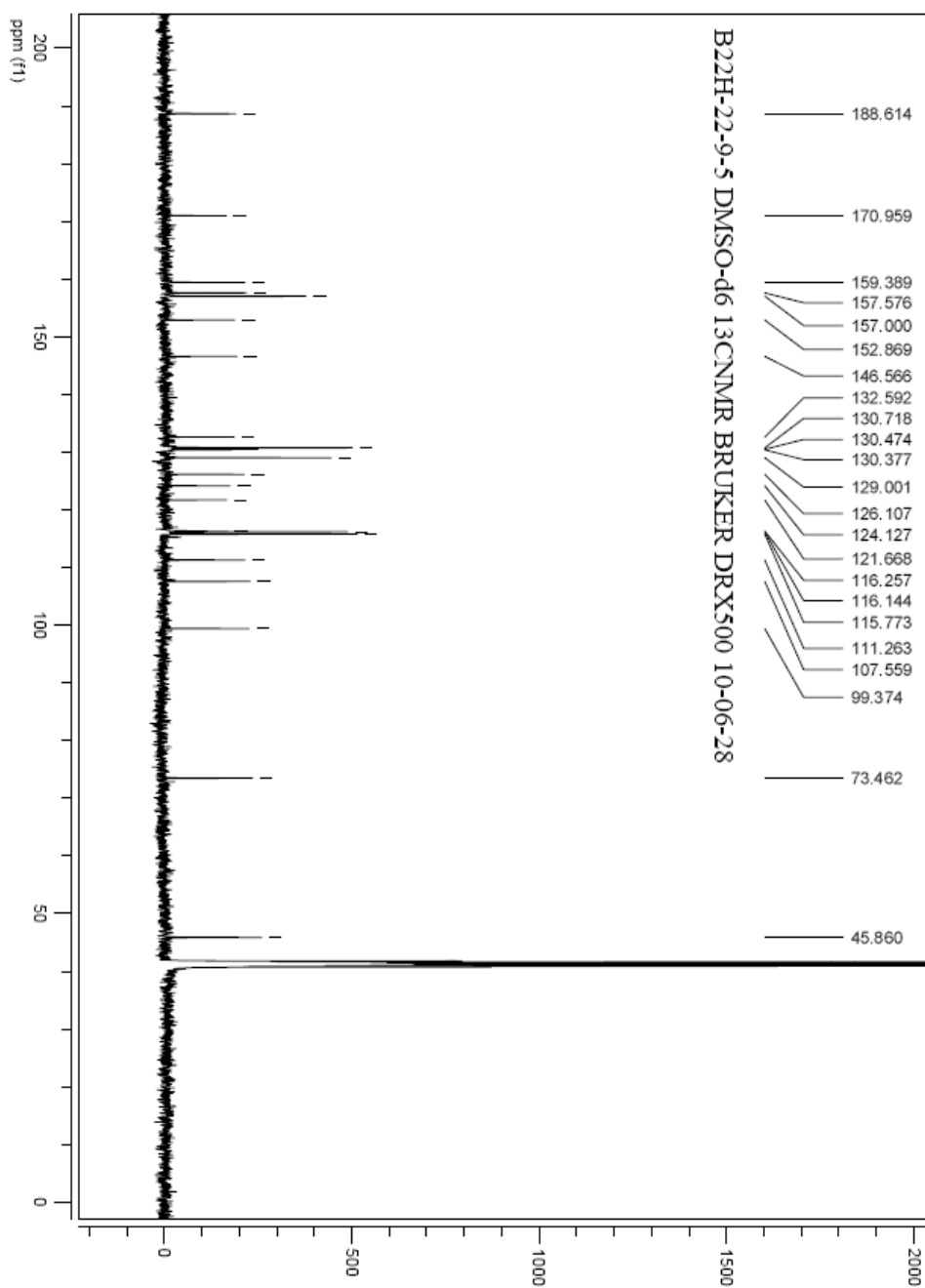


Figure S12  $^{13}\text{C}$ -NMR spectrum of compound **2** ( $\text{DMSO-}d_6$ )

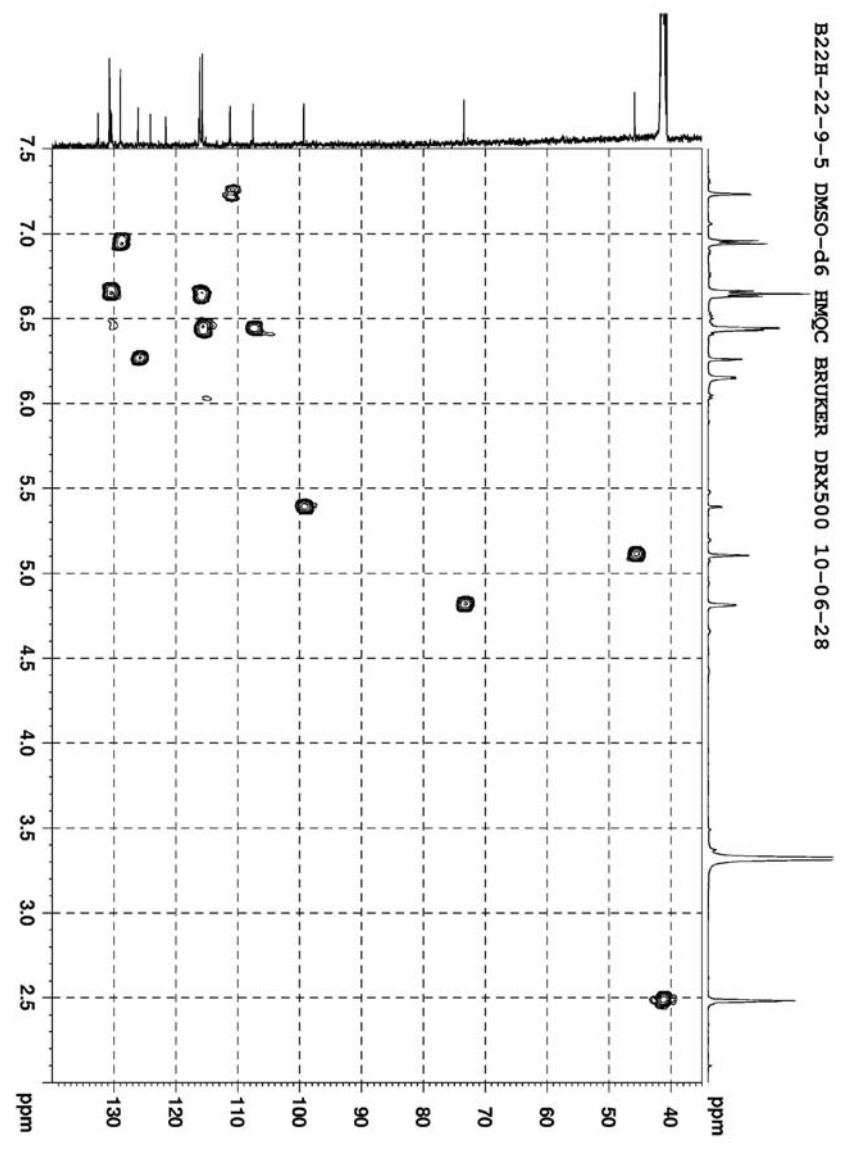


Figure S13 HMQC spectrum of compound **2** (DMSO-*d*<sub>6</sub>)

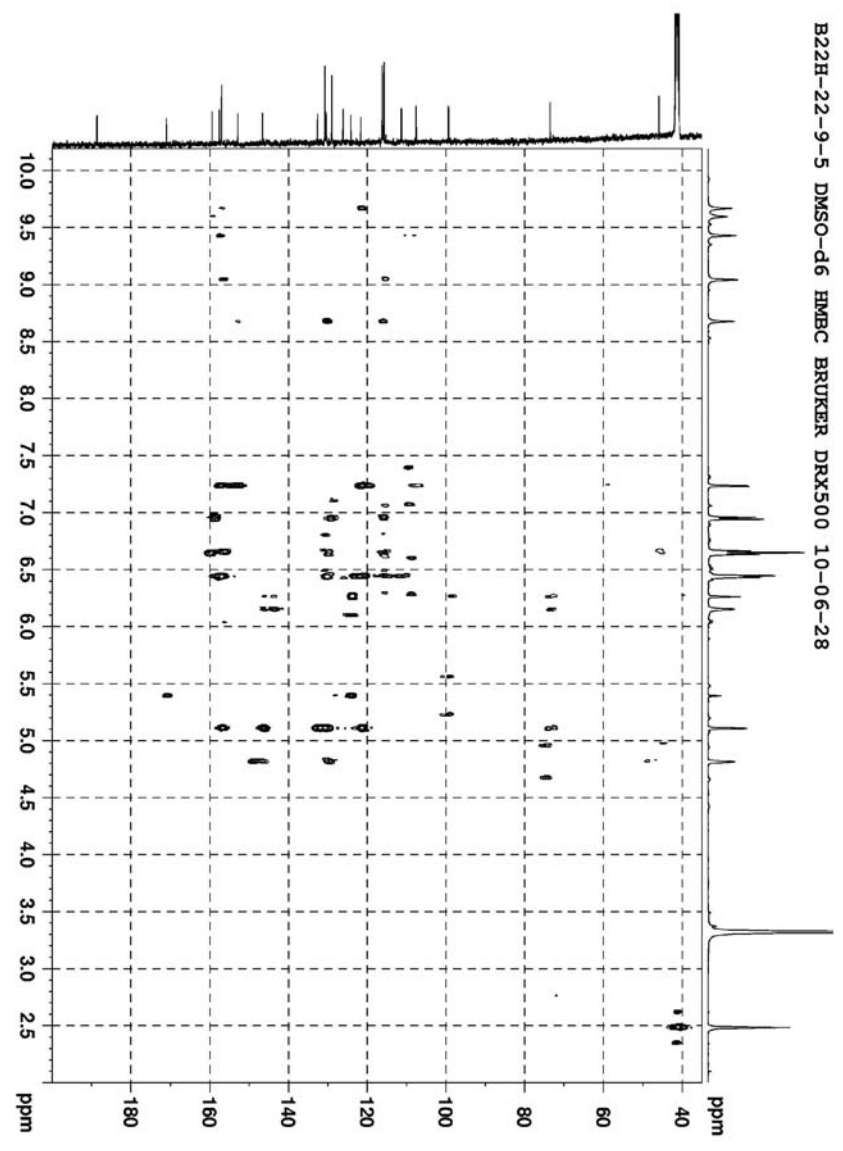


Figure S14 HMBC spectrum of compound 2 (DMSO-*d*<sub>6</sub>)



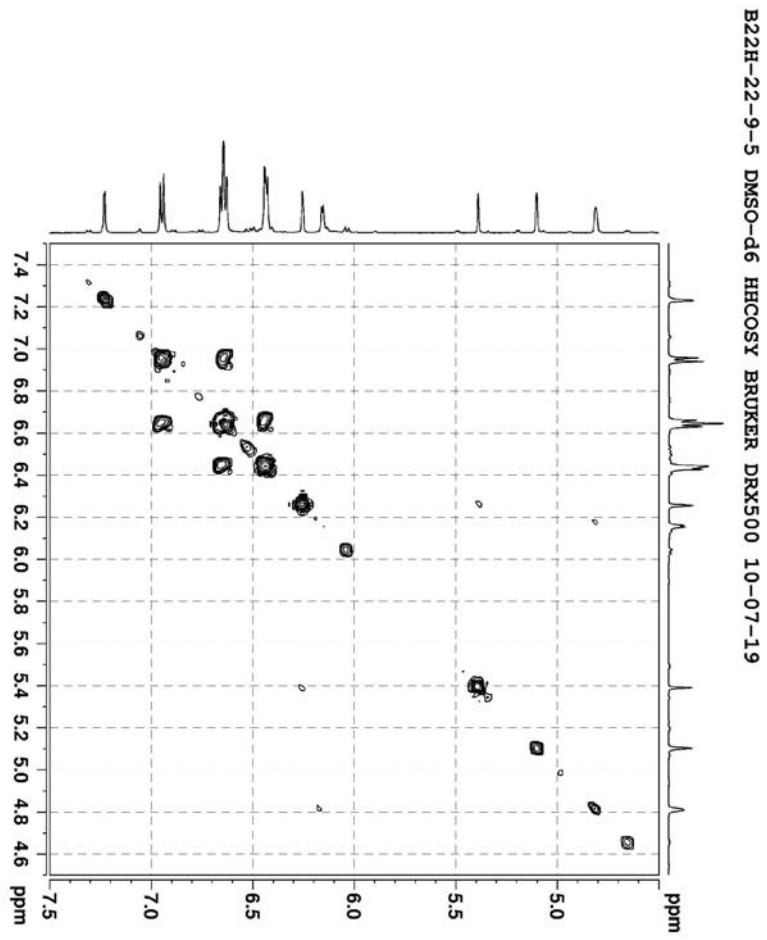


Figure S15 <sup>1</sup>H-<sup>1</sup>H COSY spectrum of compound 2 (DMSO-*d*<sub>6</sub>)

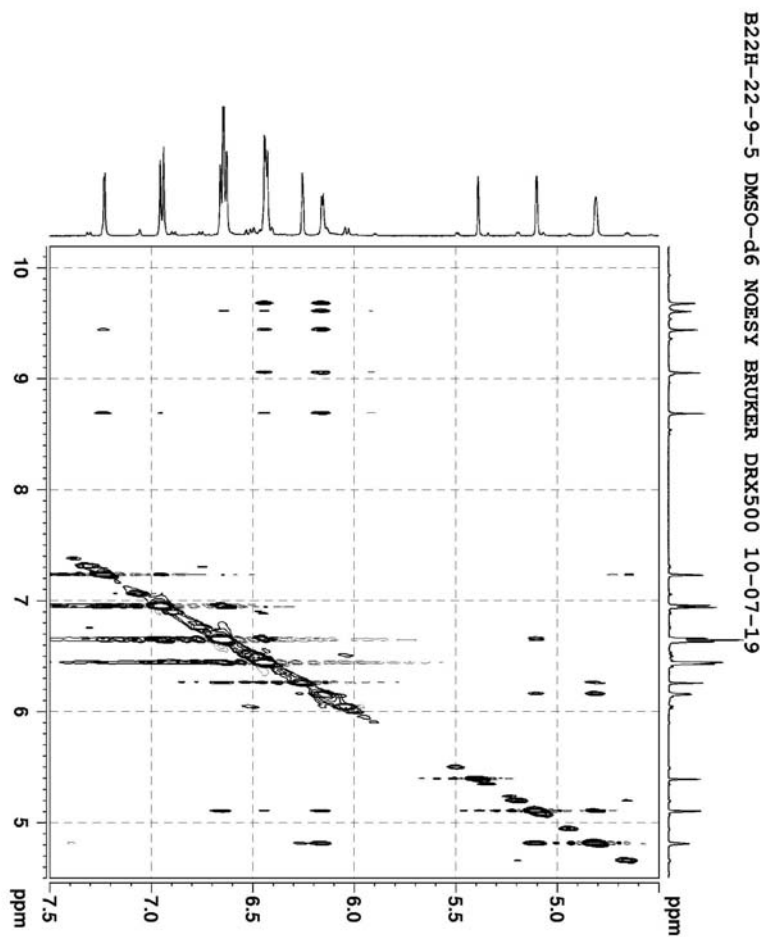
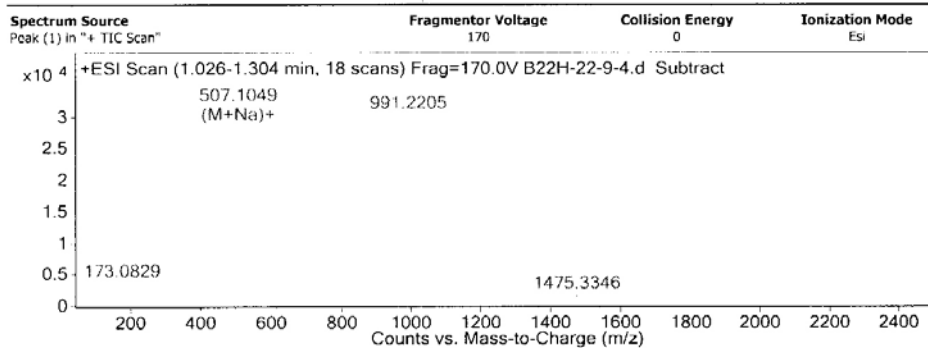


Figure S16 NOESY spectrum of compound **2** (DMSO- $d_6$ )

### Qualitative Analysis Report

<b>Data Filename</b>	B22H-22-9-4.d	<b>Sample Name</b>	B22H-22-9-4
<b>Sample Type</b>	Sample	<b>Position</b>	P1-D5
<b>Instrument Name</b>	Instrument 1	<b>User Name</b>	
<b>Acq Method</b>	TEMP2.m	<b>Acquired Time</b>	5/5/2010 6:48:07 PM
<b>IRM Calibration Status</b>	Some Ions Missed	<b>DA Method</b>	Default.m
<b>Comment</b>			

#### User Spectra



Peak List				
m/z	z	Abund	Formula	Ion
173.0829		3876		
485.1232	1	15882	C28 H21 O8	(M+H)+
486.1264	1	4303	C28 H21 O8	(M+H)+
507.1049	1	33319	C28 H20 Na O8	(M+Na)+
508.1086	1	8317	C28 H20 Na O8	(M+Na)+
955.2047	1	8060		
956.2079	1	3540		
991.2205	1	30555		
992.2241	1	16822		
993.2264	1	5094		

--- End Of Report ---

Figure S17 HRESIMS spectrum of compound **2** (positive mode)

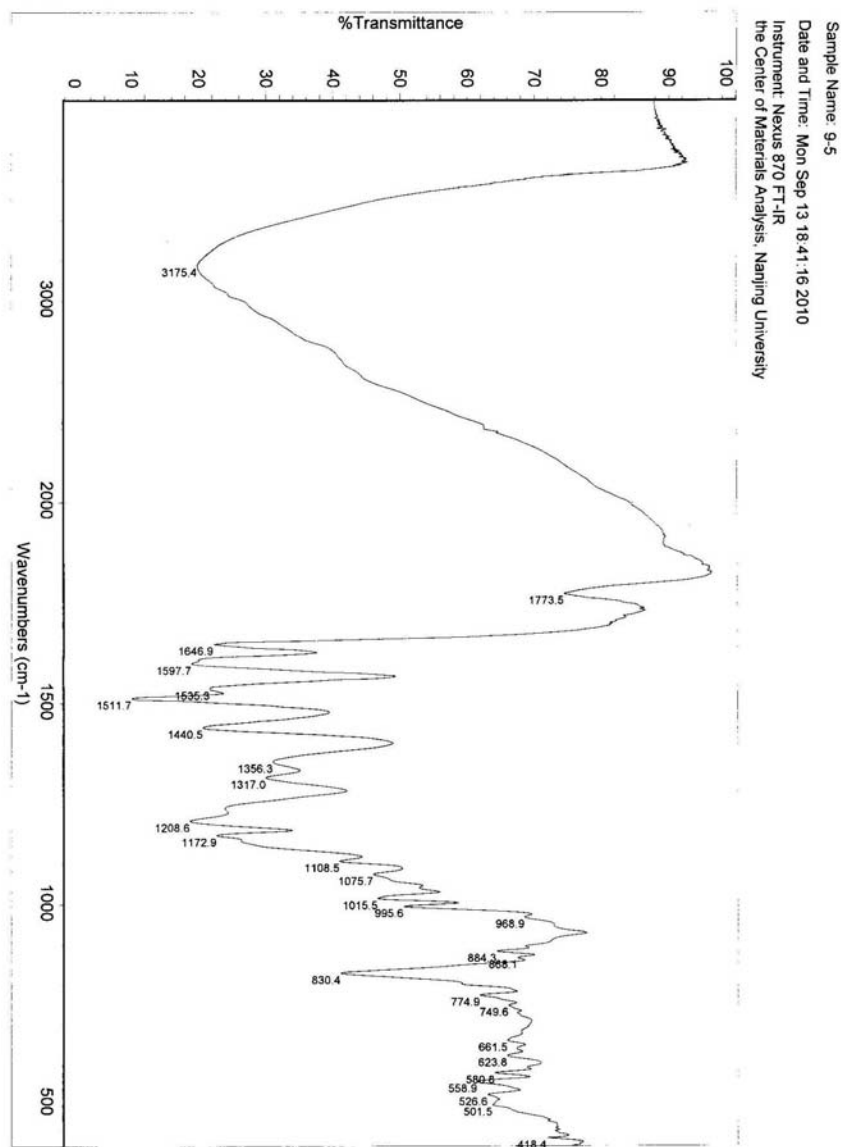


Figure S18 FT-IR spectrum of compound **2** (KBr)

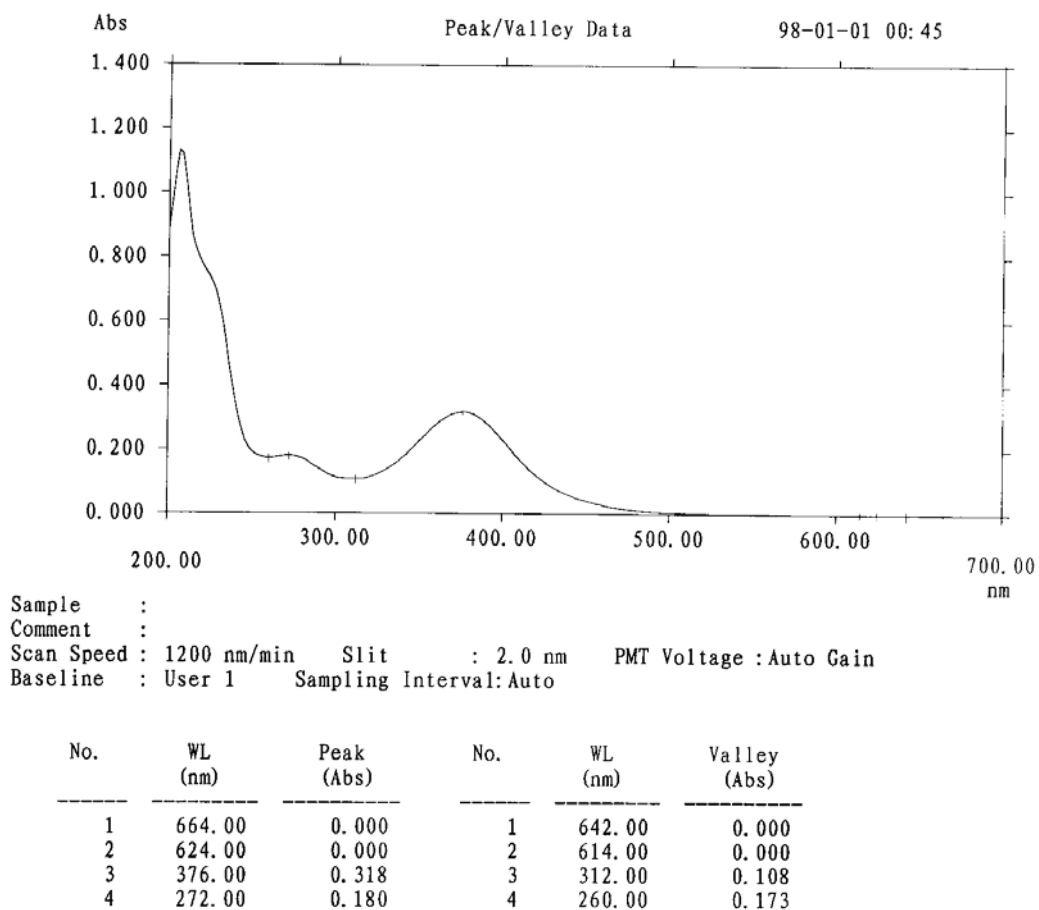


Figure S19 UV spectrum of compound 2 (CH<sub>3</sub>OH)

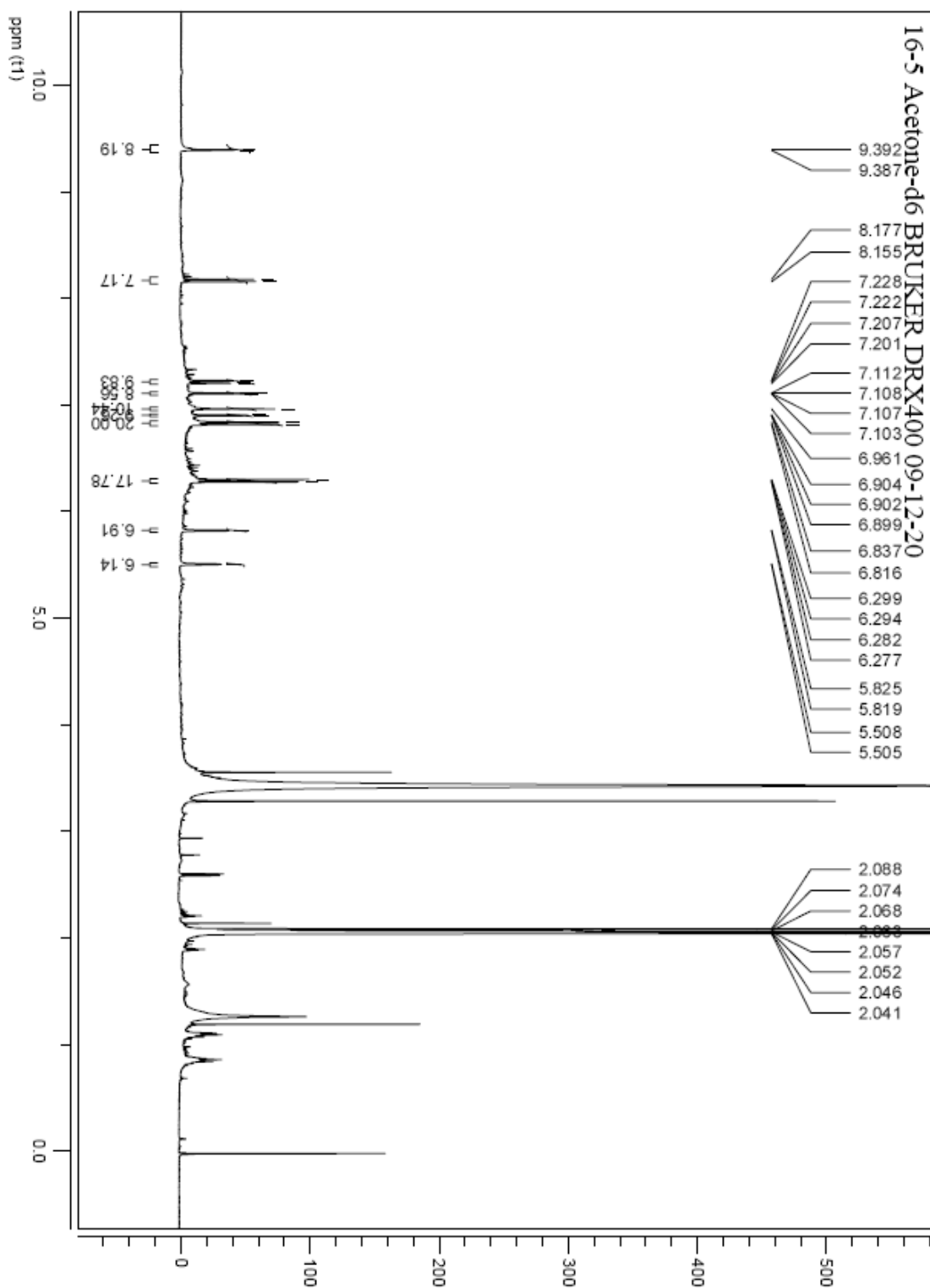


Figure S20  $^1\text{H-NMR}$  spectrum of compound **3** (acetone- $d_6$ )

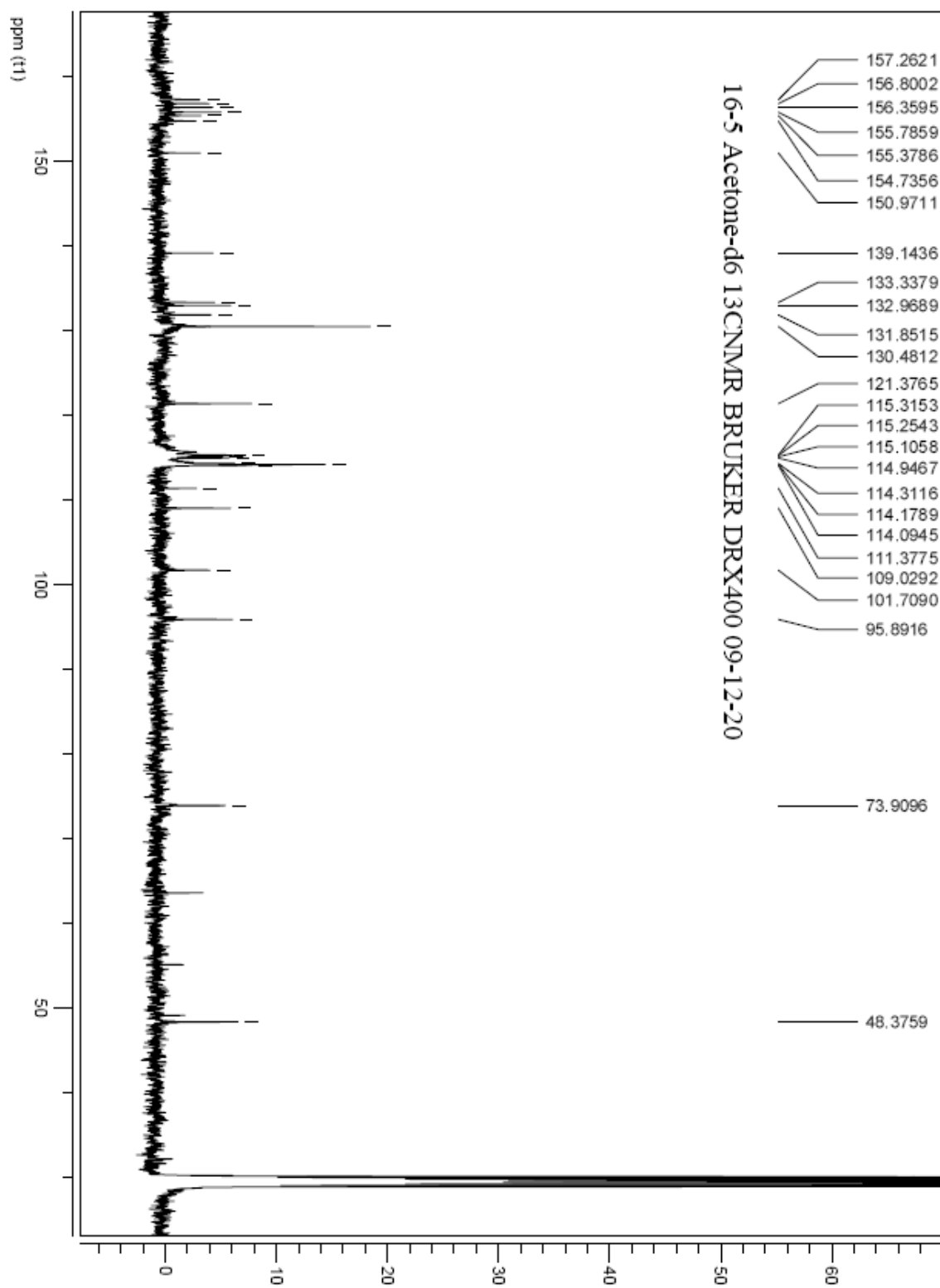


Figure S21  $^{13}\text{C}$ -NMR spectrum of compound **3** (acetone- $d_6$ )

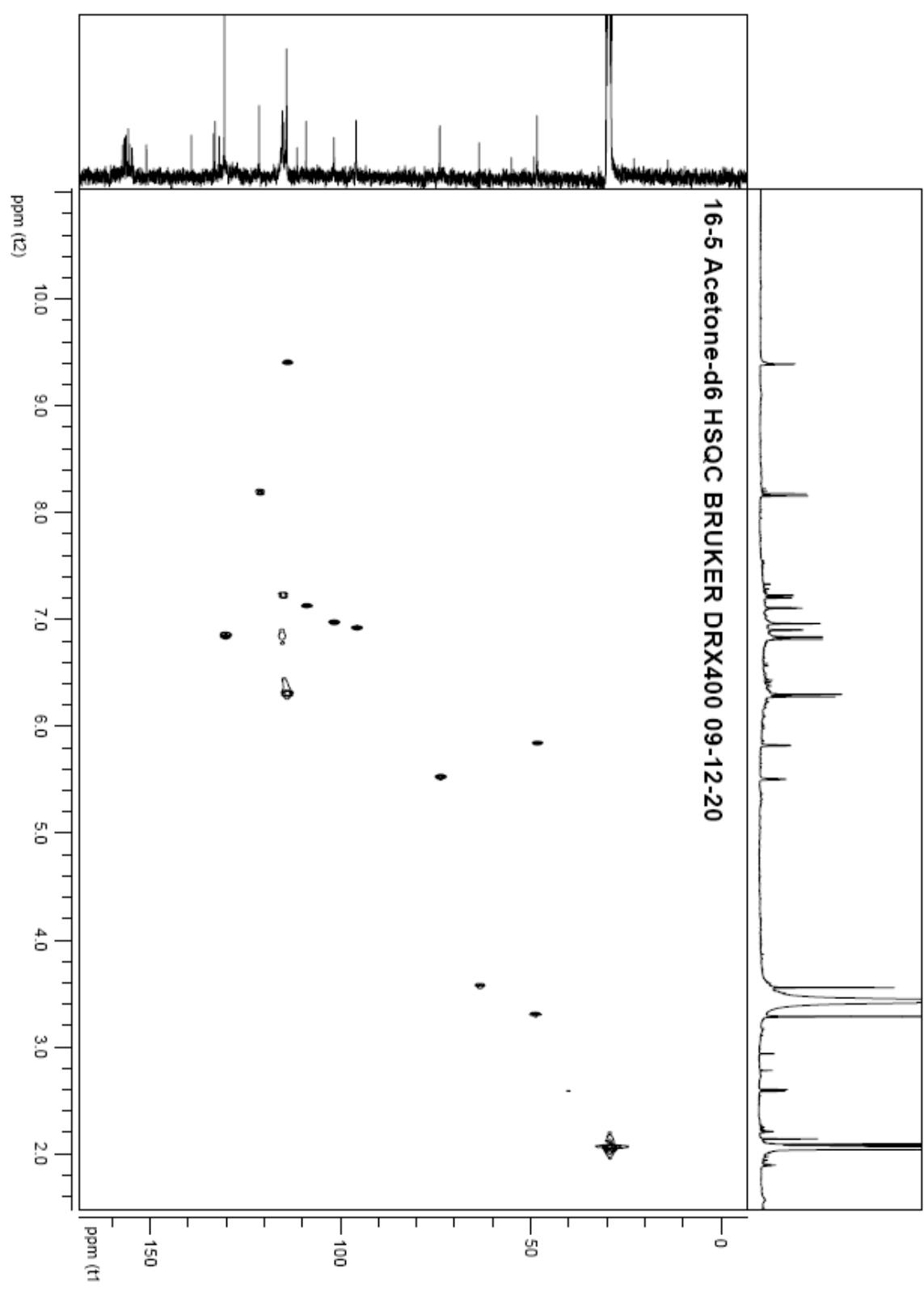


Figure S22 HSQC spectrum of compound **3** (acetone- $d_6$ )



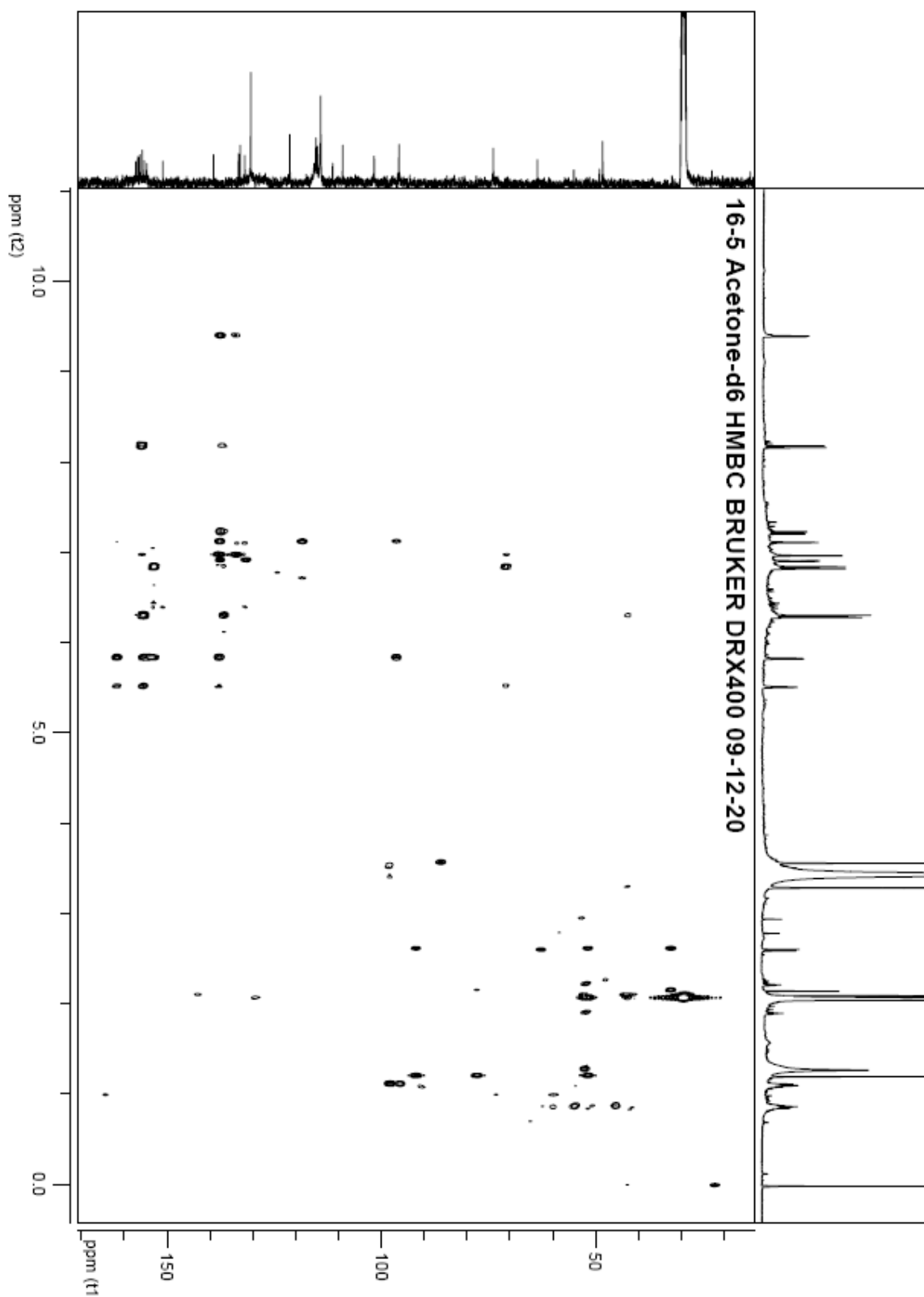


Figure S23 HMBC spectrum of compound **3** (acetone- $d_6$ )

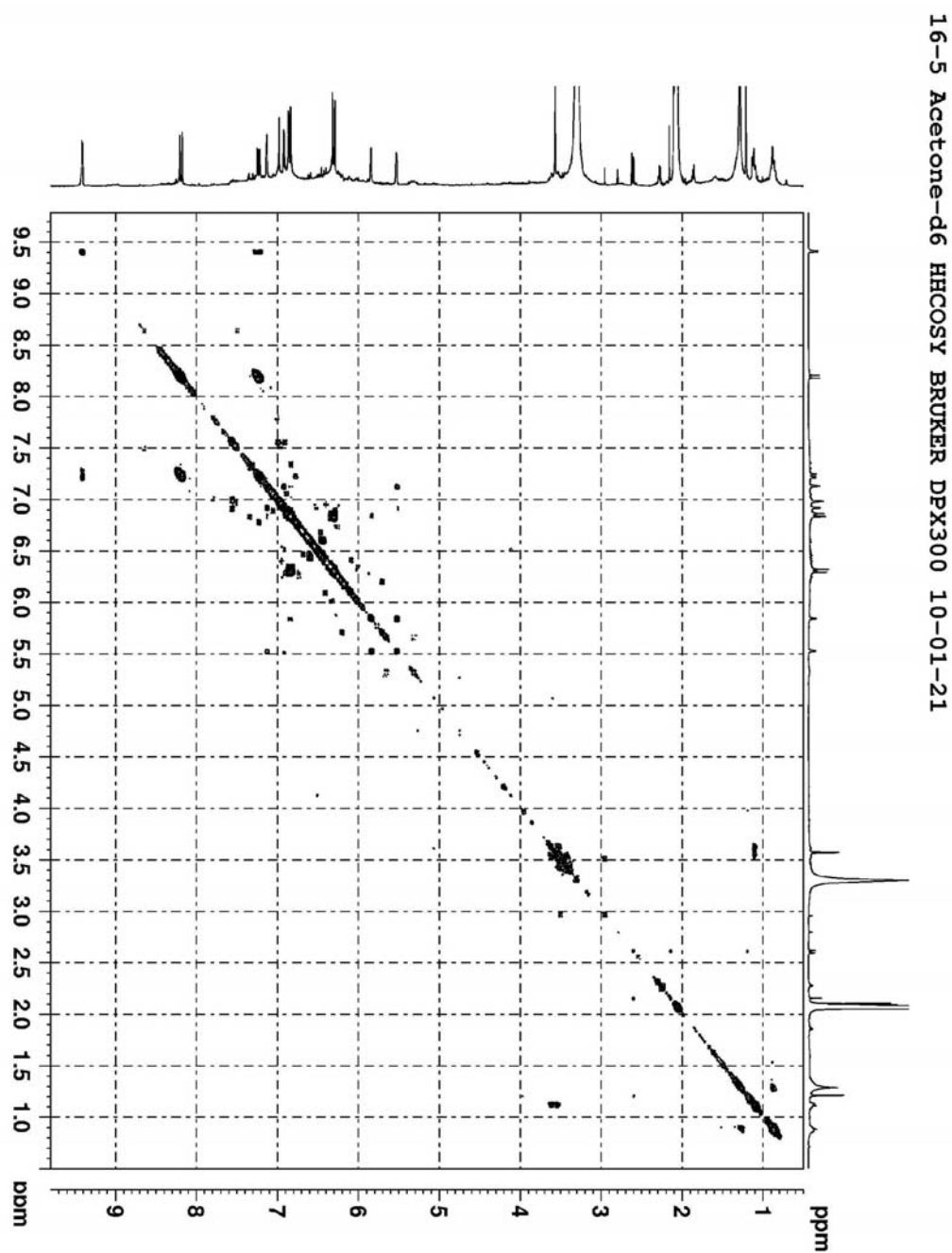


Figure S24  $^1\text{H}$ - $^1\text{H}$  COSY spectrum of compound **3** (acetone- $d_6$ )

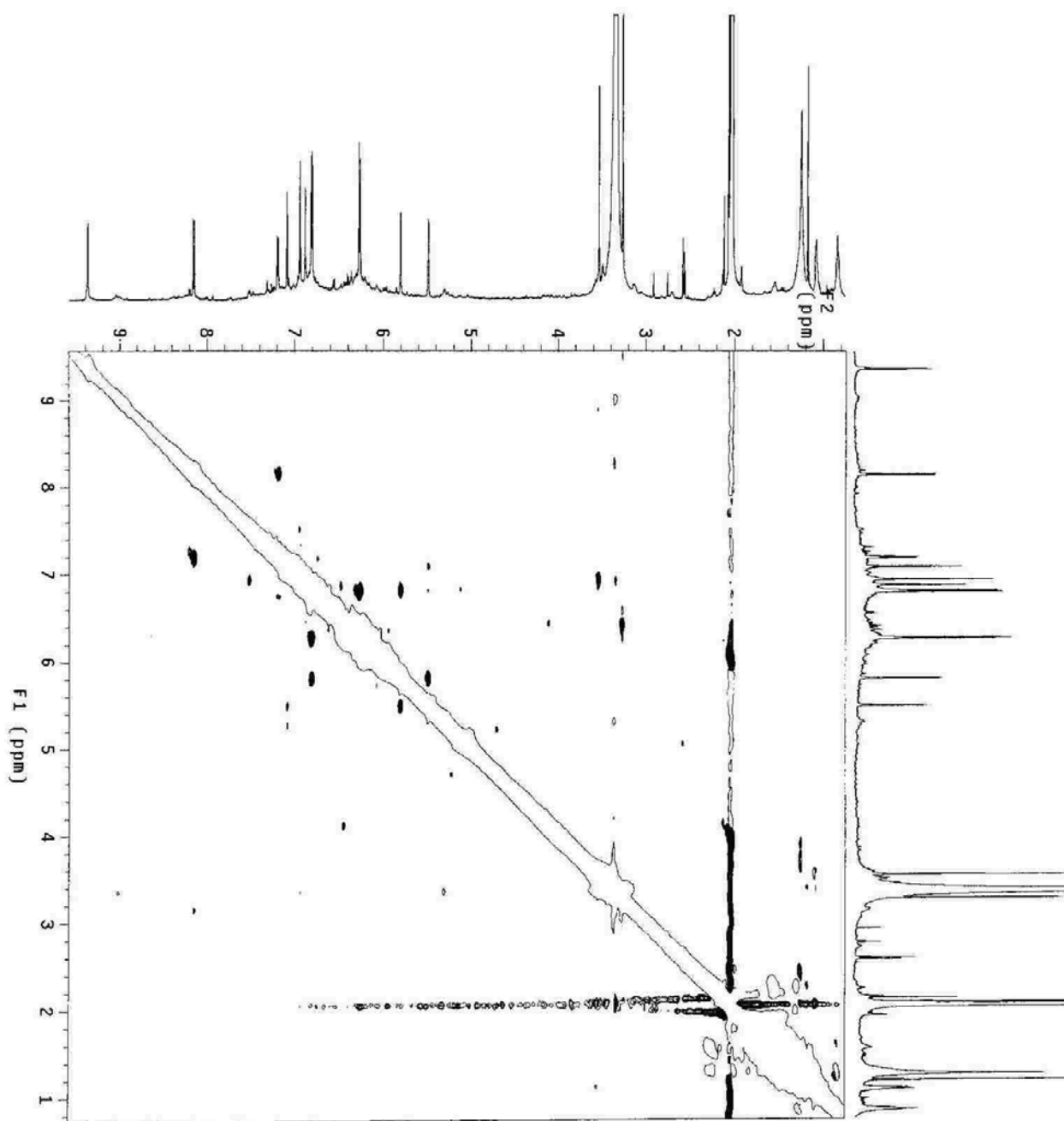
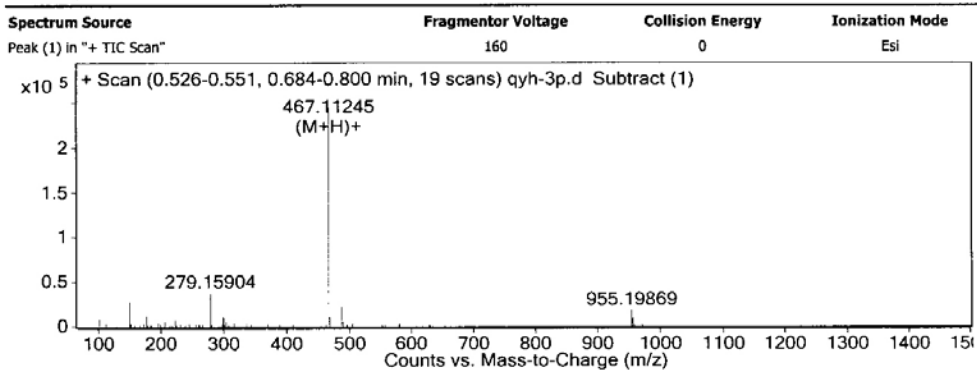


Figure S25 ROESY spectrum of compound **3** (acetone- $d_6$ )

## Qualitative Analysis Report

<b>Data Filename</b>	qyh-3p.d	<b>Sample Name</b>	qyh-3
<b>Sample Type</b>	Sample	<b>Position</b>	P1-D3
<b>Instrument Name</b>	Instrument 1	<b>User Name</b>	
<b>Acq Method</b>	positive.m	<b>IRM Calibration Status</b>	SH/04/08
<b>DA Method</b>	Default.m	<b>Comment</b>	

### User Spectra



Peak List

m/z	z	Abund.	Formula	Ion
149.02341		27281		
279.15904		37659		
467.11245	1	244253	C <sub>28</sub> H <sub>19</sub> O <sub>7</sub>	(M+H) <sup>+</sup>
467.37867		15248		
468.1153	1	67709	C <sub>28</sub> H <sub>19</sub> O <sub>7</sub>	(M+H) <sup>+</sup>
489.09437		22240		
955.19869		17157		

--- End Of Report ---

Figure S26 HRESIMS spectrum of compound **3** (positive mode)

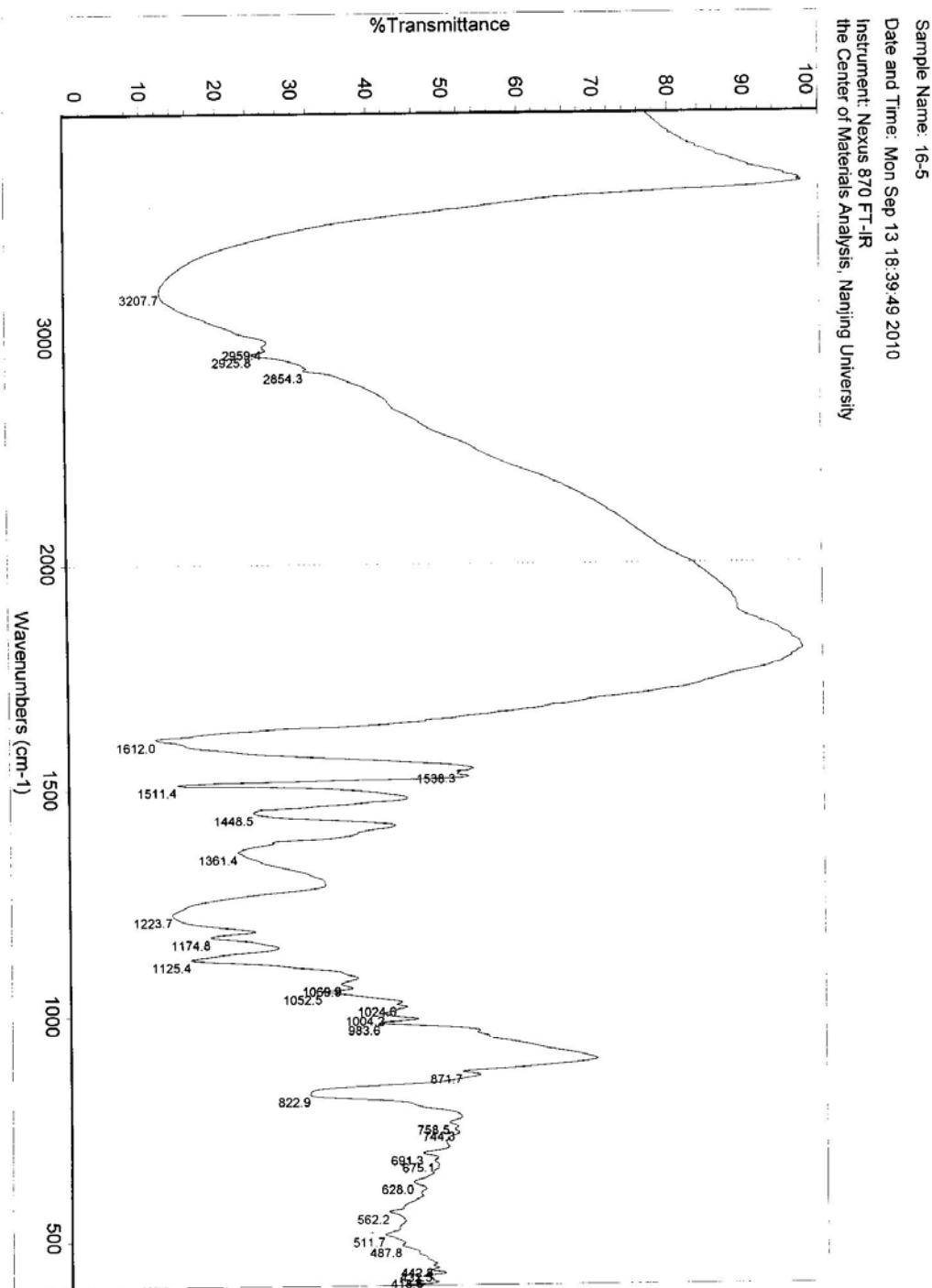
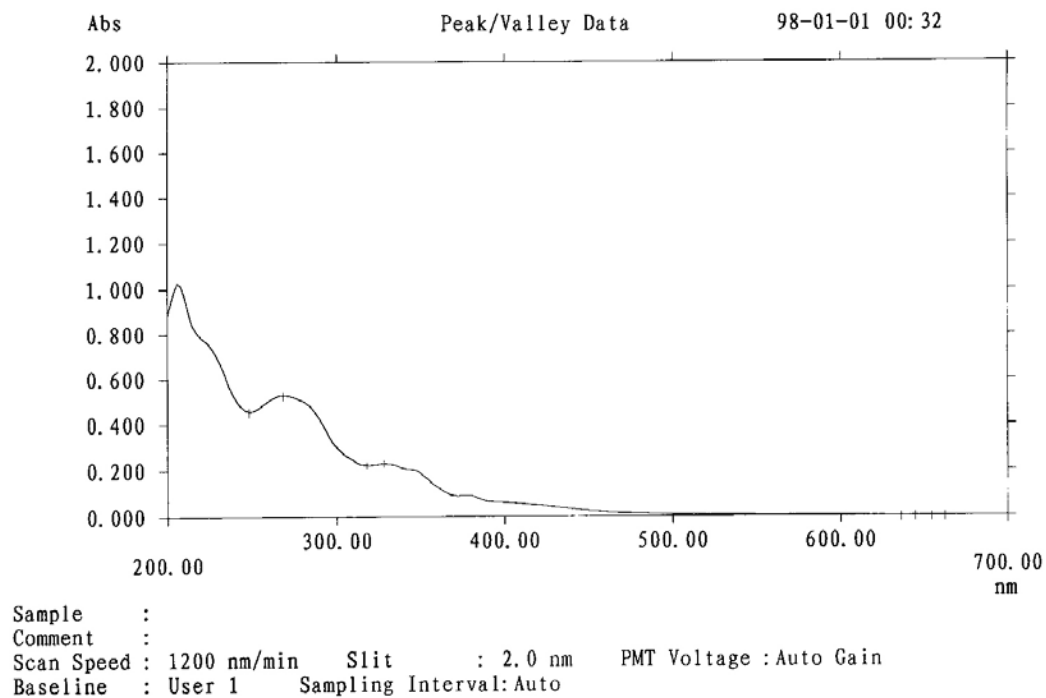


Figure S27 FT-IR spectrum of compound **3** (KBr)



No.	WL (nm)	Peak (Abs)	No.	WL (nm)	Valley (Abs)
1	662.00	0.001	1	654.00	0.001
2	644.00	0.001	2	636.00	0.001
3	328.00	0.229	3	318.00	0.222
4	268.00	0.530	4	248.00	0.457

Figure S28 UV spectrum of compound **3** (CH<sub>3</sub>OH)

Table S1 The optimized structural information for **1** within polarizable continuum model (CH<sub>3</sub>OH solvent: dielectric constant  $\epsilon = 32.63$ ) at B3LYP/6-31G(d) level.

Atoms	Coordinate X	Coordinate Y	Coordinate Z
C	0.535	-0.801	1.099
C	-0.688	-0.243	1.546
C	-1.103	-0.357	2.874
C	-0.313	-1.054	3.790
C	0.871	-1.651	3.374
C	1.280	-1.537	2.039
H	-2.024	0.090	3.222
H	1.480	-2.221	4.076
C	-3.313	-2.085	-2.505
C	-1.921	-2.145	-2.382
C	-4.070	-1.269	-1.653
C	-1.230	-1.390	-1.421
H	-1.381	-2.795	-3.067
C	-3.359	-0.519	-0.733
H	-5.154	-1.225	-1.697
C	-1.970	-0.523	-0.609
C	-1.595	0.442	0.507
C	-3.010	0.815	1.041
C	0.273	-1.623	-1.296
H	0.712	-1.502	-2.300
C	0.977	-0.624	-0.359
H	0.590	0.349	-0.672
C	2.482	-0.506	-0.607
C	3.119	0.696	-0.278
C	3.254	-1.512	-1.207
C	4.477	0.895	-0.527
H	2.543	1.499	0.177
C	4.608	-1.324	-1.472
H	2.793	-2.464	-1.445
C	5.226	-0.118	-1.130
H	4.948	1.841	-0.265
H	5.201	-2.106	-1.937
O	0.529	-2.950	-0.824
H	-0.089	-3.550	-1.270
O	-3.887	-2.857	-3.473
H	-4.847	-2.724	-3.461
O	-3.977	0.290	0.199
O	-3.315	1.505	1.972
C	-1.044	1.811	0.029
C	-0.448	2.675	0.956
C	-1.215	2.264	-1.287
C	-0.012	3.942	0.577
H	-0.327	2.357	1.986
C	-0.783	3.529	-1.677
H	-1.683	1.619	-2.024

C	-0.177	4.375	-0.744
H	0.451	4.598	1.312
H	-0.908	3.874	-2.698
O	0.231	5.605	-1.176
H	0.614	6.091	-0.429
O	-0.764	-1.126	5.079
H	-0.133	-1.639	5.607
O	2.425	-2.150	1.629
H	2.829	-2.603	2.385
O	6.562	0.014	-1.411
H	6.852	0.896	-1.130



Table S2 TDDFT results for the optimized conformer of (7bS)-1 (200nm< $\lambda$ <400nm). The TDDFT calculations are done at the level of B3LYP/6-31G(d) in the polarizable continuum model (CH<sub>3</sub>OH solvent: dielectric constant  $\epsilon = 32.63$ ).

Transition	Excitation energy <sup>a</sup> (nm)	Rotatory Strength $R^b$ (10 <sup>-40</sup> cgs)	Oscillator Strength $f^b$	Dominant Contributions <sup>c</sup>	Weight
4	283.33	30.1915	0.0550	124 → 127	0.44
5	281.26	36.4259	0.0259	123 → 127	0.46
7	273.28	1.5034	0.0621	126 → 129	0.26
10	264.87	-47.8648	0.0117	122 → 127	0.34
11	263.73	-20.6052	0.0078	125 → 129	0.32
12	261.37	52.7662	0.0296	124 → 128	0.24
16	249.32	21.2609	0.0362	126 → 132	0.24
18	247.56	15.7061	0.0346	122 → 128	0.15
19	245.65	26.7606	0.0456	123 → 129	0.30
21	243.29	-16.5655	0.0549	123 → 130	0.29
22	242.34	22.7906	0.0298	121 → 127	0.30
23	240.22	76.2092	0.0673	120 → 127	0.42
25	235.92	-40.3156	0.0327	119 → 127	0.24
26	235.41	-27.1127	0.0517	126 → 134	0.10
31	229.47	17.2238	0.0408	122 → 130	0.26
32	228.41	31.0249	0.0111	117 → 127	0.48
36	225.01	19.4478	0.0132	124 → 132	0.20
45	216.58	20.9542	0.1103	120 → 129 126 → 132	0.12 0.14
46	216.11	22.7801	0.0669	120 → 129 123 → 133	0.13 0.10
48	215.11	-37.8064	0.0617	125 → 135	0.24
51	212.27	-27.5234	0.0398	120 → 130 123 → 134	0.10 0.23
52	210.14	95.6042	0.2251	122 → 132	0.15
53	208.12	-32.6942	0.0157	119 → 130 124 → 135	0.17 0.17
55	206.53	30.5997	0.0228	119 → 129 124 → 135	0.11 0.11
56	205.11	-159.092	0.1134	121 → 132 122 → 133	0.10 0.10
57	204.88	-220.727	0.1386	123 → 134	0.09
58	204.63	-31.5183	0.0487	121 → 132 123 → 135	0.12 0.13
60	202.45	-72.4806	0.1782	120 → 131 122 → 134	0.13 0.09
61	202.00	228.1845	0.2204	122 → 134	0.19
62	201.12	236.9758	0.3268	120 → 132	0.19
63	200.06	36.0665	0.0614	116 → 127	0.39

<sup>a</sup> Excited states with  $f < 0.1$  and  $R < \pm 16.0$  were not presented. All the excitation energies are shifted 11 nm according to the UV correction.

<sup>b</sup> All the strengths were in the velocity representation.

<sup>c</sup> Configurations with weights below 0.10 were not displayed.

Table S3. The Cartesian coordinates of the optimized structure for **2** within PCM model (CH<sub>3</sub>OH solvent: dielectric constant  $\epsilon=32.63$ ) at B3LYP/6-31G(d) level.

Atom	X	Y	Z
C	-1.250	0.628	-0.748
C	1.614	0.324	-0.049
C	1.711	1.792	-0.550
H	0.906	0.358	0.789
H	2.503	2.299	0.017
O	2.048	1.772	-1.938
H	2.074	2.694	-2.241
C	2.920	-0.164	0.575
C	2.872	-1.034	1.667
C	4.180	0.273	0.141
C	4.035	-1.469	2.306
H	1.909	-1.384	2.033
C	5.348	-0.142	0.771
H	4.247	0.927	-0.723
C	5.279	-1.020	1.859
H	3.971	-2.145	3.158
H	6.321	0.197	0.431
O	6.460	-1.397	2.443
H	6.265	-1.997	3.180
C	0.945	-0.568	-1.103
C	-0.446	-0.436	-1.363
C	1.642	-1.537	-1.855
C	-1.093	-1.287	-2.282
C	0.993	-2.380	-2.761
C	-0.375	-2.253	-2.974
H	-2.144	-1.139	-2.496
H	1.552	-3.124	-3.322
O	-0.951	-3.094	-3.885
H	-1.891	-2.871	-3.963
C	0.424	2.544	-0.277
C	-0.864	1.872	-0.337
C	0.444	3.842	0.108
C	-2.042	2.609	0.097
C	-0.768	4.627	0.469
H	1.383	4.382	0.211
C	-2.041	3.907	0.470
H	-2.942	4.434	0.764
O	-0.670	5.817	0.776
O	-3.133	1.803	0.074
C	-2.766	0.512	-0.479
O	-3.463	0.354	-1.689
H	-4.410	0.349	-1.468
C	-3.127	-0.555	0.550
C	-2.941	-0.314	1.916
C	-3.624	-1.804	0.155
C	-3.245	-1.288	2.862

H	-2.569	0.650	2.249
C	-3.935	-2.785	1.092
H	-3.773	-2.015	-0.899
C	-3.745	-2.530	2.453
H	-3.102	-1.080	3.921
H	-4.325	-3.751	0.786
O	-4.066	-3.523	3.331
H	-3.891	-3.218	4.235
O	2.990	-1.654	-1.696
H	3.312	-2.363	-2.274

Table S4. TDDFT results for the optimized conformer of compound (7aR, 8aR, 7bR)-2 (200nm< $\lambda$ <600nm). The TDDFT calculations are done at the level of B3LYP/6-31G(d) in the PCM model (CH<sub>3</sub>OH solvent: dielectric constant  $\epsilon=32.63$ ).

Transition	Excitation energy <sup>a</sup> (nm)	Rotatory Strength $R^b$ (10 <sup>-40</sup> cgs)	Oscillator Strength $f^b$	Dominant Contributions <sup>c</sup>	Weight
2	468.92	20.2392	0.0151	125 → 127 126 → 127	0.11 0.26
4	388.61	-28.0819	0.1250	122 → 127 124 → 127	0.20 0.15
5	378.57	-70.7110	0.2493	122 → 127 124 → 127	0.20 0.20
6	362.02	16.6995	0.0739	121 → 127	0.36
9	296.93	-21.5154	0.0409	117 → 127	0.36
10	290.56	-22.1269	0.1405	117 → 127 119 → 127	0.13 0.22
24	233.72	-22.6919	0.0247	114 → 127 124 → 130	0.13 0.19
33	222.26	-18.2217	0.0267	124 → 131	0.29
36	218.74	-62.1837	0.0633	122 → 127	0.11
37	218.49	19.9538	0.0243	122 → 129	0.12
38	217.54	39.8335	0.0512	123 → 129	0.10
39	216.85	28.0179	0.0145	121 → 129 122 → 130	0.14 0.21
41	212.80	18.3992	0.0415	112 → 127	0.18
43	211.46	49.9723	0.0492	125 → 134	0.06
44	210.98	-53.7444	0.0444	121 → 130	0.22
45	210.72	33.4481	0.0130	111 → 127	0.13

<sup>a</sup> Excited states with  $f < 0.1$  and  $R < \pm 15.0$  were not presented.

<sup>b</sup> All the strengths were in the velocity representation.

<sup>c</sup> Configurations with weights below 0.10 were not displayed.

Table S5. The Cartesian coordinates of the optimized structure for **3** within PCM model (CH<sub>3</sub>OH solvent: dielectric constant  $\epsilon=32.63$ ) at B3LYP/6-31G(d) level.

Atom	X	Y	Z
C	1.050	0.814	-0.046
C	-1.970	1.409	0.808
C	-1.796	0.118	-0.048
C	-3.141	-0.525	-0.389
C	-4.243	-0.496	0.474
C	-3.316	-1.124	-1.643
C	-5.472	-1.039	0.104
H	-4.133	-0.058	1.462
C	-4.534	-1.682	-2.029
H	-2.480	-1.155	-2.341
C	-5.624	-1.639	-1.153
H	-6.318	-1.005	0.791
H	-4.657	-2.141	-3.008
O	-6.804	-2.188	-1.576
H	-7.482	-2.092	-0.872
C	-0.941	2.458	0.432
C	0.356	2.077	0.066
C	-1.245	3.821	0.427
C	1.301	3.066	-0.252
C	-0.279	4.791	0.086
C	1.029	4.424	-0.261
C	-0.707	-0.855	0.460
C	0.667	-0.543	0.240
C	-0.977	-2.107	1.024
C	1.734	-1.507	0.328
C	0.050	-3.047	1.204
C	1.358	-2.787	0.832
C	4.172	-2.075	-0.155
C	5.448	-1.691	-0.555
H	4.020	-3.113	0.093
C	4.712	0.574	-0.844
C	5.730	-0.355	-0.899
H	4.904	1.611	-1.107
O	-2.240	-2.425	1.408
H	-2.273	-3.365	1.694
O	2.301	-3.760	0.988
H	1.879	-4.576	1.340
H	-0.195	-4.020	1.633
C	2.348	1.135	-0.368
C	3.410	0.203	-0.449
C	3.103	-1.151	-0.078
H	6.735	-0.068	-1.206
O	6.403	-2.667	-0.599
H	7.258	-2.283	-0.896
O	2.524	2.495	-0.529

H	-1.414	0.471	-1.016
H	-2.968	1.838	0.609
O	-1.880	1.038	2.191
H	-2.119	1.828	2.720
O	-0.567	6.125	0.076
H	-1.502	6.268	0.344
H	1.773	5.173	-0.515
H	-2.251	4.151	0.688

Table S6. TDDFT results for the optimized conformer of compound (7aR, 8aR)-**3** (200nm< $\lambda$ <600nm). The TDDFT calculations are done at the level of B3LYP/6-31G(d) in the PCM model (CH<sub>3</sub>OH solvent: dielectric constant  $\epsilon=32.63$ ).

Transition	Excitation energy <sup>a</sup> (nm)	Rotatory Strength $R^b$ (10 <sup>-40</sup> cgs)	Oscillator Strength $f^b$	Dominant Contributions <sup>c</sup>	Weight
1	357.47	-23.4709	0.3012	121 → 122	0.36
2	343.16	-21.2557	0.1386	121 → 123	0.32
3	308.97	-12.1778	0.1331	120 → 122	0.31
5	287.95	22.7403	0.0440	121 → 124	0.41
6	285.12	-17.2423	0.1033	118 → 122	0.30
7	279.25	-21.1226	0.4884	120 → 123	0.28
9	271.95	17.1501	0.2131	121 → 125	0.26
11	262.49	59.0263	0.1940	118 → 123 120 → 124 121 → 126	0.14 0.33 0.15
12	255.36	0.1334	0.1380	117 → 122	0.34
13	249.96	25.8374	0.0631	120 → 124	0.23
21	238.85	-23.7552	0.0146	120 → 126	0.36
22	232.08	21.3873	0.0765	119 → 125	0.39
23	228.91	-8.8636	0.2678	118 → 124	0.25
31	215.64	37.8311	0.0254	118 → 126	0.26
40	203.83	-15.7421	0.0359	116 → 124	0.31

<sup>a</sup> Excited states with  $f < 0.1$  and  $R < \pm 15.0$  were not presented.

<sup>b</sup> All the strengths were in the velocity representation.

<sup>c</sup> Configurations with weights below 0.10 were not displayed.