Supporting Information

Hypervalent Iodine Catalyzed Transamidation of Carboxamides with Amines

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I. General Experimental Section:

All the reagents were purchased from commercial suppliers and were used as such. The solvents used in the reactions were distilled and dried prior to use. All the reactions conducted under conventional heating were performed under air atmosphere in a Schlenk flask. Microwave reactions were performed using an Anton Paar Monowave 300 mono-mode microwave reactor with a sealed 10-mL vial containing teflon-coated magnetic stir bar. The microwave system contains a single magnetron that delivers up to 850 W installed microwave power in unpulsed mode over the full power range. The sophisticated software prevents thermal overshoots and the design of the microwave applicator provides utmost field density, which allows efficient heating, even of low-absorbing solvents at any scale. A precisely adjusted IR sensor reflects the internal reaction temperature up to 300 °C. Pressure control up to 30 bar is provided by a non-invasive hydraulic piston embedded in the swiveling cover. For cooling, the cavity is flushed with compressed air automatically after the programmed experiment has been processed. Column chromatography was performed using Merck silica gel (100-200 mesh). Thin layer chromatography (TLC) was performed using silica gel $60/KieselguhrF_{254}$ pre-coated on aluminum sheets (thickness 0.2 mm), commercially available from Merck. Visualization of spots on TLC plate was accomplished with UV light and by staining in I₂ chamber.

All the melting points were determined in open capillary tubes and are uncorrected. ¹H and ¹³C NMR spectra were recorded on 300 MHz and 75 MHz JEOL AL300 FTNMR spectrometer respectively at a temperature of 300 K. NMR chemical shifts are expressed in δ values with reference to tetramethylsilane (TMS) as internal standard. Product yields refer to isolated yields after column chromatography.

II. Experimental procedures:

a) General conventional experimental procedure: Diacetoxyiodobenzene (DIB) (0.15 mmol, 48.3 mg) was added to a Schlenk flask, equipped with a rubber septum and teflon-coated stir bar, containing carboxamide (3 mmol) and amine (3 mmol). The resulting mixture was vigorously stirred at the appropriate temperature for a specified time. The progress of the reaction was monitored through TLC. After completion of the reaction, mixture was cooled to room temperature; distilled water (10 mL) was added and then extracted with ethyl acetate (3 × 10 mL). The combined organic phase was dried over Na₂SO₄, and then concentrated using rotary vacuum evaporator. The crude product was purified by column chromatography using a mixture of ethyl acetate/n-hexane as an eluent.

b) General MW experimental procedure: An oven-dried 10-mL microwave reaction vial containing a teflon-coated magnetic stir bar was charged with carboxamide (3 mmol), amine (3 mmol), and diacetoxyiodobenzene (DIB) (0.15 mmol, 48.3 mg). The vessel was sealed with a plastic microwave septum, stirred at room temperature for 5 min and then placed into the MW cavity for a specified temperature and time. After completion of reaction (TLC), the mixture was cooled to room temperature and worked-up as given in conventional method.

Fata		mol	Solvent	Conventional heating			MW heating		
Entry	Accelerator	(%)		Т	t	Yield	Т	t	Yield
				(°C)	(h)	(%)	(°C)	(min.)	(%)
1.	-	-	-	120	24	n.r	120	20	n.r
2.	<i>p</i> -TSA	10	-	120	24	60	120	20	62
3.	TFA	10	1	120	24	40	120	20	41
4.	BF ₃ ·Et ₂ O	10	-	120	24	57	120	20	57
5.	l ₂	10	-	120	24	51	120	20	52
6.	$Zn(OAc)_2 \cdot 2H_2O$	10	-	120	24	20	120	20	23
7.	CdI ₂	10	-	120	24	33	120	20	39
8.	FeCl₃	10	-	120	24	41	120	20	42
9.	NiCl ₂ ·6H ₂ O	10	1	120	24	30	120	20	30
10.	ZrOCl ₂	10	-	120	24	40	120	20	41
11.	КОН	10	-	120	24	n.r	120	20	n.r
12.	DBU	10	1	120	24	n.r	120	20	n.r
13.	TBAB	10	1	120	24	trace	120	20	trace
14.	[BmIm]BF ₄	10	1	120	24	50	120	20	55
15.	DIB	10	1	120	24	81	120	20	83
16.	DIB	5	-	120	24	81	120	20	83
17.	DIB	5	1	120	17	72	120	10	77
18.	DIB	3	1	120	24	67	120	20	69
19.	DIB	5	Toluene	120	24	18	120	20	27
20.	DIB	5	Xylene	120	24	29	120	20	30
21.	DIB	5	Chlorobenzene	120	24	20	120	20	25
22.	DIB	5	DMSO	120	24	19	120	20	21
23.	DIB	5	DMF	120	24	24	120	20	27
24.	DIB	5	H ₂ O	100	24	17	120	20	23
25.	DIB	5	n-C₅H ₁₁ OH	120	24	n.r	120	20	n.r

IV. Characterization of Products:

1. N-Benzylformamide (3a)¹:



Colorless solid; m.p.: 59-60 °C.

¹H NMR (300 MHz, CDCl₃, TMS) δ/ppm: 8.28 (s, 0.8H), 8.23 (d, *J* = 12.0 Hz, 0.2H), 7.37–7.29 (m, 5H), 5.75 (br s, 1H, NH), 4.51 (d, *J* = 6 Hz, 1.8H), 4.44 (d, *J* = 6.3 Hz, 0.2H).

¹³C NMR (75 MHz, CDCl₃, TMS) δ/ppm: 164.7, 161.3, 137.5, 128.6, 128.4, 127.7, 127.6, 127.4, 127.2, 45.4, 41.8.

2. Piperidine-1-carbaldehyde (3b)²:



Colorless oil.

¹H NMR (300 MHz, CDCl₃, TMS) δ/ppm: 7.99 (s, 1H), 3.47 (t, *J* = 5.4 Hz, 2H), 3.32 (t, *J* = 5.4 Hz, 2H), 1.69–1.54 (m, 6H).

¹³C NMR (75 MHz, CDCl₃, TMS) δ/ppm: 160.4, 46.4, 40.2, 26.2, 24.7, 24.3.

3. N-Formylmorpholine (3c)³:



Colorless oil.

¹H NMR (300 MHz, CDCl₃, TMS) δ/ppm: 8.06 (s, 1H), 3.72–3.56 (m, 6H), 3.42–3.39 (m, 2H).

¹³C NMR (75 MHz, CDCl₃, TMS) δ/ppm: 160.8, 67.1, 66.3, 45.7, 40.5.

4. N-4-Tolylformamide(3d)⁴:



Colorless solid; m.p.: 53–54 °C.

¹H NMR (300 MHz, CDCl₃, TMS) δ/ppm: 9.05 (br s, 1H), 8.63 (d, *J* = 11.1 Hz, 1H), 8.39 (s, 1H), 8.27 (s, 1H), 7.43 (d, *J* = 7.5 Hz, 2H), 7.12–6.96 (m, 6H), 2.30/2.28 (each s, 6H).

¹³C NMR (**75** MHz, CDCl₃, TMS) δ/ppm: 163.1, 159.5, 134.8, 134.4, 134.2, 134.1, 130.0, 129.3, 120.1, 118.9, 20.7, 20.6.

5. N-(4-Methoxyphenyl)formamide (3e)⁵:



Light brownish solid; m.p.: 80–83 °C.

¹H NMR (300 MHz, CDCl₃, TMS) δ/ppm: 9.01 (br s, 1H, NH), 8.53 (d, *J* = 11.1 Hz, 1H), 8.46 (br s, 1H, NH), 8.24 (s, 1H), 7.45 (d, 2H), 7.03 (d, 2H), 6.87–6.79 (m, 4H), 3.77/3.75 (s, 6H).

¹³C NMR (75 MHz, CDCl₃, TMS) δ/ppm: 163.3, 159.4, 157.3, 156.4, 130.0, 129.6, 121.8, 121.2, 114.7, 114.0, 55.3, 55.2.

IR (KBr Disc, v/Cm⁻¹**):** 3246, 3192, 3129, 3050, 3001, 2969, 2937, 2895, 1678, 1656, 1604, 1551, 1510, 1462, 1396, 1308, 1235, 1182, 1109, 1029, 835, 808, 783, 696.

6. N-(2-Methoxyphenyl)formamide (3f)⁶:



Brown solid; m.p.: 84–85 °C.

¹H NMR (300 MHz, CDCl₃, TMS) δ/ppm: 8.74 (d, *J* = 11.4 Hz, 1H), 8.44 (s, 1H), 8.37 (d, *J* = 6.9 Hz, 1H), 7.87 (s, 1H), 7.20–6.87 (m, 8H), 3.87 (s, 6H).

¹³C NMR (75 MHz, CDCl₃, TMS) δ/ppm: 161.4, 158.7, 148.7, 147.7, 126.7, 126.1, 125.1, 124.2, 121.0, 120.4, 116.6, 111.2, 110.0, 55.6.

IR (KBr Disc, \bar{v} /Cm⁻¹): 3254, 3008, 2922, 2937, 2835, 1694, 1659, 1597, 1535, 1460, 1396, 1287, 1227, 1182, 1109, 1029, 865, 813, 745.

7. N-(4-Chlorophenyl)formamide (3g)⁷:



Grey solid; m.p.: 99–100 °C.

¹H NMR (300 MHz, CDCl₃, TMS) δ/ppm: 8.88 (br s, 1H), 8.67 (d, *J* = 11.4 Hz, 1H), 8.34 (s, 1H), 7.99 (br s, 1H), 7.51 (d, *J* = 8.7 Hz, 2H), 7.33 (m, 4H), 7.05 (d, *J* = 8.4 Hz, 2H).

¹³C NMR (75 MHz, CDCl₃, TMS) δ/ppm: 162.7, 159.3, 135.4, 135.3, 130.6, 129.7, 129.0, 121.2, 119.9.

8. *N*-(4-Nitrophenyl)formamide (3h)⁵:



Yellow solid; m.p.: 197–199 °C.

¹H NMR (300 MHz, DMSO-*d*₆, TMS) δ/ppm: 10.77 (s, 1H), 9.03 (br s, 1H), 8.38 (s, 1H), 8.20 (d, *J* = 6.2 Hz, 2H), 7.80 (d, *J* = 7.8 Hz, 2H), 7.40 (br s, 1H).

¹³C NMR (**75** MHz, DMSO-*d*₆, TMS) δ/ppm: 162.7, 160.5, 144.2, 142.5, 125.4, 125.0, 119.0, 116.5.

IR (KBr Disc, \bar{v} **/Cm**⁻¹**):** 3259, 3214, 3152, 3090, 3060, 3050, 2925, 2885, 2629, 1689, 1620, 1596, 1563, 1493, 1411, 1382, 1331, 1302, 1270, 1177, 1155, 1111, 853, 829, 752, 688.

9. N-(3-Hydroxyphenyl)formamide (3i)⁸:



Brown solid; m.p.: 117–118 °C.

¹H NMR (300 MHz, DMSO-*d*₆, TMS) δ/ppm: 10.02 (br s, 1H), 9.53 (s, 1H, OH), 9.42 (s, 1H), 8.72 (d, *J* = 11.1 Hz, 1H), 8.21 (s, 1H), 7.17–6.45 (m, 4H).

¹³C NMR (**75** MHz, DMSO-*d*₆, TMS) δ/ppm: 162.5, 159.6, 158.3, 157.7, 139.5, 139.3, 130.3, 129.6, 110.9, 110.0, 108.1, 106.5, 104.8.

10. N-Methyl-N-phenylformamide (3j)⁹:



Colorless oil.

¹H NMR (300 MHz, CDCl₃, TMS) δ/ppm: 8.46 (s, 1H), 7.43–7.15 (m, 5H), 3.31 (s, 3H).

¹³C NMR (75 MHz, CDCl₃, TMS) δ/ppm: 161.4, 161.3, 141.4, 128.8, 128.1, 125.4, 125.1, 122.5, 121.3, 30.9.

11. N-(Pyridin-2-yl)formamide (3k)¹⁰:



Yellow solid; m.p.: 70-71 °C

¹H NMR (300 MHz, CDCl₃, TMS) δ/ppm: 10.68 (br s, 1H), 10.54 (br s, 1H), 9.36 (d, *J* = 9.9 Hz, 1H), 8.56 (s, 1H), 8.35–8.26 (m, 3H), 7.76–7.64 (m, 2H), 7.11–6.95 (m, 3H).

¹³C NMR (75 MHz, CDCl₃, TMS) δ/ppm: 163.2, 159.7, 151.0, 148.2, 147.1, 138.7, 138.5, 119.9, 119.6, 115.0, 110.4.

12. N-(2-Bromophenyl)formamide (3I)⁸:



Colorless solid; m.p.: 91 °C.

¹H NMR (300 MHz, CDCl₃, TMS) δ/ppm: 8.72 (d, 1H), 8.49 (s, 1H), 8.40 (d, *J* = 8.1 Hz, 1H), 7.67 (br s, 1H), 7.61–7.01 (m, 8H).

¹³C NMR (75 MHz, CDCl₃, TMS) δ/ppm: 161.6, 158.9, 134.7, 133.4, 132.3, 128.6, 128.4, 126.3, 125.6, 122.1, 118.9, 113.0.

13. N-Phenethylformamide (3m)¹¹:



Colorless oil.

¹H NMR (300 MHz, CDCl₃, TMS) δ/ppm: 8.08 (s, 1H), 7.89 (d, *J* = 11.7 Hz, 1H), 7.33–7.15 (m, 5H), 5.85 (br s, 1H, NH), 3.59–3.49 (m, 2H), 3.44 (t, *J* = 6.6 Hz, 2H), 2.85–2.78 (m, 2H).

¹³C NMR (**75** MHz, CDCl₃, TMS) δ/ppm: 164.4, 161.2, 138.4, 128.7, 128.7, 128.6, 126.8, 126.5, 43.1, 39.1, 37.6, 35.4.

14. N-Benzylacetamide (3o)¹²:



Brown solid; m.p.: 60-61 °C.

¹H NMR (300 MHz, CDCl₃, TMS) δ/ppm: 7.29–7.21 (m, 5H), 6.61 (br s, 1H), 4.34 (d, *J* = 3.9 Hz, 2H), 1.93 (s, 3H).

¹³C NMR (75 MHz, CDCl₃, TMS) δ/ppm: 170.1, 138.2, 128.4, 127.6, 127.2, 43.4, 22.9.

IR (KBr Disc, \bar{v} **/Cm**⁻¹**):** 3295, 3063, 3033, 2928, 1647, 1548, 1499, 1454, 1374, 1357, 1221, 1162, 1097, 1033, 907, 741, 696.

15. N-Phenylacetamide (3p)¹³:



Colorless solid; m.p.: 113–114 °C.

¹H NMR (300 MHz, CDCl₃, TMS) δ/ppm: 8.50 (br s, 1H), 7.52 (d, *J* = 7.8 Hz, 2H), 7.28–7.23 (m, 2H), 7.08–7.06 (m, 1H), 2.11 (s, 3H).

¹³C NMR (75 MHz, CDCl₃, TMS) δ/ppm: 169.2, 138.0, 128.7, 124.1, 120.2, 24.1.

16. N-(4-Chlorophenyl)acetamide (3q)¹⁴:



Yellow solid; m.p.: 177–178 °C.

¹**H NMR (300 MHz, DMSO-***d*₆, **TMS)** δ/ppm: 10.03 (s, 1H), 7.60 (d, *J* = 6.3 Hz, 2H), 7.31 (d, *J* = 6.3 Hz, 2H), 2.03 (s, 3H).

¹³C NMR (75 MHz, DMSO-*d*₆, TMS) δ/ppm: 168.5, 138.3, 128.5, 126.6, 120.5, 24.0.

17. N-(3-Hydroxyphenyl)acetamide (3r)¹⁵:



Colorless solid; m.p.: 148-149 °C.

¹H NMR (300 MHz, DMSO-d₆, TMS) δ/ppm: 9.77 (br s, 1H), 9.34 (s, 1H), 7.17 (s, 1H), 7.03 (t, 1H), 6.91 (d, *J* = 7.8 Hz, 1H), 6.42 (d, *J* = 6.9 Hz, 1H), 1.99 (s, 3H).

¹³C NMR (75 MHz, CDCl₃, TMS) δ/ppm: 168.3, 157.6, 140.4, 129.3, 110.2, 109.9, 106.3, 24.1.

18. N-(4-Tolyl)acetamide (3s)¹⁶:



Colorless solid; m.p.: 149–150 °C.

¹H NMR (300 MHz, CDCl₃, TMS) δ/ppm: 7.83 (br s, 1H), 7.38 (d, *J* = 8.4 Hz, 2H), 7.10 (d, *J* = 8.1 Hz, 2H), 2.29 (s, 3H), 2.12 (s, 3H).

¹³C NMR (75 MHz, CDCl₃, TMS) δ/ppm: 168.6, 135.4, 133.8, 129.3, 120.1, 24.3, 20.8.

19. N-(4-Methoxyphenyl)acetamide (3t)¹⁷:



Colorless solid; m.p.: 126–128 °C.

¹H NMR (300 MHz, CDCl₃, TMS) δ/ppm: 8.04 (s, 1H), 7.39 (d, *J* = 8.7 Hz, 2H), 6.81 (d, *J* = 8.7 Hz, 2H), 3.76 (s, 3H), 2.10 (s, 3H).

¹³C NMR (75 MHz, CDCl₃, TMS) δ/ppm: 168.7, 156.3, 131.1, 122.0, 113.9, 55.3, 24.1.

20. N-(2-Methoxyphenyl)acetamide (3u)¹⁸:



Colorless solid; m.p.: 87-88 °C.

¹H NMR (300 MHz, CDCl₃, TMS) δ/ppm: 8.35 (d, *J* = 7.8 Hz, 1H), 7.77 (br s, NH), 7.04–6.84 (m, 4H), 3.86 (s, 3H), 2.19 (s, 3H).

¹³C NMR (75 MHz, CDCl₃, TMS) δ/ppm: 168.1, 147.6, 127.6, 123.5, 120.9, 119.7, 109.8, 55.5, 24.8.

21. N'-Phenylacetohydrazide (3v)¹⁹:



Pale brown solid; m.p.: 129–130 °C.

¹H NMR (300 MHz, CDCl₃, TMS) δ/ppm: 7.72 (br s, 1H, NH), 7.26–7.18 (m, 5H), 6.93–6.71 (m, 6H), 6.20 (br s, 1H, NH), 5.85 (br s, 1H, NH), 2.08/2.01 (s, 6H).

¹³C NMR (75 MHz, CDCl₃, TMS) δ/ppm: 177.0, 170.3, 147.8, 147.0, 129.4, 129.1, 121.2, 121.1, 113.4, 112.4, 20.8, 19.1.

IR (KBr Disc, \bar{v} /Cm⁻¹): 3287, 3234, 3031, 2928, 1665, 1643, 1597.

22. N,2-Diphenylacetamide (3x)²⁰:



Colorless solid: m.p.: 115–116 °C.

¹**H NMR (CDCl₃, 300 MHz, TMS)** δ/ppm: 7.75 (br s, 1H), 7.43 (d, *J* = 7.8 Hz, 2H), 7.32–7.02 (m, 8H), 3.64 (s, 2H).

¹³C NMR (CDCl₃, **75** MHz, TMS) δ/ppm: 169.4, 137.7, 134.5, 129.3, 128.9, 128.8, 127.4, 124.3, 119.9, 44.5.

IR (KBr Disc, \bar{v} **/Cm**⁻¹**):** 3285, 3256, 3060, 1657, 1601, 1556, 1496, 1442, 1167, 755, 724, 691.

23. N-Benzylbenzamide (3y)²¹:



Colorless solid; m.p.: 101–102 °C.

¹H NMR (300 MHz, CDCl₃, TMS) δ/ppm: 7.80 (d, *J* = 6.9 Hz, 2H), 7.50–7.25 (m, 8H), 6.37 (br s, 1H), 4.67 (d, *J* = 5.4 Hz, 2H).

¹³C NMR (**75** MHz, CDCl₃, TMS) δ/ppm: 167.3, 138.2, 134.3, 131.4, 128.7, 128.5, 127.8, 127.5, 126.9, 44.0.

IR (KBr Disc, v̄/Cm⁻¹): 3291, 3061, 3030, 2926, 2853, 1638, 1602, 1550, 1452, 1417, 1362, 1260, 1151, 1056, 1027, 928, 726, 695.

24. N-Cyclohexylbenzamide (3z)²²:



Colorless solid; m.p.: 156–157 °C.

¹H NMR (300 MHz, CDCl₃, TMS) δ/ppm: 7.76–7.74 (m, 2H), 7.47–7.38 (m, 3H), 6.09 (br s, 1H, NH), 4.03–3.91 (m, 1H, NHCH), 2.04–1.17 (m, 10H).

¹³C NMR (75 MHz, CDCl₃, TMS) δ/ppm: 166.6, 135.0, 131.1, 128.4, 126.8, 48.6, 33.1, 25.5, 24.9.

IR (KBr Disc, \bar{v} **/Cm**⁻¹**):** 3239, 3078, 2928, 2851, 1639, 1626, 1564, 1488, 1453, 1330, 1295, 1262, 1241, 1152, 1083, 1025, 1001, 970, 891, 849, 804, 700, 670.

25. N-Phenethylbenzamide (3aa)²²:



Colorless solid; m.p.: 117–118 °C.

¹H NMR (300 MHz, CDCl₃, TMS) δ/ppm: 7.66 (d, *J* = 7.2 Hz, 2H), 7.47–7.19 (m, 8H), 6.11 (br s, 1H, NH), 3.72–3.66 (m, 2H), 2.90 (t, 2H).

¹³C NMR (75 MHz, CDCl₃, TMS) δ/ppm: 167.5, 138.8, 134.5, 131.3, 128.7, 128.5, 128.4, 126.7, 41.1, 35.6.

26. N-Cyclopropylbenzamide (3ab)²³:



Colorless solid; m.p.: 94–95 °C.

¹H NMR (300 MHz, CDCl₃, TMS) δ/ppm: 7.74 (d, *J* = 6.9 Hz, 2H), 7.50–7.38 (m, 3H), 6.34 (br s, 1H, NH), 2.91–2.88 (m, 1H), 0.87 (br s, 2H), 0.62 (br s, 2H).

¹³C NMR (75 MHz, CDCl₃, TMS) δ/ppm: 168.9, 134.4, 131.4, 128.6, 128.5, 127.2, 126.8, 125.8, 31.9, 23.1, 6.7.

IR (KBr Disc, v̄/Cm⁻¹): 3303, 2924, 2852, 1641, 1531, 1488, 1310, 1050, 1027, 862, 799, 693, 667.

27. N-Hexylbenzamide (3ac)²⁴:



Colorless liquid.

¹H NMR (300 MHz, CDCl₃, TMS) δ/ppm: 7.78 (d, *J* = 7.2 Hz, 2H), 7.49–7.36 (m, 3H), 6.47 (br s, 1H, NH), 3.45–3.38 (m, 2H), 1.61–1.54 (m, 2H), 1.31 (br s, 6H), 0.88 (t, *J* = 6.3 Hz, 3H).

¹³C NMR (75 MHz, CDCl₃, TMS) δ/ppm: 167.5, 134.8, 131.1, 128.4, 126.8, 40.0, 31.4, 29.5, 26.6, 22.5, 13.9.

IR (KBr Disc, \bar{v} **/Cm**⁻¹**):** 3321, 2929, 2857, 1632, 1578, 1541, 1491, 1310, 695.

28. *N*-Benzylnicotinamide (3ad)²⁵:



Pale yellow solid; m.p.: 72–73 °C.

¹H NMR (300 MHz, CDCl₃, TMS) δ/ppm: 8.93 (br s, 1H), 8.61 (br s, 1H), 8.12 (br s, 1H), 7.32–7.31 (br s, 6H), 4.61 (s, 2H).

¹³C NMR (75 MHz, CDCl₃, TMS) δ/ppm: 165.5, 151.9, 147.8, 137.7, 135.3, 130.0, 128.7, 127.8, 127.6, 123.5, 44.0.

29. 2-Benzylisoindoline-1,3-dione (3ae)²⁶:



Colorless crystals; m.p.: 116–117 °C.

¹H NMR (300 MHz, CDCl₃, TMS) δ/ppm: 7.85–7.81 (m, 2H), 7.71–7.67 (m, 2H), 7.44–7.41 (m, 2H), 7.33–7.23 (m, 3H), 4.84 (s, 2H).

¹³C NMR (75 MHz, CDCl₃, TMS) δ/ppm: 168.0, 136.3, 133.9, 132.1, 128.6, 128.5, 127.7, 123.3, 41.5.

IR (KBr Disc, v̄/Cm⁻¹): 3459, 3086, 3060, 3035, 2947, 2852, 1775, 1765, 1715, 1611, 1603, 1583, 1491, 1466, 1391, 1331, 1298, 1276, 1184, 1101, 792, 763.

30. 2-Cyclohexylisoindoline-1,3-dione (3af)²⁷:



Colorless solid; m.p.: 169–171 °C.

¹**H NMR (300 MHz, CDCl₃, TMS)** δ /ppm: 7.82–7.79 (dd, J = 3.3 Hz, J = 3 Hz, 2H), 7.70–7.67 (dd, J = 3 Hz, J = 3.3 Hz, 2H), 4.15–4.06 (m, 1H), 2.26–2.14 (m, 2H), 1.89–1.71 (m, 5H), 1.44–1.25 (m, 3H).

¹³C NMR (75 MHz, CDCl₃, TMS) δ/ppm: 168.4, 133.6, 132.0, 122.9, 50.8, 29.8, 26.0, 25.1.

31. 2-Hexylisoindoline-1,3-dione (3ag)²⁸:



Colorless liquid.

¹H NMR (300 MHz, CDCl₃, TMS) δ/ppm: 7.84–7.81 (m, 2H), 7.72–7.69 (m, 2H), 3.67 (t, *J* = 7.5 Hz, 2H), 1.69–1.65 (m, 2H), 1.31 (br s, 6H), 0.89-0.85 (m, 3H).

¹³C NMR (75 MHz, CDCl₃, TMS) δ/ppm: 168.1, 133.6, 132.0, 122.9, 37.8, 31.2, 28.3, 26.3, 22.3, 13.8.

32. 2-Cyclopropylisoindoline-1,3-dione (3ah)²⁹:



White solid; m.p.: 134–136 °C.

¹H NMR (CDCl₃, 300 MHz, TMS) δ/ppm: 7.83–7.80 (m, 2H), 7.72–7.69 (m, 2H), 2.75–2.68 (m, 1H), 1.04–1.01 (m, 4H).

¹³C NMR (CDCl₃, **75** MHz, TMS) δ/ppm: 168.8, 133.9, 131.7, 123.0, 20.8, 5.1.

IR (KBr Disc, \bar{v} **/Cm**⁻¹**):** 3026, 1767, 1713, 1699, 1610, 1463, 1402, 1140, 946, 699.

33. N-tert-Butylphthalimide (3ai)³⁰:



White solid; m.p.: 204–205 °C.

¹H NMR (300 MHz, CDCl₃, TMS) δ/ppm: 7.76 (br s, 2H), 7.68 (br s, 2H), 1.70/1.64 (s, 9H).

¹³C NMR (75 MHz, CDCl₃, TMS) δ/ppm: 169.5, 133.6, 132.1, 122.5, 57.8, 29.1.

34. 2-(Phenethyl)isoindole-1,3-dione (3aj)³¹:



Colorless solid; m.p.: 130 °C.

¹H NMR (300 MHz, CDCl₃, TMS) δ/ppm: 7.84–7.81 (m, 2H), 7.71–7.68 (m, 2H), 7.26 (br s, 5H), 3.92 (t, *J* = 7.8 Hz, 2H), 2.99 (t, *J* = 7.2 Hz, 2H).

¹³C NMR (**75** MHz, CDCl₃, TMS) δ/ppm: 168.1, 138.0, 133.9, 132.0, 128.8, 128.5, 126.6, 123.2, 39.2, 34.6.

IR (KBr Disc, v/Cm⁻¹**):** 3020, 1770, 1710, 1614, 1496, 1429, 1396, 1359, 1083, 989, 869, 755, 633.

V. References:

1. M. Suchỳ, A. A. H. Elmehriki and R. H. E. Hudson, Org. Lett., 2011, **13**, 3952.

2. G.-L. Li, K. K.-Y. Kung and M.-K. Wong, Chem. Commun., 2012, 48, 4112.

3. T. B. Nguyen, J. Sorres, M. Q. Tran, L. Ermolenko and A. Al-Mourabit, *Org. Lett.*, 2012, 14, 3202.

4. G. B. Villeneuve and T. H. Chan, *Tetrahedron Lett.*, 1997, **38**, 6489.

5. L. Di Nunno and A. Scilimati, *Tetrahedron*, 1986, **42**, 3913.

6. A. Ghorbani-Choghamarani and Z. Akbaripanah, Chinese Chem. Lett., 2012, 23, 450.

7. H. Tumma, Nagaraju N and K. V. Reddy, J. Mol. Cat. A: Chem., 2009, **310**, 121.

8. M. Hosseini-Sarvari and H. Sharghi, J. Org. Chem., 2006, 71, 6652.

9. J.-G. Kim and D.-O. Jang, Synlett, 2010, 8, 1231.

10. S. Ko, H. Han and S. Chang, Org. Lett., 2003, 5, 2687.

11. M. C. Elliott and E. Williams, Org. Biomol. Chem., 2003, 1, 3038.

12. J. E. Taylor, M. D. Jones, J. M. J. Williams and S. D. Bull, J. Org. Chem., 2012, 77, 2808.

13. (a) D. R. Stuart, M. Bertrand-Laperle, K. M. N. Burgess and K. Fagnou, *J. Am. Chem. Soc.*, 2008, **130**, 16474; (b) M. Li, L. Hu, X. Cao, H. Hong, J. Lu and H. Gu, *Chem. -Eur. J.*, 2001, **17**, 2763.

14. B. T. Gowda, K. M. Usha and K. L. Z. Jayalakshmi, Naturforsch., A: Phys. Sci., 2003, 58, 801.

15. S. R. Narahari, B. R. Reguri and K. Mukkanti, Tetrahedron Lett., 2011, 52, 4888.

16. A. R. Katritzky, C. Cai and S. K. Singh, J. Org. Chem., 2006, 71, 3375.

17. (a) C. J. Bennett, S. T. Caldwell, D. B. McPhail, P. C. Morrice, G. G. Duthie and R. C. Hartley, *Bioorg. Med. Chem.*, 2004, **12**, 2079; (b) B. P. Fors, P. Krattiger, E. Strieter and S. L. Buchwald, *Org. Lett.*, 2008, **10**, 3505.

18. B. S. Kim, C. Jang, D. J. Lee and S. W. Youn, Chem. -Asian. J., 2010, 5, 2336.

19. (a) S. Xun, G. LeClair, J. D. Zhang, X. Chen, J. P. Gao and Z. Y. Wang, *Org. Lett.*, 2006, **8**, 1697;
(b) K. Hisler, A. G. J. Commeureuc, S.-Z. Zhou and J. A. Murphy, *Tetrahedron Lett.*, 2009, **50**, 3290.

20. L. U. Nordstrom, H. Vogt and R. Madsen, J. Am. Chem. Soc., 2008, 130, 17672.

21. J. Pan, N. O. Devarie-Baez and M. Xian, Org. Lett., 2011, 13, 1092.

22. S. Das, B. Join, K. Junge and M. Beller, Chem. Commun., 2012, 48, 2683.

23. N. D. Kokare, R. R. Nagawade, V. P. Rane and D. B. Shinde, Synthesis, 2007, 766.

24. T. Ohshima, T. Iwasaki, Y. Maegawa, A. Yoshiyama and K. Mashima, *J. Am. Chem. Soc.*, 2008, **130**, 2944.

25. T. H. Graham, W. Liu and D.-M. Shen, Org. Lett., 2011, 13, 6232.

26. H. Cao and H. Alper, Org. Lett., 2010, 12, 4126.

27. D. N. Sawant, Y. S. Wagh, K. D. Bhatte and B. M. Bhanage, Eur. J. Org. Chem., 2011, 6719.

28. M.-L. Wang and W.-H. Chen, React. Kinet. Catal. Lett., 2006, 89, 377.

29. S. Bénard, L. Neuville and J. Zhu, J. Org. Chem., 2008, 73, 6441.

30. D. Marosvölgyi-Haskó, A. Petz, A. Takács and L. Kollér, Tetrahedron, 2011, 67, 9122.

31. Y.-P. Li, F.-X. Ning, M.-B. Yang, Y.-C. Li, M.-H. Nie, T.-M. Ou, J.-H. Tan, S.-L. Huang, D. Li, L.-Q. Gu and Z.-S. Huang, *Eur. J. Med. Chem.*, 2011, **46**, 1572.

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