Azaindole-1,2,3-triazole coujugate as selective fluorometric sensor for dihydrogenphosphate

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Figure 1S. (a) Absorption and (b) emission spectra of 3 ($c = 3.09 \times 10^{-5}$ M) in different solvents.



Figure 2S. Change in emission of (a) **1** ($c = 3.23 \times 10^{-5} M$) and (b) **3** ($c = 4.25 \times 10^{-5} M$) upon titration with DMSO in CH₃CN containing 0.01%DMSO.



Figure 3S. Change in emission of (a) **1** ($c = 3.25 \times 10^{-5}$ M) and (b) **3** ($c = 3.09 \times 10^{-5}$ M) upon titration with CH₃OH in CH₃CN containing 0.01%DMSO.



Figure 4S. Change in fluorescence ratio of **1** ($c = 4.52 \times 10^{-5}$ M) at 370 nm upon addition of 15 equiv. amounts of different guests in DMSO.



Figure 5S. Change in fluorescence ratio of **3** ($c = 3.45 \times 10^{-5}$ M) at 370 nm upon addition of 15 equiv. amounts of different guests in DMSO.



Figure 6S. Change in emission of (a) **1** (c = 4.52×10^{-5} M) and (b) **3** (c = 3.63×10^{-5} M) when titrated with H₂PO₄⁻ (c = 1×10^{-3} M) in DMSO.



Figure 7S. Change in emission of **1** ($c = 2.51 \times 10^{-5} \text{ M}$) upon addition of a) Cl⁻, b) Br⁻, c) F⁻, (d) Γ , e) HSO₄⁻, (f) HP₂O₇⁻³⁻, (g)AcO⁻, (h) NO₃⁻ in CH₃CN containing 0.01% DMSO.



Figure 8S. Change in absorbance of 1 ($c = 2.51 \times 10^{-5} \text{ M}$) upon addition of a) Cl⁻, b) Br⁻, c) F⁻, (d) I⁻, e) HSO₄⁻, (f) HP₂O₇⁻³⁻, (g)AcO⁻, (h) NO₃⁻ in CH₃CN containing 0.01% DMSO.



Figure 9S. Photographs of color change of (a) $\mathbf{1}(c = 2.5 \times 10^{-5} \text{ M})$ and with 15 equiv. amounts of (b) $\text{HP}_2\text{O}_7^{-3-}$ ($c = 1 \times 10^{-3} \text{ M}$) and (c) $\text{H}_2\text{PO}_4^{--}$ ($c = 1 \times 10^{-3} \text{ M}$) under UV illumination.



Figure 10S. Change in emission of receptor **2** ($c = 2.21 \times 10^{-5}$ M) upon addition of 15 equiv. .amounts of H₂PO₄⁻ ($c = 1 \times 10^{-3}$ M) in CH₃CN containing 0.01% DMSO.



Figure 11S. Partial ¹H NMR (400 MHz) of **1** ($c = 5.01 \times 10^{-3}$ M) in the absence (A) and presence of 1 equiv (B) and 2 equiv (C) amounts of tetrabutylammonium dihydrogenphosphate in d₆-DMSO.



Figure 12S. Change in emission of **3** ($c = 5.01 \times 10^{-5}$ M) upon addition of (a) Cl⁻, (b) Br⁻, (c) F⁻, (d) l⁻, (e) HSO₄⁻, (f) HP₂O₇⁻³⁻, (g) AcO⁻, (h) NO₃⁻ in CH₃CN containing 0.01% DMSO.



Figure 13S. Change in absorbance of **3** ($c = 5.01 \times 10^{-5}$ M) upon addition of (a) Cl⁻, (b) Br⁻, (c) F⁻, (d) I⁻, (e) HSO₄⁻, (f) HP₂O₇⁻³⁻, (g)AcO⁻, (h) NO₃⁻, (i) H₂PO₄⁻ in CH₃CN containing 0.01% DMSO.



Figure 14S. Nonlinear binding constant curve for tetrabutylammonium dihydrogenphosphate ($c=1x10^{-3}M$) with receptor **1** ($c=2.25x10^{-5}$ M) at 363 nm in CH₃CN containing 0.01% DMSO.



Figure 15S. Nonlinear binding constant curves for (a) tetrabutylammonium hydrogenpyrophosphate (b) tetrabutyl ammonium dihydrogenphosphate, (c) tetrabutylammonium fluoride (c = 1×10^{-3} M) with receptor **3** (c = 5.01×10^{-5} M) at 366 nm in CH₃CN containing 0.01% DMSO.



Figure 16S. PM6 optimized geometries of the (a) *anti*, (b) *syn* and (c) $H_2PO_4^-$ complexed forms of **1**. Numerical values in the complex (c) indicate the hydrogen bond distances in Å and the *anti* form is stable by 3.01 kcal/mol over the *syn* form.

¹H NMR (400 MHz, d₆-DMSO)



¹³C NMR (100 MHz, d₆-DMSO)



HRMS



¹H NMR (400 MHz, d₆-DMSO)



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¹³C NMR (100 MHz, d₆-DMSO)



HRMS



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¹H NMR (400 MHz, d₆-DMSO)



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¹³C NMR (100 MHz, d₆-DMSO)



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