

Electronic Supplementary Information

Quantitative detection of HO• generated in high temperature H₂O₂ bleaching system with a novel fluorescent probe benzenepentacarboxylic acid

Fang Si,^{ab} Xuan Zhang^{*a} and Kelu Yan^{*ab}

^a School of Chemistry, Chemical Engineering and biotechnology, Donghua University, Shanghai 201620, P. R. China. E-mail: xzhang@dhu.edu.cn; klyan@dhu.edu.cn; Tel.: +86 021 67792619; Fax: +86 021 67792619

^b National Engineering Research Center for Dyeing and Finishing of Textiles, Donghua University, Shanghai 201620, P. R. China

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1. ESI-MS spectra of reaction mixtures of BA (left) and TA (right) with H₂O₂ under alkaline and high temperature conditions

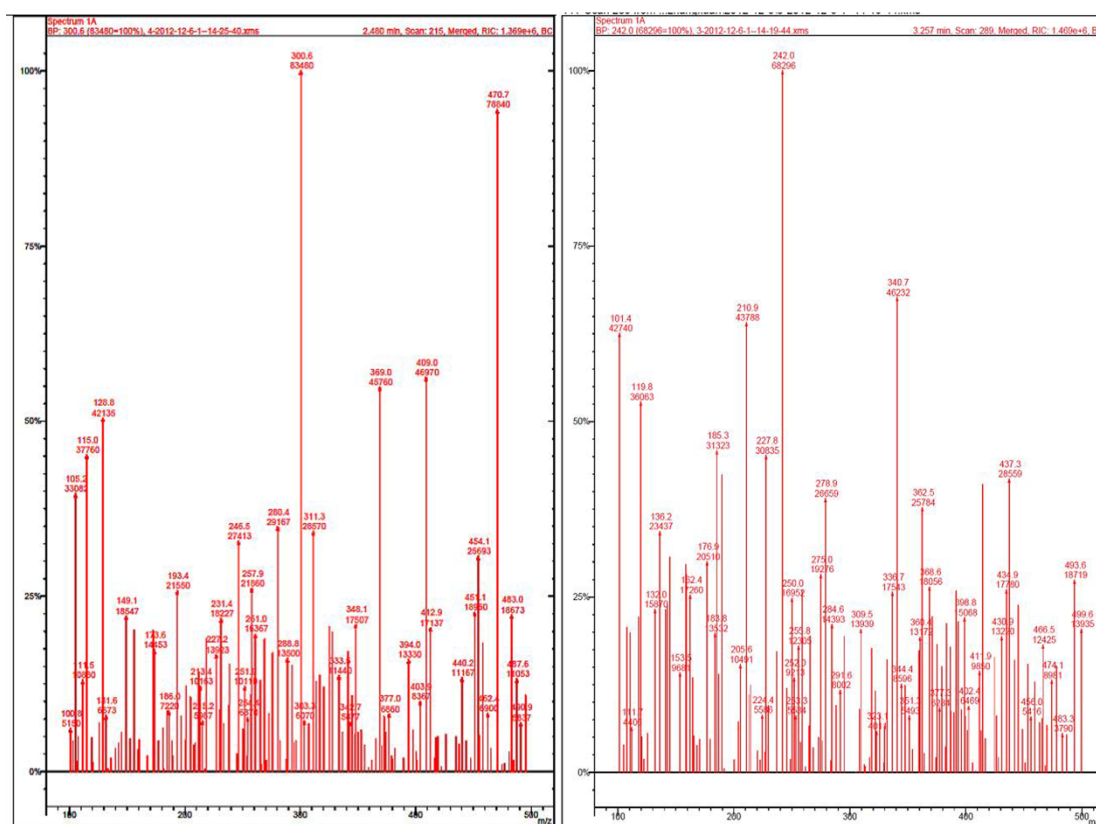


Fig. S1 ESI-MS spectra of reaction mixtures of BA (left) and TA (right) with H₂O₂ under alkaline and high temperature conditions. BA (4×10^{-4} mol l⁻¹); TA (4×10^{-4} mol l⁻¹); H₂O₂ (2.2×10^{-4} mol l⁻¹); T=80°C; pH = 10.0; t=60 min.

2. Scavenging effect of HO• with DMSO

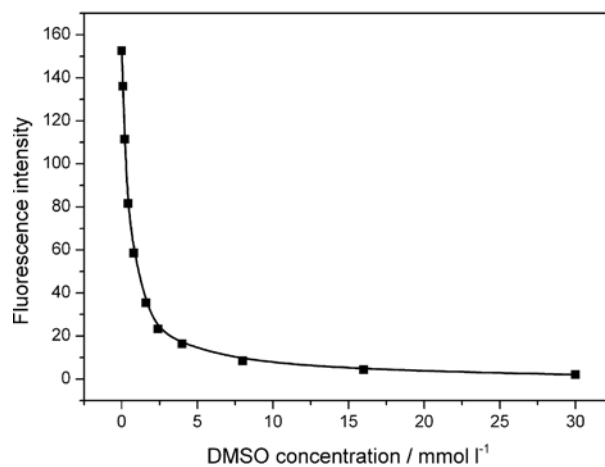


Fig. S2. Scavenging effect of HO• with DMSO. DMSO (0.08-30 mmol l⁻¹); BA (4×10^{-4} mol

Γ^{-1}); H_2O_2 ($2.2 \times 10^{-4} \text{ mol } \Gamma^{-1}$); $T=80^\circ\text{C}$; $\text{pH} = 10.0$; $t=60 \text{ min}$.

3. Comparison of $\text{HO}\cdot$ generation in H_2O_2 alkali system and Fenton buffer system

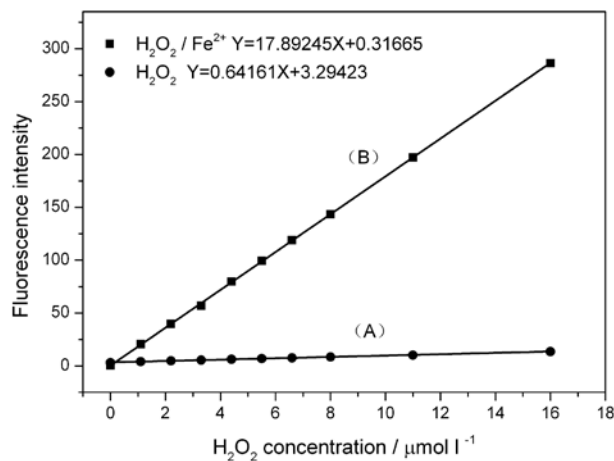


Fig. S3. Comparison of $\text{HO}\cdot$ generation in H_2O_2 alkali system and Fenton buffer system. (A) H_2O_2 alkaline system: BA $4 \times 10^{-4} \text{ mol } \Gamma^{-1}$, $T=80^\circ\text{C}$, $\text{pH} = 10.0$, $t=60 \text{ min}$. (B) $\text{Fe}^{2+}/\text{H}_2\text{O}_2$ buffer system: BA ($20 \mu\text{mol } \Gamma^{-1}$), Fe(II) ($8 \mu\text{mol } \Gamma^{-1}$), $\text{KH}_2\text{PO}_4\text{-K}_2\text{HPO}_4$ ($0.02 \text{ mol } \Gamma^{-1}$, $\text{pH} 7.4$), Room temperature, $t=30 \text{ min}$. H_2O_2 (1×10^{-6} - $1.6 \times 10^{-5} \text{ mol } \Gamma^{-1}$).

4. ^1H NMR spectra of HBA

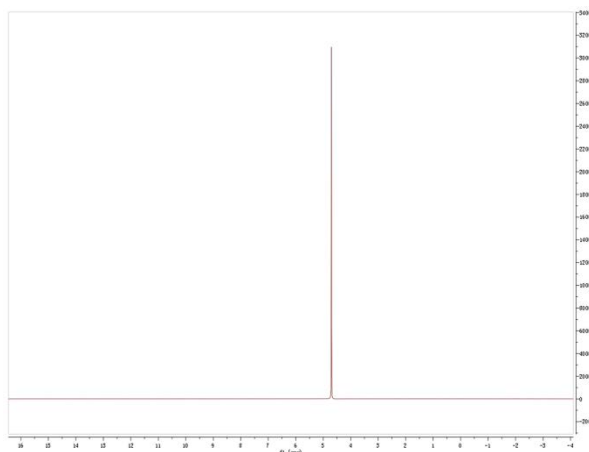


Fig. S4. ^1H NMR spectrum of HBA.

5. ^{13}C NMR spectra of HBA

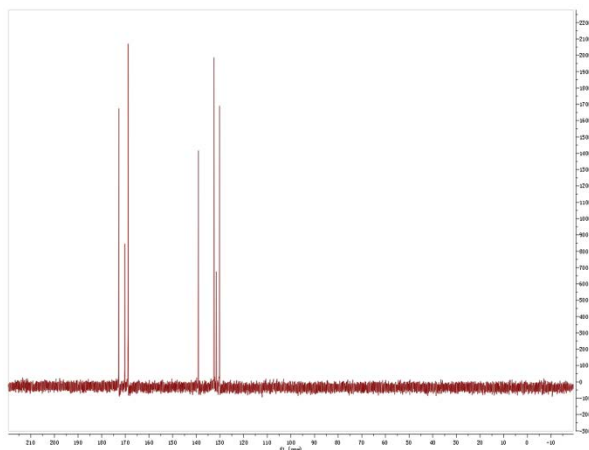


Fig. S5. ^{13}C NMR spectrum of HBA.

6. ESI-MS spectra of HBA

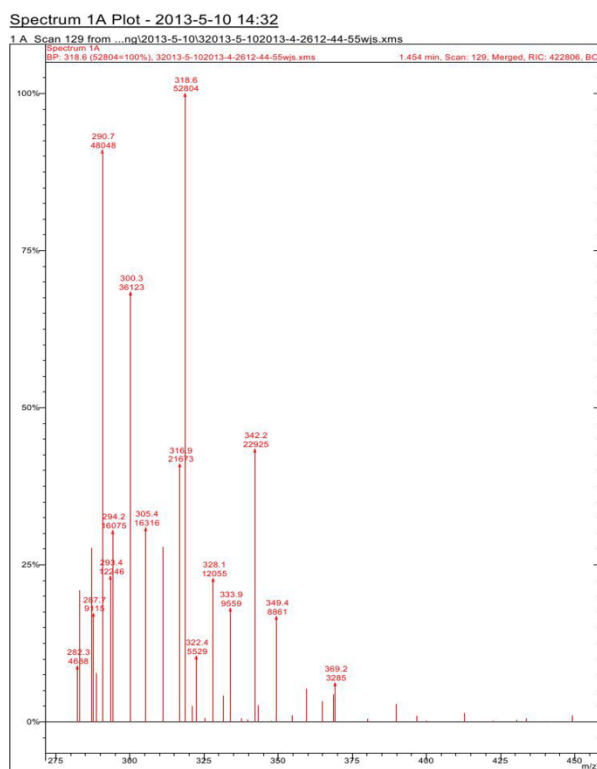


Fig. S6. ESI-MS spectrum of HBA.