Supporting Information

A single colorimetric sensor for multiple target ions: the simultaneous detec tion of Fe²⁺ and Cu²⁺ in aqueous media

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Figure S1. UV-vis absorption spectra of only Fe^{3+} (60 μ M) and a mixture of 1 (20 μ M) with Fe^{3+} (60 μ M) in Bis-tris buffer/DMF (8/2, v/v).



Figure S2. Positive-ion electrospray ionization mass spectrum of **1** ($1.0 \times 10^{-4} \text{ M}$) upon additi on of **2** equiv of Fe²⁺.



Figure S3. Li's equation plot of 1, assuming 1:2 stoichiometry for association between 1 and Fe^{2+} .



Figure S4. Change in the ratio of absorption intensity of 1 with Fe^{2+} .



Figure S5. Job plot of Cu^{2+} complex formation. The total concentration of 1 with Cu^{2+} was 4 0 μ M.



Figure S6. Li's equation plot of 1, assuming 1:2 stoichiometry for association between 1 and Cu^{2+} .



Figure S7. Change in the ratio of absorption intensity of 1 with Cu^{2+} .



Figure S8. Color of 1 (20 μ M) in the presence of both Fe²⁺ (60 μ M) and Cu²⁺ (60 μ M).



Figure S9. (a) Reversible changes in absorbance of **1** after the sequential addition of Cu^{2+} an d EDTA in Bis-tris buffer/DMF (8/2, v/v). (b) Reversible color changes of **1** after the sequent ial addition of Cu^{2+} and EDTA in Bis-tris buffer/DMF (8/2, v/v). (c) UV-vis absorption spectr a of **1**- Cu^{2+} and Cu^{2+} -EDTA, respectively.