

Supporting Information

For

Substituent effect on fluorescence signalling of the HSO₄⁻ receptors through single point to ratiometric response in green solvent[†]

Manjira Mukherjee,^a Siddhartha Pal,^a Buddhadeb Sen,^a Somenath Lohar,^a Samya Banerjee,^b Snehasis Banerjee^c and Pabitra Chattopadhyay^{a*}

Department of Chemistry, Burdwan University, Golapbag, Burdwan-713104, West Bengal, India

E-mail: pabitracc@yahoo.com

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Materials and Instruments

All of the solvents were of analytical grade. The elemental analyses (C, H and N) were performed on a Perkin Elmer 2400 CHN elemental analyzer. A Shimadzu (model UV-1800) spectrophotometer was used for recording electronic spectra. IR spectra were recorded using Prestige-21 SHIMADZU FTIR spectrometer preparing KBr disk. ¹H NMR spectrum of organic moiety was obtained on a Bruker Avance DPX 300 spectrometer using DMSO-d₆ solution. Electrospray ionization (ESI) mass spectra were recorded on a Qtof Micro YA263 mass spectrometer. A Systronics digital pH meter (model 335) was used to measure the pH of the solution and the adjustment of pH was done using either 50 mM HCl or KOH solution. Steady-state fluorescence emission and excitation spectra were recorded with a Hitachi-4500 spectrofluorimeter. Time-resolved fluorescence lifetime measurements were performed using a HORIBA JOBIN Yvon picosecond pulsed diode laser-based time-correlated single-photon counting (TCSPC) spectrometer from IBH (UK) at $\lambda_{\text{ex}} = 370$ nm and MCP-PMT as a detector. Emission from the sample was collected at a right angle to the direction of the excitation beam maintaining magic angle polarization (54.71°). The full width at half-maximum (FWHM) of the instrument response function was 250 ps, and the resolution was 28.6 ps per channel. Data were fitted to multi exponential functions after de convolution of the instrument response function by an iterative re convolution technique using IBH DAS 6.2 data analysis software in which reduced χ^2 and weighted residuals serve as parameters for goodness of fit.

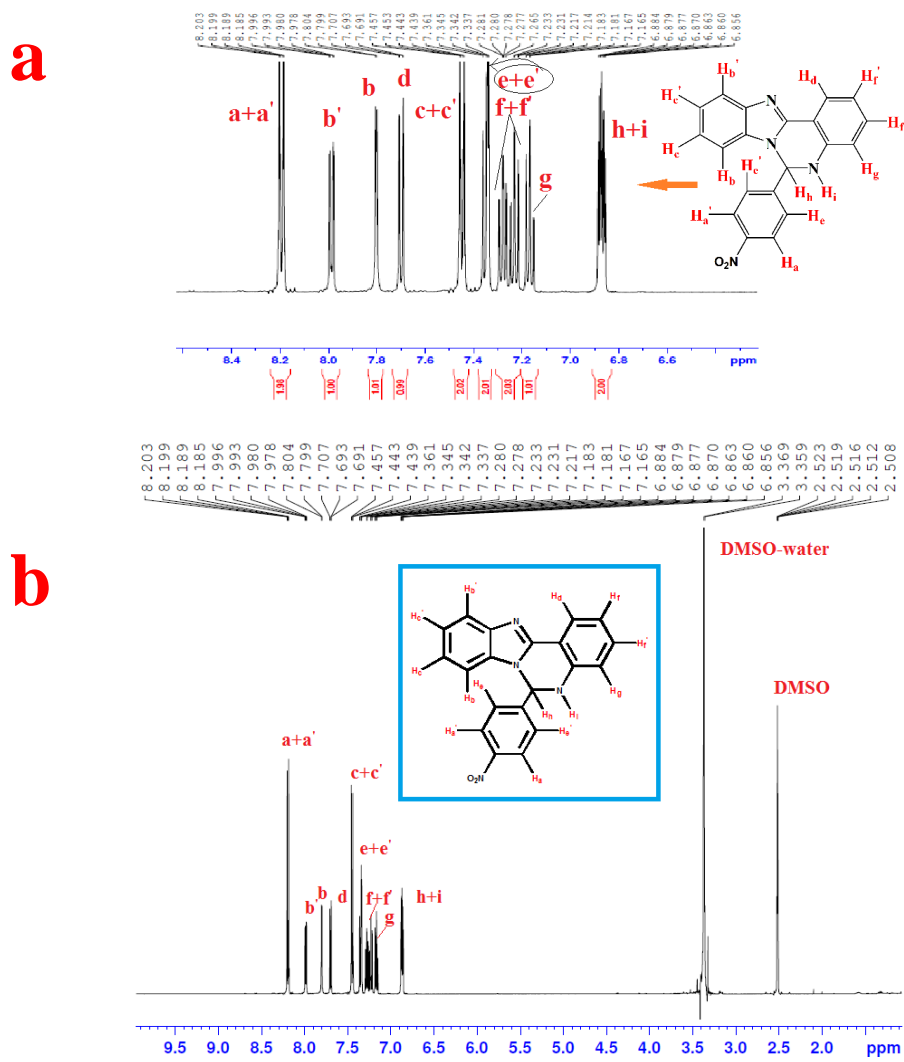


Fig. S1 a) Expansion of aromatic region b) ^1H NMR of L_1H

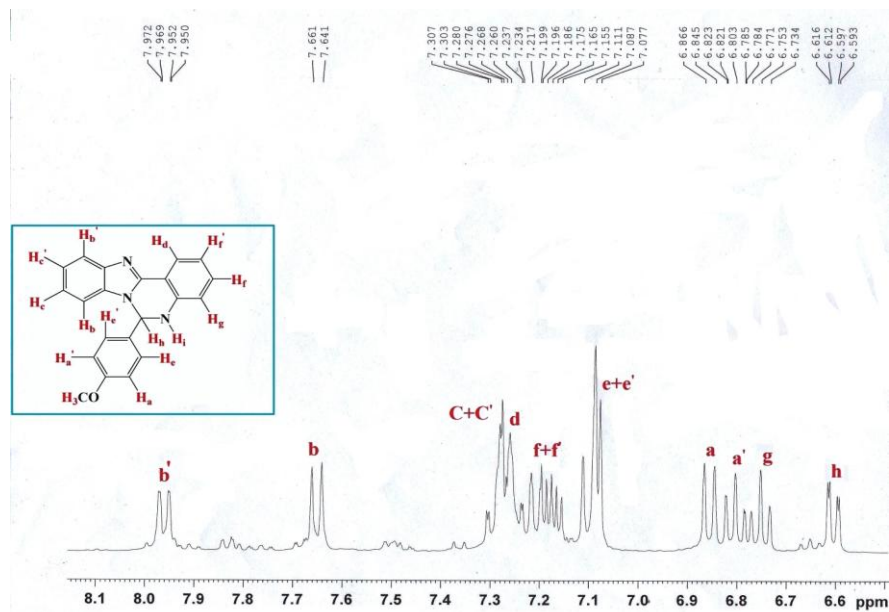
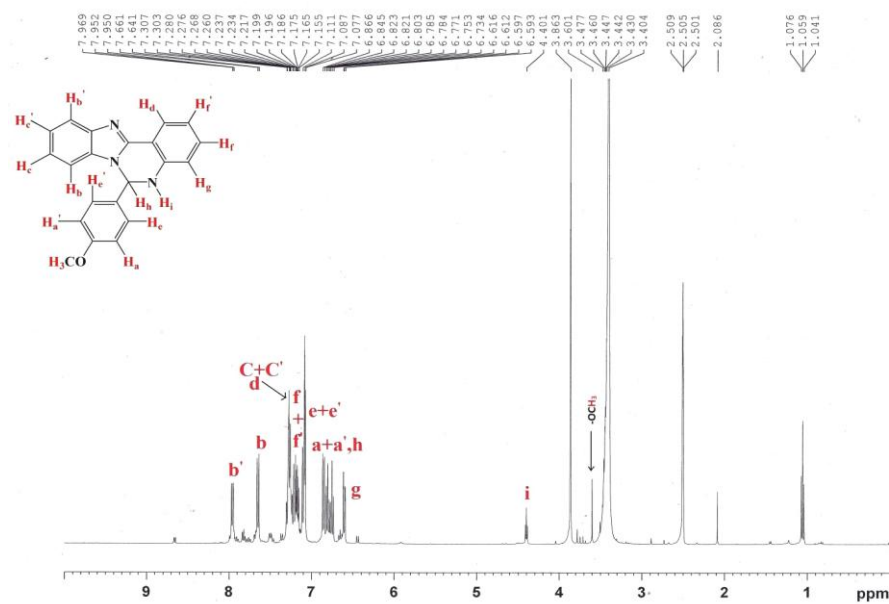
a**b**

Fig. S2 a) Expansion of aromatic region b) 1H NMR of L_2H

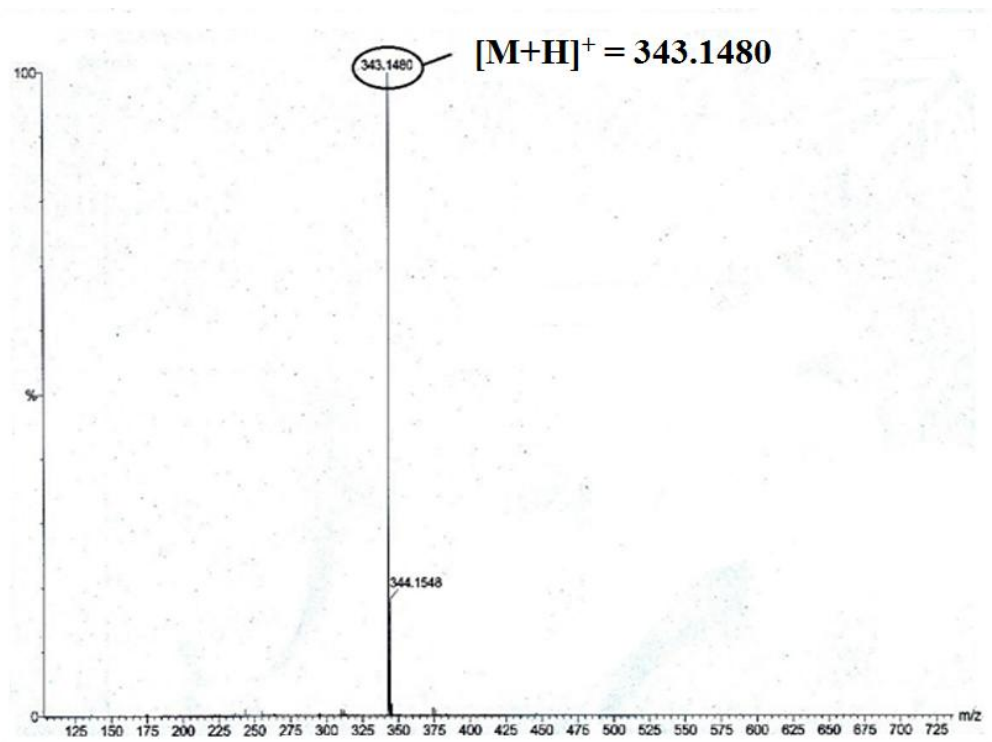


Fig. S3. Mass spectrum of L_1H

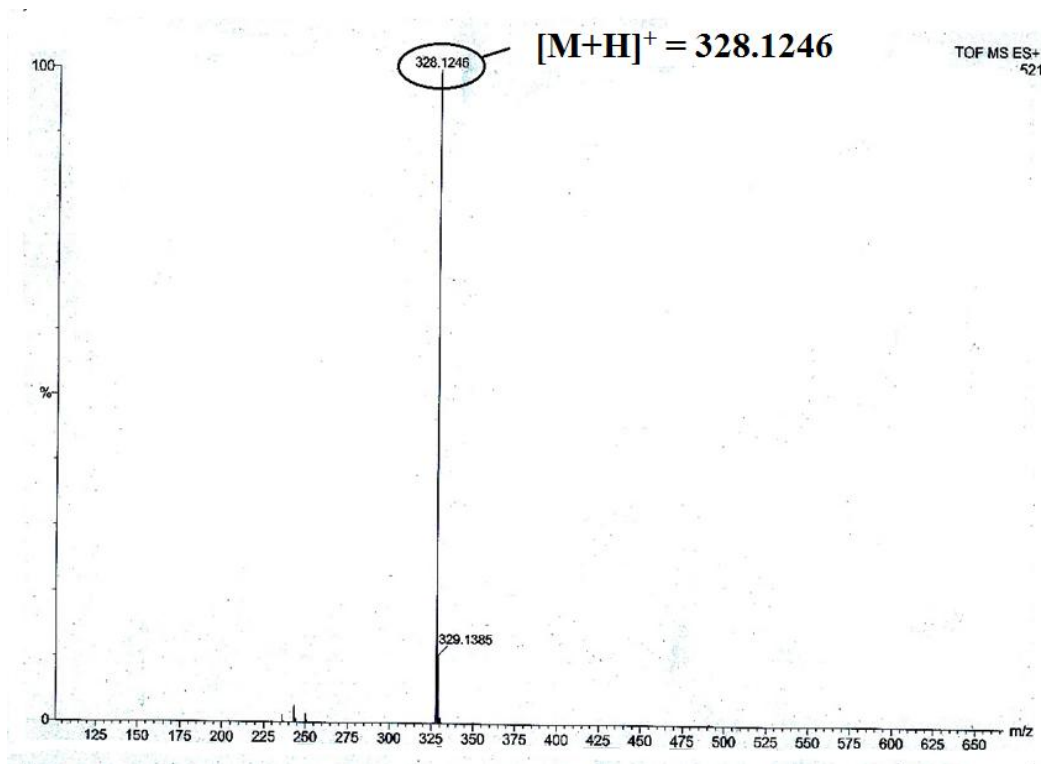


Fig. S4 Mass spectrum of L_2H

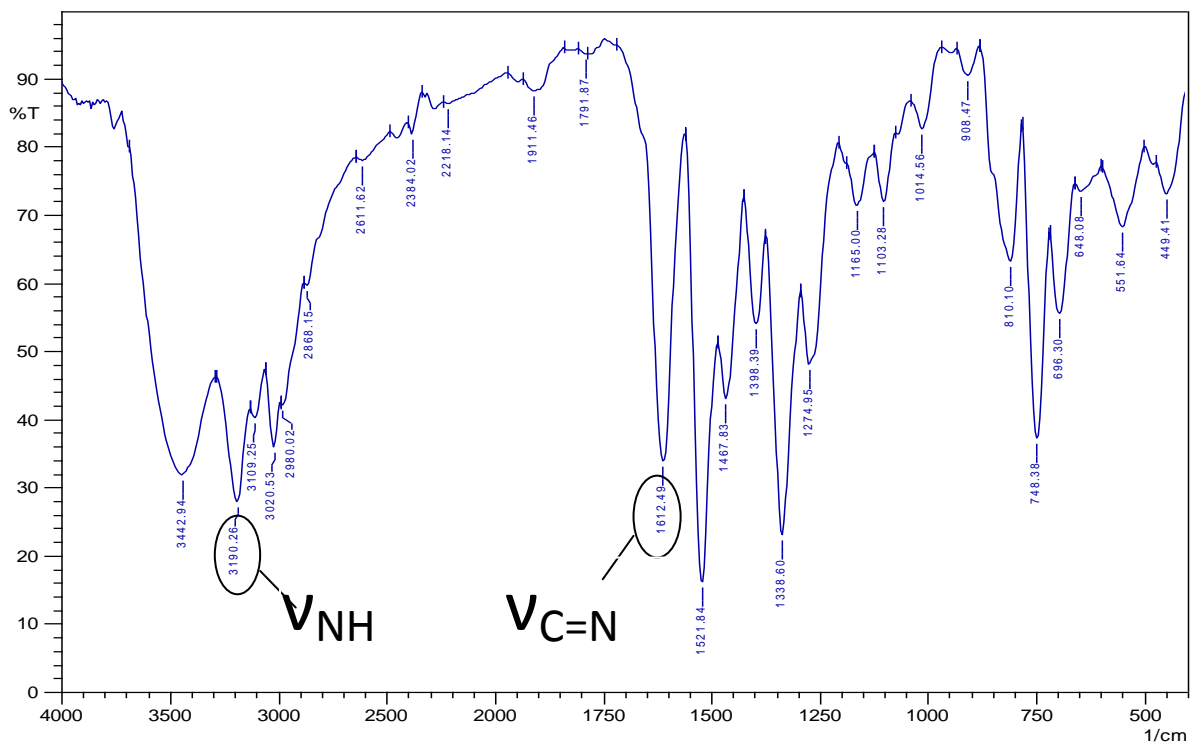


Fig. S5 IR spectrum of **L₁H**

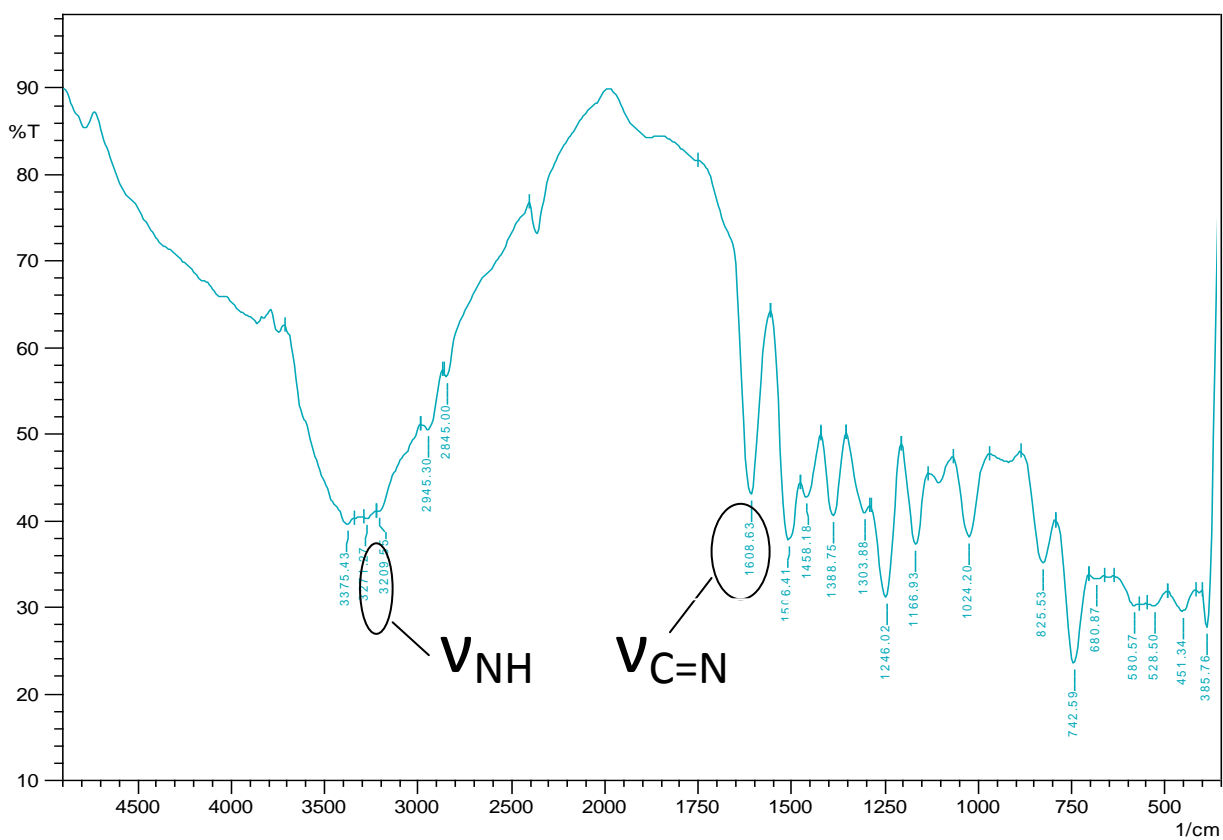


Fig. S6 IR spectrum of **L₂H**

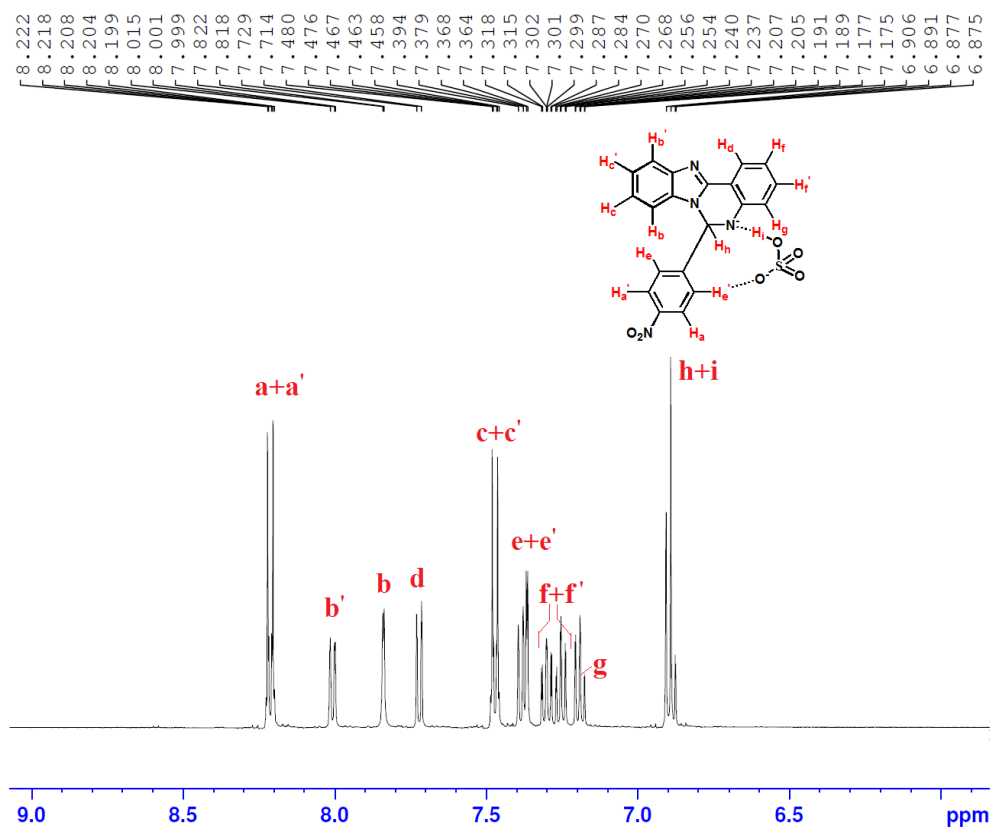
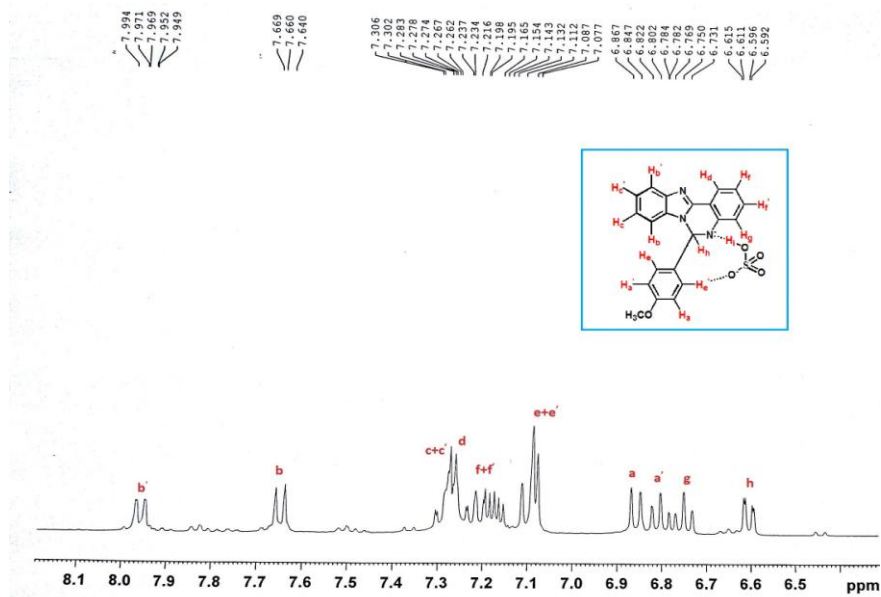


Fig. S7 ^1H NMR expansion of $\text{K}[\text{L}_1\text{H-HSO}_4]$

a)



b)

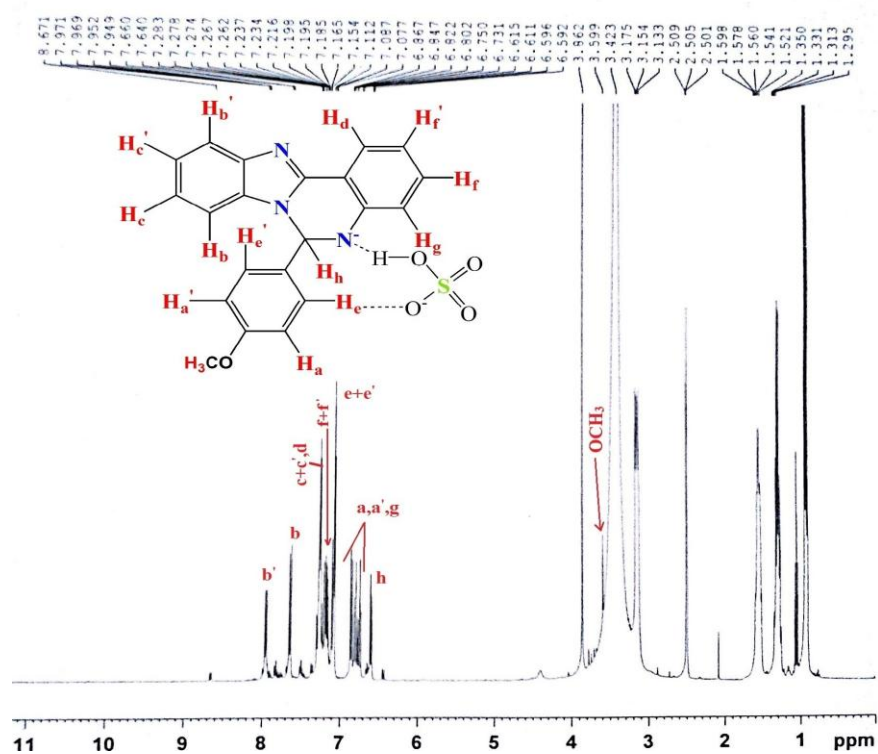


Fig. S8 a) Expansion of aromatic region b) ^1H NMR of $\text{K}[\text{L}_2\text{H-HSO}_4]$

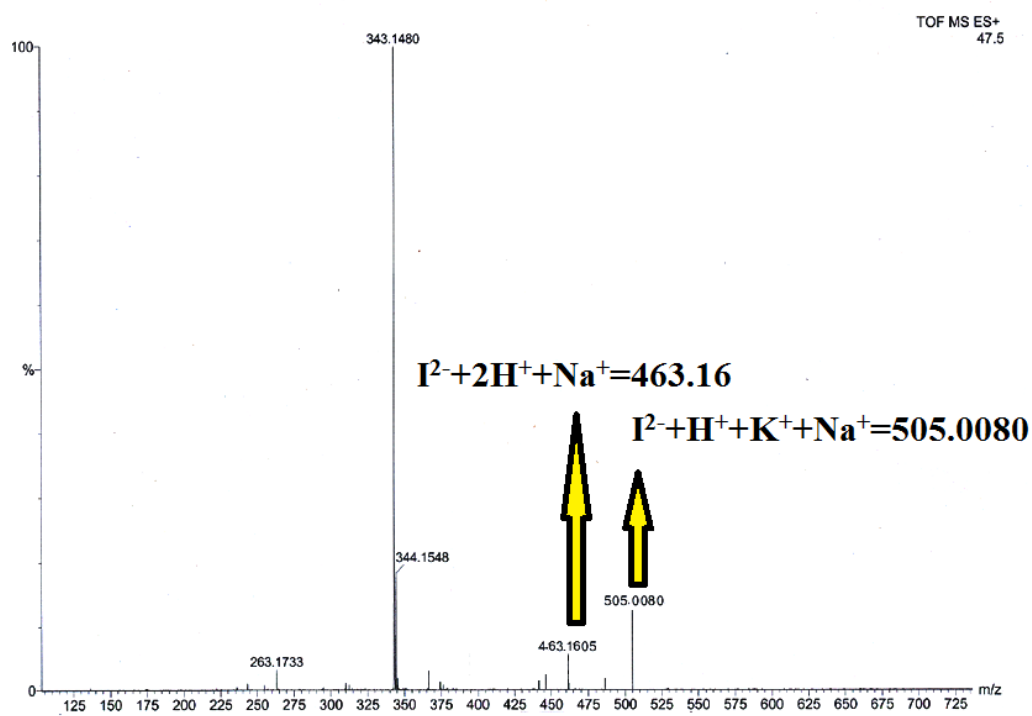


Fig. S9 Mass spectrum of $K[L_1H-HSO_4]$

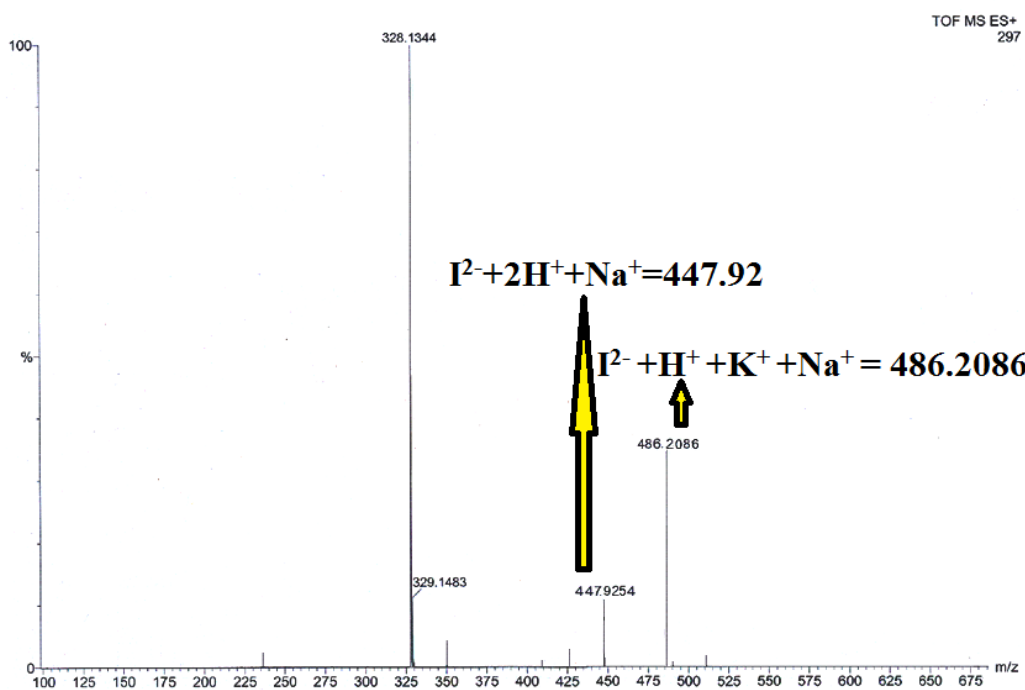


Fig. S10 Mass spectrum of $K[L_2H-HSO_4]$

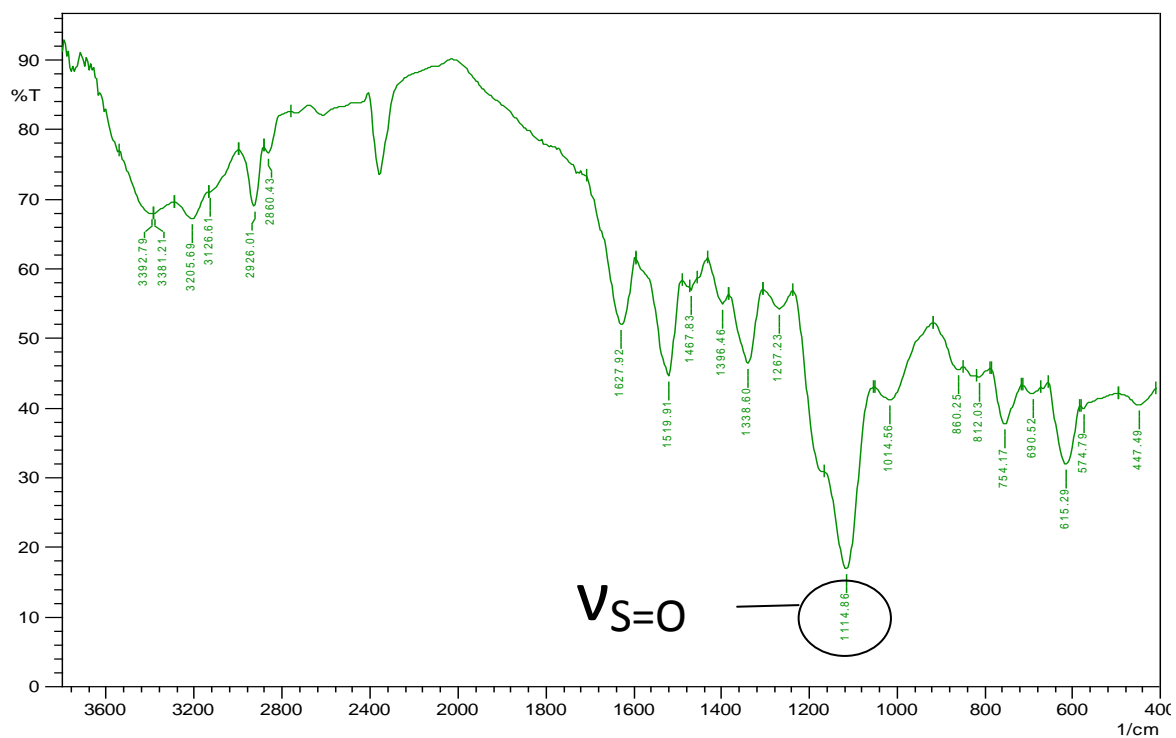


Fig. S11 IR spectrum of $K[L_1H-HSO_4]$

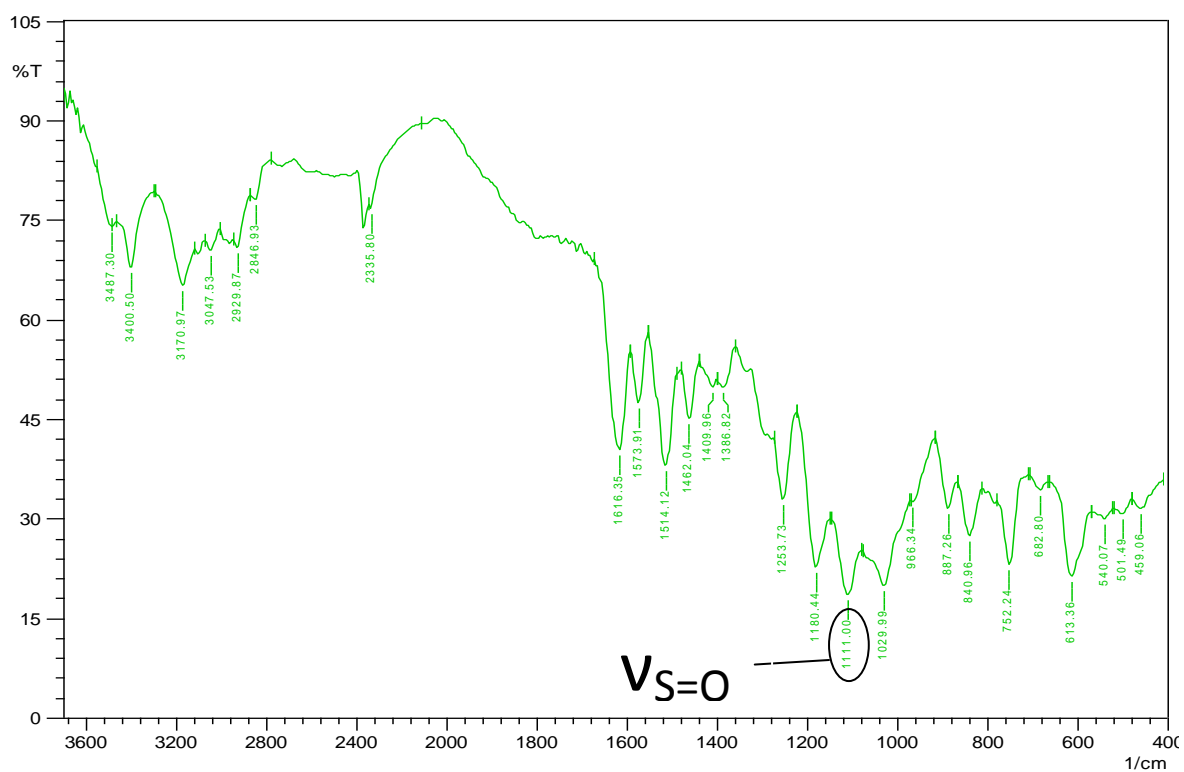
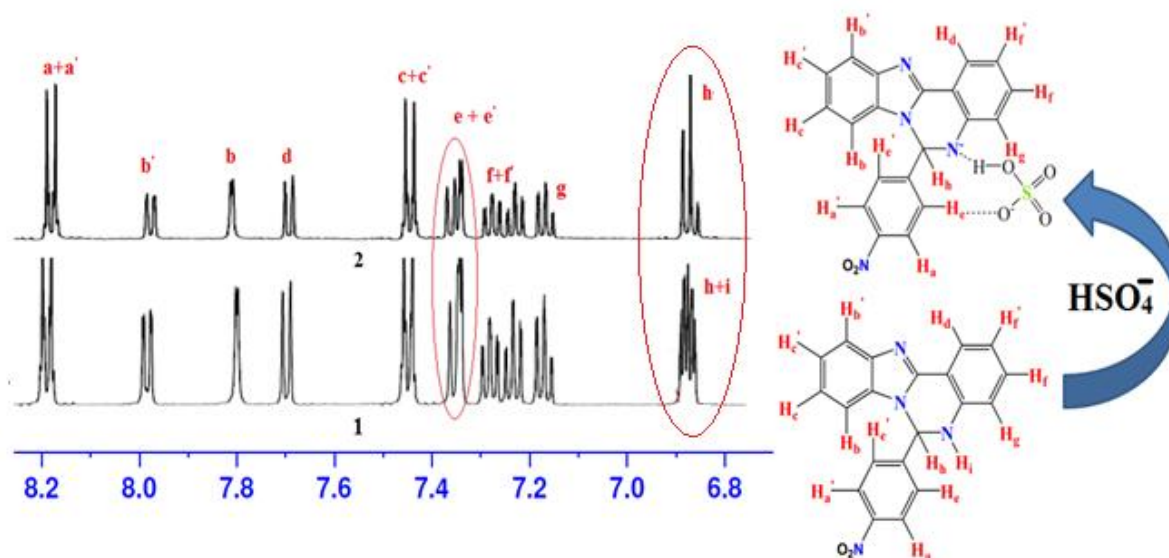


Fig. S12 IR spectrum of $K[L_2H-HSO_4]$

Table S1. ^1H nmr titration data of L_1H

	$\text{H}_a, \text{H}_{a'}$ (2H,d-d)	$\text{H}_{b'}$ (1H,d-d)	H_b (1H,d)	H_d (1H,d-d)	$\text{H}_c, \text{H}_{c'}$ (2H,m)	$\text{H}_e, \text{H}_{e'}$ (2H,m)	$\text{H}_f, \text{H}_f', \text{H}_g$ (3H,m)	H_h, H_i (2H,m)
L_1H	8.194	7.986	7.801	7.704	7.457- 7.439	7.361,7.345 7.342,7.337	7.281- 7.165	6.884,6.879, 6.877,6.870, 6.863,6.860, 6.856
L_1^- HSO_4^-	8.213	8.007	7.82	7.720	7.480- 7.458	7.394,7.39 7.368,7.364	7.318- 7.170	6.906,6.891, 6.877,6.875

**Fig. S13** ^1H NMR titration of L_1H **Table S2.** ^1H NMR titration data of L_2H

	$\text{H}_{b'}$ (1H,d-d)	H_b (1H,d)	$\text{H}_d, \text{H}_c, \text{H}_{c'}$, $\text{H}_e, \text{H}_{e'}, \text{H}_f,$ H_f' (7H,m)	$\text{H}_a, \text{H}_{a'}, \text{H}_g$ (3H,m)	H_h (1H,d-d)	H_i (1H,s)	$-\text{OCH}_3$ (3H,s)
L_2H	7.96	7.651	7.307- 7.077	6.866- 6.734	6.604	4.401	3.601
L_2^- - HSO_4^-	7.9605	7.645	7.306- 7.077	6.867- 6.731	6.594	-	3.601

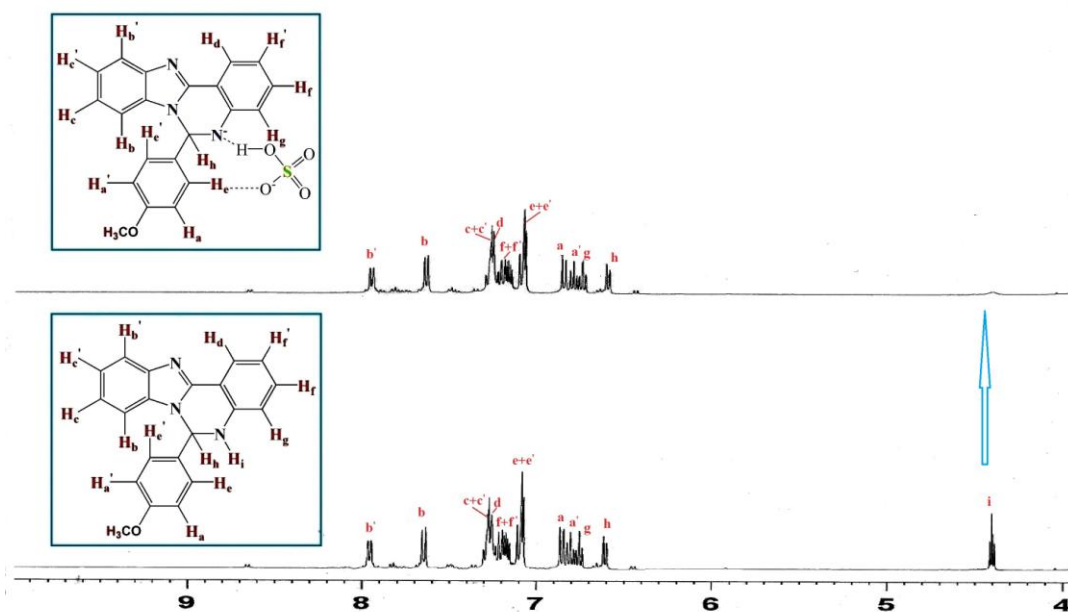


Fig. S14 ^1H NMR titration of L_2H

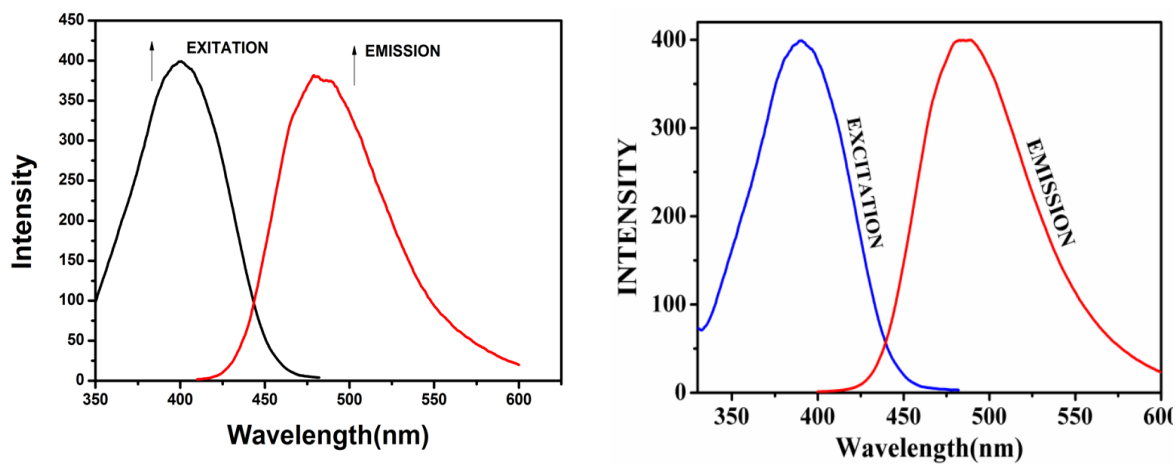


Fig. S15 Absorption and emission spectra of 25 μM of the L_1H (left) and L_2H (right) in 100 mM HEPES buffer (ethanol/water 1:5, v/v) at 25°C

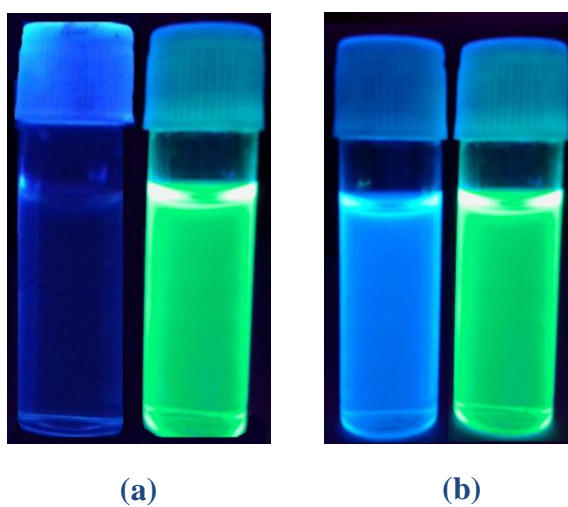


Fig. S16 (a).Fluorescence colour of the L_1H in absence (left) and presence(right) of HSO_4^- ion. (b). Fluorescence colour of the L_2H in absence (left) and presence(right) of HSO_4^- ion.

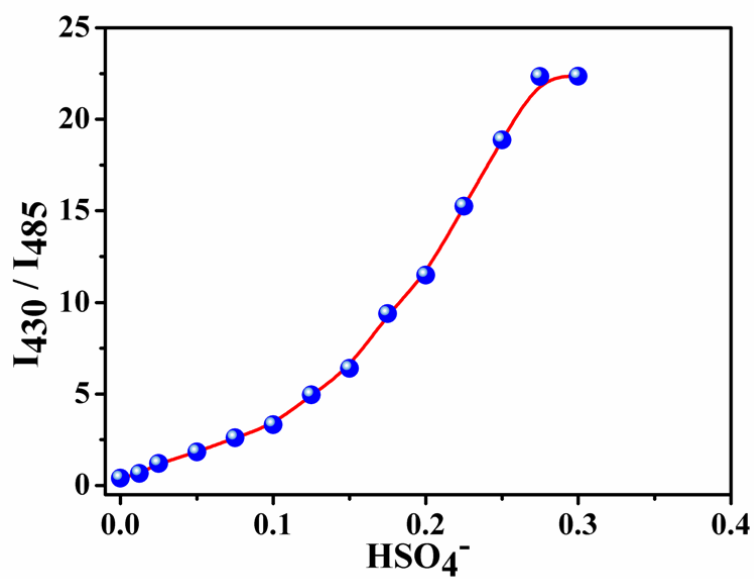


Fig. S17 Ratiometric signalling output of L_2H

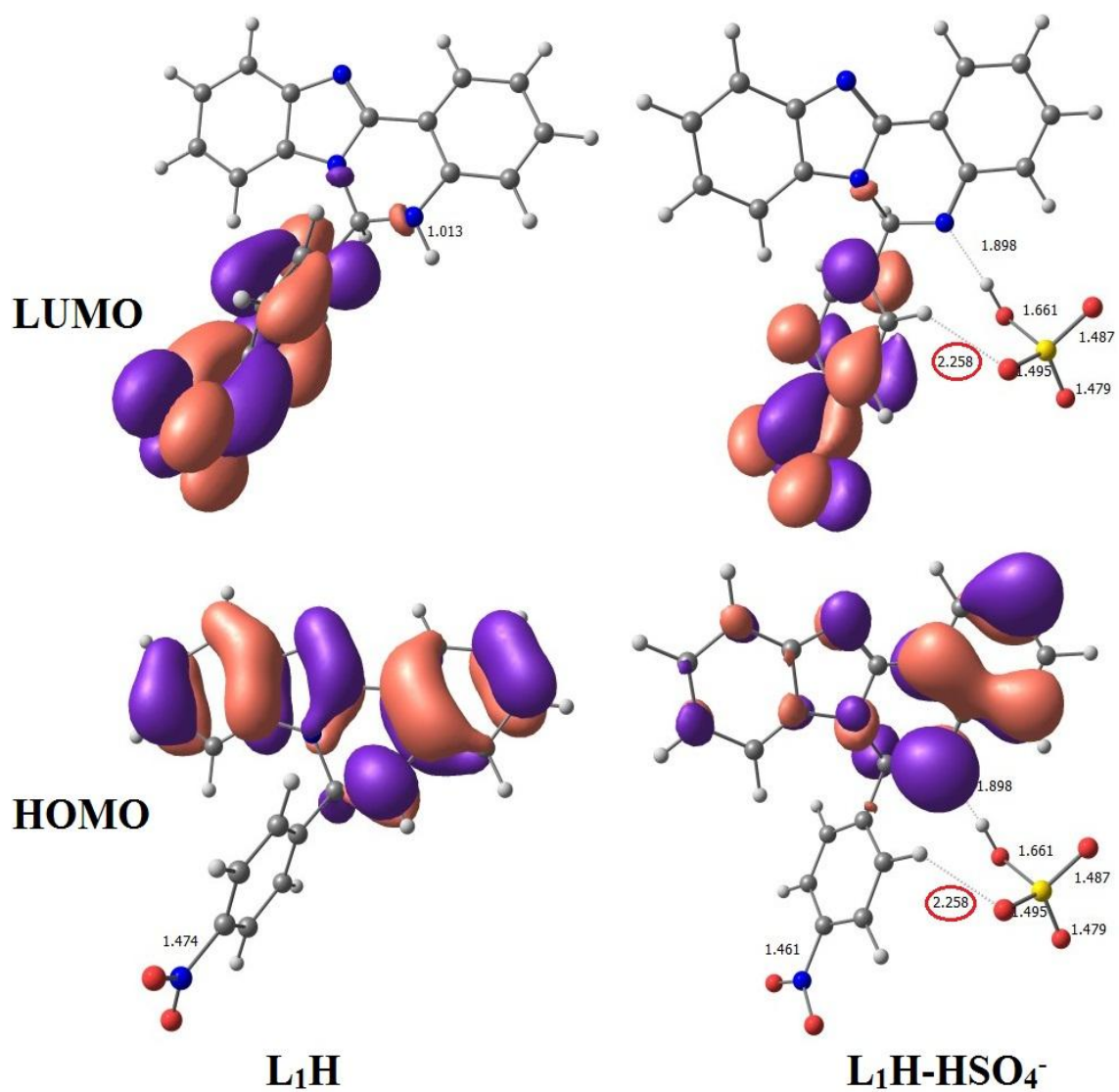


Fig. S18 Computational study of probe L_1H and it's adduct with HSO_4^-

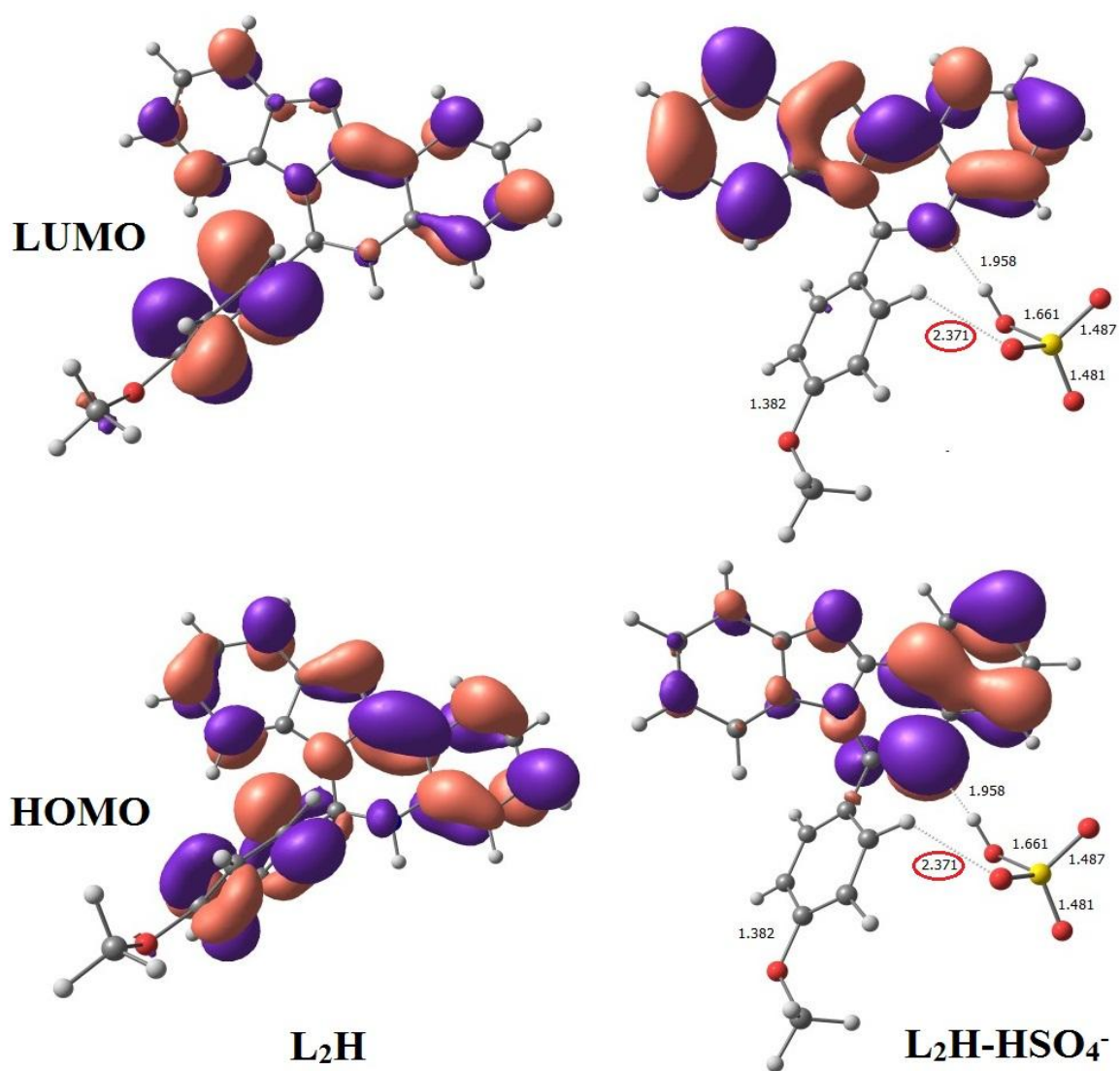


Fig. S19 Computational study of probe **L₂H** and it's adduct with **HSO₄⁻**

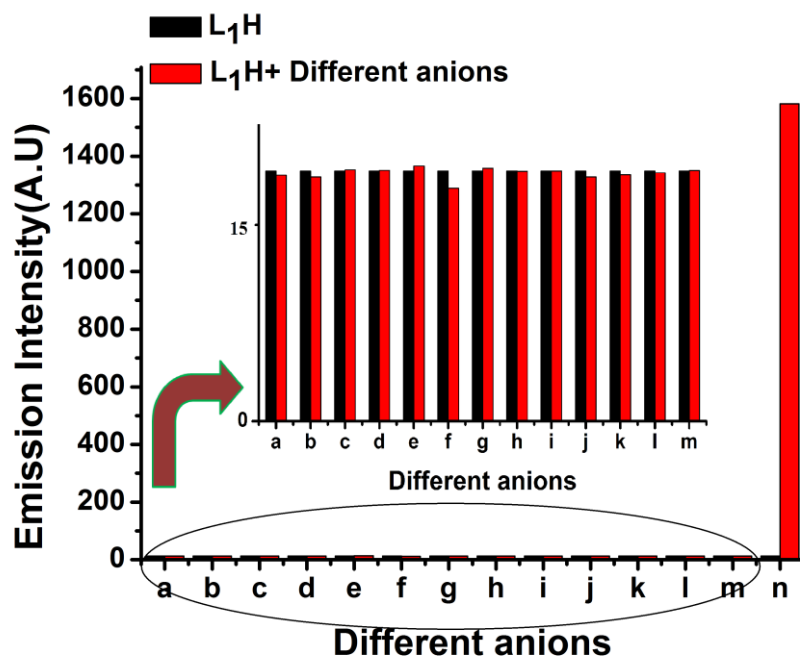


Fig. S20. Fluorescence intensity of **L₁H** in presence of different anions in HEPES buffer (100 mM, pH 7.4; ethanol/water: 1/5, v/v) at 25 °C, (a) OAc⁻, (b) F⁻, (c) I⁻, (d) H₂PO₄⁻, (e) ClO₄⁻, (f) N₃⁻, (g) Br⁻, (h) H₂AsO₄⁻, (i) Cl⁻, (j) SO₄²⁻, (k) S²⁻, (l) CN⁻, (m) NO₃⁻, (n) HSO₄⁻

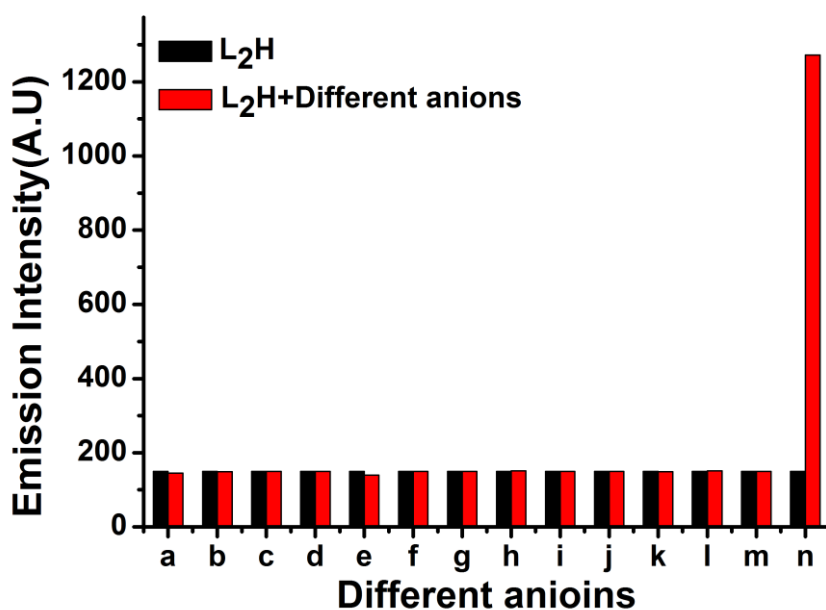


Fig. S21. Fluorescence intensity of **L₂H** in presence of different anions in HEPES buffer (100 mM, pH 7.4; ethanol/water: 1/5, v/v) at 25 °C, (a) Cl⁻, (b) Br⁻, (c) I⁻, (d) F⁻, (e) OAc⁻, (f) H₂PO₄⁻, (g) N₃⁻, (h) ClO₄⁻, (i) H₂AsO₄⁻, (j) SO₄²⁻, (k) S²⁻, (l) CN⁻, (m) NO₃⁻, (n) HSO₄⁻

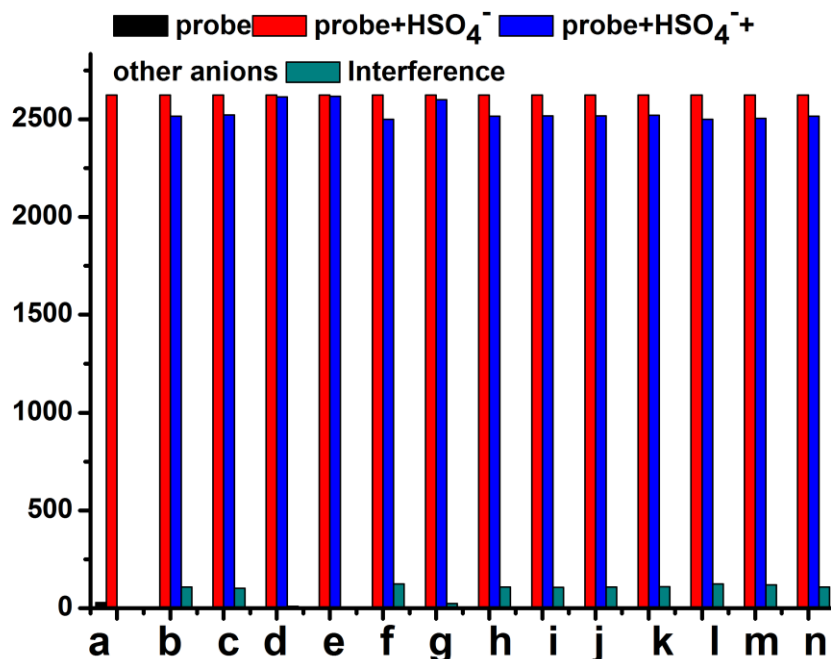


Fig. S22 Change of relative fluorescence intensity profile of **L₁H** in presence of different anions in ethanol: water (1: 5, v/v) at room temperature ($\lambda_{\text{ex}}= 400 \text{ nm}$) where (a) HSO_4^- , (b) OAc^- , (c) F^- , (d) I^- , (e) H_2PO_4^- , (f) ClO_4^- , (g) N_3^- , (h) Br^- , (i) H_2AsO_4^- , (j) NO_3^- , (k) SO_4^{2-} , (l) S^{2-} , (m) CN^-

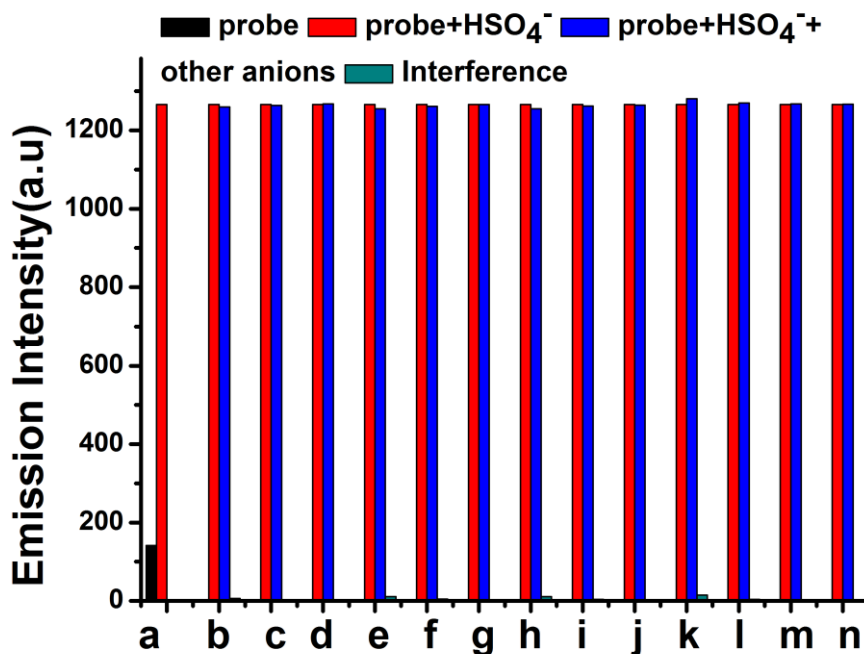


Fig. S23 Change of relative fluorescence intensity profile of **L₂H** in presence of different anions in ethanol: water (1: 5, v/v) at room temperature ($\lambda_{\text{ex}}= 390 \text{ nm}$). (a) HSO_4^- , (b) Cl^- , (c) Br^- , (d) I^- , (e) F^- , (f) OAc^- , (g) H_2PO_4^- , (h) H_2AsO_4^- , (i) ClO_4^- , (j) N_3^- , (k) SO_4^{2-} , (l) S^{2-} , (m) CN^- , (n) NO_3^-

Table S3. Life time detail of **L₁H**

	B₁	B₂	T₁(ns)	T₂(ns)	T_{av}(ns)	χ²	φ	K_r	K_{nr}
(L₁H)	5.07	94.93	1.74	8.67	8.32	1.082	0.0055	0.0006 6	0.119 5
L₁+HSO₄⁻ (1:0.5)	29.28	70.72	1.93	12.53	9.42	1.028			
L₁+HSO₄⁻ (1:1)	18.98	81.02	1.86	12.84	10.75	1.064	0.355	0.033	0.06

Table S4. Life time detail of **L₂H**

	B₁	B₂	T₁(ns)	T₂(ns)	T_{av}(ns)	X²	φ	K_r	K_{nr}
(L₂H)	33.61	66.39	7.30	10.96	8.26	1.068	0.065	0.00785	0.113
L₂+HSO₄⁻ (1:0.5)	14.67	85.33	6.76	8.89	8.58	1.031	-	-	-
L₂+HSO₄⁻ (1:1)	6.55	93.45	3.65	12.36	11.79	1.078	0.48	0.04	0.044 11

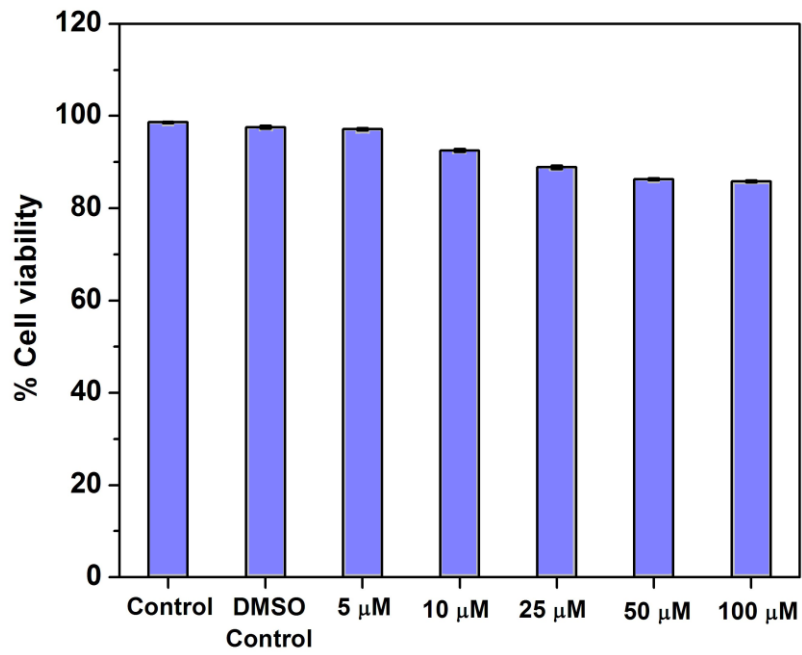


Fig. S24 Cytotoxic effect of **L₁H** (5, 10, 25, 50 and 100 μ M) in HeLa cells incubated for 8 h

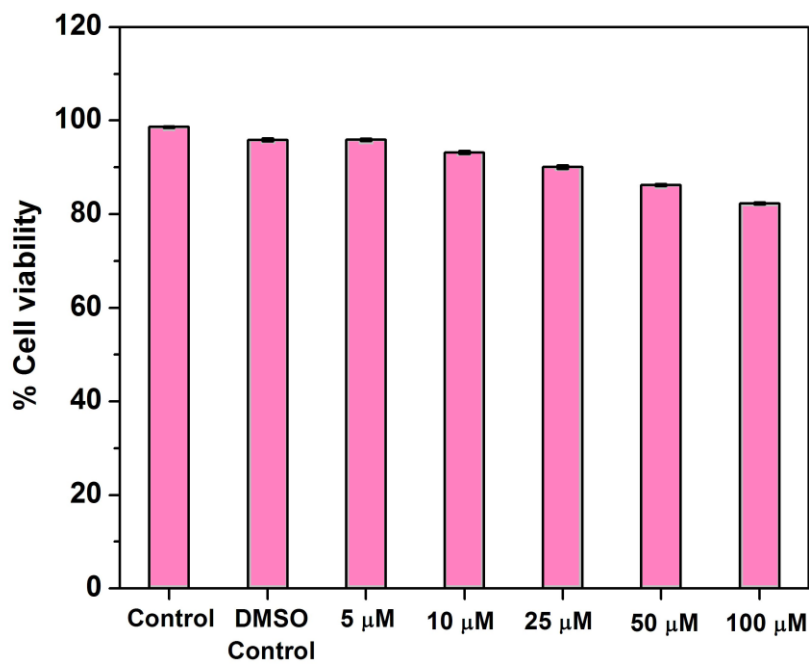


Fig. S25 Cytotoxic effect of **L₂H** (5, 10, 25, 50 and 100 μ M) in HeLa cells incubated for 8 h