A smart copper (II) responsive binuclear gadolinium (III) complex based magnetic resonance imaging contrast agent

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Fig. S1 ¹H NMR spectrum of Compound 3 in DMSO (600 MHz).



Fig. S2 ¹³C NMR spectrum of Compound 3 in CDCl₃ (151 MHz).





Fig. S3 HRMS (ESI) spectrum and expanded view of Compound 3. The peaks at m/z = 661.92584, 672.91690, and 683.90682 correspond to $[M + 2H]^{2+}$ (calc. 661.92650), $[M + H + Na]^{2+}$ (calc.672.91747), and $[M + 2Na]^{2+}$ (calc. 683.90844) respectively.



Fig. S4 ¹H NMR spectrum of Compound 4 in DMSO (600 MHz).



Fig. S5¹³C NMR spectrum of Compound 4 in D₂O (151 MHz).



Fig. S6 MALDI-TOF-MS spectrum of **Compound 4**. The peak at m/z = 984.7 correspond to [M - H]⁻ (calc. 984.46).



Fig. S7 MALDI-TOF-MS spectrum of $[Gd_2(DO3A)_2BMPNA]$. The peak at m/z = 1281.4 correspond to $[M + H]^+$ (calc. 1281.46).



Fig. S8 MALDI-TOF-MS spectrum of $[Tb_2(DO3A)_2BMPNA]$. The peak at m/z = 1284.4 correspond to $[M + H]^+$ (calc. 1284.3).



- Fig. S9 (a) Luminescence decay of [Tb₂(DO3A)₂BMPNA] in H₂O and D₂O without addition of Cu²⁺;
 - (b) Luminescence decay of [Tb₂(DO3A)₂BMPNA] in H₂O and D₂O with addition of Cu²⁺.



Fig. S10 Luminescence spectra of $[Tb_2(DO3A)_2BMPNA]$ (20 µM) upon addition of different concentrations of Cu²⁺ (0-2 equiv.) in HEPES-buffered (pH 7.4, 100 mM) aqueous solutions. Excitation wavelength: 295 nm. Inset shows the luminescence at 545 nm as a function of Cu²⁺ concentration suggesting a 1:1 binding ratio between $[Tb_2(DO3A)_2BMPNA]$ and Cu²⁺.



Fig. S11 ¹H NMR spectrum of Compound 1 in CDCl₃ (600 MHz).



Fig. S12 ¹H NMR spectrum of Compound 2 in CDCl₃ (600 MHz).



Fig. S13 MALDI-TOF-MS spectrum of $[Gd_2(DO3A)_2BMPNA]$ in the presence of Zn^{2+} at room temperature after 7 days. No metal-ion exchange was observed.



Fig. S14 Luminescence decay of $[Tb_2(DO3A)_2BMPNA]$ in H₂O (black) and D₂O (red) with addition of Cu²⁺ and phosphate.