

Supplementary information

Scalable production of transition metal disulphide/graphite nanoflake composites for high-performance lithium storage

Zhi-Qiang Duan ^{a,1}, Yan-Chun Sun ^{b,1}, Yi-Tao Liu ^{a,*}, Xu-Ming Xie ^{a,*}, Xiao-Dong Zhu ^{c,*}

^a*Laboratory of Advanced Materials (Ministry of Education), Department of Chemical Engineering,*

Tsinghua University, Beijing 100084, China

^b*Heilongjiang River Fishery Research Institute, Chinese Academy of Fishery Sciences,*

Harbin 150070, China

^c*Academy of Fundamental and Interdisciplinary Sciences, Harbin Institute of Technology,*

Harbin 150080, China

*Corresponding authors. Tel.: +86 10 62773607; fax: +86 10 62784550.

E-mail addresses: liu-yt03@mails.tsinghua.edu.cn (Y.-T. Liu), xxm-dce@mail.tsinghua.edu.cn (X.-M. Xie), zxd9863@163.com (X.-D. Zhu).

¹These authors contributed equally to this work.

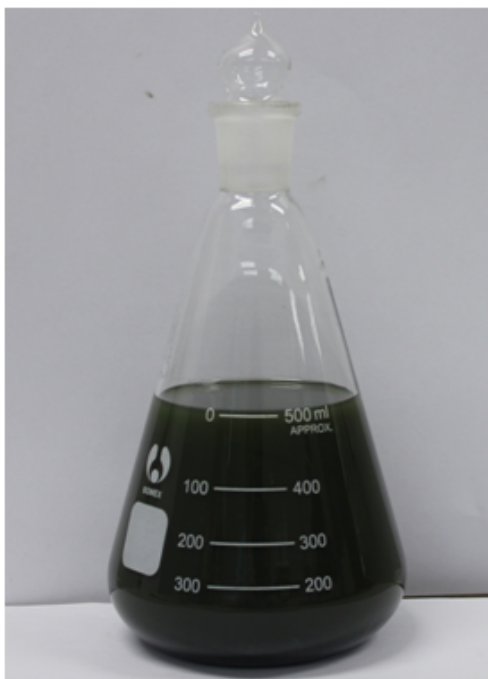


Fig. S1. Digital image of a MoS₂ nanoflakes/DMF solution (10 mg mL⁻¹) after the ball milling and short-time sonication processes.



Fig. S2. Digital image of a WS₂ nanoflakes/DMF solution (10 mg mL⁻¹) after the ball milling and short-time sonication processes.



Fig. S3. Digital image of a graphite nanoflakes/DMF solution (10 mg mL^{-1}) after the ball milling and short-time sonication processes.

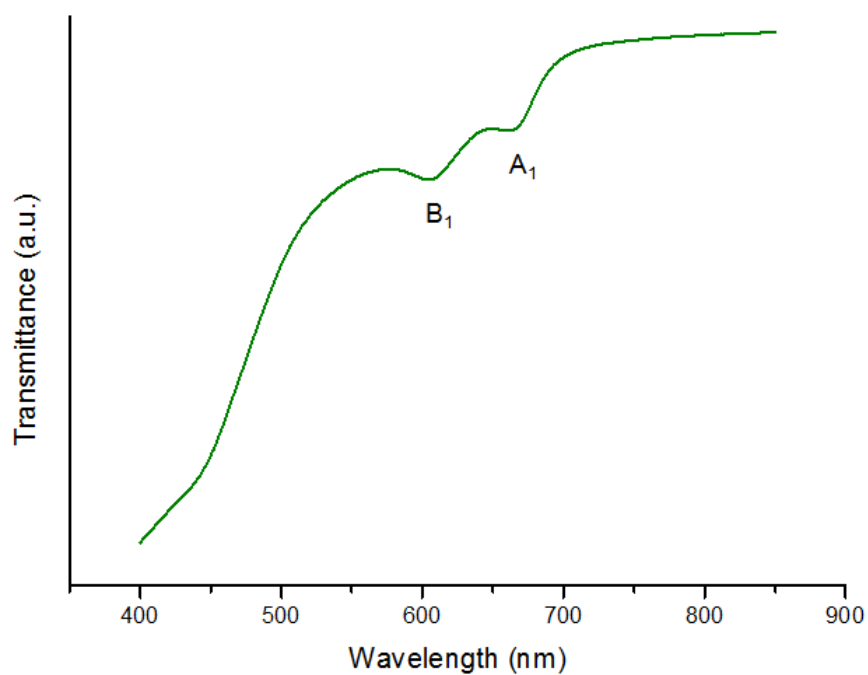


Fig. S4. UV/vis spectrum of a MoS_2 nanoflakes/DMF solution. The spectrum has two characteristic peaks at 674 and 622 nm, respectively, which represent the successful exfoliation of MoS_2 nanoflakes in DMF.

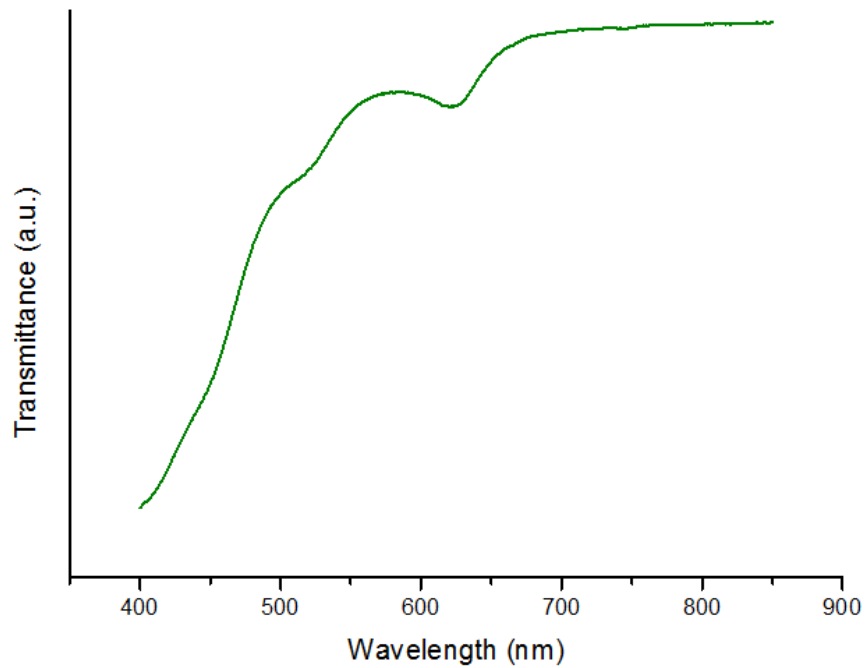


Fig. S5. UV/vis spectrum of a WS₂ nanoflakes/DMF solution. The spectrum has one characteristic peak at 621 nm, which represents the successful exfoliation of WS₂ nanoflakes in DMF.

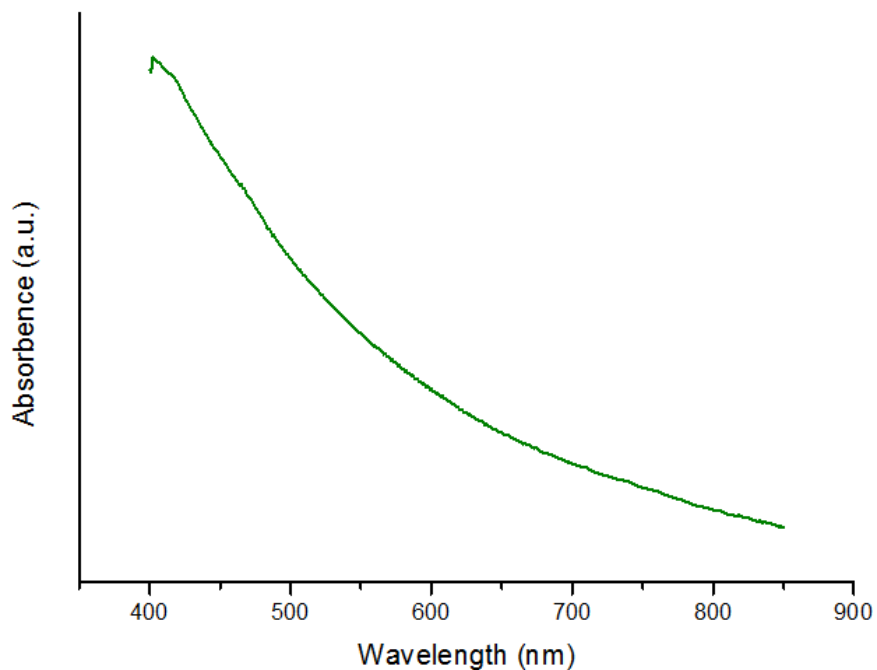


Fig. S6. UV/vis spectrum of a graphite nanoflakes/DMF solution. Note that the UV/vis spectrum is featureless within this wavelength range.

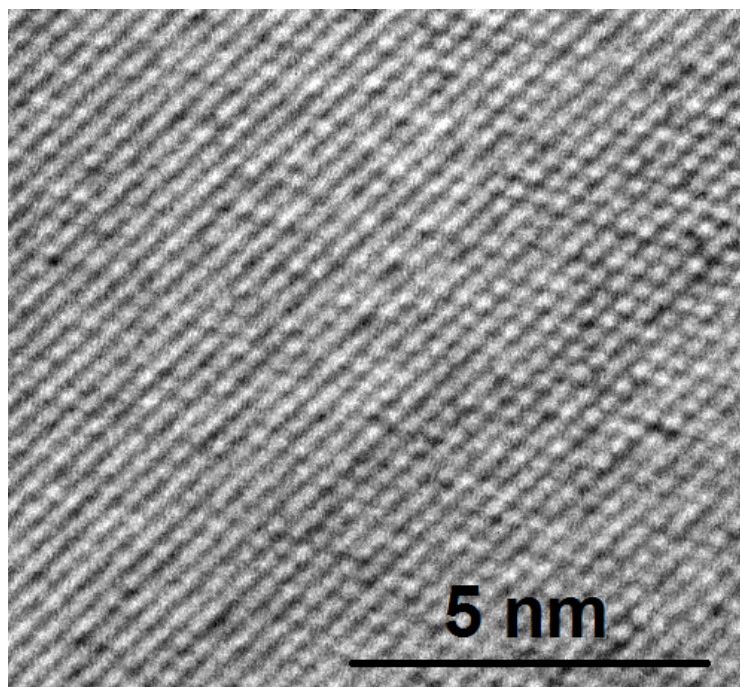


Fig. S7. Atomically resolved HRTEM image of a MoS₂ nanoflake showing a high degree of hexagonal symmetry.

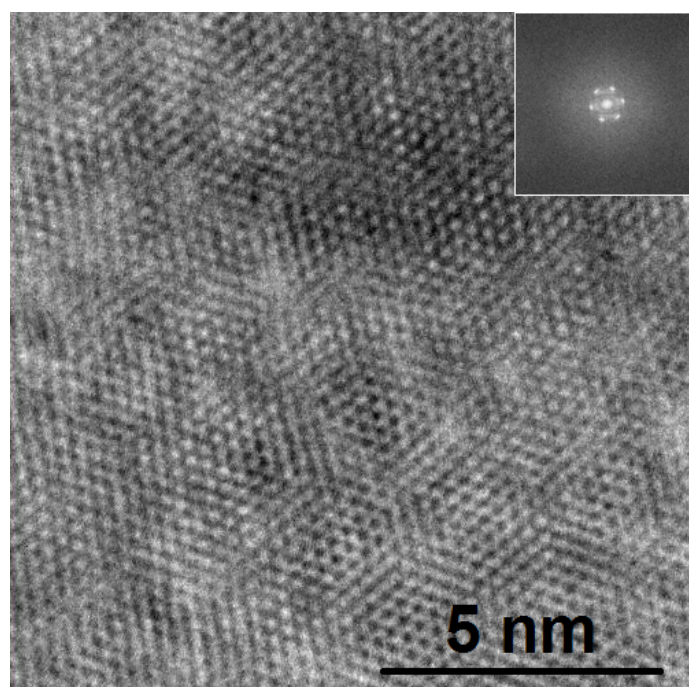


Fig. S8. Atomically resolved HRTEM image of a WS₂ nanoflake showing a high degree of hexagonal symmetry.

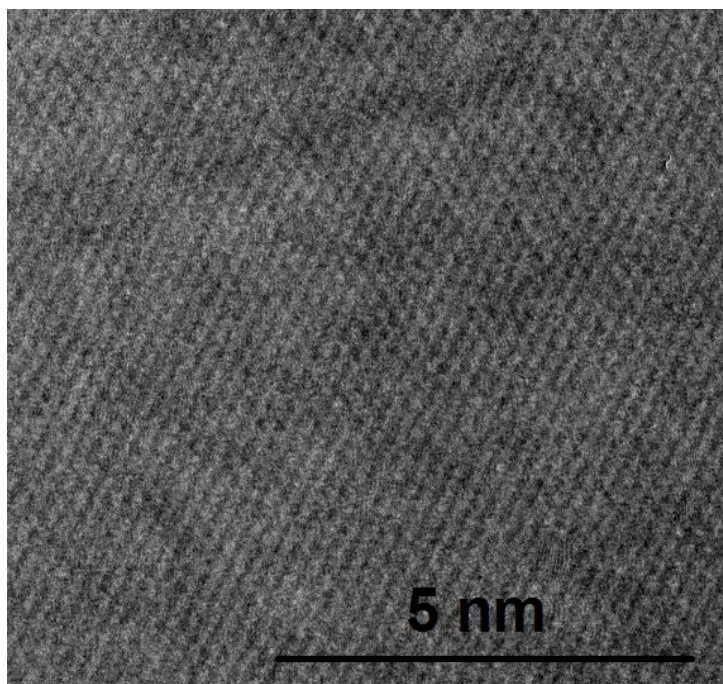


Fig. S9. Atomically resolved HRTEM image of a graphite nanoflake showing a high degree of hexagonal symmetry.

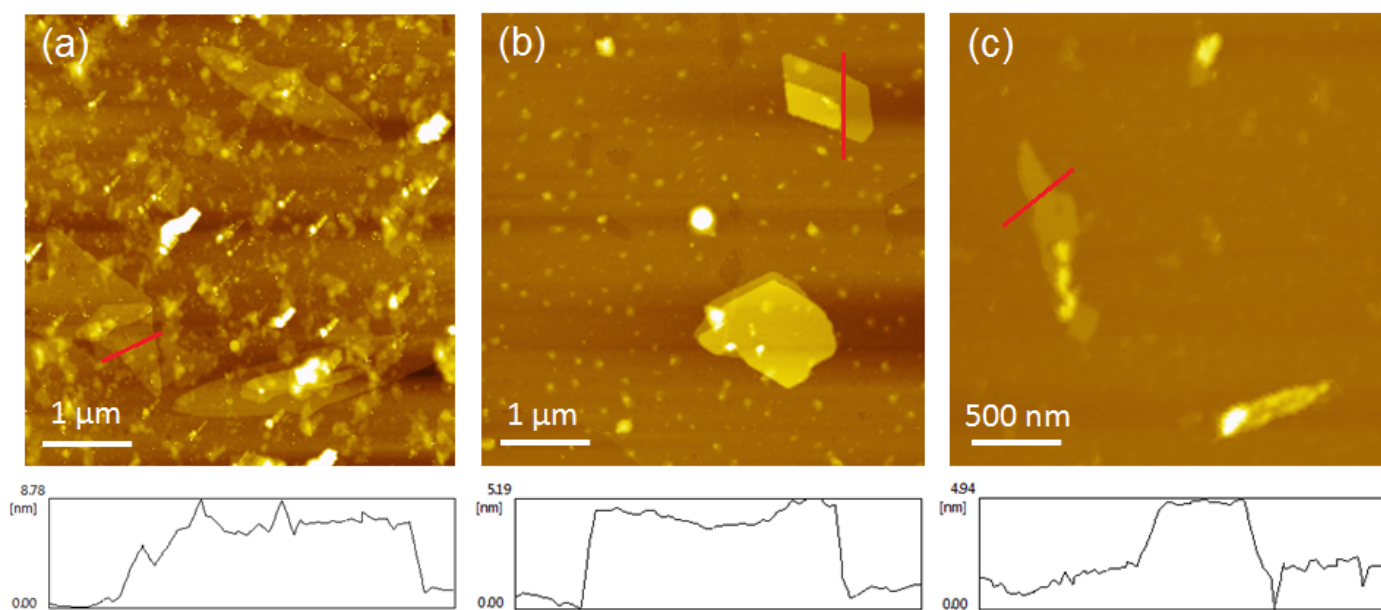


Fig. S10. SPM images of (a) graphite, (b) MoS₂ and (c) WS₂ nanoflakes and the height profiles corresponding to the red lines.

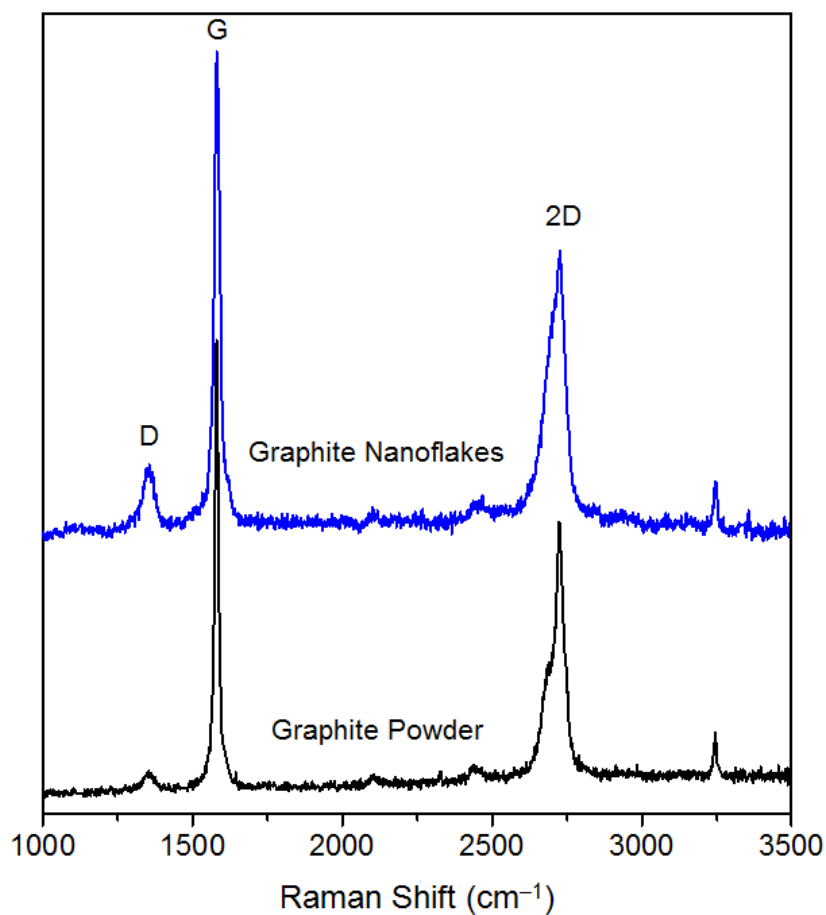


Fig. S11. Raman spectra of graphite powder and graphite nanoflakes.

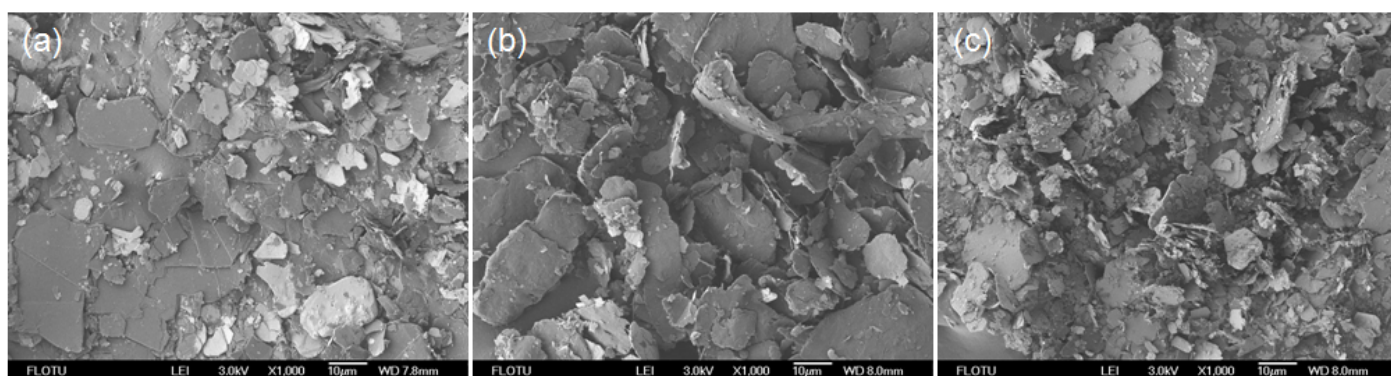


Fig. S12. SEM images of (a) graphite, (b) MoS₂ and (c) WS₂ nanoflakes.

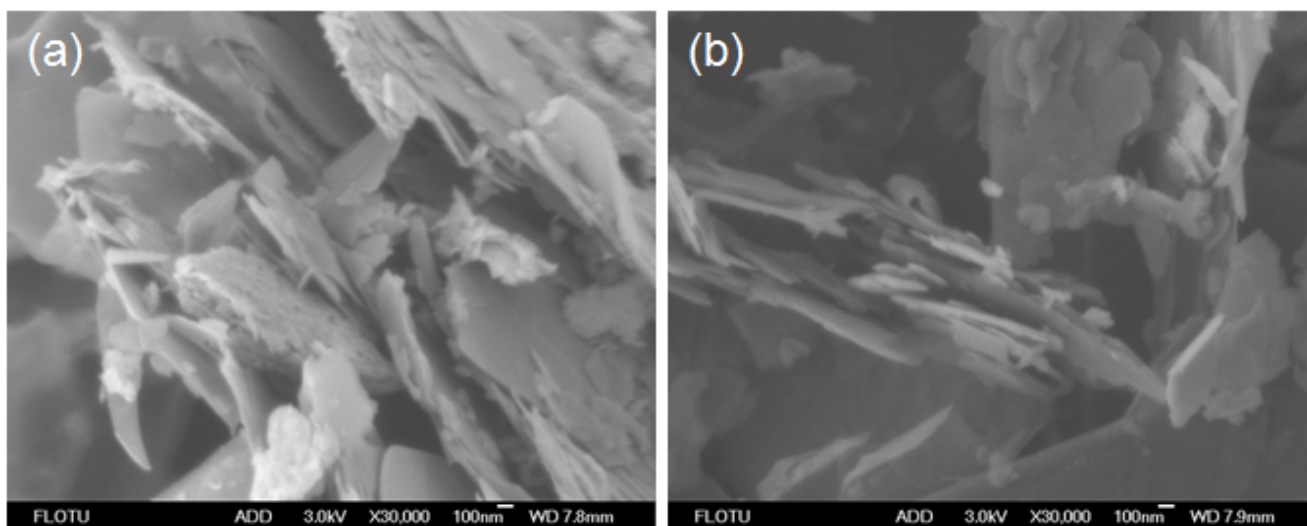


Fig. S13. High-magnification SEM images of (a) MoS₂/graphite and (b) WS₂/graphite nanoflake composites at a wt ratio of 50/50.

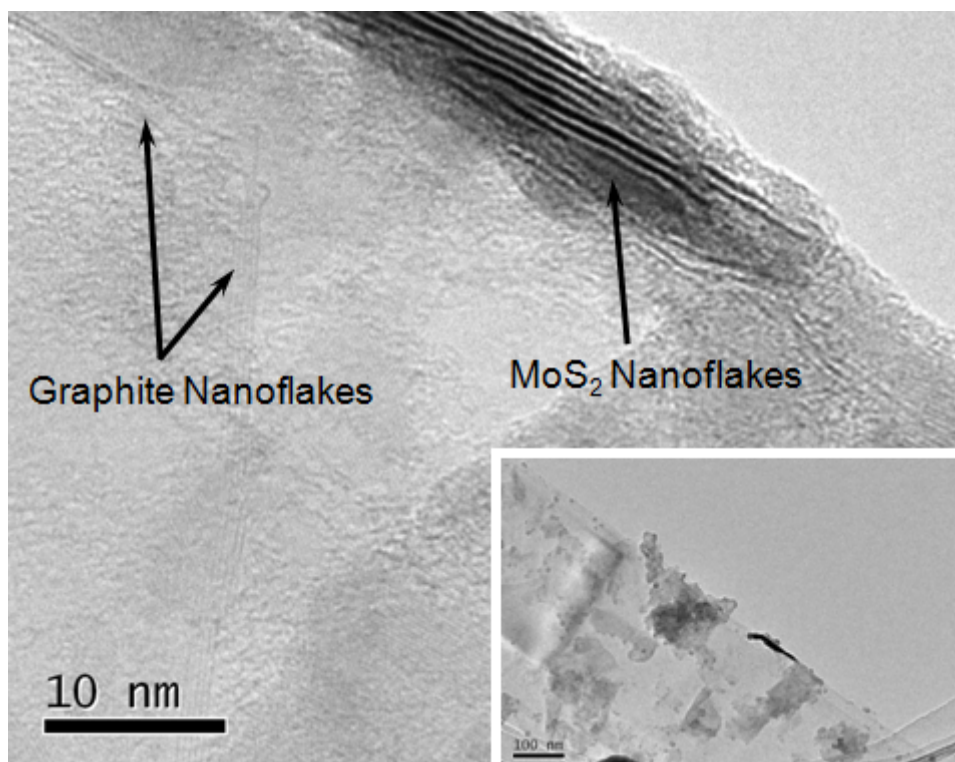


Fig. S14. HRTEM image of MoS₂/graphite nanoflake composite at a wt ratio of 50/50. The insert is the corresponding low-magnification TEM image. In the HRTEM image we can see that the MoS₂ and graphite nanoflakes are mixed uniformly, and their edges can be clearly distinguished.

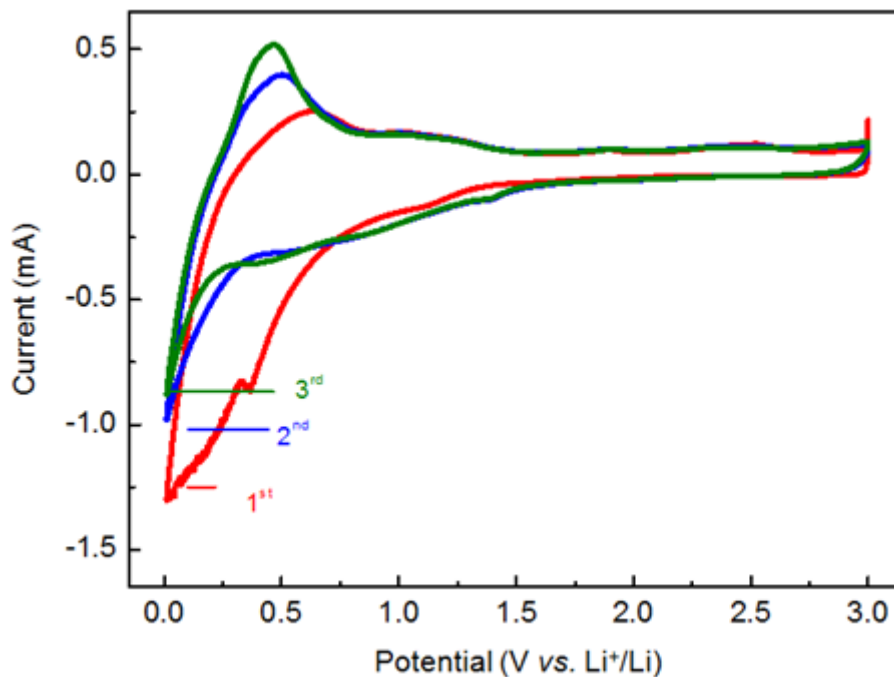


Fig. S15. CV (scan rate = 0.1 mV s⁻¹) curves of graphite nanoflakes.

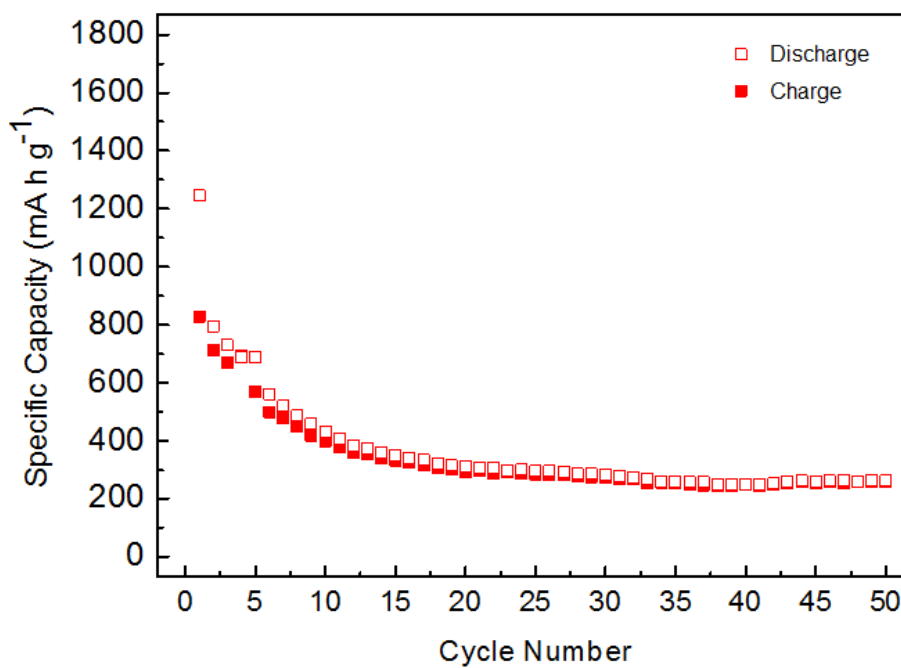
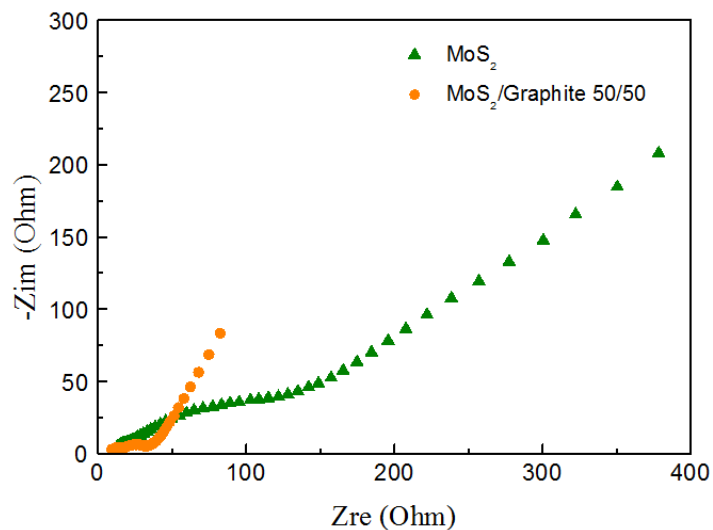
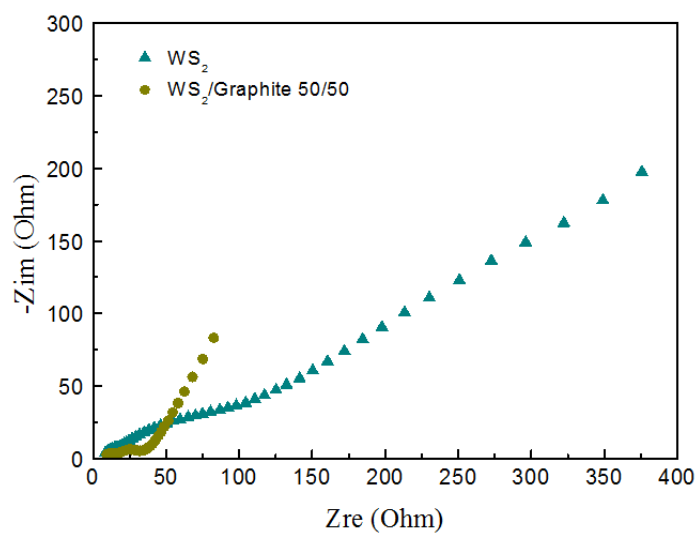


Fig. S16. Cycle behaviour (current density = 100 mA g⁻¹) of graphite nanoflakes.



Electrode	R_{Ω} (Ω)	R_f (Ω)	R_{ct} (Ω)
MoS ₂ /Graphite 50/50	7.0	9.7	15.5
MoS ₂ Nanoflakes	9.0	16.5	57.1

Fig. S17. Nyquist plots of MoS₂ nanoflakes and MoS₂/graphite nanoflake composites at a wt ratio of 50/50.



Electrode	R_{Ω} (Ω)	R_f (Ω)	R_{ct} (Ω)
WS ₂ /Graphite 50/50	8.1	18.1	13.6
MoS ₂ Nanoflakes	12.6	20.2	109.2

Fig. S18. Nyquist plots of WS₂ nanoflakes and WS₂/graphite nanoflake composites at a wt ratio of 50/50.