Supplementary Data for

Synthesis of corrole-fullerene dyad *via* [4+2] cycloaddition reaction

Chengjie Li, ^{a, †} Jie Zhang, ^{a,b, †} Xiujun Liu, ^a Yongzhu Zhou, ^a Dongming Sun, ^c Ping Cheng, ^c Bao Zhang ^{a,*} and Yaqing Feng ^{a, b, *}

^a School of Chemical Engineering and Technology, Tianjin University, Tianjin 300072, P. R. China

^b Synergetic Innovation Center of Chemical Science and Engineering, Tianjin 300072, P. R. China

^c School of Environmental and Chemical Engineering, Shanghai University, Shanghai 200444, P.R. China

baozhang@tju.edu.cn, yqfeng@tju.edu.cn

Supporting Information

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I. General experimental methods:

Pyrrole and TFA are distilled before use, the other starting materials are purchased and used as received; THF, n-C₆H₁₄, PhCH₃ and petroleum ether are used without further purification, but CH₂Cl₂ is distilled and o-DCB is filtered through basic Al₂O₃ before use.

Column chromatography (CC): silica gel (300-400 mesh). UV-Vis spectra: Shimadzu UV-1800 10 mm quartz cell; Fluorescence emission spectra: a Varian Cary Eclipse Spectrometer, emission wavelengths λ in nm; Fluorescence lifetime: HORIBA JY Fluorolog, single electron photoelectric counter (argon saturated CH₂Cl₂, excitation at 366 nm, detection at 667 nm); Nuclear magnetic resonance (¹H-) spectra: Bruker AV400 and Varian Unityplus 500 chemical shifts (δ) in ppm, with δ (CHCl₃) = 7.26 ppm and δ (CD₂HSOCD₃) = 2.50 ppm for ¹H NMR and δ (CDCl₃) = 77.16 ppm for ¹³C NMR; Mass spectra: HR-ESIMS Bruker microTof-QII with positive ion mode and Maldi-Tof Bruker Autoflex Tof/Tof III.

Number system in terms of 1H- and 13C-NMR assignment:



Scheme S1 Numbering of corrole derivatives in this study.

II. Experimental Section:

Synthesis of dipyrromethane (1a-c)

Compound **1a-c** was prepared according to the reported methods [ARKIVOC, 2007, (x): 307-324].

Synthesis of Ald 1

To the solution of anthracene-9-carbaldehyde (100 mg, 0.485 mmol) in THF (25 mL) was added NaBH₄ (92.2 mg, 2.426 mmol, 5 eq.). The reaction mixture was refluxed until TLC revealed the total consuming of the starting materials. H₂O (50 mL) was added to quench the reaction. The product was extracted by $CH_2Cl_2(3 \times 20 \text{ mL})$. The organic layer was filter through hot cotton and concentrated to dryness under reduced pressure. The product anthracen-9-ylmethanol was obtained by re-crystallization in $CH_2Cl_2 / n-C_6H_{14}$.

The anthracen-9-ylmethanol (100 mg, 0.481 mmol), 4-formylbenzoic acid (300 mg, 2 mmol, 4.2 eq) and DCC (618 mg, 3 mmol, 6.2 eq) were dissolved in CH_2Cl_2 (50 mL). The reaction mixture was stirred at

room temperature for 48 h. The formed DCU was removed by filtration. The filtrate was washed with H₂O (3 × 30 mL) and the organic phase was filtered through dried cotton and evaporated to dryness under reduced pressure. The residue was separated by silica gel column (CH₂Cl₂/ petroleum ether = 2 / 1, v / v). The second fraction was collected. The product was obtained as yellow needle shape by 132.8 mg (82 %) after evaporation of the solvent. UV-Vis (in CH₂Cl₂): λ_{max} (nm) 257.0 (1.00), 333.5 (0.02), 349.5 (0.03), 368 (0.05), 388.5 (0.046); ¹H NMR (400 MHz, CDCl₃, 25 °C): δ 10.04 (s, 1H, CHO), 8.56 (s, 1H, An10H), 8.43 (d, *J* = 8.9 Hz, 2H, An1H & An8H), 8.15 (d, *J* = 8.2 Hz, 2H, *m*-PhH), 8.07 (d, *J* = 8.4 Hz, 2H, An4H & An5H), 7.87 (d, *J* = 8.2 Hz, 2H, *o*-PhH), 7.61 (t, *J* = 7.0 Hz, 2H, An2H & An7H), 7.52 (t, *J* = 7.1 Hz, 2H, An3H & An6H), 6.44 (s, 2H, OCH₂); ESI-MS (C₂₃H₁₆O₃: exact mass = 340.1099): calcd. for [M + H]⁺: 341.1172, found: 341.3.

Synthesis of reference compound AnC₆₀

The solution of Ald1 (5.1 mg, 15 μ mol) and C₆₀ (21.6 mg, 30 μ mol) in acid-free o-DCB (2.5 mL) was left under darkness in the presence of Ar for 3 days. Then the reaction mixture was filtered through sizeexclusion column to remove the unreacted fullerene using toluene as the eluent. The collected green fraction was concentrated to dryness under reduced pressure and the green residue was separation by silica gel column with CH₂Cl₂ as eluent. The brawn fraction was collected and the solvent was removed under reduced pressure and the residue precipitated by CH_2Cl_2 / n -hexane. Finally, the product AnC₆₀ was obtained as dark brawn powder by 9.5 mg (60 %). UV/Vis spectra (in CH₂Cl₂): λ_{max} (nm, rel. intensity) 257.5 (1.00), 314.0 (0.25); ¹H NMR (500 MHz, CDCl₃ + CS₂, 0 °C) δ 10.10 (s, 1H, CHO), 8.21 (br. s, 2H, C103H & C105H), 7.94 (br. s, 2H, C102H & C106H), 7.83 (d, J = 7.1 Hz, 2H, An4H & An5H), 7.75 (br. s, 2H, An1H & An8H), 7.54 (t, J = 6.7 Hz, 2H, An3H & An6H), 7.49 (td, J = 6.7 Hz, 2H, An2H & An7H), 6.35 (br.s, 2H, OCH₂), 5.89 (s, 1H, An10H). ¹³C NMR (125MHz / 500MHz,, CDCl₃ + CS₂, 0 °C, based on HSQC and HMBC): 190.7 (CHO), 155.4 (saturated C of fullerene), 140.7 (An12C & An13C), 139.4 (An11C & An14C),130.2 (C103 & C105), 129.3 (C102 & C106), 127.2 (An2C, An3C, An6C & An7C), 125.8 (An4C & An5C), 124.0 (An1C & An8C), 74.2 (unsaturated C of fullerene), 65.3 (OCH₂), 58.5 (An10C). ¹³C NMR (125 MHz / 500 MHz, CDCl₃ + CS₂, 0 °C, based on ¹³C NMR): 191.1, 155.6, 147.5, 147.4, 146.4, 146.3, 146.2, 146.1, 145.4, 145.3, 144.6, 144.4, 143.0, 142.9, 142.6, 142.1, 141.9, 141.8, 141.6, 140.1, 139.4, 134.2, 130.6, 129.7, 127.6, 127.5, 126.1, 77.4, 65.6, 58.8; Maldi-Tof ($C_{83}H_{16}O_3$: exact mass = 1060.0133): calcd. for [M]⁺ 1060.0133, found: 1061.3.

Synthesis of corrole (2a-c)

Ald 1 (6.8 mg, 20.0 μ mol) and the corresponding dipyrromethane (40.3 μ mol) was dissolved in the solution of TFA (28 mM) in CH₂Cl₂ (0.3 mL) and the reaction mixture was stirred under darkness in the presence of Ar for 10 h. Then the mixture was diluted to 10 mL with CH₂Cl₂ and oxidized by addition of the solution of DDQ (11.8 mg, 52.7 μ mol) in toluene (0.2 mL). After 2 h, the reaction mixture was diluted with 20 mL CH₂Cl₂ and washed with NaHCO₃ aq. (3 × 30 mL) and H₂O (2 × 20 mL). The organic phase was filtered through the dried cotton and evaporated to dryness under reduced pressure. The residue was

separated by silica gel column (CH₂Cl₂/ petroleum ether = 2 / 1, v / v for **2a-b** and CH₂Cl₂ for **2c**). The main green fraction was collected. The corrole was obtained after evaporation of the solvent and crystallization in CH₂Cl₂/n-C₆H₁₄ = 1 / 3 (v / v).

Corrole **2a** (green): 1.3 mg (8 %); UV-Vis (in CH₂Cl₂): λ_{max} (nm) 250 (0.74), 258.0 (1.36), 350.0 (0.14), 368.5 (0.20), 388.5 (0.31), 420.5 (1.00), 519.0sh (0.06), 576.5 (0.15), 620.0 (0.12), 653.5 (0.10); ¹H NMR (400 MHz, CDCl₃, 25 °C): δ 8.93 (d, J = 4.0 Hz, 2H, β -H), 8.84 (d, J = 4.0 Hz, 2H, β -H), 8.61 (d, J = 8.0 Hz, 3H, An1H, An8H & superimposed by An10H), 8.57 (d, J = 4.0 Hz, 2H, β -H), 8.42 (d, J = 4.0 Hz, 2H, β -H), 8.38 (d, J = 7.9 Hz, 2H, C103H & C105H), 8.22 (d, J = 7.6 Hz, 3H, C56H, C152H & C156H), 8.17 (d, J = 7.9 Hz, 2H, C102H & C106H), 8.10 (d, J = 8.3 Hz, 2H, An4H & An5H), 7.68 (t, J = 8.0 Hz, 2H, An2H & An7H), 7.61 (d, J = 7.6 Hz, 4H, C52H, C55H, C153H & C155H), 7.56 (t, J = 8.0 Hz, 3H, An3H, An6H & superimposed by C53H), 6.58 (s, 2H, OCH₂), 2.66 (s, 6H, CH₃); ¹H NMR (500 MHz, DMSO-d⁶): δ 9.00 (br.s, 2H, β -H), 8.35 (br.s, 4H, 2 β -H, C102H & C103H), 8.28 (d, J = 7.1 Hz, 1H, C105H), 8.20 (dd, J = 2.3, 8.4 Hz, 3H, An4H, An5H & superimposed by C106H), 8.17 (d, J = 6.6 Hz, 3H, C56H, C152H & C156H), 7.96 (d, J = 6.4 Hz, 1H, C52H), 7.73 (t, J = 7.9 Hz, 2H, An2H & An7H), 7.65 (br.s, 3H, C55H, C153H & C155H), 7.62 (td, J = 2.0, 7.9 Hz, 2H, An3H & An6H), 7.57 (d, J = 6.5 Hz, 1H, C53H), 6.58 (s, 2H, OCH₂), 2.61 (s, 6H, CH₃), -2.65 (br.s, NH); HRMS (ESI) (C₅₅H₄₀N₄O₂: exact mass = 788.3151): calcd. for [M+H]⁺ 789.3224, found: 789.3224.

Corrole **2b** (green): 1.6 mg (10 %); UV/Vis (in CH₂Cl₂) λ_{max} (nm) 249.5 (0.64), 258.0 (1.25), 350.5 (0.11), 369 (0.17), 389 (0.28), 420.5 (1.00), 522sh (0.04), 577.5 (0.16), 619.0 (0.11), 651.5 (0.08); ¹H NMR (500 MHz, CDCl₃, 25 °C): δ 8.94 (br.s, 2H, β -H), 8.83 (br.s, 1H, β -H), 8.79 (br.s, 1H, β -H), 8.59 (m, 3H, An1H, An8H, & An10H), 8.53 (m, 3H, β -H), 8.44 (d, J = 7.0 Hz, 2H, C103H & superimposed by β -H), 8.38 (d, J = 6.7 Hz, 1H, C105H), 8.35 (d, J = 7.0 Hz, 1H, C102H), 8.25 (br.s, 3H, C56H, C152H & C156H), 8.15 (d, J = 6.7 Hz, 1H, C106H), 8.10 (m, 2H, An4H & An5H), 8.06 (d, J = 6.3 Hz, 1H, C52H), 7.79 (br.s, 3H, C55H, C153H & C155H), 7,72 (d, J = 6.3 Hz, 1H, C53H), 7.68 (t, J = 7.9 Hz, 2H, An2H & An7H), 7.56 (m, 2H, An3H & An6H), 6.57 (s, 2H, CH₂), -2.30 (NH); HRMS (ESI) (C₅₃H₃₄Cl₂N₄O₂: exact mass = 828.2059): calcd. for [M+H]⁺ 829.2132, found: 829.2128.

Corrole **2c** (green): 1.3 mg (7.4 %); UV-Vis (in CH₂Cl₂) λ_{max} (nm) 250.0 (0.77), 257.5 (1.35), 350.0 (0.15), 368.5 (0.19), 388.5 (0.26), 426.5 (1.00), 522.5sh (0.05), 582.5 (0.18), 621.0 (0.12), 653.5 (0.08); ¹H NMR (500 MHz, CDCl₃, 25 °C): 8.90 (s, 2H, β -H), 8.82 - 8.78 (m, 2H), 8.58 - 8.55 (m, 3H, An1H, An8H & An10H), 8.55 - 8.49 (m, 2H), 8.49 - 8.43 (m, 6H), 8.43 - 8.35 (m, 5H), 8.32 (d, J = 7.0 Hz, 1H), 8.21 (d, J = 7.0 Hz, 1H), 8.14 (d, J = 7.0 Hz, 1H), 8.07 (t, J = 8.0 Hz, 2H, An4H & An5H), 7.66 (m, 2H, An2H & An7H), 7.53 (m, 2H, An3H & an6H), 6.55 (br.s, 2H, CH₂), 4.09 (br.s, 6H, COOCH₃); HRMS (ESI) (C₅₅H₄₀N₄O₆: exact mass = 876.2948): calcd. for [M+H]⁺ 877.3021, found: 877.3020.

Synthesis of corrole-fullerene dyad 3a-c

The solution of 2a-c (2 µmol) and C₆₀ (10.8 mg, 15 µmol) in acid-free *o*-DCB (1.5 mL) was left under

darkness in the presence of N₂ for 5 days. Then the reaction mixture was filtered through size-exclusion column to remove the unreacted fullerene using toluene as the eluent. The collected green fraction was concentrated to dryness under reduced pressure and the green residue was separation by silica gel column with toluene as eluent. The first fraction was collected and the solvent was removed under reduced pressure to obtain the corrole-fullerene dyad **3a** (1.4 mg, 46 %). UV/Vis spectra (in CH₂Cl₂): λ_{max} (nm) 258.5 (1.01), 314.0 (0.41), 421 (1.00), 522.5sh (0.06), 577.0 (0.15), 618.0 (0.12), 652.5 (0.10), 705; ¹H NMR (500 MHz, CDCl₃ / CS₂, 0 °C) δ 8.92 (br.s, 2H, β-H), 8.86 (br.s, 2H, β-H), 8.57 (br.s, 3H, 2β-H & C103H), 8.50 (br. s, 2H, & H & C105H), 8.42 (br.s, 2H, & H & C102H), 8.22 (br.s, 4H, C52H, C56H, C152H & C156H), 8.03 (d, J = 7.0 Hz, 1H, C106H), 7.95 (br.s, 2H, An1H & An8H), 7.85 (d, J = 4.7 Hz, 2H, An4H & An5H), 7.63 (br.s, 4H, C53H, C55H, C153H & C155H), 7.56 (m, 4H, An2H, An3H, An6H & An7H), 6.45 (br.s, 2H, CH₂), 5.89 (s, 1H, An10H), 2.69 (s, 6H, CH₃); ¹³C NMR (125 / 500 MHz, CDCl₃, 0 °C, based on HSQC and HMBC): 141.3 (An12C&An13C), 139.6 (An11&An14), 137.1 (C54 &C154), 136.4 (C51 & C151), 134.9 (C52, C56, C152 & C156), 134.7 (C102 & C106), 128.7 (C53, C55, C153 & C155), 127.6 (An2C, An3C, An6C & An7C), 126.8 (&C), 126.0 (An4C & An5C), 125.8 (&C), 125.2 (&C), 124.9 (An1C & An8C), 116.3 (β-C), 74.6 (unsaturated C of fullerene), 65.2 (OCH₂), 58.8 (An10C), 21.5 (CH₃); Maldi-Tof $(C_{115}H_{40}N_4O_2: exact mass = 1508.3151): calcd. for [M+H]^+ 1509.3224, found: 1509.3.$

3b (1.3 mg, 42 %). UV/Vis spectra (in CH₂Cl₂): λ_{max} (nm) 258 (1.00), 314 (0.39), 420.5 (1.00), 521sh (0.06), 579.5 (0.14), 618.5 (0.10), 651.5 (0.07), 705; ¹H NMR (500 MHz, CDCl₃ / CS₂, 5 °C) δ 8.95 (s, 2H), 8.84 (s, 2H), 8.60 ~ 8.35 (overlap, 6H), 8.25 (br. s, 4H), 8.11 (m, 2H), 7.94 (br.s, 2H), 7.85 ~ 7.75 (m, 6H), 7.58 (m, 4H), 6.46 (br.s, 2H, OCH₂), 5.89 (s, 1H, An10H), -3.43 (br. s); Maldi-Tof (C₁₁₃H₃₄Cl₂N₄O₂: exact mass = 1548.2059): calcd. for [M]⁺ 1548.2059, found: 1548.1.

3c (0.96 mg, 30 %). UV/Vis spectra (in CH₂Cl₂): λ_{max} (nm) 258 (1.01), 314.5 (0.44), 426 (1.00), 526sh (0.07), 582.5 (0.17), 622.0 (0.12), 655.5 (0.08), 707; ¹H NMR (500 MHz, CDCl₃, 5 °C) δ 9.03 (s, 2H), 8.89 (m, 2H), 8.61 (s, 2H), 8.56 (s, 2H), 8.49 ~ 8.45 (br. s, 10H), 8.26 (s, 2H), 7.98 (br.s, 2H), 7.87 (s, 2H), 7.58 (m, 4H), 6.49 (br.s, 2H, OCH₂), 5.93 (s, 1H, An10H), 4.10 (s, 6H, COOCH₃); Maldi-Tof (C₁₁₇H₄₀N₄O₆: exact mass = 1596.2948): calcd. for [M]⁺ 1596.2948, found: 1596.1.

S5

III. NMR spectra of Ald1, 2a-c and 3a-c:



Fig. S1 ¹H NMR spectra of Ald1 (400 MHz, CDCl₃, 25 °C)



Fig. S2 ^1H (up) and ^{13}C (down) NMR spectra of AnC_{60} (500 / 125 MHz, CDCl_3, 0 °C)



Fig. S3 ¹H NMR spectra of 2a (400 MHz, CDCl₃, 25 °C)



Fig. S4 ¹H NMR spectra of 2a (500 MHz, DMSO-d⁶, 25 °C)



Fig. S5 ¹H NMR spectra of 2b (500 MHz, CDCl₃, 25 °C)





Fig. S6 ¹H NMR spectra of 2c (500 MHz, CDCl₃, 25 °C)



Fig. S7 $^1\!\mathrm{H}$ NMR spectra of 3a (500 MHz, CDCl_3/ CS2, 0 °C)



Fig. S8 ^1H NMR spectra of 3b (500 MHz, CDCl_3/ CS_2, 5 °C)



Fig. S9 $^1\mathrm{H}$ NMR spectra of 3c (500 MHz, CDCl_3, 5 $^{\circ}\mathrm{C}$)

IV. Maldi-Tof spectrum of 2b



Fig. S10 Maldi-Tof spectrum of 2b

V. Optimal structure of 3a



Fig. S11 The optimal structure of 3a obtained from TD-DFT calculation (one front view and two side views)