Supplementary information

Supplementary 1

Determination of weight content of HMS-301 in the microcapsule by ¹H NMR

According to the information provided by the manufacturer, the HMS-301 has a molecular weight of 2,000g/mol and an average chain length of 28 Si atoms. Eight of these are CH₃HSiO groups. Meanwhile, the PMMA used in the experiment has a molecular weight of 15,000g/mol, consisted of 150 repeating units of – $(CH_2C(CH_3)(COOCH_3)-$. Hence, each PMMA molecule contains 450 protons from the methyl ester group. The mol ratio between HMS-301 and PMMA can be obtained by calculating the ratio between 450 times the area of hydride group of the signals at δ =4.7ppm and eight times the area of methyl groups of the signals at δ =3.6ppm by means of the following equation³⁰:

$$\frac{n_{HMS-301}}{n_{PMMA}} = \frac{450 \cdot A_1}{8 \cdot A_2}$$

where A_1 is area of the signal due to eight protons of the hydride groups in each HMS-301 molecule, A_2 is area of the signal due to 450 protons of the methyl groups in each PMMA molecule. Thus, the weight fraction of HMS-301 in the microcapsule can be obtained by the following equation:

$$\%HMS - 301(wt) = \frac{\frac{450 \cdot A_{1}}{8 \cdot A_{2}} \cdot M_{w_{HMS-301}}}{\frac{450 \cdot A_{1}}{8 \cdot A_{2}} \cdot M_{w_{HMS-301}} + M_{w_{PMMA}}}$$

where Mw $_{\rm HMS-301}$ and Mw $_{\rm PMMA}$ are the molecular weight of HMS-301 and PMMA, respectively

Supplementary 2

Rheological behaviours PMMA/HMS-301 microcapsule+V35 (Figure S2a), HMS-301+empty PMMA capsule+V35 (Figure S2b) and HMS-301+V35 (Figure S2c) from time sweep rheological measurement. All the measurements were performed with a strain of 2% and a frequency of 1Hz.



Figure S2a. Evolution of storage modulus of mixture containing PMMA/HMS-301 microcapsule+V35 at 120 $^{\circ C}$ in a period of 5h.



Figure S2b. Evolution of storage modulus of mixture containing HMS-301+empty PMMA capsule+V35 at 120 $^{\circ C}$ in a period of 2h.



Figure S2c. Evolution of storage modulus of mixture containing HMS-301+V35 at 120°C in a period of 2h.