Supporting Information

Extraction mechanism and γ-radiation effect on the removal of Eu³⁺ by a novel BTPhen/[C_nmim][NTf₂] system in the presence of nitric acid

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Page No.	Contents
S – 2	Fig. S1 Influence of the oscillation time on E_{Eu} in BTPhen/[C ₂ mim][NTf ₂] system.
S – 3	Table S1. D_{Eu} in BTPhen/[C_n mim][NTf ₂] systems depending on the initial nitric acid
	concentration.
S-4	Table S2. D_{Eu} in BTPhen/1-octanol system depending on the initial nitric acid
	concentration.
S – 5	Fig. S2 ¹ H NMR spectra of BTPhen before and after irradiation.
S – 6	Fig. S3 Micro-FTIR spectra of BTPhen before and after irradiation
S – 7	Fig. S4 Influence of γ -radiation on the D_{Eu} in BTPhen/1-octanol system.
S – 8	Fig. S5 19 F NMR spectra of [C ₂ mim][NTf ₂] before and after irradiation.
S – 9	Table S3. Core level binding energy for sediment, $Eu_2(SO_3)_3$ and EuF_3



Fig. S1 Influence of the oscillation time on E_{Eu} in BTPhen/[C₂mim][NTf₂] system. Initial nitric acid concentration: 0.1 M.

$[\mathbf{HNO}] (\mathbf{mol} \mathbf{J}^{-1})$	$D_{ m Eu}$			
	[C ₂ mim][NTf ₂]	[C ₄ mim][NTf ₂]	[C ₈ mim][NTf ₂]	
0.01	52	21	0.55	
0.05	4.7	1.0	0.32	
0.1	1.9	0.34	0.23	
0.5	0.80	0.07	0.02	
1	0.55	0.06	0.04	
2	0.33	0.07	0.05	
3	0.40	0.12	0.06	

Table S1. D_{Eu} in BTPhen/[C_nmim][NTf₂] systems depending on the initial nitric acid concentration.

[HNO ₃] (mol L ⁻¹)	$D_{ m Eu}$	
0.1	0.56	
1	1.66	
3	2.56	

Table S2. D_{Eu} in BTPhen/1-octanol system depending on the initial nitric acid concentration.



Fig. S2 ¹H NMR spectra of BTPhen before and after irradiation.



Fig. S3 Micro-FTIR spectra of BTPhen before and after irradiation.



Fig. S4. Influence of γ -radiation on the D_{Eu} in BTPhen/1-octanol system. Initial nitric acid concentration: 1 M.



Fig. S5 19 F NMR spectra of [C₂mim][NTf₂] before and after irradiation.

Correland	Binding energy (eV)			
Core level	Sediment	Eu ₂ (SO ₃) ₃	EuF ₃	
O 1s	531.69	531.24	-	
F 1s	684.67	-	684.19	
S 2p	168.85	167.33	-	
Eu 4d	136.90	136.62	136.57	

Table S4. Core level binding energy for sediment, $Eu_2(SO_3)_3$ and EuF_3