Electronic Supplementary Information (ESI) for

Controlling microstructure of MFI zeolites with Mg(OH)₂ nanocrystals to

improve their catalytic performances

Huijuan Wei,^a Ning Zhang,^a Tian Zhao,^a Yangqing Liu,^a Yiqiang Wen,^a Xiangyu Wang^{*a} and Baojun Li^{*ab}

- ^a Institute of Industrial Catalysis and College of Chemistry and Molecular Engineering, Zhengzhou University, 100 Science Road, Zhengzhou 450001, P R China. E-mail: xiangyuwang@zzu.edu.cn and lbjfcl@zzu.edu.cn
- ^b Department of Chemistry, Tsinghua University, Beijing 100084, P R China. E-mail: lbjfcl@tsinghua.edu.cn



Fig. S1 TEM images of Mg(OH)₂ nanocrystals.



Fig. S2 TEM images of TS-1-0.50Mg samples before and after removing Mg(OH)₂ (a) TS-1-0.50Mg before removing Mg(OH)₂ (b) TS-1-0.50Mg.



Fig. S3 XRD pattern of Mg(OH)₂ nanocrystals.

Sample	$n_{\rm Si}/n_{\rm Ti}$ (atomic)	$n_{\rm Si}/n_{\rm Mg}$ (atomic)
TS-1	51.8	ŝ
TS-1-0.05Mg	56.4	∞
TS-1-0.10Mg	70.5	∞
TS-1-0.30Mg	84.6	28.4
TS-1-0.50Mg	73.9	32.8
S-1	∞	∞
S-1-0.05Mg	∞	∞
S-1-0.10Mg	∞	113.0
S-1-0.15Mg	∞	33.7
S-1-0.25Mg	œ	4.27

Table S1. Chemical compositions of TS-1-xMg $\,$ and S-1-xMg.