## Synthesis, structures, surface photovoltage and luminescent properties

## of two new nickel(II) carboxyphosphonates with a 3D framework

## structure

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Supplementary Materials



Fig. S1 The simulated XRD pattern of compound 1 (down) and experimental powder XRD pattern of compound 1 (up)



Fig. S2 The simulated XRD pattern of compound 2 (down) and experimental powder XRD pattern of compound 2 (up)



Fig. S3 The IR spectrum of compound 1.





Fig. S4 The IR spectrum of compound 2.

Fig. S5 The TG curve of compound 1.



Fig. S6 The TG curve of compound 2.

Fig. S7 The X-ray powder diffraction pattern of the final product in the thermal decomposition for compound 1. The final product is Ni<sub>2</sub>P<sub>2</sub>O<sub>7</sub> (JCPDS 01–074–1604).



Fig. S8 The X-ray powder final product in for compound 2. Ni<sub>2</sub>P<sub>2</sub>O<sub>7</sub> (JCPDS 01–074–1604).

diffraction pattern of the the thermal decomposition The final product is



Fig. S9 Solid–state emission spectrum of 4,4'-bipy at room temperature.



Fig. S10 Solid–state emission spectrum of  $H_5L$  at room temperature.

 Table
 S1
 Selected
 bond
 angles
 (°)
 for compounds 1 and 2

Compound **1** O(4)–Ni(1)–O(1) O(4)–Ni(1)–O(3)#1

91.18(13) 178.45(13)

O(9)-Ni(2)-O(10)#3 O(2)-Ni(2)-O(10)

90.67(14) 87.69(12)

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O(8)#2–Ni(2)–N(3)	95.9	P(2) = O(4) = Ni(1)	114.5(3)
N(3)#2-Ni(2)-N(3)	179.75(10)	1(2) O(4) IU(1)	

<sup>a</sup> Symmetry transformations used to generate equivalent atoms: #1 - x + 1/2, -y + 3/2, -z; #2 - x + 1/2, y - 1/2, -z + 1/2; #3 - x + 1/2, -y + 1/2, -z; #4 - x + 1/2, y + 1/2, -z + 1/2 for 1; #1 - x + 1/2, y + 1/2, -z + 1/2; #2 - x, y, -z + 1/2; #3 - x + 1/2, y - 1/2, -z + 1/2 for 2.