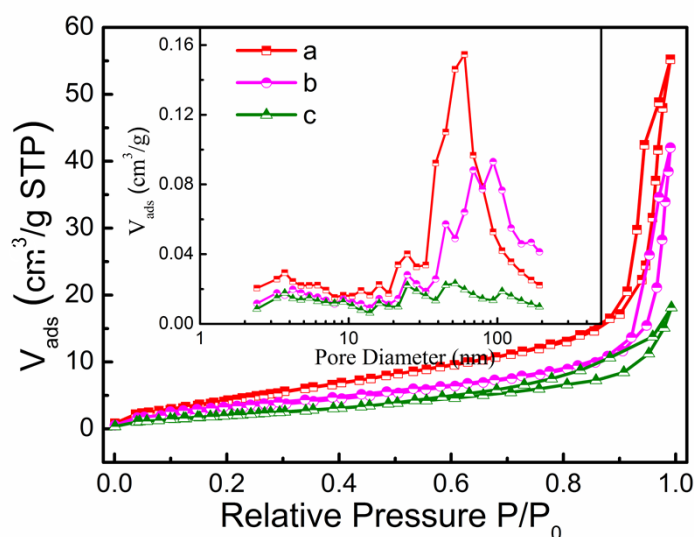


# Enhanced visible light photocatalytic degradation of Rhodamine B by Bi/Bi<sub>2</sub>MoO<sub>6</sub> hollow microsphere composites

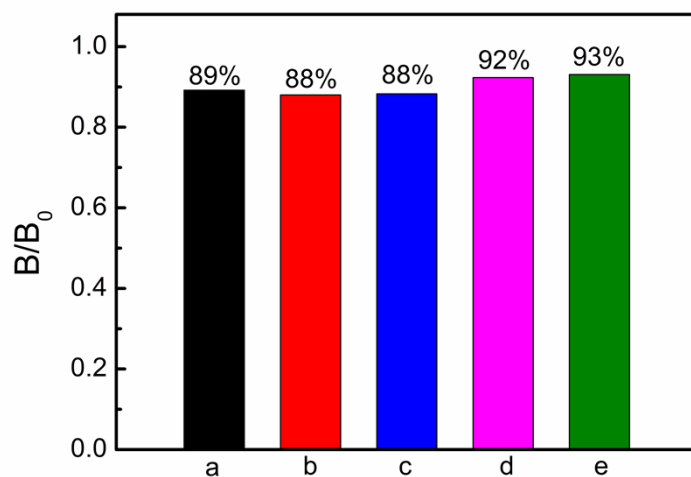
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**Fig. S1** Nitrogen adsorption-desorption isotherms and corresponding pore size distribution curves (inset) of (a)BMO-1, (b)BMO-3 and (c)BMO-4.



**Fig. S2** Bar plot showing the remaining RhB in the solution after reaching the adsorption equilibrium in the dark by using (a)  $\text{Bi}_2\text{MoO}_6$ , (b) BMO-1, (c) BMO-2, (d) BMO-3 and (e) BMO-4 in 30 min.

Fig. S2 shows the result of RhB adsorption experiments. The normalized temporal concentration changes ( $B/B_0$ ) of RhB during the adsorption process are proportional to the normalized maximum absorbance ( $A/A_0$ ), which can be derived from the change in the RhB absorption profile during the adsorption process.