

Electronic Supplementary Information

Functionalized mesoporous silica material and anionic dye adsorption: MCM-41 incorporated with amine groups for competitive adsorption of Acid Fuchsine and Acid Orange II

Yunhai Wu^{ab*}, Meili Zhang^b, Huaiyang Zhao^b, Shengxin Yang^b, Aynigar Arkin^b

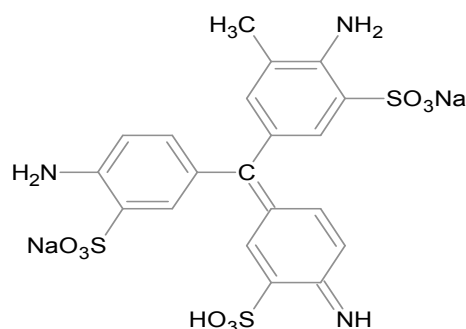
^a Key Laboratory of Integrated Regulation and Resources Development of Shallow

Lakes, Ministry of Education, Hohai University, 1st Xikang Road, Nanjing 210098

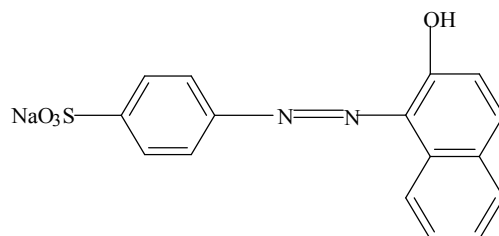
(China). Fax: +86-25-83786697; Tel: +86-25-83786697. E-mail: smilehhu@sina.com

^b College of Environment, Hohai University, 1st Xikang Road, Nanjing 210098

(China).



a: AF, C₂₀H₁₇N₃Na₂O₉S₃, molecular weight: 585.5



b: AO, C₁₆H₁₁N₂NaO₄S, molecular weight: 350.3

Fig. S1 Molecular structure of the selected basic anionic dyes for the current adsorption study. a:

AF, b: AO.

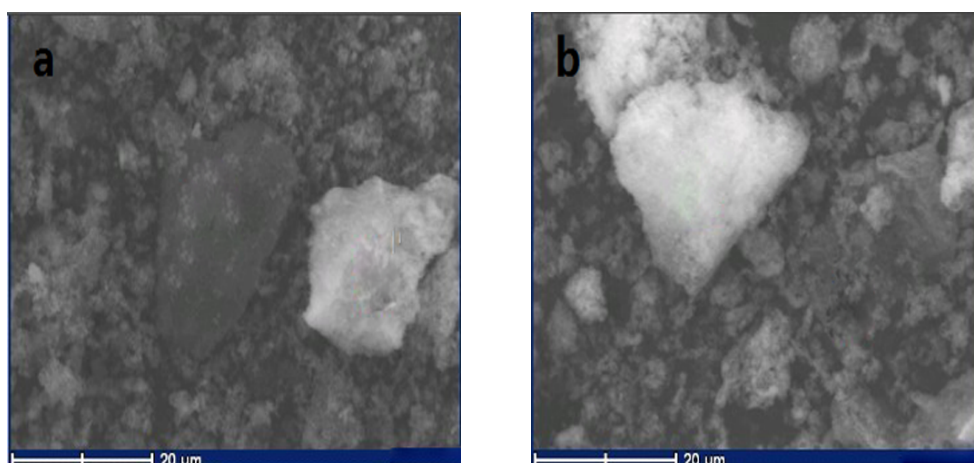


Fig. S2 SEM surface morphology. a: MCM-41, b: NH₂-MCM-41.

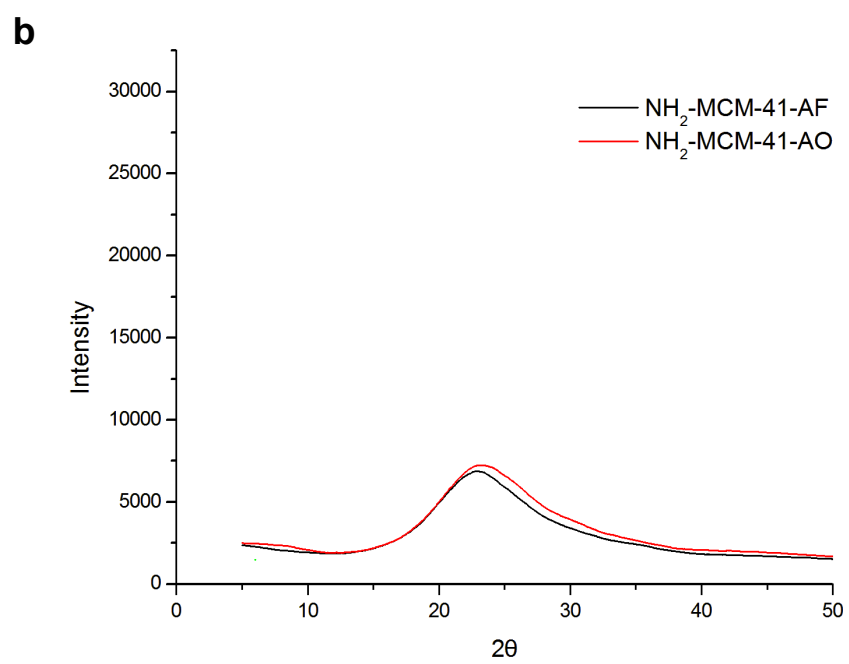
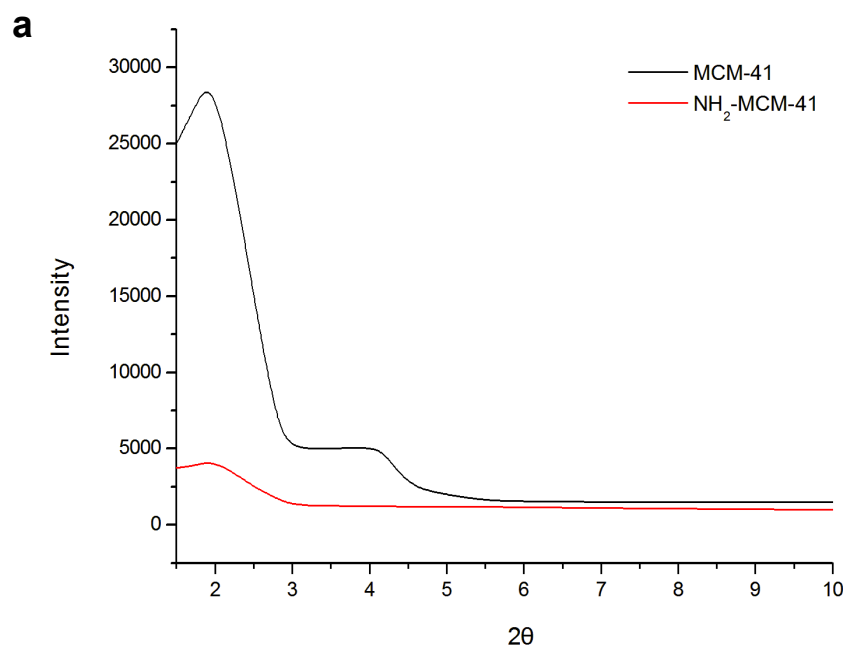


Fig. S3 (a) XRD pattern of MCM-41 and NH₂-MCM-41.

(b) XRD pattern of NH₂-MCM-41-AF and NH₂-MCM-41-AO.

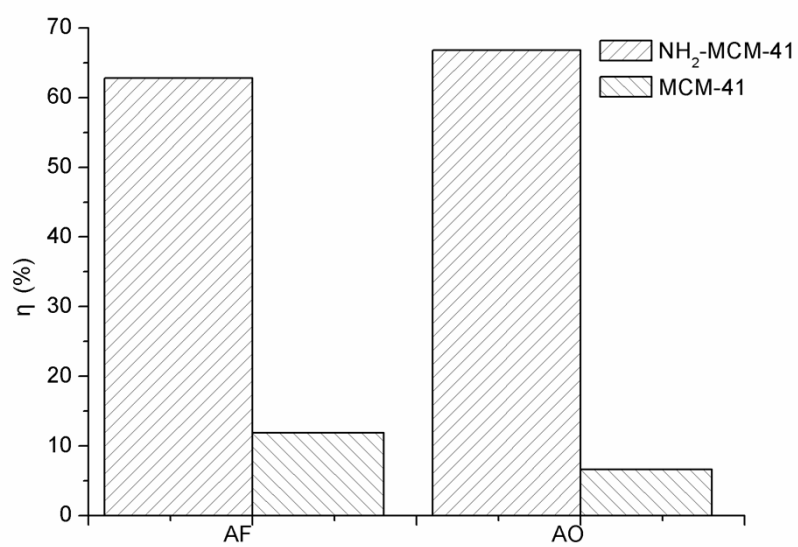


Fig. S4 The adsorptive removal ratios of AO and AF on MCM-41 before and after amine functionalization.

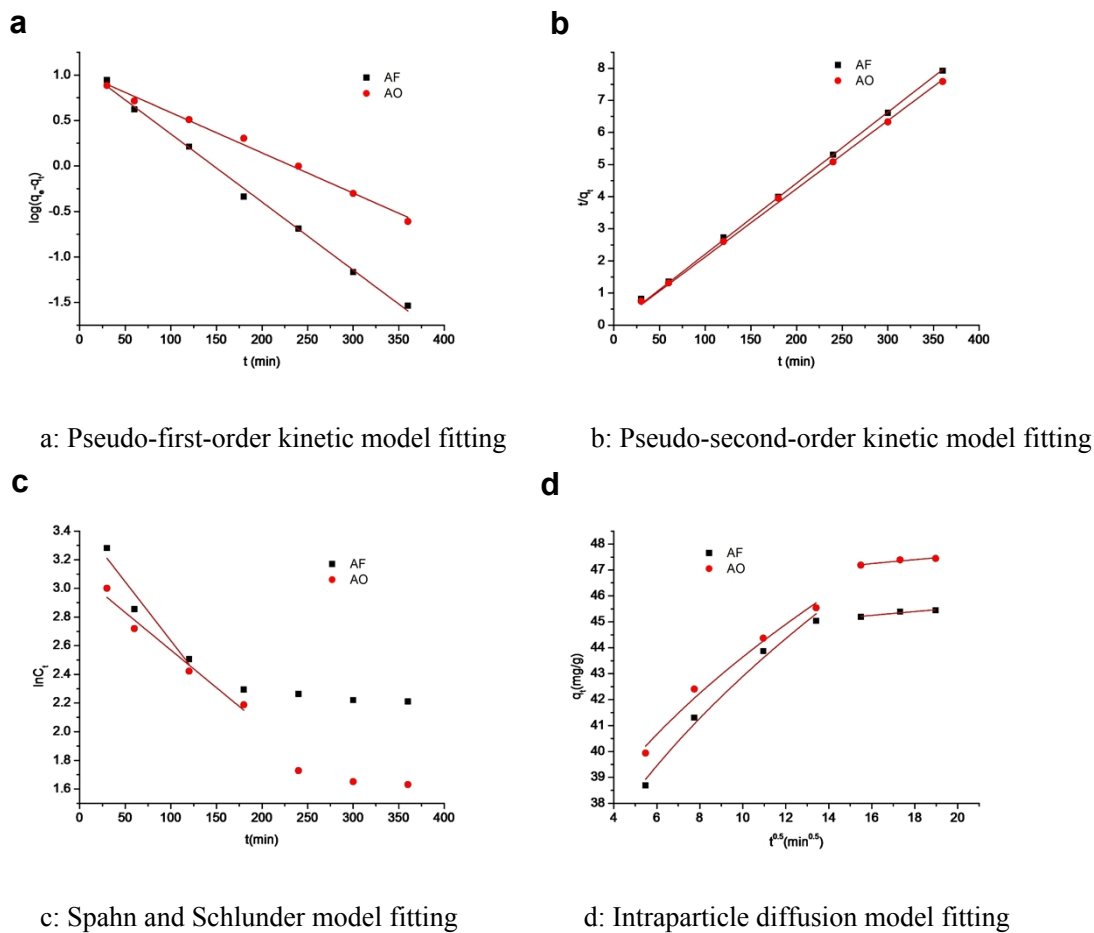


Fig. S5 Adsorption kinetics of AF and AO onto NH₂-MCM-41 in single component systems. a: Pseudo-first-order kinetic model fitting, b: Pseudo-second-order kinetic model fitting, c: Spahn and Schlunder model fitting, d: Intraparticle diffusion model fitting. (Conditions: pH 2.0 for AF and pH 3.0 for AO, 25 °C, NH₂-MCM-41 dosage 2.0 g/L.)

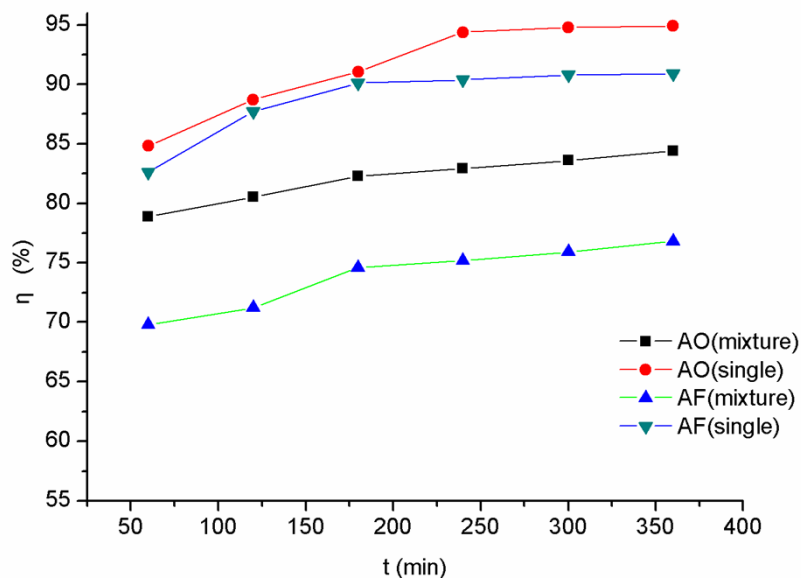


Fig. S6 Effect of contact time on adsorption of the two anionic dyes in binary component systems.

(Conditions: pH 2.0, 25 °C, NH₂-MCM-41 dosage 2.0 g/L, dye concentration 100mg/L.)

Table S1 Adsorption isotherm constants for adsorption of AO and AF onto NH₂-MCM-41 at various temperatures in binary component systems.

| Dyes | Temperature(°C) | Langmuir | | | Freundlich | | |
|------|-----------------|----------|--------|--------|------------|--------|--------|
| | | q_m | K_L | R^2 | K_F | n | R^2 |
| AF | 25 | 75.4620 | 0.0365 | 0.9986 | 5.6680 | 1.7916 | 0.9811 |
| AO | 25 | 75.8369 | 0.0786 | 0.9903 | 10.6100 | 2.1176 | 0.9245 |

Table S2 Kinetic parameters for adsorption of AF and AO onto NH₂-MCM-41 in binary component systems.

| Dyes | Pseudo-first-order kinetic model | | | Pseudo-second-order kinetic model | | | Spahn and Schlunder model | | Intraparticle diffusion model | | | |
|------|----------------------------------|--------|--------|-----------------------------------|---------|--------|---------------------------|--------|-------------------------------|--------|-----------|--------|
| | k_1 | q_e | R^2 | k_2 | q_e | R^2 | k_{ext} | R^2 | $k_{p,2}$ | R^2 | $k_{p,3}$ | R^2 |
| AF | 0.0056 | 1.6291 | 0.9678 | 0.0672 | 18.9268 | 0.9974 | 0.0014 | 0.8778 | 1.3036 | 0.7724 | 0.9682 | 0.9818 |
| AO | 0.0048 | 1.4580 | 0.9889 | 0.0307 | 20.8688 | 0.9988 | 0.0015 | 0.9964 | 0.9559 | 0.9660 | 0.8756 | 0.9858 |