

Electronic Supplementary Information _ESI

Multi-analyte, ratiometric and relay recognition of a 2,5-diphenyl-1,3,4-oxadiazole-based fluorescent sensor through modulating ESIPT

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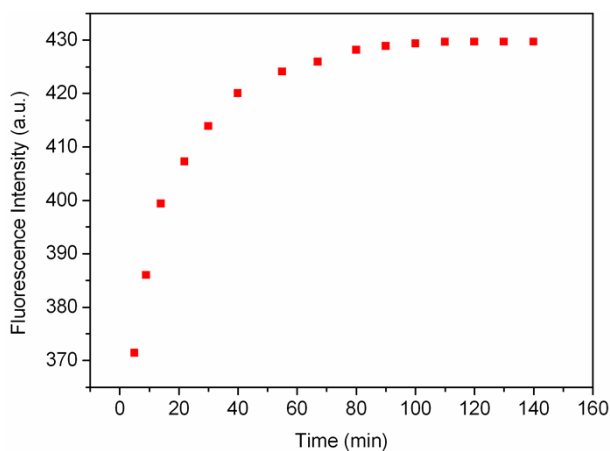


Fig. S1. Fluorescence intensity changes of **L** (10 μ M) in HEPES buffered (10 mM, pH = 7.4) water solution against time.

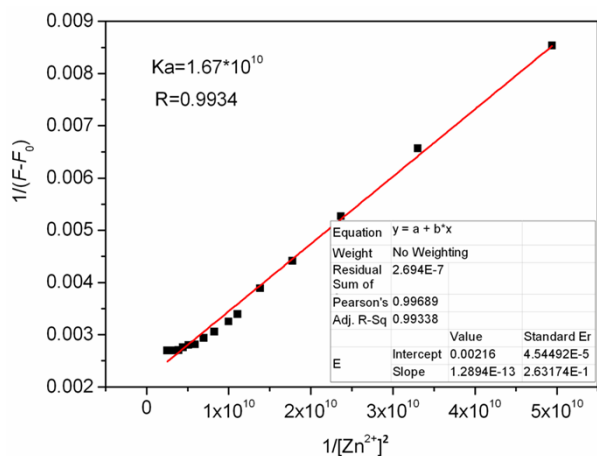


Fig. S2. Benesi-Hildebrand plot of **L** assuming 1:2 stoichiometry with Zn^{2+} . $\lambda_{\text{em}} = 445 \text{ nm}$.

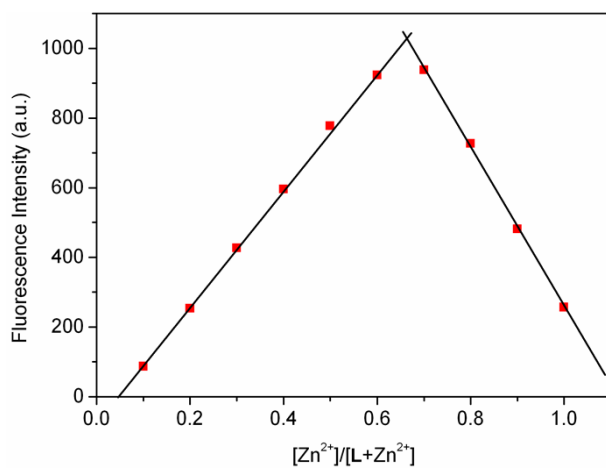


Fig. S3. Job's plot of **L** with Zn^{2+} (at 445 nm). The total concentration of **L** and Zn^{2+} is $5.0 \times 10^{-5} \text{ M}$.

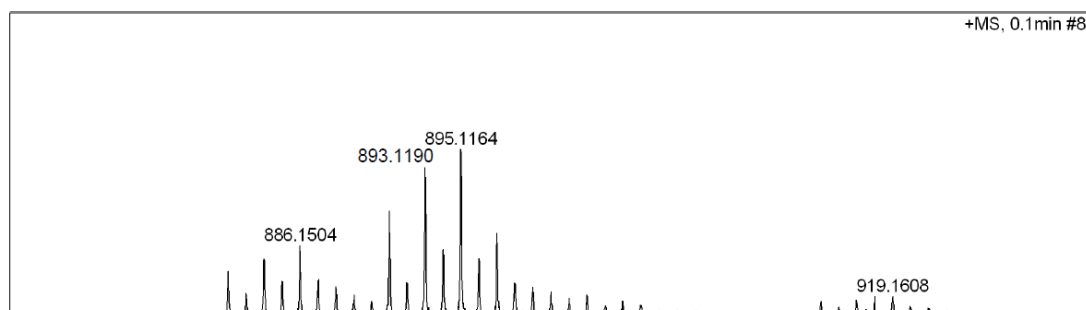


Fig. S4. HRMS spectrum of **L** in the presence of Zn^{2+} .

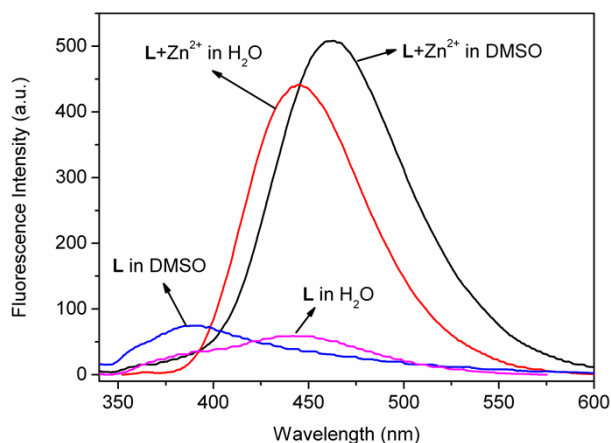


Fig. S5. Fluorescence spectra of **L** with and without Zn^{2+} in DMSO and water solution (HEPES 10 mM, pH = 7.4).

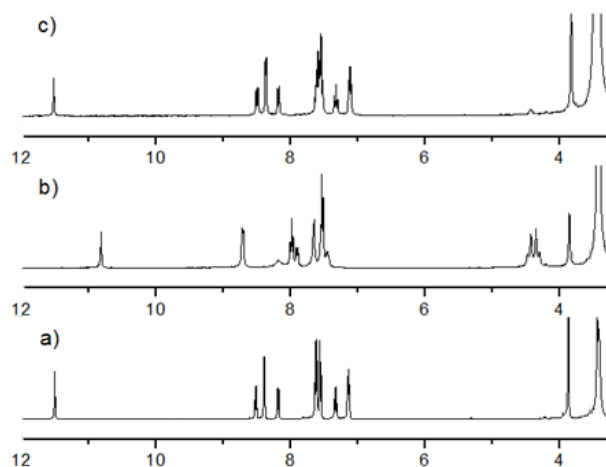


Fig. S6. ^1H NMR spectrum of **L** (a), **L**+ Zn^{2+} (b) and **L**+ Zn^{2+} + S^{2-} (c) in $\text{DMSO}-d_6$.

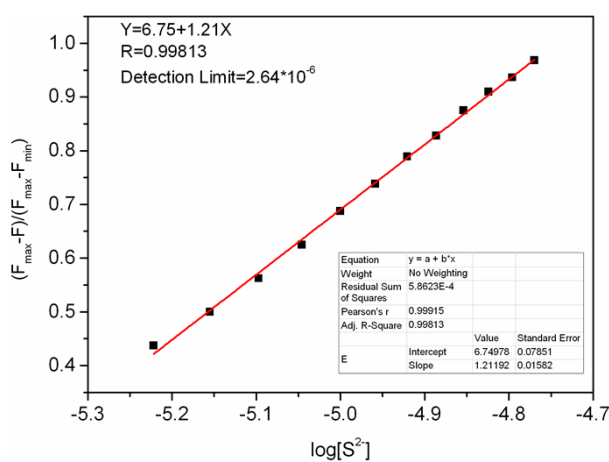


Fig. S7. Normalized fluorescence intensity $(F - F_{\min}) / (F_{\max} - F_{\min})$ of **L**- 2Zn^{2+} ($10 \mu\text{M}$) to $\text{Log}[\text{S}^{2-}]$ in water solution (HEPES 10 mM, pH = 7.4). $\lambda_{\text{em}} = 445 \text{ nm}$.

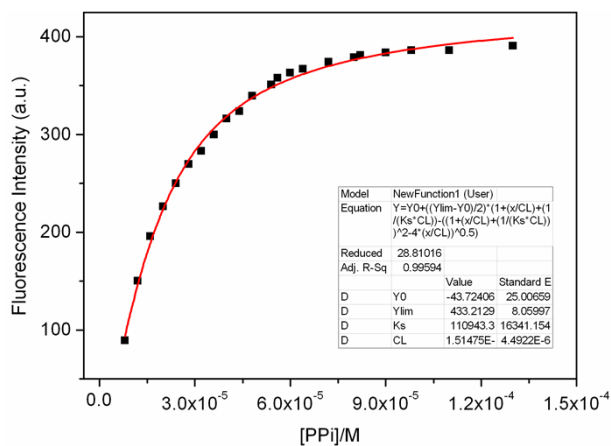


Fig. S8. Non-linear least squares fitting of fluorescence intensity of (at 396 nm) L-2Zn²⁺ solution with different concentrations of PPI employing a 1:1 binding equation.

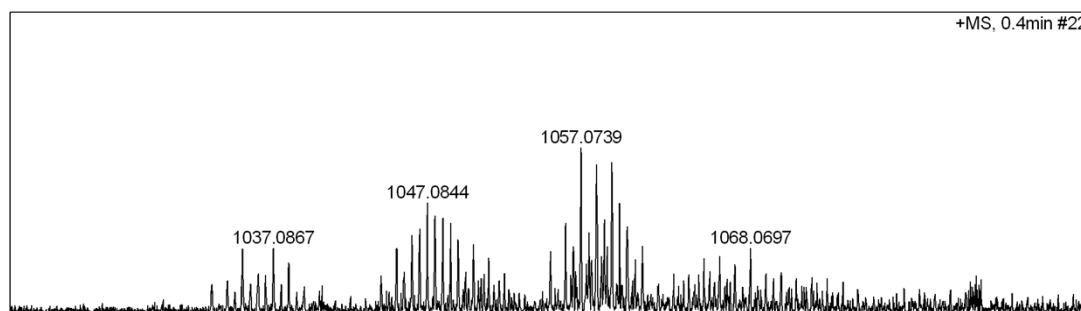


Fig. S9. HRMS spectrum of L-2Zn²⁺ in the presence of PPI.

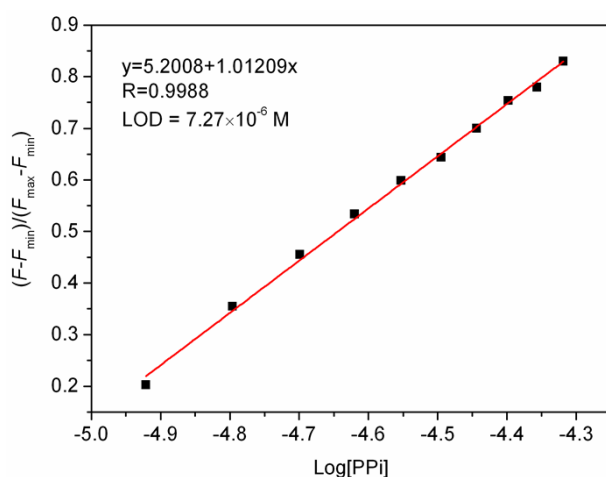


Fig. S10. Normalized fluorescence intensity $(F-F_{\min})/(F_{\max}-F_{\min})$ of L-2Zn²⁺ (10 μM) to Log[PPI] in water solution (HEPES 10 mM, pH = 7.4). $\lambda_{\text{em}} = 396$ nm.

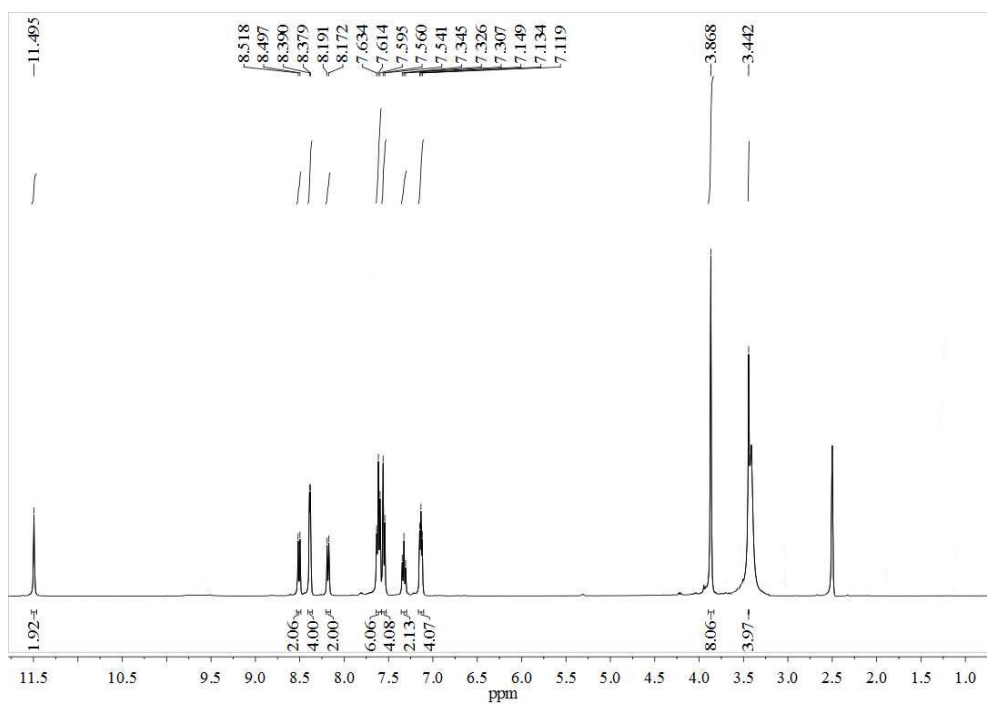


Fig. S11. ^1H NMR spectrum of sensor **L** in $\text{DMSO-}d_6$.

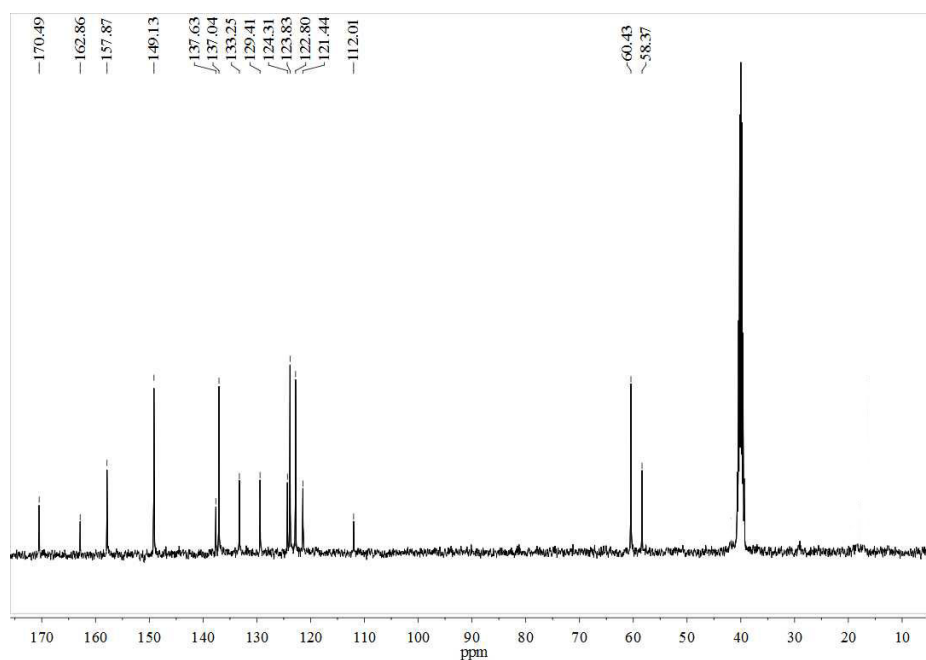


Fig. S12. ^{13}C NMR spectrum of sensor **L** in $\text{DMSO-}d_6$.

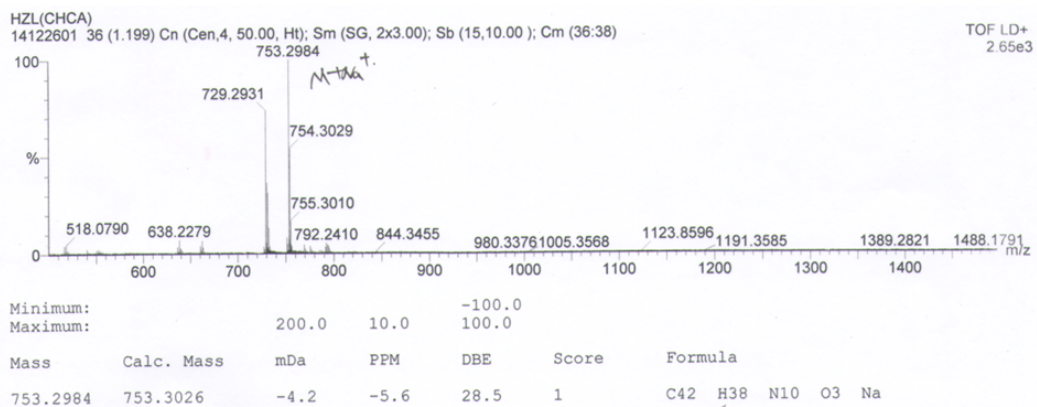


Fig. S13. MALDI-TOF-MS (positive) spectrum of sensor of **L**.