Sustainable Synthesis of Hollow Cu-loaded Poly(*m*phenylenediamine) particles and Their Application for Arsenic Removal

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ESI-1 Equipment









ESI-3 Total TOC and Cu amounts in filtrates after polymerization

PmPD-name	TOC in filtrate (mg L ⁻¹)	Total Cu in filtrate (mg)
Cu1:1	8.7±0.4	864±44
Cu <i>1:0.5</i>	9.65±0.6	445±38
Cu1:0.25	122.5±4	65±2

ESI-4 SEM and TEM images of PmPD-Cu1:1 in different time.



(a) PmPD-Cu1:1-before air introduction



(b) PmPD-Cu1:1 -after reaction for 6h

ESI-5 Raman spectra of PmPD particles.



ESI-6 Parameters of Langmuir and Freundlich models simulated by liner fit for the adsorption of arsenic on PmPD particles.

D DD	Langmuir model		
PmPD-name	K _L	$Q_m(mg \cdot g^{-1})$	R ²
Cu <i>1:1</i>	0.0295±0.0004	24.51±0.12	0.9784±0.0032
Cu <i>1:0.5</i>	0.0382±0.0013	24.60±0.33	0.9845 ± 0.0004
Cu <i>1:0.25</i>	0.0448±0.0039	28.13±0.12	0.9793±0.0051
Cu <i>1:0.125</i>	0.0298 ± 0.0005	16.63±0.10	$0.9853 {\pm} 0.0071$
Cu <i>1:0.1</i>	0.0399±0.0021	13.12±0.19	0.9897±0.0045
Cuθ	0.0089±0.0011	11.24±0.69	0.9913±0.0001

	Freundlich model		
PmPD-name	K _F	1/n	R ²
Cu1:1	3.0191±0.0719	0.3613 ±0.0068	0.9962 ±0.0008
Cu1:0.5	3.8971 ± 0.3005	0.3216 ±0.0122	0.9956 ±0.0015
Cu <i>1:0.25</i>	5.9635±0.0027	0.2699 ±0.0002	0.9951 ±0.0026
Cu <i>1:0.125</i>	2.0266 ±0.0487	0.4009 ±0.0119	0.9896 ±0.0051
Cu1:0.1	1.7481±0.0569	0.3334 ±0.0090	0.9888 ±0.0041
Cu0	0.1030 ±0.0162	0.8142 ±0.0420	0.9563 ±0.0048



PmPD-name		Cu1:0.25
Q _{e,exp}		6.55±0.19
Pseudo-first-order model	<i>k</i> (min ⁻¹)	- 0.00058±0.00012
	Q _{e,cal} (mg·g ⁻¹)	1.247±0.029
	R ²	0.8662 ± 0.1037
– Pseudo-second-order model	<i>h</i> (min ⁻¹)	2.522 ± 0.538
	Q _{e,cal} (mg·g ⁻¹)	6.081±0.0278
	R ²	0.99996±0.00003

ESI-7 Kinetic parameters of Pseudo-first and -second models for the adsorption of arsenic on PmPD-Cu1:0.25.

ESI-8 Effect of coexisting ions on AsO₄³⁻ adsorption with PmPD-Cu*1:0.25*.



Molar Proportion is the molar proportion of competing anions to AsO_4^{3-} (from 1:1 to 4:1).