

**Electronic Supplementary Information**

## Characterization of gelable PS-PAPTES diblock copolymer micelles before and after hydrolysis-condensation by SAXS and SANS

Cé Guinto Gamys<sup>a</sup>, Emmanuel Beyou<sup>a\*</sup>, Elodie Bourgeat-Lami<sup>b</sup>, Laurent David<sup>a</sup>, Julian Oberdisse<sup>c,d</sup>

<sup>a</sup> Université de Lyon, Lyon, F-69003, France, Université Lyon 1, Lyon, F-69003, France, CNRS UMR5223, Ingénierie des Matériaux Polymères IMP@Lyon1, 15 boulevard Latarjet, F-69622 Villeurbanne, France

\*Email : beyou@univ-lyon1.fr

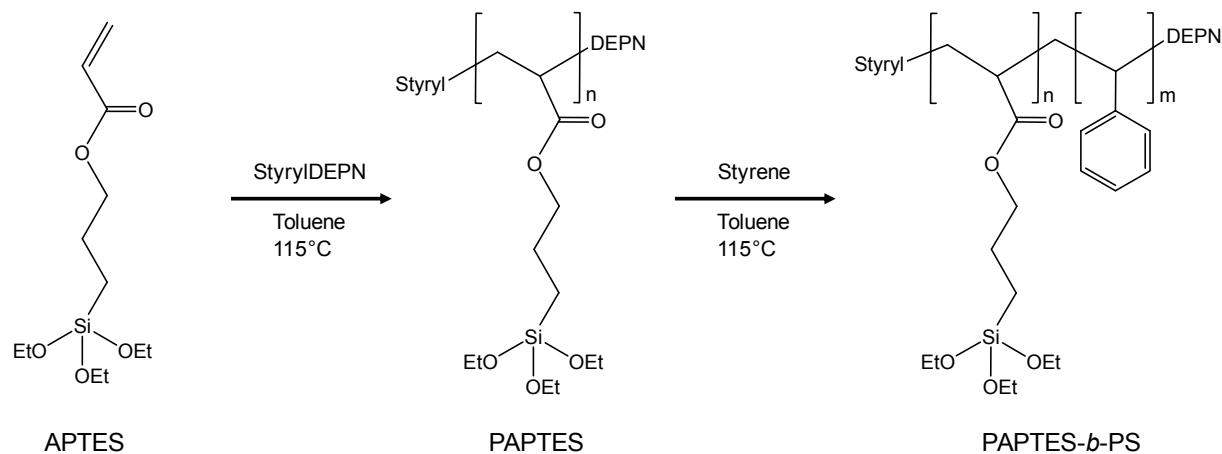
<sup>b</sup>Université de Lyon, Université Lyon 1, CPE Lyon, CNRS, UMR 5265, Laboratoire de Chimie Catalyse Polymères et Procédés (C2P2), LCPP group, 43, Bd. du 11 Novembre 1918, F-69616 Villeurbanne, France

<sup>c</sup>Université Montpellier 2/CNRS, Laboratoire Charles Coulomb UMR 5221, F-34095 Montpellier, France

<sup>d</sup>Laboratoire Léon Brillouin, UMR 12 CEA/CNRS, CEA Saclay, F-91191 Gif sur Yvette, France

### Synthesis of PS and PAPTES-based diblock copolymers

PAPTES and PS homopolymers were used as macroinitiators for the synthesis of diblock copolymers by Nitroxide Mediated Polymerization as described in previous studies<sup>1-2</sup> (Scheme S1).



**Scheme S1:** Synthesis of PAPTES-PS block copolymers by NMP from PAPTES-DEPN macroinitiator

Using this procedure, well-defined PAPTES-b-PS and PS-b-PAPTES diblock copolymers with various chain lengths and compositions were successfully obtained from PAPTES<sub>91</sub>-DEPN, PAPTES<sub>113</sub>-DEPN and PS<sub>272</sub>-DEPN macroinitiators, respectively (**Table S1**).

**Table S1:** Molecular parameters of a series of PS-PAPTES diblock copolymers obtained by NMP at 115°C using respectively PAPTES<sub>91</sub>-DEPN, PAPTES<sub>113</sub> - DEPN and PS<sub>272</sub>-DEPN as macroinitiators. Detailed polymerization conditions are available in the literature<sup>1-2</sup>.

Sample	Time (h)	Conversion <sup>a</sup> (%)	M <sub>n,theo</sub> <sup>b</sup> (g.mol <sup>-1</sup> )	M <sub>n,SEC</sub> (g.mol <sup>-1</sup> )	M <sub>w/M<sub>n</sub></sub>	d <sub>n/dc</sub> <sup>c</sup> (mL.g <sup>-1</sup> )
PAPTES <sub>91</sub> -PS <sub>34</sub>	0.25	4.3	28740	31070	1.44	0.065
PAPTES <sub>91</sub> -PS <sub>175</sub>	1	21.9	43380	45830	1.37	0.110
PAPTES <sub>91</sub> -PS <sub>317</sub>	2.5	39.6	58110	57250	1.40	0.130
PAPTES <sub>113</sub> -PS <sub>489</sub>	10	61.1	82050	84940	1.33	0.122
PS <sub>272</sub> -PAPTES <sub>130</sub>	16	26.1	64300	59990	1.18	0.100
PS <sub>272</sub> -PAPTES <sub>224</sub>	32	44.9	90250	83310	1.21	0.077
PS <sub>272</sub> -PAPTES <sub>290</sub>	64	58.1	108470	101250	1.30	0.064
PS <sub>272</sub> -PAPTES <sub>336</sub>	128	67.2	121000	110800	1.32	0.056

<sup>a</sup>Calculated by <sup>1</sup>H NMR. <sup>b</sup>M<sub>n,theo</sub> = [M]<sub>0</sub>/[I]<sub>0</sub> x Conversion x M(monomer). <sup>c</sup>Determined by interferometric refractometry.

## References

- 1 Gamys CG.; Beyou, E. and Bourgeat-Lami, E., *J. Polym. Sci., PartA : Polym. Chem.*, 2010, **48**, 784.
- 2 Gamys CG.; Beyou, E.; Bourgeat-Lami, E., David, L. and Alcouffe, P. *J. Polym. Sci., PartA : Polym. Chem.*, 2011, **49**, 4193.